# Economic Feasibility Study of an Enzymatic Hydrolysis-Based Ethanol Plant with Prehydrolysis Pretreatment

# **A Subcontract Report**

Chem Systems, Inc. Tarrytown, NY

Prepared under Subcontract No. XX-3-03097-2



SERT/STR-231-3135

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Solar Energy Research Institute

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**April 1987** 

SERI Technical Monitor: J. D. Wright

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# TABLE OF CONTENTS

					Page
I.	C. Proces D. Plant				1 1 3 7
II.	INTRODUCT	ION			13
III.	<ul><li>B. Pretro</li><li>C. Enzymo</li><li>D. Fermer</li></ul>	Chemistry eatment e Hydrolysis			15 15 19 23 25 28
IV.	1. Pi	ESIGN ss Description retreatment/Prehydrolysis ugar Separation and Neutralization			31 31 31
	3. Ei 4. Ei 5. Si	ngar Separation and Neutralization of nzyme Production nzyme Hydrolysis ugar Concentration ermentation			33 36 39 41 44
	7. 0: 8. E <sup>.</sup> 9. Fi	arbon Dioxide Recovery  thanol Purification  urfural Production  eat Generation			46 48 51 54
	11. Wa 12. O B. Design	iste Treatment Ffsites 1 Basis			56 58 61
	2. Si 3. Er 4. Er 5. Si 6. Fe	retreatment/Prehydrolysis ugar Separation and Neutralization uzyme Production uzyme Hydrolysis ugar Concentration ermentation urbon Dioxide Recovery			63 64 65 66 67
	9. Et 9. Et 10. He 11. Wa	chanol Purification  orfural Production  eat Generation  iste Treatment  al Balance			69 70 71 73 73
	D. Proces	s Utility Summaries eat Balance Summary			75
	2. St	eat Balance Summary Leam Requirements Poling Water Requirements	·		75 77 78

# TABLE OF CONTENTS (Continued)

		<u>Pa q</u> e
	4. Process Water Requirements 5. Boiler Feedwater Requirements 6. Power Requirements E. Plot Plan	79 80 80 84
	SITING CONSIDERATIONS A. Selection of Michigan Site 1. Introduction 2. Feedstock Availability and Pricing 3. Gasoline Pool Analysis 4. Tax Incentives B. Demand and Pricing Analyses 1. Ethanol 2. Furfural 3. Carbon Dioxide	86 86 86 93 97 105 105 114
VI.	ECONOMIC ANALYSIS A. Base Case Economics 1. Economic Assumptions 2. Cost of Production Analysis 3. Cash Flow Analysis B. Optimization Alternatives	134 134 134 134 139
VII.	DISCUSSION  A. Sources and Limitations of Data  B. Critical Technology Issues  C. Commercial Potential  D. Recommendations for Future R&D	158 158 165 166 168
III.	REFERENCES AND THE PROPERTY OF	170
IX.	ACKNOWLEDGMENTS	172
Χ.	APPENDIX  A. Detailed Material Balance  B. Major Equipment Specifications and Costs	173 173 201

#### I. EXECUTIVE SUMMARY

#### A. Background

Chem Systems has been involved in preliminary design and evaluation of enzyme hydrolysis processes for converting lignocellulose to ethanol, based primarily on its work for SERI in earlier subcontracts. In this earlier work, Chem Systems developed a process design and techno-economic evaluation of a prototype enzyme hydrolysis facility, as well as a computer simulation program which integrated material and utility balances, capital cost estimates and cost of production analyses. A number of parametric studies were performed with the enzyme hydrolysis simulation in an effort to optimize the pretreatment and post hydrolysis options. As a result of these parametric analyses, a process based on enzyme hydrolysis incorporating enzyme production for the conversion of cellulose to hexose sugars seemed to have commercial potential.

Dilute acid prehydrolysis was determined to be one of the most economic pretreatments for enzyme hydrolysis. This concept is the subject of a patent issued to Hans E. Grethlein and assigned to Dartmouth College (U.S. Patent No. 4,237,226, Dec. 2, 1980) entitled "Process for Pretreating Cellulosic Substrates and for Producing Sugar Therefrom." Extensive data covering a wide range of acid prehydrolysis and enzyme hydrolysis conditions were developed by the Thayer School of Engineering at Dartmouth College utilizing the Rutgers C-30 cellulase enzyme. While considerable work is still necessary to optimize the acid prehydrolysis/enzyme hydrolysis configuration and to refine a kinetic model for this configuration, Chem Systems proposed that this concept warranted a detailed feasibility study.

## Site and Feedstock Selection

One aspect of this feasibility study is the selection of a Midwest site for the ethanol facility and the selection of the wood feedstock based on

available forest types. This selection process involved screening ethanol consumption patterns, gasoline demand, applicable tax laws, wood resource acreage and cost elements for delivered wood including both current and future trends.

After a review of the above issues, it was determined that Michigan is probably the most advantageous state within the Upper Midwest and North Central states in which to locate the proposed ethanol plant. The state has the advantages of an industrial base, significant gasoline demand (about 4 percent of national demand and a higher percentage of unleaded gasoline), proximity to other major state markets, adequate wood resources, favorable tax and investment climate, and lack of significant local alcohol production (only 8 million gallons per year), as well as a relatively high state tax exemption for ethanol usage.

The proposed plant would be located in Michigan's Northern lower peninsula, which contains about 1.8 million acres of aspen forest as well as significant quantities of other suitable hardwoods. If it is assumed that 30 percent of the plant feedstock will be hardwoods from aspen forests and the remainder from maple/birch forests, the average feedstock composition will be 57 percent aspen, 20 percent maple and 23 percent other hardwoods. On a moisture-free basis, the plant feed will contain approximately 47.2 percent cellulose, 31.3 percent hemicellulose and 18.5 percent lignin. The hemicellulose can be further broken down to 7.9 percent hexosans, 16.5 percent pentosans and 6.9 percent others.

Gladwin County/Bay City, Midland area has been chosen as the specific location for this facility. This county contains 7 percent of the total state aspen forest land which, by itself, could supply approximately half the required plant feedstock. When coupled with four surrounding counties, the total aspen forest resource base would be sufficient for a facility of twice the capacity of this project. Furthermore, this site provides convenient access to product and by-product markets within the state.

The major by-products from this facility are carbon dioxide and furfural. It is anticipated that the carbon dioxide can be sold locally for use in beverages or enhanced oil recovery. Furfural is currently used primarily in the Midwest and Southwest to produce furfural alcohol, furan resins and THF. Low-cost furfural would open up possibilities of additional THF capacity as well as adipic acid and 1,4-butanediol manufacture. A wide variety of volume chemicals could be produced from these building blocks if the furfural could be priced at approximately half its current market value. Such a scenario is feasible when the furfural is produced as a by-product from an ethanol-from-wood complex.

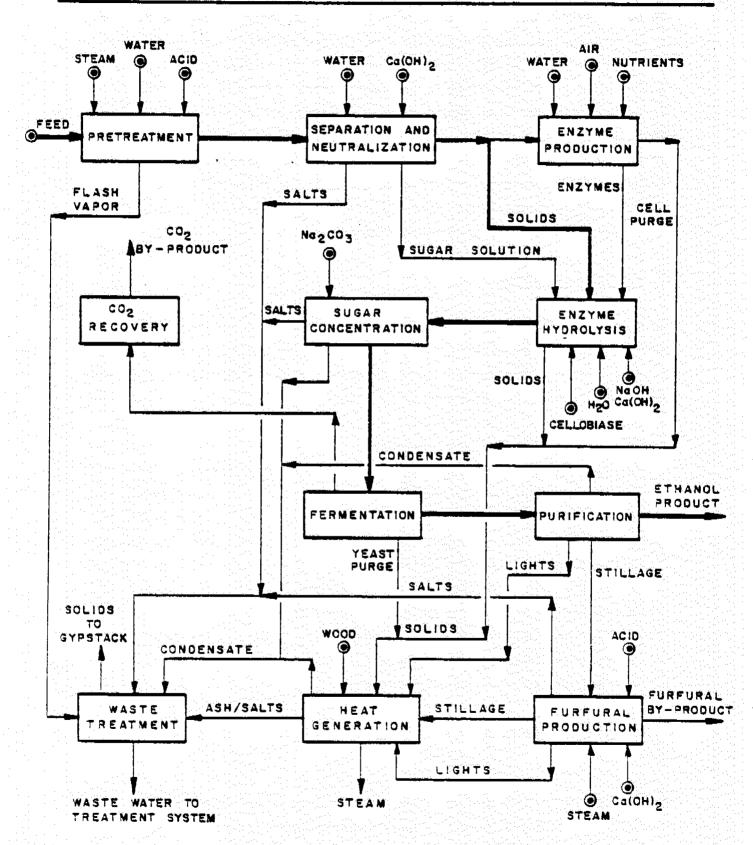
#### C. Process Description

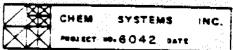
The ethanol facility is designed to produce 25 million gallons per year of anhydrous fuel-grade material from mixed hardwood chips, primarily aspen. Major by-products are furfural and carbon dioxide. The plant is divided into eleven processing sections as follows:

- Section 100 -- Pretreatment
- Section 200 -- Sugar separation and neutralization
- Section 300 -- Enzyme hydrolysis
- Section 400 -- Enzyme production
- Section 500 -- Sugar concentration
- Section 600 -- Fermentation
- Section 700 -- Ethanol purification
- Section 800 -- Carbon dioxide recovery
- Section 900 -- Furfural production
- Section 1000 -- Heat generation
- Section 1100 -- Waste treatment

A block diagram depicting the facility by plant section is shown in Figure I-C-1.

Wood chips are transferred from the storage area via a conveyor to a disk refiner where particle size is reduced sufficiently for compatibility with CELLULOSE TO ETHANOL VIA ENZYME HYDROLYSIS





prehydrolysis and hydrolysis equipment. This machine requires wood steaming to soften the wood before it is refined into fibers or particles. The steaming of the wood also serves as a preheat for prehydrolysis. Thus, the refiner effluent is fed directly to the prehydrolysis reactor without intermediate pumping. The required acid (0.75 weight percent) and dilution steam/water are mixed with this solids stream to give 15 weight percent solids at the entrance to the plug flow prehydrolysis reactor. Within this reactor the hemicellulose fraction and amorphous cellulose are converted to their respective sugars. The reactor product is flashed in three stages to remove some water/furfural vapor and then proceeds to the centrifuge. Solids from this centrifuge are sent to enzyme hydrolysis and enzyme production. Hot liquids from the centrifuge are sent to a polishing filter to remove the remaining cellulose. The liquid is then neutralized while hot and the precipitated calcium sulfate is filtered out. The liquid stream proceeds to enzyme hydrolysis.

In enzyme hydrolysis, the solids are diluted with water and with the prehydrolysis sugar solution to 10 weight percent and pH adjusted prior to the hydrolyzers. Enzymes from enzyme production are added as well as purchased QM329 cellobiase enzyme (Aspergillus phoenicius). After 24 hours residence time in the cascade reactor system, the yield of glucose on cellulose is 90 percent. The resultant sugar solution contains some unconverted cellulose, which is filtered and washed, removing the bulk of the unconverted cellulose which is pressed and sent to heat generation. The filtrate is pumped to concentration.

Approximately 12 percent of the prehydrolyzed wood is diverted to the enzyme production section where it is used as the carbon source for the RUT C-30 cellulase enzyme (<u>Trichoderma reesei</u>). The quantity of the RUT C-30 required is approximately ten times as large as the QM329 on a volume basis. The RUT C-30 is produced in a fed-batch cycle at a net solids concentration of 15 weight percent taking approximately 12 days. In addition to cellulose, air and nutrients are introduced to the enzyme production tanks. The product from these tanks is sent to a cell centrifuge to remove most of the mycellium, which is then repulped,

filtered and washed. Part of the filter cake is recycled to enzyme production to maintain the initial cell concentration in the production tank. The remainder of the filter cake is sent to heat generation. The centrifuge overflow and filtrate are sent to enzyme hydrolysis.

The sugar solution from enzyme hydrolysis is concentrated to 15 percent sugar in a multi-effect evaporator, pH adjusted and sent directly to continuous cascade fermentation. The continuous fermentation step includes yeast separation and recycle with a purge to heat generation. Carbon dioxide produced during fermentation is also recovered for sale. The ethanol product from fermentation is sent to purification, which consists of a beer still and azeotropic dehydration. The ethanol is concentrated to the water-ethanol azeotropic composition in the beer still, then dehydrated by distillation with an azeotroping agent to produce the anhydrous ethanol product. Aqueous stillage from alcohol purification is sent to furfural production.

In the furfural production section, the stillage is heated to approximately 220°C and sulfuric acid is added. After providing sufficent residence time, the reactor product is flashed and neutralized and the precipitated calcium sulfate is filtered out. The flash vapors are cooled by exchange in the furfural-water azeotrope column reboiler, combined with the neutralized liquids and fed to the azeotrope column. The furfural is further concentrated in the dehydration column. A lights column is utilized to remove low boiling material from the decanted aqueous layer originating from the azeotrope column overheads.

Hydrolysis solids purge, furfural production stillage, waste yeast, lights and waste enzymes are sent to a multi-effect evaporator prior to being used as fuel in the steam boiler where 1,200 psia steam is raised. Additional wood chips are fed to this steam boiler to bring the plant steam requirements into balance. Ash from the boiler and salts from neutralization are sent to a gypstack.

The condensate from various process concentration steps is combined with dilute aqueous waste streams and sent to a waste treatment system. Treated water is reused in the process.

The offsites included in the design are storage facilities, a cooling water system, a wood-fired steam boiler with electricity cogeneration facilities, boiler feedwater pretreatment, buildings, piping, pollution control, general utilities and land site development.

#### D. Plant Economics

An economic evaluation has been performed for the wood-based ethanol plant described in the preceding section. The capital cost for a 25 million gallon per year plant is estimated at 112.0 million dollars based on mid-1984 costs at a Michigan site location. This estimate includes 66.3 million dollars for inside battery limits cost (ISBL) and 45.7 million dollars for offsites. Summaries of the ISBL and offsites costs are presented in Tables I-D-1 and I-D-2, respectively.

A 1984 manufacturing cost summary for the 25 million gallons per year ethanol plant, shown in Table I-D-3, serves as the basis for a cash flow analysis over the project life.

The raw material costs is based on a current-hardwood price of 18 dollars per wet ton which is determined based on an average hauling distance of 50 miles.

The total cash cost of production for ethanol is estimated at 36 cents per gallon. The net cost of production for ethanol, including depreciation, is 157 cents per gallon and the instantaneous mid-1984 ethanol sales price is 206 cents per gallon with a 10 percent DCF return on investment.

TABLE I-D-1

ISBL\* CAPITAL COST SUMMARY, MID-1984

Section	<u>Name</u>	Total Field Installed Cost, \$
100 200 300 400 500 600 800	Pretreatment/prehyrolysis Sugar separation and neutralization Enzyme production Enzyme hydrolysis Sugar concentration Fermentation Ethanol purification	3,104,000 3,901,000 6,376,000 7,069,000 5,941,000 4,209,000 4,111,000
900 1000	Furfural production Heat generation	5,166,000 3,311,000
Bare plant Conting Contrac	ring ction overhead   cost ency tors fee   charges lity cost	43,188,000 4,180,000 4,850,000 52,218,000 5,222,000 1,509,000 60,490,000 5,800,000 66,290,000

<sup>\*</sup>Section 1100, Waste treatment, is included in offsites costs.

# TABLE I-D-2 OFFSITES CAPITAL COSTS, MID-1984

Total	Installed Cost, \$
Total storage	1,928,000
Steam boiler	11,700,000
Generator/turbine/switch gear	6,600,000
Wood handling system	3,100,000
Cooling water	2,760,000
Electrical system	982,000
Boiler feedwater system	490,000
Pollution control	
Pond and gypstack (Section 1100)	2,560,000
Sanitary waste	300,000
- Buildings	1,815,000
General utilities	3,024,000
Site development	3,630,000
Piping - Property of the Prope	1,815,000
Total offsites	45,704,000

#### TABLE I-D-3

#### ETHANOL MANUFACTURING COST SUMMARY Mid-1984, Michigan Site

	cents per Gailoi
Raw materials (1) By-product credit (2)	80.97 (62.04)
Utilities.	19.89
Operating costs	22.65
Overhead expenses	24.47
Cash cost of production	85.95
Depreciation	<u>71.33</u>
Net cost of production	157.28
Ethanol sales price at 10 percent DCF return	206.0

(1) Wood feedstock at \$18 per wet ton.(2) Furfural and carbon dioxide by-product values at 30 and 2.8 cents per pound, respectively.

Using the 1984 cost data as the basis, a cash flow analysis was performed to determine the internal rate of return (IRR) for the project. I-D-4 highlights the economic parameters used in this analysis.

#### TABLE I-D-4

#### ECONOMIC PARAMETERS FOR THE CASH FLOW ANALYSIS

Plant operation (after start-up): 1987 through 2001 Construction period: 3 years

Capacity build-up: 60 percent first year

80 percent second year

100 percent remaining years

Interest rates (for debt financing): 13 percent

MSARD: Six percent of revenues

Depreciation: Straight line method over five years for ISBL capital

and ten years for offsites

Investment tax credit: 10 percent of total capital credited against

current or future tax liabilities

Corporate tax: 45 percent of profit Federal/state tax incentive: 9 cents per gallon for 10 percent

ethanol blends in gasoline

Since the cash flow is based on current dollars the cost elements were escalated over the project life. These escalation rates are presented in Table I-D-5. Revenues are generated from ethanol along with by-product furfural and carbon dioxide. The ethanol market value over the life of the project is forecast based on the regional demand for gasoline by year, grade and distance from the Gladwin County plant site. The federal tax incentive for 10 percent ethanol blends in gasoline was projected at six cents per gallon and the state tax incentive was assumed to be three cents per gallon for the project life. Combining these factors, ethanol values at the plant gate were estimated for the 1987-2001 time period. Table I-D-6 summarizes this ethanol price forecast.

TABLE I-D-5
ESCALATION RATES, 1984-2001
(Percent per Year)

Period (	GNP Deflator (	<u>l)</u> <u>Capit</u>	cal Cost (2)	Utility Costs
1984-1990				
1990-2001	4.5		5.5	7.5 6.9

(1)Used for feedstock, labor and direct overhead costs. (2)Used for capital, maintenance, general plant overhead and insurance.

TABLE I-D-6
ETHANOL PRICE FORECAST

	A Commence of the Commence of		
<u>Year</u>		Price,	Cents/Gallon
e e		e*	
1987			200.3
1988			207.4
1989			214.4
1990			221.5
1991			
1992			227.7
1993			234.2
			240.7
1994			247.5
1995			254.5
1996			261.9
1997			269.5
1998			277.3
1999			235.3
2000			293.6
2001		eles file type to lyng. Ny oratrona ao ao ao	302.1
			304.1

For a plant that is 100 percent equity financed, the IRR for this project is 5.9 percent. If the plant is based on 30 percent equity and 70 percent debt financing the IRR becomes negative since the after tax cost of money is greater than the equity return.

#### E. Conclusions and Recommendations

The enzyme hydrolysis based ethanol facility evaluated in this study is a feasible project from both technical and economic points of view. The strengths and weaknesses of this project are discussed in this section.

The technology base for this facility is relatively sound. The major area where further research is necessary involves enzyme production using prehydrolyzed wood. While this has been done, the yields and process conditions assumed in this study have not yet been attained. Reasonable conditions have been assumed based on experimental evidence and it is expected that research with pretreated wood will reveal methods of handling that equal or even exceed the expectations of this study.

The remainder of the process steps through enzyme hydrolysis itself have been demonstrated individually on a small scale. It remains to verify that these steps can be carried out in an integrated fashion in industrial-type equipment. This is especially true of prehydrolysis and the various solid/liquid separations encountered. Integration of process operations will reveal if there are any inhibitory effects from minor constituents formed in pretreatment and, if so, how to deal with them.

The rest of the process, from glucose fermentation through furfural production, is based on commercially proven technology. Other than verifying that sugars from enzyme hydrolysis are fermentable, no further development of this half of the process is necessary.

In addition to 25 million gallons per year of fuel-grade ethanol, this facility coproduces 160 million pounds per year of liquefied carbon dioxide and 36 million pounds per year of furfural. Revenues from these by-products help to offset the effects of the substantial capital investment (112 million dollars in mid-1984). Furthermore, virtually all

waste organics are utilized to provide steam. Additional heat is provided by additional wood chips to bring the plant into thermal balance. More costly fuels are avoided. Most of the plant electricity requirements are met by cogeneration from plant steam. Thus, utility costs are minimal.

Evaluating this facility on a mid-1984 basis with wood chips at 18 dollars per wet ton and by-product furfural at 30 cents per pound results in a required ethanol sales price of 206 cents per gallon at 10 percent DCF return on investment. On the same economic basis a 50 million gallon per year ethanol plant based on whole corn kernel milling would yield a sales price of 219 cents per gallon. Thus, the innovative enzyme hydrolysis based facility should be more than competitive with current commercial fermentation technology.

The choice of Gladwin County in Michigan for the plant site results in an ethanol market almost entirely within a 200-mile radius. Furthermore, there are adequate hardwood feedstock supplies within a 50-mile radius of the plant site. An ethanol pricing scenario incorporating a combined federal and state tax incentive of nine cents per gallon for 10 percent ethanol blends in gasoline was developed for Michigan over the project life and then used to generate the project cash flow analysis. This resulted in an internal rate of return of 5.9 percent based on 100 percent equity financing. Although profitable, this return is probably not high enough to convince investors to go ahead with such a venture at this time. However, optimization studies have pointed out reasonable targets for improvement in enzyme production and enzyme hydrolysis. If some of these improvements could be achieved through additional research and development, the proposed ethanol facility would become commercially interesting.

In conclusion, it seems that an enzymatic hydrolysis based ethanol plant using prehydrolysis pretreatment is a viable process concept. Research to optimize the enzyme production and hydrolysis sections of the facility is still required and then the front half of the process must be demonstrated in industrial-type equipment to provide proof of concept. When these steps are successfully completed, the process would be ready for commercialization. A central Michigan location would be ideal for the first plant of this kind.

#### II. INTRODUCTION

Chem Systems has been involved in preliminary design and evaluation of enzyme hydrolysis processes for converting lignocellulose to ethanol, based primarily on its work for SERI in earlier subcontracts. In this earlier work, Chem Systems developed a process design and techno-economic evaluation of a prototype enzyme hydrolysis facility. Using the design as a basis, Chem Systems developed a computer simulation program which integrated material and utility balances, capital cost estimates and cost of production analyses. The user was able to choose pretreatement options and most of the operating parameters. A number of parametric studies were performed with the enzyme hydrolysis simulation in an effort to optimize the pretreatment and post hydrolysis options.

As a result of parametric analyses using the simulation program, a process based on enzyme hydrolysis incorporating enzyme production for the conversion of cellulose to hexose sugars seemed to have commercial potential. There are many potentially interesting process configurations incorporating enzyme hydrolysis. However, the most viable configurations seem to be those which involve minimial pretreatment steps and still maximize the recovery of useable by-products.

Dilute acid prehydrolysis has been determined by Chem Systems to be one of the most economic pretreatments for enzyme hydrolysis. This concept is the subject of a patent issued to Hans E. Grethlein and assigned to Dartmouth College (U.S. Patent No. 4,237,226, Dec. 2, 1980) entitled "Process for Pretreating Cellulosic Substrates and for Producing Sugar Therefrom." Extensive data covering a wide range of acid prehydrolysis and enzyme hydrolysis conditions were correlated and incorporated into the simulation. These data were developed by the Thayer School of Engineering at Dartmouth College utilizing the Rutgers C-30 cellulase enzyme. Further parametric studies with the model verified that acid prehydrolysis is indeed an economic pretreatment for enzyme hydrolysis since a portion of the hexosan is converted to sugar in this step and the remaining cellulose becomes very accessible to enzyme attack.

While considerable work is still necessary to optimize the acid prehydrolysis/enzyme hydrolysis configuration and to refine the kinetic model for this configuration. Chem Systems proposed that this concept warranted a detailed feasibility study.

#### III. PROCESS CHEMISTRY

#### A. Wood Chemistry

Wood consists chiefly of hollow interconnected fibers axially oriented in the tree. These support the tree and form the conduits for the transport of water from the roots to the leaves where, under the catalytic action of chlorophyll, the water reacts with carbon dioxide to form sugars and other organic materials. Growth of the tree results from the return of the solution of these substances through the inner bark to further react and polymerize at the cambium at the interface between wood and bark, thus forming new cells which in turn form a further ring in the fibrous conduit system.

The conifers, of which spruce is an example, are generally referred to as softwoods. The broad-leaved deciduous woods which are morphologically and chemically distinct from the conifers are commonly referred to as hardwoods in spite of the fact that certain species, such as basswood and poplar, have woods which are relatively soft.

The structural components of wood, which comprise some 90 percent of the weight of the wood, consist of the three natural polymers cellulose, lignin, and hemicelluloses. The 3 to 10 percent nonstructural components of wood, such as terpenes and waxes, are referred to as extractives.

Cellulose is a highly oriented, crystalline, linear polymer of glucose units. Values for the degree of polymerization of cellulose chains in wood range from 7,000 to 10,000. Cellulose can be hydrolyzed to glucose by acids whereupon the glucose formed by cellulose hydrolysis can be fermented to ethanol under the influence of enzymes from yeast.

Lignin is a 3-dimensional random polymer formed from phenylpropane units. It acts as a cement between the cellulose fibers and as a stiffening agent within the fibers. Its molecular weight is over 10,000.

Hemicelluloses are polysaccharides which are short or branched polymers of 5-carbon sugars such as xylose or 6-carbon sugars other than glucose. Hardwoods contain about 30 percent hemicelluloses, of which the principal constituent sugars in decreasing abundance are xylose, galactose, and mannose. Degrees of polymerization range from less than 100 to about 200 sugar units. Hemicelluloses are readily hydrolyzable to xylose and other simple sugars by acids. Mannose and other hexoses can be fermented to ethanol, while xylose and other pentoses are readily dehydrated to furfural by acids.

The wood cell wall is then a fiber-reinforced plastic with cellulose fibers embedded in an amorphous matrix of hemicelluloses and lignin.

The structure and composition of bark are very different from that of wood. Typically, the cellulose content of bark is only 20 to 30 percent. Bark contains a significant proportion of extractives, generally 20 to 40 percent. Extractive-free bark contains two components not found in wood, suberin and phenolic acids, as well as cellulose, hemicellulose, and lignin. Suberin is a hydroxy acid complex consisting of esters of higher aliphatic hydroxy acids and phenolic acids. Phenolic acids are high molecular weight phenols that differ from lignin in their lower molecular weight, high carboxyl content, and lower methoxyl content.

In hardwoods, the principal hemicellulose is 4-0-methylglucurono xylan acetate. This is made up of 1 part 4-0-methyl-alpha-D glucuronic acid, 10 parts beta-D xylose, and 7 parts 0-acetyl-beta-D-xylose.

Cellulose is a long chain polymer of beta-D-glucose in the pyranose form, linked together by 1,4-glycosidic bonds to form cellubiose residues that are the repeating units in the cellulose chaining. The beta-linkage requires that the alternate glucose units must be rotated through 180 degrees. An important implication of this structure is a marked tendency for the individual cellulose chains to come together to form bundles, of a crystalline nature, held together by hydrogen bonds.

Lignins are 3-dimensional network polymers formed from phenylpropane units that have randomly grown into a complicated large molecule with many different kinds of linkages between the monomers. Although hardwood ligning differ somewhat in composition from softwood ligning, chiefly in methoxyl substitution and the degree of carbon-carbon linkage between phenyl groups, the common structural features predominate. This includes an aromatic and phenolic character as well as covalent carbon-carbon bonding that prevents reversion to monomers by mild processing. random structure arises from an enzymatically initiated free radical polymerization of lignin precursors in the form of p-hydroxy cinnamyl alcohols. In conifers, the precursor is principally coniferyl alcohol (3-methoxy-4-hydroxycinnamyl alcohol). Coniferyl alcohol yields the so-called quaiacyl lignin. In hardwoods, additional precursors such as sinapyl alcohol (3,5-dimethoxy-4-hydroxycinnamyl alcohol) and coumaryl alcohol (4-hydroxycinnamyl alcohol) are also present resulting in the so-called quaiacyl-syringyl lignins. Because the quaiacyl unit has an additional potential reactive site instead of the extra methoxyl group in the syringyl unit, a higher degree of cross-linking exists in quaiacyl lignins. The lower apparent molecular weight and easier dissolution of hardwood lignins are manifestations of their lower degree of cross-linking. The small quantities of methanol produced as a by-product of the wood hydrolysis process arise from the methoxyl groups (-OCH3) in lignin.

The specific composition of wood varies from species to species. Table III-A-1 shows the typical composition of several woods. It can be seen from the table that, in general, hardwoods have a greater holocellulose (cellulose and hemicellulose) content than softwoods, although the hexosan content is approximately equivalent for both, due to a lower percentage of xylans in the softwood hemicellulose.

Table III-A-2 indicates the potential reducing sugars that can be produced from hardwoods and softwoods. It shows that aspen has the greatest potential for producing reducing sugars which can subsequently be fermented. Populus Tremuloides (poplar) or trembling aspen has high availability in North America and can be grown as a high yield, short

TABLE III-A-1

PERCENTAGE COMPOSITION OF CERTAIN WOODS

(Extractive Free, Ash Free)

	<u>Lignia</u>	Holo- cellulose	Alpha- cellulose	Hemi- <u>cellulose</u>	<u>Pentosans</u>	Uronic Acid Anhydride	<u>Acetyl</u>	Methoxyl in Carbohydrate
Softwoods								
White Spruce	26.6	73.3	49.5	23.8	10.9	2.68	2.35	0.70
Red Spruce	26.6	72.9	48.3	24.6	11.6	3.20	2.50	0.92
Eastern Hemlock	31.5	68.5	48.2	20.3	10.0	3.40	1.87	0.81
Balsam Fir	30.1	69.9	44.0	25.9	10.3	3.08	2.24	0.41
Jack Pine	27.2	72.5	49.5	23.0	12.8	2.92	1.92	0.75
Hardwoods								
Aspen	17.3	82.5	50.7	31.8	23.5	4.28	4.65	0.93
Willow	22.0	78.3						
Maple	23.5	76.3	50.0	26.3				
White Oak	24.1	75.4	49.5	25.9				

rotation crop in many parts of the northern United States. These reasons make aspen an excellent major feedstock for the enzymatic hydrolysis-based ethanol process described herein.

TABLE III-A-2

YIELD OF POTENTIAL REDUCING SUGARS AND FERMENTABLE SUGARS
FROM SAMPLES OF REPRESENTATIVE HARDWOODS AND SOFTWOODS

Hardwoods	Potential	Ferment-	Potential
	Reducing	ability,	Fermentable
	Sugars, %	%	Sugars, %
American Beech	70.1	75.1	52.6
Aspen	75.1	76.3	57.3
Birch	69.9	67.8	47.4
Maple	68.2	71.0	48.4
Red Oak	63.6	63.0	40.2
Sweetgum	66.4	73.8	49.0
Yellow Poplar	70.9	76.1	54.0
Softwoods Douglas Fir Eastern White Pine Hemlock Ponderosa Pine Redwood Sitka Spruce Southern Yellow Pine Sugar Pine	66.6	86.2	57.4
	66.5	86.3	57.4
	66.1	88.2	58.3
	68.0	82.2	55.9
	52.4	77.1	40.4
	70.1	85.3	59.8
	64.8	82.0	53.2
	64.8	82.4	53.0

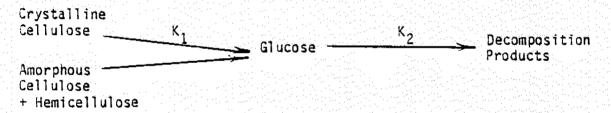
### B. <u>Pretreatment</u>

Two methods of pretreating wood feedstock, steam explosion and dilute acid prehydrolysis, have been proposed recently as efficient methods of rendering cellulose accessible to enzymatic attack thus resulting in high glucose yields. The latter pretreatment, based on the developmental work of the Thayer School of Engineering at Dartmouth College (1), is the basis for this study.

Dilute acid prehydrolysis is a pretreatment designed primarily to attack and convert the hemicellulose and amorphous cellulose fractions of wood.

Because of the relatively low reaction temperature, only a small fraction of the crystalline cellulose is converted to glucose. Furthermore, the amount of degradation products formed from the resulting hexose and pentose sugars is minimal. Thus, the net result of prehydrolysis is the formation of soluble sugars from the hemicellulose fraction which leaves a residual higher porosity solid material that is more susceptable to enzymatic attack.

A plug-flow reactor kinetic model has been developed at Dartmouth for this system (2,3). These kinetics assume that the acid-catalyzed hydrolysis of crystalline cellulose to glucose can be described as a homogeneous pseudo-first-order reaction and that the acid-catalyzed decomposition of glucose is also a first-order reaction. Amorphous cellulose and hemicellulosic hexosans hydrolyze immediately to glucose and can be taken as an initial glucose fraction. Thus, the hydrolysis can be described by:



The integrated kinetic expressions describing these reactions are:

$$G(t) = C(o)$$
  $\left[\frac{K_1}{K_1 - K_2}\right] \left[\exp(-K_2 t) - \exp(-K_1 t)\right] + G(o) \exp(-K_1 t)$  (1)

$$C(t) = C(o) \exp(-K_1 t)$$
 (2)

where

$$K_i = P_i A^{Ni} \exp(-E_i/RT)$$

C is the fraction of potential glucan remaining as crystalline cellulose, G is the fraction of potential glucan present as glucose and A is the weight percent sulfuric acid in the aqueous phase. The cellulose decomposition product is assumed to be hydroxymethyl furfural although further reaction will result in some levulinic acid and formic acid. The

activation energies, exponents and pre-exponential factors are tabulated for different wood feedstocks in Table III-B-1. It can be seen that some variation in cellulose decomposition rate is expected for different wood feedstocks.

TABLE III-B-1

CELLULOSE DECOMPOSITION KINETIC CONSTANTS

	GLUCOSE FOR	MATION (2,3)	
	$p_1$ , min <sup>-1</sup>	<u>N</u> 2	$E_1$ , cal/g mol
Poplar	$6.12 \times 10^{15}$	0.99	35,150
White Pine	$7.80 \times 10^{13}$	0.96	30,170
Mixed hardwoods (90% birch,			
10% maple)	$8.98 \times 10^{20}$	1.55	47,100

	GLUCUSE I	DECOMPOSITION	
	$\underline{P_2}$ , min <sup>-1</sup>	<u>N</u> 2 <u>E</u> 2.	, cal/g mol
Glucose	3.96 x 10 <sup>8</sup>	0.57	21,000

Decomposition kinetics of the hemicellulosic xylan have also been investigated in a plug-flow reactor system at  $Dartmouth^{(4)}$ . This hydrolysis sequence can be described by:

The rate equations describing these reactions are:

$$\frac{dH}{dt} = -K_0H \tag{3}$$

$$\frac{dX}{dt} = K_0 H - K_3 X \tag{4}$$

$$\frac{dD}{dt} = -K_5 F$$

$$U = 0; \frac{dU}{dt} = 0 \tag{6}$$

$$\frac{dF}{dt} = K_3 X - K_5 F - K_7 X F Q_0 \tag{7}$$

where

$$K_7 = K_3 K_4 / K_6$$

and

H = Mole fraction of potential xylose as xylan.

X = Mole fraction of potential xylose as xylose.

U = Mole fraction of potential xylose as unstable intermediates.

F = Mole fraction of potential xylose as furfural.

D = Mole fraction of potential xylose as decomposed furfural.

C = Mole fraction of potential xylose as condensation products.

Qo = Concentration in moles/liter of potential xylose in feed.

The activation energies, exponents and pre-exponential factors for xylan decomposition in mixed hardwood are shown in Table III-B-2.

TABLE III-8-2

KINETIC PARAMETERS FOR MIXED HARDWOOD XYLAN DECOMPOSITION (4)

			- I	< <sub>0</sub>	K <sub>3</sub>		К <sub>5</sub>	K	7
F (	ca	l/gmo]	27.	,826	27,130	15	.279	23.	943
Ln	P	(min <sup>-1</sup>	): 31	.76	28.48	1	3.23	26	.74
N		and the second	]	17	0.688	0	.579	0.	517

Solving these equations for an aspen/mixed hardwood feed to a 12-second plug flow reactor at 200°C with 0.75 weight percent sulfuric acid results in the conversion of 5 percent of the crystalline cellulose. Since all the hemicellulose hexosans are converted, the total C<sub>6</sub> hydrolyzed is 13.6 mole percent with a selectivity to glucose of 98.5 percent. On the other hand, approximately 71 percent of the xylan is converted with a selectivity to xylose of 93.5 percent and a selectivity to furfural of 6.4 percent. Thus, these conditions result in 27 weight percent of the cellulose and hemicellulose converted, primarily to sugars with only minor amounts of further degradation products.

In order to feed the wood to the pressurized prehydrolysis reactor, the chips must be reduced in size to aproximately 1 millimeter. This can be accomplished in a disk refiner which operates most efficiently when the wood chips are steam presoaked. Thus, a digester with a 1-5 minute residence time preceeds the disk refiner. The chip steaming treatment produces acetic acid by the cleavage of acetyl groups in the wood, thereby improving the digestibility of the wood due to the hydrolytic effect of the acetic acid. The effectiveness of that treatment varies with the ash content of the wood, which has a buffering effect on the acid. It has been assumed that 25 percent of the available hemicellulose acetyl groups are converted to acetic acid in 1 minute at 200°C.

Cleavage of methoxy groups in the lignin fraction occurs at these conditions. Based on measurements made at Dartmouth(5), a 0.05 weight percent methanol yield based on dry wood is taken for the prehydrolysis product. Also, some of the lignin is rendered soluble by steam treatment and prehydrolysis. This has been estimated at 10 percent.

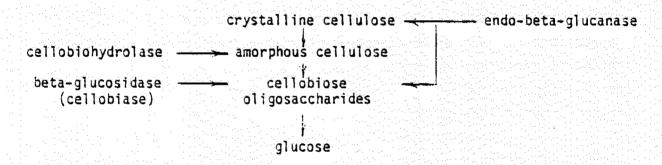
## C. Enzyme Hydrolysis

Recent advances in enzyme technology, specifically in enzyme productivity, have made enzyme hydrolysis an attractive route for cellulose conversion. Steam pretreatment of cellulose substrates and mild acid prehydrolysis have enabled glucose yields via enzyme hydrolysis to approach theoretical in a relatively short residence time of around 24 hours. Enzyme

production costs, once prohibitively high, have been markedly decreased recently by significant increases in enzyme productivity utilizing new mutant strains of the fungus, Trichoderma reesei.

Genetic researchers at Rutgers University have developed a mutation of  $\underline{T}$ . reesei, RUT C-30, which has significantly higher enzyme productivity than the previously used strain, QM9414. In addition, experimentation with various types of nutrient media for enzyme production has enabled cheaper ingredients, such as corn steep liquor and steam exploded wood, to be substituted for the more expensive ingredients previously used such as proteose peptone and delignified cellulose.

The mechanism of enzyme hydrolysis by <u>T. reesei</u> enzymes occurs when crystalline cellulose is attacked by the enzymes endo-beta-glucanase and cellobiohydrolase forming oligosaccharides, including cellobiose. The cellobiose is then acted upon by endo-beta-glucanase and beta-glucosidase to form glucose. This is represented as:



Therefore the enzyme mixture necessary to obtain optimum saccharification is a complex mixture of several enzymes in various proportions. RUT-C-30 has high enzyme productivity, but produces essentially all endo-beta-glucanase with little beta-glucosidase. Natick labs<sup>(6)</sup> has experimented with various mutant strains and has developed QM329 from Aspergillus phoenicis, which produces beta-glucosidase at very high productivities. Ideally, the optimum enzyme mixture would be a combination of endo-beta-glucanase from RUT C-30 and beta-glucosidase from QM329 in a ratio of cellulase:cellobiase of about 15:1.

Enzyme hydrolysis following dilute acid prehydrolysis pretreatment has been thoroughly investigated at Dartmouth (7). Based on the prehydrolysis conditions described earlier, a glucose yield of 90 percent of theoretical can be achieved in 24 hours at  $50^{\circ}$ C with a pH of 4.8. The initial solids concentration can be 10 weight percent. A cellulase loading of 15 international units per gram of solids is required while the cellobiase loading is 0.9. It is assumed that all the xylan present is converted to xylose.

#### D. Fermentation

The fermentation process involves the conversion of simple hexose sugars to ethanol and carbon dioxide. The many possible hexose sugars are distinguished from one another by subtle structural differences. Relatively few hexose sugars occur naturally in large quantities, but of these the most important for fermentation are glucose, fructose, mannose and galactose.

A great deal of research has been devoted to investigating the metabolism of sugar by yeast. As a result, the complex transformations that take place and the enzymes responsible for these changes are now better understood than for any other organism. The yeast fermentation reactions and their interrelationships are usually designated by the Embden-Meyerhof-Parnas scheme, which shows the importance of organic phosphates (Figure III-D-1).

Most industrially important microorganisms metabolize carbohydrates by means of the EMP pathway to pyruvic acid. The overall reaction may be represented by the following equation:

$$c_6^{H}_{12}^{O}_{6}$$
 + 2 DPN + 2 ADP + 2 P<sub>1</sub> 2 CH<sub>3</sub>CO COOH + 2 H<sub>2</sub>DPN + 2 ATP

Since both DPN and ADP are present in only small quantities in the microbial cell, they must be regenerated if metabolism is to continue. The terminal phosphate bond of the ATP is used by energy-requiring reactions in the cell, thus liberating ADP for further phosphorylation. Alternatively, it can by hydrolyzed by ATPase to ADP and inorganic phosphate.

#### FIGURE III-D-1

#### EMBDEN - MEYERHOF - PARNAS PATHWAY

```
30 Tysacchar ide
                                           = Glucose
                                                * + ATP (Hexok inase)
         ± Hg PO4(Phosphorylase)
Glucose -1-Phosphate
                                              Glucosa-6-Phosphate + ADP
                      (Phosphoglucomutase)
   (Cori Ester)
                                                  Robinson Ester)
                                                (Phosonomexose Isomerase)
                                             Fructose-6-Phosonate
                                                (Neuberg Ester)
                                                    + ATP Phosphonexok (nase)
                                              Fructose-1.6-0 tonosonate - 40P
                                                (Harden-Young Ester)
                                                     (Aldolase)
                                                  Hz - OPN
                               CH-, CHOH-CHO
CH2+ CO+CH2OH ...
                                                             CH- CHOH-CH-OH
3P03HZ
                              OPO3HZ
                                                           3P0<sub>3H2</sub>
                √(somerase)
                                                    + JPN
                               3 - Giyceraldenyde
Dinydroxyacetone

    Glycerophosonate

   Phosphate
                                                                   + ก่∍ปี (Phosphatase:
                                    ± H<sub>3</sub>204
                                                              SHISH CHOMPCHISH - HIPS.
                                        CPC_H2
                               снь снон-сной
                                                             Glycardi
                               JPC<sub>3</sub>HZ
                               1.3-Diphosphoglyceraldenyde
                                - OPN (Trioseonosonate Denydrogenase
                               JPÖ<sub>3</sub>H<sub>2</sub>
                               1.3-Otomosomoglyceric acta
                               + ADP
                               CH4 CHOH COOH - ATP
                               OPO<sub>3</sub>H<sub>Z</sub>
                               3-Phosomoglyceric acid
                               Phosonoglycaromutasa:
                               сн<sub>э</sub>он-сн-соон
сро<sub>зна</sub>
                               1-Phosomogiyceris acis
                               * = 4,0 'Eno'ase
                               520545
250<sup>2</sup>45
                               ENOL - Phosphopyruvic acid
                               - 10b
       Lact's semycrogenase; + +2-0PN
                               0Hg+00+00CH - 4TP
Shighthon-cook -
                              Pymuvic acid

$ PPT Campoxy ase

Dmg+CHO + DDg
   _actit_acti
```

TMG— on MM— lons required

170 Adenosine triphosphate

120 Adenosine triphosphate

1304 Inorganic phosphate

1304 Ophosphopyridine rucleotide (cozymase)

12 OPN Reduced triphosphopyridine rucleotide

12 OPN Reduced triphosphopyridine rucleotide

13 OPN Reduced triphosphopyridine rucleotide

14 OPN Reduced triphosphopyridine rucleotide

The pyruvic acid formed by the EMP pathway is converted to ethanol and carbon dioxide by the following reactions:

• The enzyme carboxylase catalyzes a decarboxylation of pyruvic acid to acetaldehyde:

• The enzyme alcohol dehydrogenase, working in reverse, catalyzes a reduction of acetaldehyde to ethanol at the expense of reduced DPN:

The fermentation reaction is generally catalyzed by enzymes produced by special strains of yeast developed for efficient fermentation. Although there is much research at present on the production of ethanol using fungi and bacteria, all commercial grain ethanol fermentation processes are presently based on yeast. The most commonly used are strains of Saccharomyces cerevisiae. The criteria for good distillery yeasts are high alcohol and sugar tolerance, efficient conversion at temperatures of at least 32°C, and general hardiness to fairly extreme changes in conditions such as pH, temperature, and osmotic pressure. Fortunately, a relatively high percentage of yeast strains meets these criteria.

In a continuous cascade system with yeast recycle and an initial yeast concentration of 385 x  $10^6$  cells per milliliter it is assumed that all the glucose is converted. This requires an 18-hour residence time at  $30^{\circ}$ C with an initial glucose concentration of 13.2 weight percent. Adjusting the initial pH to 4.0 will result in the following selectivity structure:

 Conversion of 4.9 mole percent to glycerol, acetaldehyde and carbon dioxide via:

$$C_{6}H_{12}O_{6} \longrightarrow C_{3}H_{8}O_{3} + C_{2}H_{4}O + CO_{2}$$

- Conversion of 0.1 mole percent to fusel oils.
- Remaining 2.0 mole percent utilized for yeast growth with equal weight of carbon dioxide evolution.

The net result is approximately 47.5 pounds of ethanol produced and 47.6 pounds of carbon dioxide produced per 100 pounds of glucose.

#### E. Furfural Production

The formation of furfural is accomplished by dilute acid hydrolysis of xylose. As explained in Section III-B, this reaction is accompanied by condensation products such as resinous tars and decomposition products such as formic acid. The overall kinetics are dependent on the concentration of potential xylose in the feed with higher furfural yields obtainable at lower xylose concentrations.

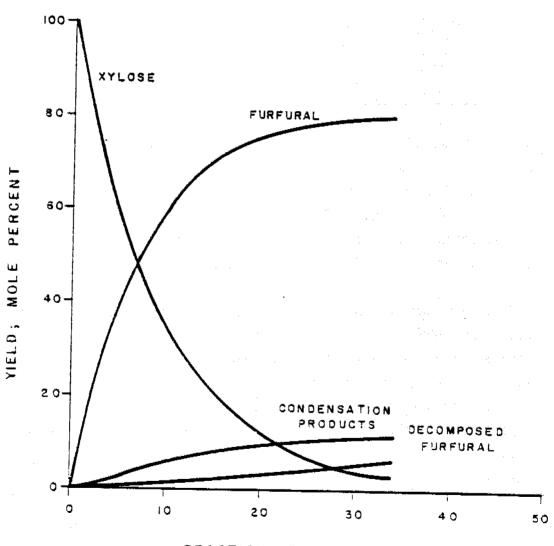
Furfural production takes place in the liquid phase at temperatures in excess of 200°C with sulfuric acid concentrations of 0.5-1.5 weight percent. The residence time can vary from 10 seconds to 10 minutes, depending on reactor temperature. The effect of residence time on furfural and by-product yields is shown in Figure III-E-1 for a 1 weight percent xylose solution; based on equations 3-7 in Section III-B. At this xylose concentration, furfural yields of nearly 80 mole percent are possible.

A more concentrated xylose stream of nearly 5 weight percent, such as encountered in the beer still bottoms in the current design, will result in a 47.6 mole percent furfural yield. This can be accomplished with a

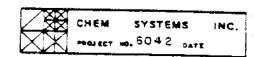
#### FIGURE III-E-1

# XYLOSE DECOMPOSITION AT 240°C WITH 1% ACID

1% XYLOSE



REACTION TIME; SECONDS



70-second residence time at 220°C and a sulfuric acid concentration of 1 weight percent. Obviously, the furfural yield could be increased by dilution, but the plant would be penalized by increased steam requirements.

#### IV. PROCESS DESIGN

#### A. Process Description

This section provides a process description of a plant designed to produce 25 million gallons per year of anhydrous ethanol from mixed hardwood. Figures IV-1 through IV-11 are process flowsheets for the plant and should be referred to with the following process descriptions. The detailed material balance is located in Section X-A and the design basis for the major processing units is detailed in Section IV-B.

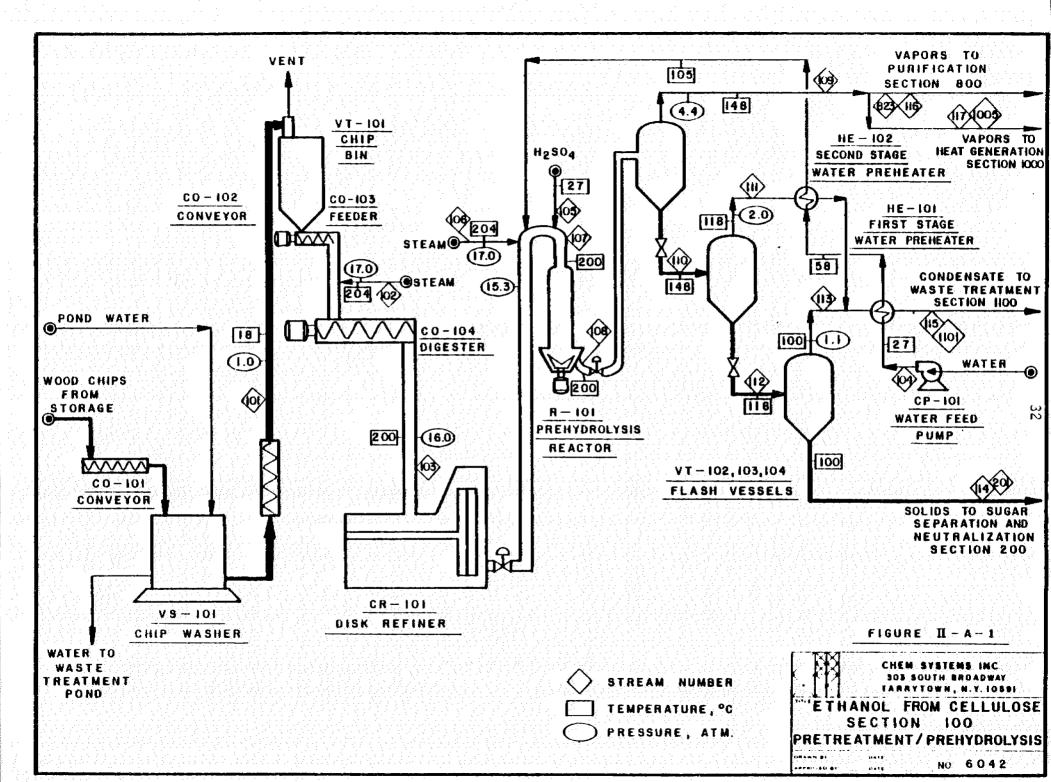
# 1. Pretreatment/Prehydrolysis (Section 100) (Figure IV-A-1)

The pretreatment/prehydrolysis section can be subdivided into four subsections:

- Chip washing
- Disk refining
- Prehydrolysis
- Reactor effluent flashing

Mixed hardwood chips are transferred from the storage area via a conveyor to a chip washer, VS-101, where any entrained dirt accompanying the wood chips is removed. Untreated water from the waste pond is used for this purpose and the effluent is returned to the waste pond. The washed wood chips are transferred to chip bin VT-101 via a conveyor (Stream 101).

Wood chips are fed from the chip bin to digester CO-104 where they are mixed with 17 atm steam (Stream 102). The digester operates at 16 atm and 200°C with a residence time of one minute. This softens the wood and provides a pressure head for further processing. Thus, the softened wood chips (Stream 103) are discharged into single disk refiner CR-101 which reduces the particles to nominal 1 millimeter diameter short fibers. After size reduction, the wood particles are discharged through a ball valve into the prehydrolysis reactor.

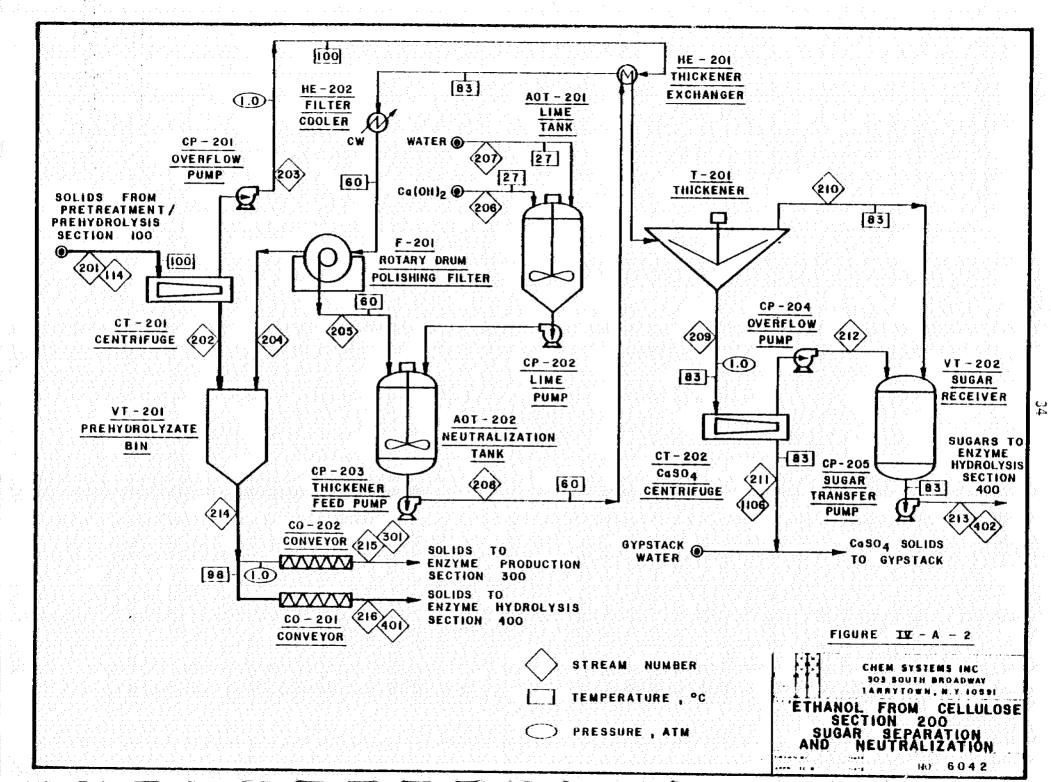


The prehydrolysis reactor, R-101, is designed for a 12-second residence time at 200°C and 15.3 atm. The wood particles are mixed with water (Stream 104) and steam (Stream 106) to achieve a 15 percent solids concentration and then sulfuric acid is added (Stream 105) to a concentration of 0.75 weight percent based on the water content. The water feed (Stream 104) is preheated by exchange with flash vapors in first stage water preheater HE-101 and in second stage water preheater H-102 to reach a temperature of 105°C prior to injection into the prehydrolysis reactor. All remaining heat is supplied by saturated 17 atm steam in Stream 106. Within the prehydrolysis reactor most of the hemicellulose fraction and the amorphous cellulose are converted to their respective sugars and some degradation products such as furfural. The reactor product (Stream 108) is flashed in three stages to remove some water/furfural vapor and simultaneously recover heat from the prehydrolysis reactor.

The first flash vessel, VT-102, operates at 4.4 atm and 148°C. Liquid/solid discharge (Stream 110) proceeds to the second flash vessel, VT-103, operating at 2.0 atm. Discharge from this flash (Stream 112) is further let down to 1.1 atm in VT-104. The effluent from the third flash (Stream 114), containing approximately 14 weight percent solids, passes to the sugar separation centrifuge in Section 200. Flash vapors (Stream 109) from VT-102 are sent to the multi-effect evaporator steam chest in Section 1000 (Stream 117) and to purification in Section 800 (Stream 116) as beer still preheat. Stream 111 from VT-103 is condensed against prehydrolysis feed water in second stage water preheater HE-102. The condensate is let down and combined with Stream 113 from the third stage flash. This stream is condensed against prehydrolysis feed water in first stage water preheater HE-101. The combined furfural/water condensate (Stream 115) proceeds to waste treatment in Section 1100.

## 2. Sugar Separation and Neutralization (Section 200) (Figure IV-A-2)

The sugar separation and neutralization section can be subdivided into three subsections:



- Sugar separation
- Neutralization
- Precipitated salts separation

Prehydrolyzed wood slurry from Section 100 is fed to centrifuge CT-201 to be separated into solid and liquid streams. The overflow (Stream 203) which contains most of the solubles and approximately 5 percent of the solids is pumped to rotary drum polishing filter F-201 via thickener exchanger HE-201 and filter cooler HE-202 which cool the stream to 60°C. The bottoms from CT-201 (Stream 202) containing 35 weight percent solids, are accumulated in prehydrolyzate bin VT-201. In the polishing filter, approximately 95 percent of the solids in Stream 203 are recovered and sent to VT-201 via Stream 204. The clean sugar solution (Stream 205) proceeds to neutralization. Prehydrolyzate solids (Stream 214) are split with 12 percent conveyed to enzyme production in Section 300 via Stream 215 and the remainder (Stream 216) conveyed to enzyme hydrolysis in Section 400.

In neutralization tank AOT-202, the acid in the sugar solution is neutralized with calcium hydroxide solution from lime tank AOT-201 (Streams 206 and 207). The neutralized sugar solution containing insoluble calcium sulfate (Stream 208) is pumped to a thickener via thickener exchanger HE-201 which reheats the stream to  $83^{\circ}$ C.

In thickener T-201, the calcium sulfate solids are concentrated to 20 weight percent and fed to a solid bowl centrifuge via Stream 209. The clear thickener overflow (Stream 201) is accumulated in sugar receiver VT-202. Calcium sulfate centrifuge CT-202 produces a high solids content salt (Stream 211) which is dumped into a slurry transfer line and transported to the gypstack. Drainage water from the gypstack is accumulated and recirculated through the slurry transfer line to transport the calcium sulfate solids. Overflow from CT-202 (Stream 212) is pumped to sugar receiver VT-202. The accumulated sugar solution from VT-202 is pumped via Stream 213 to enzyme hydrolysis in Section 400.

# 3. Enzyme Production (Section 300) (Figure IV-A-3)

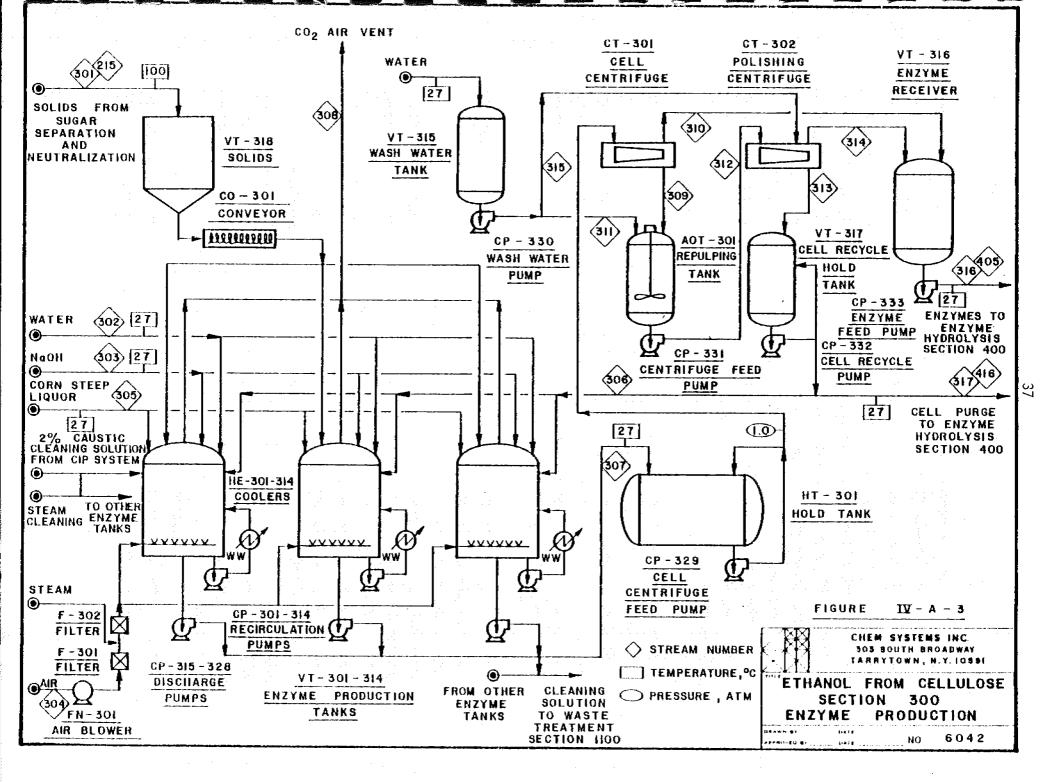
The cellulase enzyme complex, which is used for enzymatic hydrolysis of cellulose, is produced by the microorganism RUT C-30 in a fed batch production system. A portion of the prehydrolyzed wood feedstock is used as the carbon source for enzyme growth. The fed batch system differs from a conventional batch system in that fresh solid feed is added to the enzyme production tanks periodically over the course of the total batch. In this way the actual total solids concentration being processed is much higher than that obtained for a conventional batch system, since fresh solids are periodically added as soon as the previously added solids are consumed.

The enzyme production section can be subdivided into two subsections:

- Enzyme fermentation
- Enzyme separation and cell recycle

The fed batch production system is designed such that the solids feed stream is essentially continuous. Thirteen separate enzyme fermentation tanks are used, with one additional tank being out of operation as a reserve. The total batch cycle time for each tank is 13 days or 312 hours, including emptying and sterilization for the next batch. One batch is started each day with filling taking four hours. Fresh solids are added to each tank approximately every 48 hours or six times during the entire batch cycle. One batch is emptied each day after 296 hours on stream. Emptying takes 4 hours and then the remaining 12 hours before starting the next batch are used for sterilization and cleaning. Dilution water, nutrients, and cell recycle are added only once for each cycle during the initial filling of the tank. Fresh solids and caustic for neutralization are added six times during each cycle.

Solids from sugar separation and neutralization (Section 200) which serve as the carbon source for enzyme growth, are continuously fed to solids bin, VT-318 (Steam 301) prior to being fed to the individual enzyme production tanks. The solids are then fed to the enzyme production tanks,



VT-301-313, in the manner previously described, along with nutrient corn steep liquor (Stream 305), recycle cells (Stream 306), caustic (Stream 303), and dilution water (Stream 302). The aerobic fermentation oxygen requirement is met by sterilized air (Stream 304) being continuously sparged through the enzyme production tanks via air blower FN-301 and filters F-301 and F-302 which also provides agitation in the tanks. Enzyme production pH is controlled at 5.0 by the addition of the caustic, and temperature is controlled at 28°C by an external cooling line through heat exchangers HE-301-314. Excess air, nitrogen and the CO2 formed during enzyme production are vented from the tanks to the atmosphere (Stream 308).

Initial cellulose concentration in the fermenters is about 1.7 weight percent, and total solids concentration, including recycle solids, is about 9.4 percent. Cell density is initially maintained at 7 grams per liter, and inital corn steep liquor concentration is 3 weight percent. Final enzyme concentration is 15 IU/ml which translates to an enzyme productivity of about 52 IU/liter/hour.

The enzyme fermenter effluent (Stream 307) is discharged from the tanks via pumps CP-315-328 to hold tank, HT-301. The enzyme solution is then fed to the solid bowl cell centrifuge, CT-301, where the mycellium and other solids are removed from the enzyme solution. The centrifuge bottoms (Stream 309), which is 15 weight percent solids, is repulped to 10 percent solids by addition of water (Stream 311) in repulping tank, AOT-301. The repulped solids (Stream 312) are sent to a washing polishing centrifuge, CT-302, where solids are once again removed as the bottoms (Stream 313), and sent to the cell recycle hold tank, VT-317. The overflow from the polishing centrifuge (Stream 314) is combined with the overflow from the cell centrifuge (Stream 310) in the enzyme receiver, VT 316, and sent to enzyme hydrolysis, Section 400 (Stream 316). The cells are discharged from the cell recycle hold tank via pump CP-332, part of which is recycled to the enzyme production tanks (Stream 306). The remaining cells are purged (Stream 317) to the dewatering press in enzyme hydrolysis, Section 400. The cell recycle rate is approximately 55 percent of the total cell stream.

## 4. Enzyme Hydrolysis (Section 400) (Figure IV-A-4)

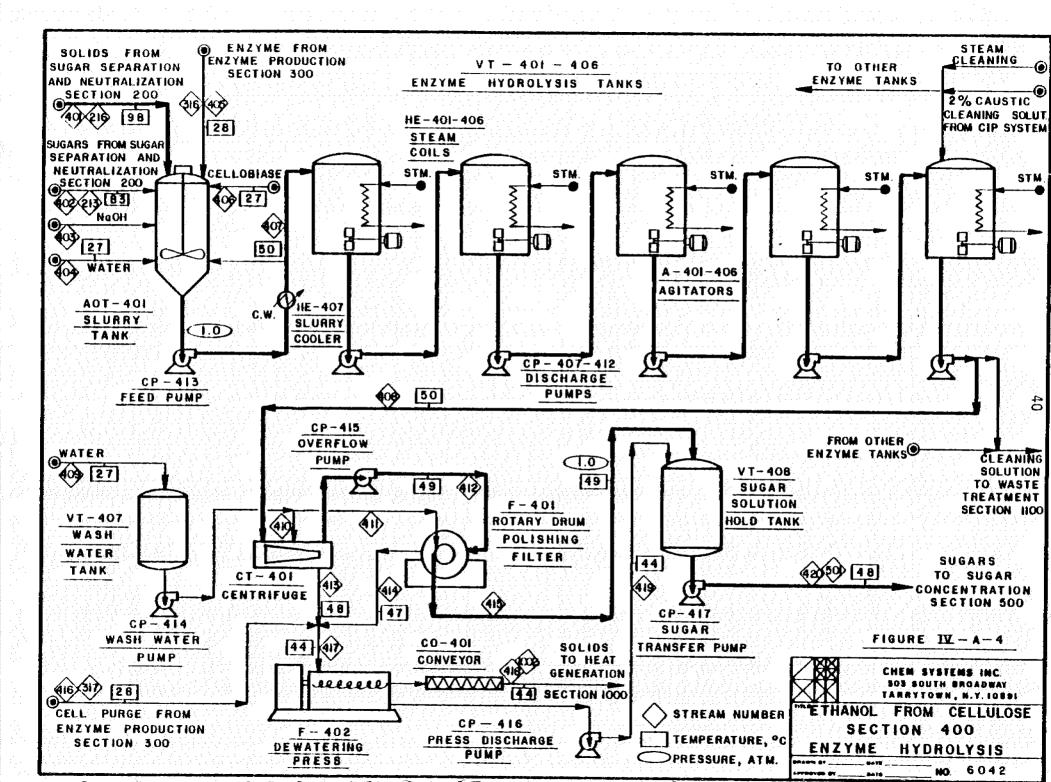
The enzyme hydrolysis section can be subdivided into two subsections:

- Enzyme hydrolysis
- Sugar separation

The solid stream (Stream 401) and sugar stream (Stream 402) from Section 200, sugar separation and neutralization, are fed to slurry tank, A0T-401. Slurry pH is controlled at 4.8 by addition of sodium hydroxide/calcium hydroxide (Stream 403) such that the final slurry is slightly below the saturation point of calcium sulfate. Cellulase enzymes from Section 300, enzyme production, (Stream 405) and make up NOVO cellobiase enzyme (Stream 406) are also added to the slurry, along with sufficient dilution water (Stream 404) to bring the total solids concentration to 10 weight percent. Temperature is controlled at  $50^{\circ}$ C via an external cooling line on the slurry tank through HE-407.

Enzyme hydrolysis takes place in a continuous cascade system in which partial hydrolysis occurs in each of a series of five tanks, until hydrolysis is complete in the last tank. The pH controlled enzyme hydrolysis feed (Stream 407) is pumped from the slurry tank to the enzyme hydrolysis tanks, VT-401-406. Enzyme hydrolysis takes place at 50°C with a cellulase loading of 15 IU per gram of solids and supplemented by a cellobiase loading of 0.9 IU per gram of solids. Total hydrolysis residence time in the five tanks is 24 hours. Agitation is provided in the tanks by internal side-mounted agitators and temperature is controlled by internal heating/cooling coils. The enzyme hydrolysis conditions give a 90 percent of theoretical glucose yield with 10 percent cellulose remaining unconverted. All remaining xylan present in the hydrolysis feed is assumed to be converted to xylose.

The enzyme hydrolysis effluent (Stream 408), which contains about 6.9 percent glucose is sent to a series of centrifuging and filtering steps. The first separation takes place in a washing centrifuge, CT-401, where most of the lignin and other remaining solids are removed in the



centrifuge bottoms (Stream 413). The centrifuge overflow (Stream 412) is sent to a rotary drum polishing filter, F-401, where most of the remaining solids are removed. The bottoms from the centrifuge and filter, which contain about 35 percent solids, are combined with the cell purge (Stream 416) from enzyme production, Section 300, and sent to dewatering press F-402 (Stream 417). The dewatering press squeezes out additional liquid to bring the solid concentration to about 55 percent. This solid stream (Stream 418) is sent to heat generation, Section 1000. The filtrate from the polishing filter (Stream 415) and the liquid from the dewatering press (Stream 419) are sent to the sugar solution hold tank, VT-408, prior to concentration. The feed to the sugar concentrator, Section 500 (Stream 420) contains about 6.8 percent glucose.

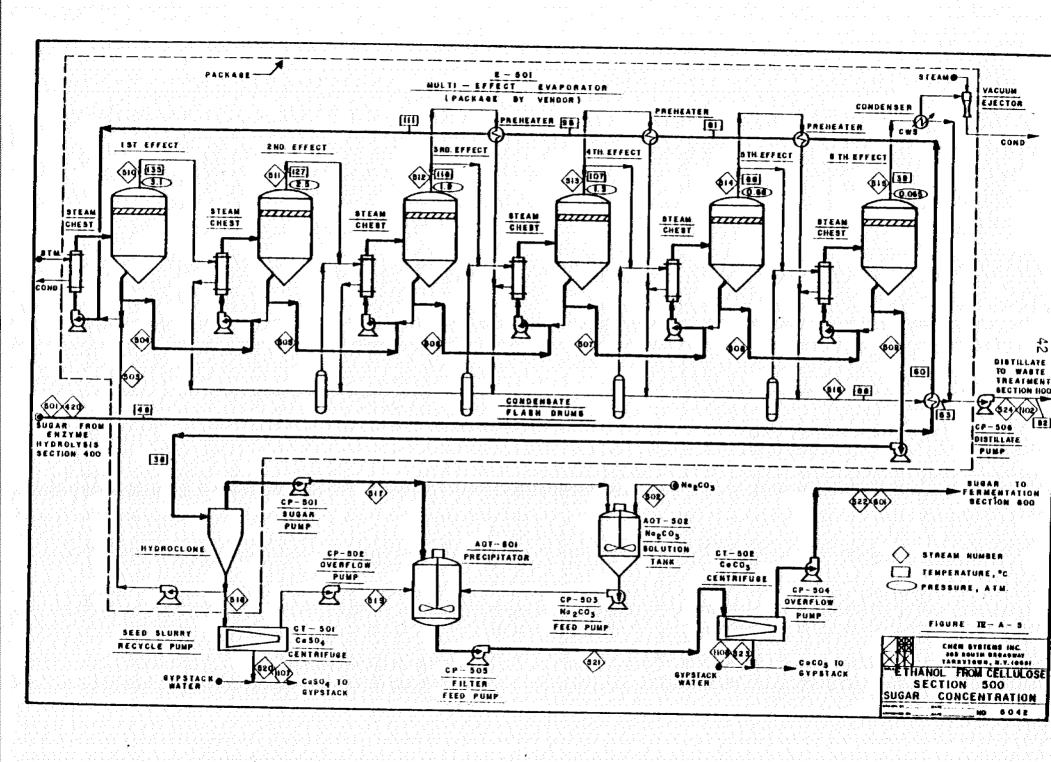
## 5. Sugar Concentration (Section 500) (Figure IV-A-5)

The sugar concentration can be subdivided into two subsections:

- Sugar concentration
- Salt separation

The 6.8 weight percent sugar solution from enzyme hydrolysis is concentrated to 15 weight percent in multi-effect evaporator E-501. This evaporator is a six-effect, external forward forced fed system, constructed of 304 SS based on vendor recommendations for this service. The first effect operates at 3.1 atm and the final effect under vacuum at 0.065 atm.

Several heat recovery steps are employed to minimize the overall steam duty. These include the flash of the steam chest condensate after each effect such that total distillate from effects two through five has been reduced to  $38^{\circ}\text{C}$  (0.66 atm). Also the cold feed (50°C) is preheated in a series of exchangers to  $111^{\circ}\text{C}$ . The flowsheet depicts a complete vendor package for the evaporator including heat recovery equipment. In the operation of the concentrator system the feed (Stream 501) is first heat exchanged against the  $88^{\circ}\text{C}$  distillate (Stream 516) then further heated to  $111^{\circ}\text{C}$  using an overhead vapor bleed from the fifth, fourth,



and third effects, respectively. This scheme serves to reduce the amount of unrecoverable heat from the steams (i.e., the 6th effect distillate, Stream 515, is vapor at  $38^{\circ}$ C).

The hot feed then enters the first effect. In each of the next effects the exchanger is driven by the overhead of the previous effect. Also, the condensed distillate from each steam chest is flashed and fed to the next heater for the purpose of heat recovery.

Since the feed stream is saturated with calcium sulfate, this salt will precipitate as the liquor is concentrated. To minimize fouling of the exchangers, a crystal seeding process is employed. This is accomplished by feeding the liquor out of the sixth effect to a hydroclone where a clear solution is taken overhead (Stream 517) and a 40 weight percent calcium sulfate stream is taken off the bottom. A portion of the bottoms is recycled back to the first effect such that the final effect on the evaporator contains an 8 weight percent slurry of calcium sulfate salt. This provides seeds on which the precipitating solids build rather than plating on the equipment walls. As this system does not totally prevent fouling, the heat exchangers must be cleaned periodically.

A slipstream (Stream 518) from the hydroclone bottoms, containing the net amount of calcium sulfate in the feed (Stream 501), is sent to the calcium sulfate centrifuge, CT-501. The solids stream containing 77 weight percent organic and inorganic solids (Stream 520) is sent to the gypstack.

The liquid from CT-501 (Stream 519) along with the clear liquid from the hydroclone (Stream 517) is pumped to the precipitator, AOT-501, where sodium carbonate is added such that the calcium sulfate is converted to calcium carbonate. The reason for calcium carbonate conversion is that this salt is much less soluble than calcium sulfate. Since both have reverse solubilities, there will be less calcium salts that can plate out in the downstream beerstill as the stream is heated. The insoluble salts are separated in the calcium carbonate centrifuge, CT-502. The solids bottoms containing 90 weight percent calcium carbonate (Stream 523) are sent to the gypstack while the liquor is sent to fermentation in Section 600 (Stream 522).

## 6. Fermentation (Section 600) (Figure IV-A-6)

The 15 percent sugar solution from sugar concentration in Section 500 (Stream 601) is fed to concentrated sugar receiver VT-608. Sterile aeration is provided to this vessel through air blower FN-602 with filters F-601 and F-602 preventing external contamination. The sugar solution (Stream 603) is pumped through sugar cooler HE-607 into the fermenters. Recycle yeast is also added to the fermenters via stream 605.

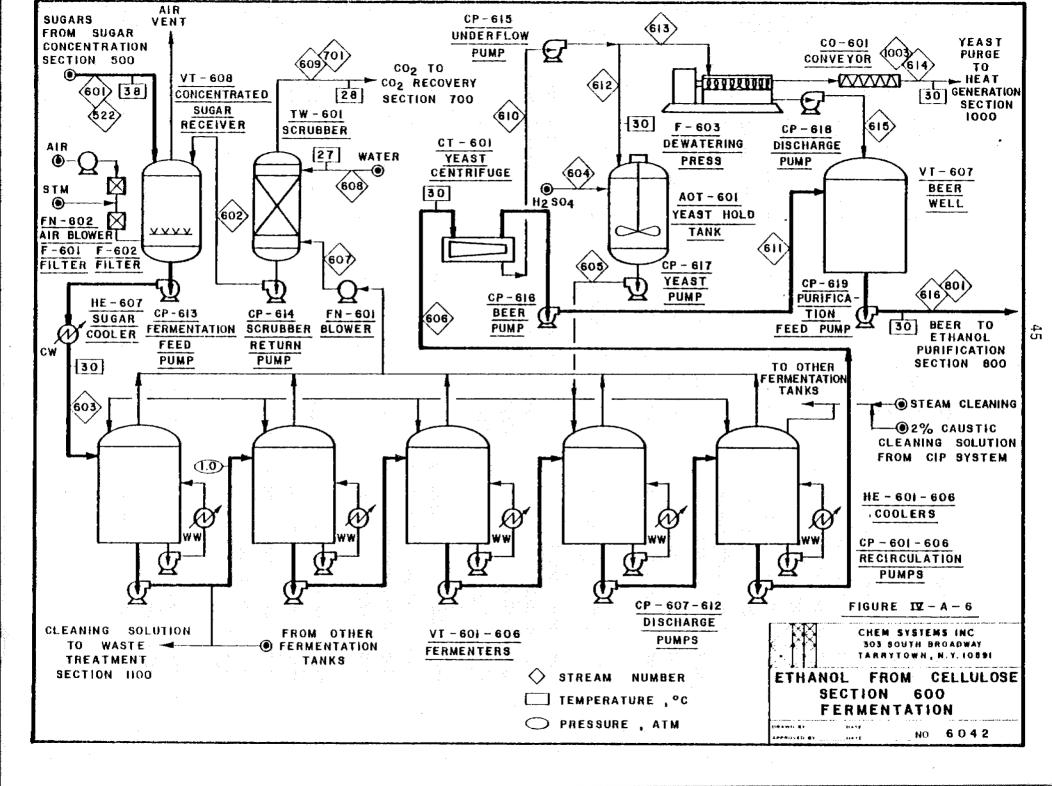
Fermentation takes place in a continuous cascade system consisting of six tanks, VT-601-606, with five tanks active at any one time and the additional tank down for sterilization. Partial fermentation occurs in each tank and is completed in the last tank. The total fermentation time is 18 hours. The yeast, Saccharomyces cerevisiae, is fed to fermentation at a concentration of 385 x  $10^6$  cells per millimeter or approximately 0.07 pounds per gallon. The pH in fermentation is adjusted to 4.0 primarily by addition of sulfuric acid to the yeast recycle stream. Well water cooling is provided to each fermenter through external coolers, HE-601-606 to maintain a temperature of  $30^{\circ}$ C.

In fermentaion, all of the glucose is converted with 93 percent selectivity to ethanol and carbon dioxide; 4.9 percent selectivity to glycerol, acetaldehyde, and carbon dioxide; 0.1 percent selectivity to fusel oils; and 2 percent selectivity to yeast growth and carbon dioxide production. Thus, the ethanol productivity is 0.475 pounds per pound of glucose and 0.476 pounds of carbon dioxide are produced per pound of glucose. Yeast growth is approximately 18 percent per pass. Thus, the entire yeast inventory can be changed out in less than a week, even without fresh yeast addition. Under these conditions, no yeast makeup is necessary.

Carbon dioxide evolved during fermentation (Stream 607) passes through scrubber TW-601 where entrained ethanol is absorbed in water from stream 608, pumped via stream 602 to concentrated sugar receiver VT-608 and then returned to the fermenters. The scrubbed carbon dioxide (Stream 609) is sent to carbon dioxide recovery in Section 700.







The beer exiting the fermentation tanks (Stream 606) has an ethanol concentration of 7.0 weight percent. This stream is sent to yeast centrifuge CT-601 where a 15 weight percent yeast (Stream 610) is separated from the ethanol containing beer (Stream 611). The beer is pumped to beer well VT-607. Approximately 90 percent of the yeast (Stream 611) is sent to yeast hold tank AOT-601 where sulfuric acid is added via stream 604 to bring the pH to 3. The yeast is held at this condition for four hours to kill off bacteria and then recycled to the fermenters via stream 605. A yeast purge (Stream 613) is sent to dewatering press F-603 where the solids concentration is raised to 55 weight percent. These solids (Stream 614) are conveyed to heat generation in Section 1000. The dewatered liquids (Stream 615) are sent to the beer well. From VT-607 the beer (Stream 616) is pumped to purification in Section 800.

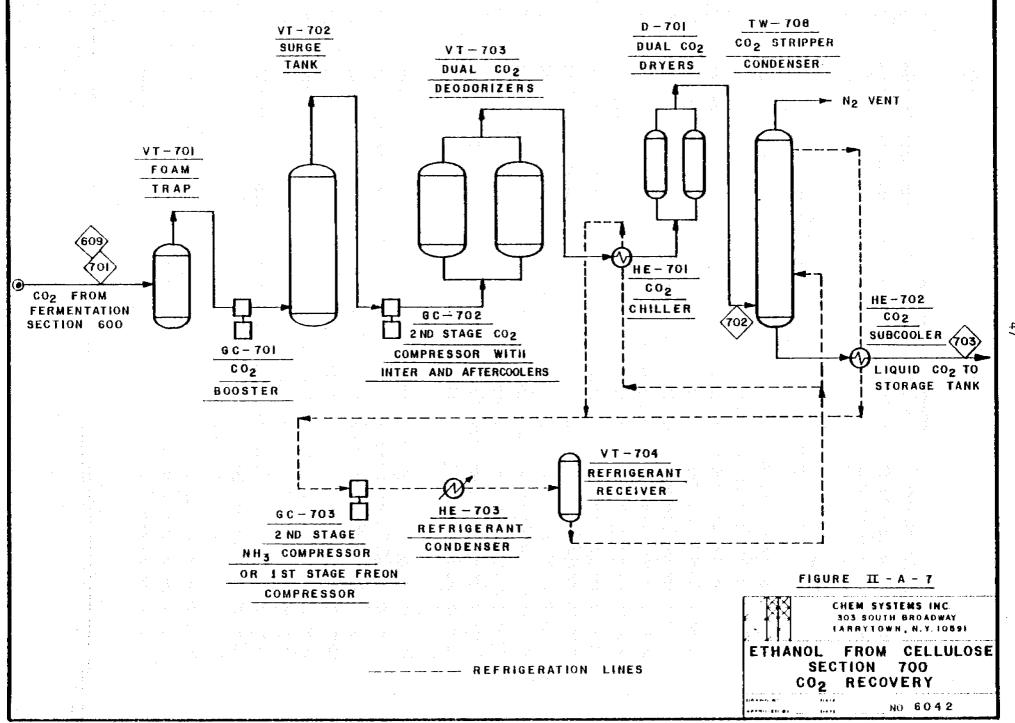
## 7. Carbon Dioxide Recovery (Section 700) (Figure IV-A-7)

When sugars are fermented to ethanol, large quantities of carbon dioxide are produced as a by-product of cell respiration. As the fermenters are usually closed vessels, it is possible to collect the off-gas and recover liquefied carbon dioxide for sale.

The carbon dioxide generated in the fermentation vessels (Stream 701) passes through a foam trap, boosted in pressure with a rotary positive displacement compressor and discharged to a surge vessel. The gas is then compressed to 21.4 atm in a nonlubricated, reciprocating compressor.

The compressed gas is deodorized in a twin-tower, activated-carbon absorption system to remove remaining impurities. The carbon beds are periodically regenerated using live steam. The purified gas is then chilled and dried in a conventional alumina bed system to a dew point of  $-60^{\circ}$ C.

The dry gas passes to a low-temperature stripper-condenser system, where the carbon dioxide is liquefied and separated from the non-condensible gases, mainly nitrogen, which are vented to the atmosphere. The pure



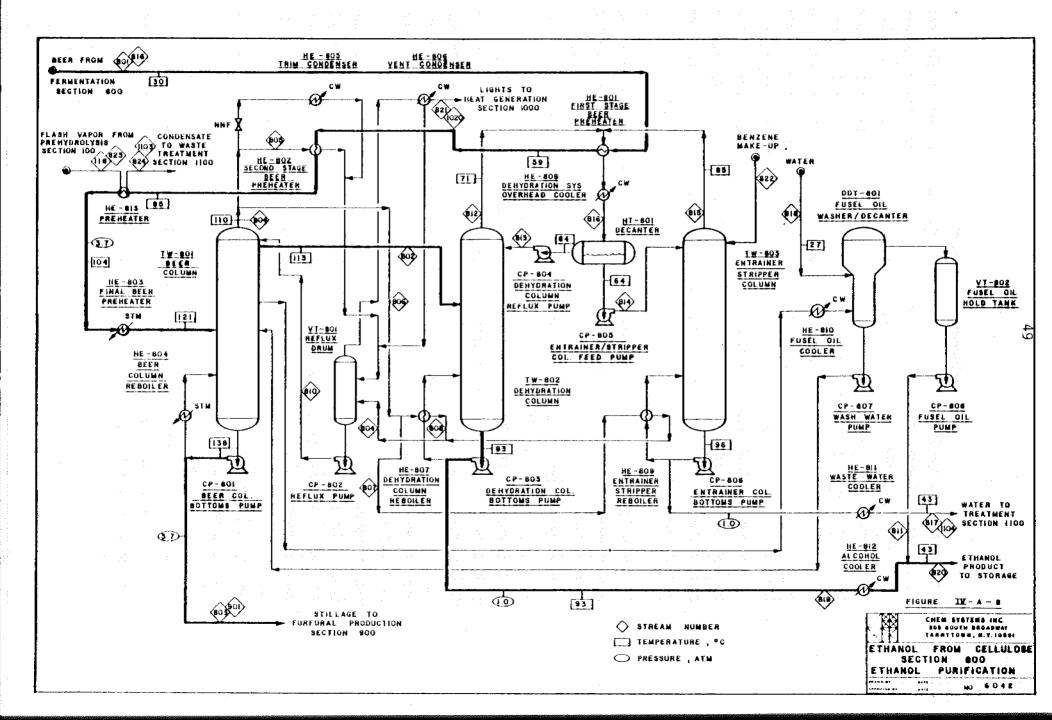
liquid carbon dioxide from the base of the stripper-condenser is then subcooled and sent to storage (Stream 703), where it is maintained under a pressure of about 21.4 atm.

## 8. Ethanol Purification (Section 800) (Figure IV-A-8)

The dilute beer from fermentation (Stream 801) undergoes a series of preheating steps that heat the feed from 30°C to 121°C. The first preheating step utilizes the combined overhead vapors from the dehydration column, TW-802, and from the entrainer stripper column, TW-803, in the first-stage beer preheater, HE-801. Approximately 28 percent of the total preheating is accomplished here, bringing the beer up to 59°C. The warmed beer then enters the second-stage beer preheater, HE-802, which uses the beer column overheads to supply another 30 percent of preheat, raising the beer temperature to 85°C. The next 22 percent of the preheat is supplied in preheater HE-813 by exchange against flash vapors from Section 100 (Stream 823) which raises the temperature to 104°C. From here the beer is preheated with steam in final beer preheater HE-803. This supplies the final 20 percent of beer preheat, raising the temperature to 121°C. All four stages of preheating use condensing vapors.

The hot, saturated dilute beer then enters the beer column, TW-801, which operates at 3.7 atm at the bottom. The column is composed of 60 sieve trays. The nonvolatile solubles and any suspended solids work their way down through the column and are removed from the bottom of the tower in a dilute aqueous stream. The temperature of the bottoms is 138°C. The hot stillage (Stream 803) proceeds to the reactor in furfural production (Section 900). Heat is supplied to the beer column by condensing steam in beer column reboiler HE-804.

The alcohol-rich vapors in the overhead of the beer column (Stream 804) exit as the binary azeotrope (94 percent ethanol and 6 percent water) at a temperature of about  $110^{\circ}$ C and a pressure of 3.4 atm. These vapors are



used to preheat the beer in the second-stage beer preheat (Stream 805) and as a source of heat for the reboilers of the dehydration tower (Stream 807) and entrainer stripper column (Stream 808), HE-807 and HE-809, respectively. Of the total overhead vapor generated, 40 percent is used for beer preheating, 57 percent is used in HE-807, and 13 percent is used in HE-809. The total condensate from these operations is collected in reflux drum VT-801 and returned to the top tray of the beer column (Stream 810) as reflux. The light (noncondensable) products of fermentation (Stream 821) such as aldehydes are removed from the system by venting from the reflux drum and sent to heat generation in Section 1000. These vapors enter vent condenser HE-806 to recover any alcohol entrained with them before they exit the distillation system.

The upper trays of the beer column operate under total reflux with the liquid product removed as a side draw about five trays from the top of the tower. This stream, which contains 92.9 percent ethanol and 7.1 percent water, then enters the mid-section of dehydration tower TW-802.

Also removed from the beer column are the fusel oils produced in fermentation. These higher alcohols are more volatile than ethanol in dilute aqueous solutions but less volatile than ethanol in concentrated alcohol solutions. As a result, they tend to concentrate on a tray in the beer column and must be removed. They are removed as a side draw, cooled in fusel oil cooler HE-810 and washed with cold water in the fusel oil washer/decanter, DDT-801. In this operation, the alcohol content of the fusel oils is washed off by countercurrent contact of the cooled fusel oils with a stream of cold water. The heavy aqueous stream, containing the recovered alcohol, is returned to the lower section of the beer column. The light organic (fusel oil) stream passes into fusel oil hold tank VT-802 for later blending with the ethanol product since it is to be used as a motor-grade fuel.

The dehydration system consists of a dehydration column, TW-802, and a small entrainer stripper column, TW-803. The entrainer is benzene and both columns operate at essentially atmospheric pressure.

The dehydration tower, TW-803, contains 50 sieve trays. The alcohol product bottoms (Stream 819) contains 99.5 percent ethanol and 0.5 percent water. This stream is cooled to 43°C in alcohol cooler HE-812, combined with the recovered fusel oils (Stream 811), and pumped to product storage (Stream 820).

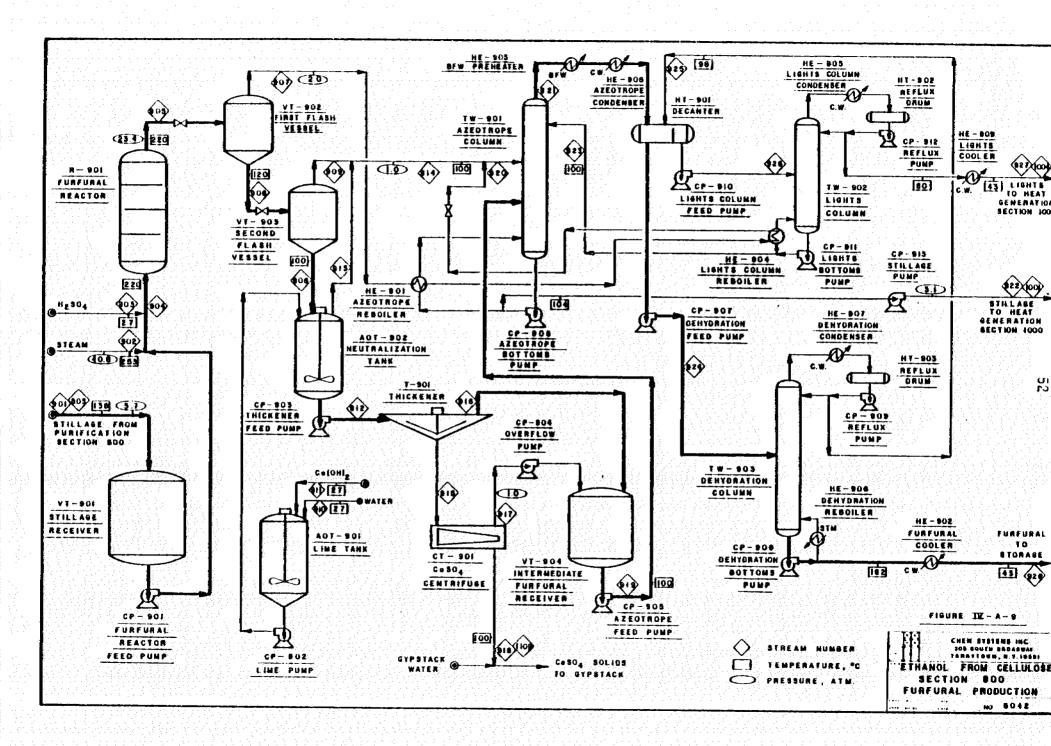
The overheads from dehydration tower TW-802 (Stream 812) is the tertiary minimum boiling azeotrope consisting of 71 percent ethanol, 17 percent water and 12 percent benzene at about 71°C. These vapors are condensed in HE-801 with the overhead vapors of entrainer stripper column TW-803 (Stream 815). These condensed vapors (Stream 816) are cooled to 64° in dehydration system overheads cooler HE-808 and enter decanter HT-801 where they separate into two layers. The upper layer (Stream 813) is hydrocarbon-rich and is pumped back to TW-802 as reflux. The lower layer (Stream 814) is the aqueous layer containing alcohol and some benzene. It is pumped to TW-803 as feed. Here, the remaining alcohol and benzene are stripped overhead and recycled to decanter HT-801. The entrainer stripper bottoms (Stream 817), essentially water, is cooled and sent to waste treatment in Section 1100. Heat is supplied to the dehydration system by condensing a portion of the overhead vapors of the beer column in reboilers HE-807 and HE-809, respectively.

## 9. Furfural Production (Section 900) (Figure IV-A-9)

The furfural production section can be subdivided into three subsections:

- Furfural production
- Neutralization
- Furfural recovery

Hot stillage containing approximately 5 weight percent xylose is fed from the beer still in Section 800 to stillage receiver VT-901. The stillage (Stream 901) is at  $138^{\circ}$ C and 3.7 atm. This stream is mixed with steam (Stream 902) to raise its temperature to  $220^{\circ}$ C and with sulfuric acid (Stream 903) prior to injection into furfural reactor R-901. The sulfuric acid concentration in the reactor feed is 1 weight percent.



The furfural reactor is a plug flow column with a 70-second residence time at 220°C. At these conditions approximately 94 percent of the xylose is converted with a 48 percent selectivity to furfural. The remainder is further converted to condensation and degradation products.

The liquid reactor effluent (Stream 905) is let down to 2 atm in first flash vessel VT-902. Flash vapors from this vessel (Stream 907) are condensed in azeotrope column reboiler HE-901 and lights column reboiler HE-904. The liquids from the first flash vessel (Stream 906) are further let down to 1 atm in second flash vessel VT-903. Flash vapors from this vessel (Stream 909) are combined with neutralization vapors (Stream 913) to form Stream 914. These vapors are combined with let down condensate from reboilers HE-901 and HE-904 and sent to the azeotrope column (Stream 920). The liquids from the second flash vessel (Stream 908) are sent to neutralization.

In neutralization tank AOT-902, calcium hydroxide is added from lime tank AOT-901 to form calcium sulfate. The heat of neutralization produces some vapors which are sent to the azeotrope column in Stream 913. The neutralized liquid proceeds to thickener T-901 where the soldis are concentrated to 20 weight percent and sent to calcium sulfate centrifuge CT-901 (Stream 915). The clarified liquid overflow (Stream 916) from T-901 is sent to intermediate furfural receiver VT-904 along with the liquid overflow from CT-901 (Stream 917). The calcium sulfate solids (Stream 918) from CT-901 at approximately 70 weight percent solids are slurried in drainage water from the gypstack and pumped to the gypstack.

Furfural solution (Stream 919) is pumped from the intermediate furfural receiver to azeotrope column TW-901. The total azeotrope column feed including Stream 920 contains approximately 1.5 weight percent furfural. This column operates at essentially atmospheric pressure and is composed of 69 sieve trays with a reflux ratio of 15.9. The furfural/water azeotrope is taken overhead along with some additional water. The overhead vapor (Stream 921), containing 99 percent of the furfural fed to recovery and composed of 23 weight percent furfural and 77 percent water, is condensed in BFW preheater HE-903 and azeotrope column condenser HE-908 at 98°C and sent to decanter HT-901 where it separates into two layers.

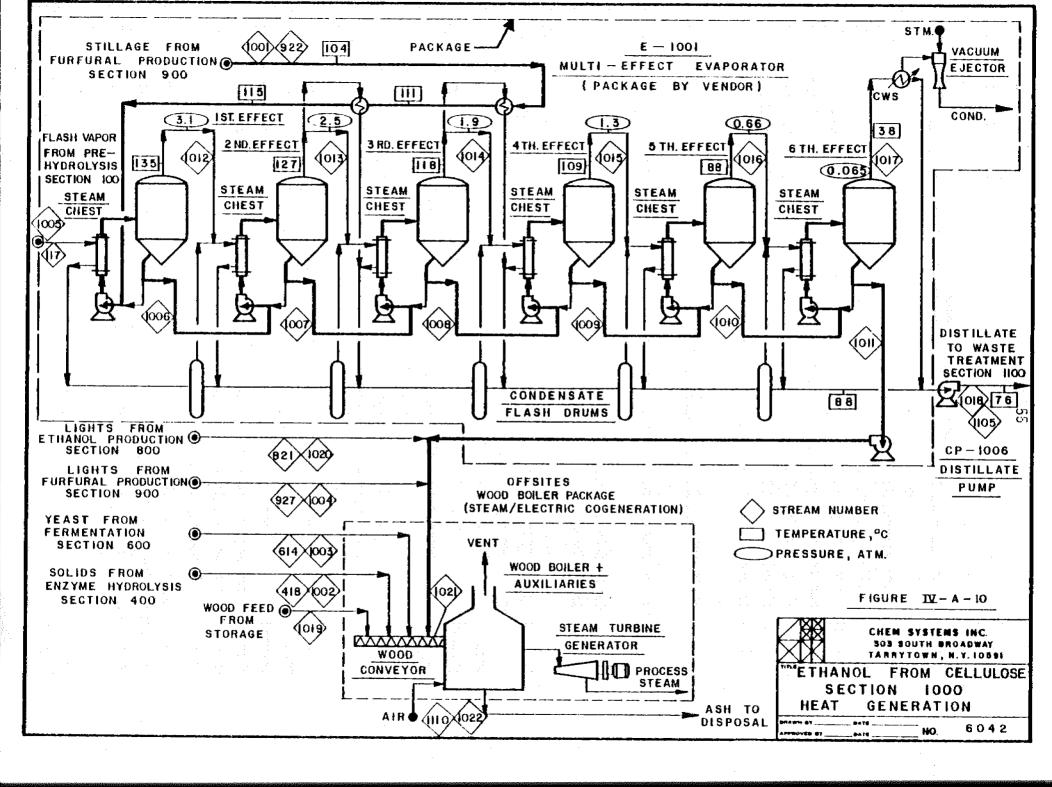
The lower organic layer composed of 84 percent furfural and 16 percent water is sent via Stream 924 to dehydration column TW-903. The upper layer composed of 18 percent furfural and 82 percent water (and possibly some lights) is sent to lights column TW-902 via Stream 926. Azeotrope column bottoms (Stream 922) is pumped to heat generation in Section 1000. Azeotrope column reboiler duty is provided via condensation of furfural reactor flash vapors in HE-901.

In lights column TW-902, any low boilers such as ethanol and methanol are removed in the overhead (Stream 927) and sent to heat generation in Section 1000. The column bottoms (Stream 923), which is essentially the aqueous layer from the decanter, is returned to the azeotrope column as reflux. This column operates at essentially atmospheric pressure.

The furfural-rich organic layer from the decanter enters dehydration column TW-903 where the remainder of the water is removed. This column operates at atmospheric pressure, has a reflux ratio of 0.34 and requires 10 sieve trays. The furfural product (Stream 928), which forms the column bottoms, is cooled to 43°C and sent to furfural storage. Its composition is 99.5 percent furfural and 0.5 percent water. The dehydration column overheads (Stream 925), which is the water/furfural azeotrope, is condensed and recycled to the decanter.

## 10. Heat Generation (Section 1000) (Figure IV-A-10)

The stillage from furfural production in Section 900 (Stream 1001) is concentrated to a 50 weight percent solution of organics in a forward feed multi-effect evaporator, E-1001. This evaporator consists of six effects operating with a forced circulation of the liquor. The system is designed with feed preheaters and condensate flash between effects to optimize the heat duty. The evaporator is a package system provided by a vendor and is constructed of carbon steel. In this system flash vapor from prehydrolysis in Section 100 (Stream 1005) is used as the heat source in the first effect.



The distillate (Stream 1018) is sent to waste treatment in Section 1100.

The concentrated organics from the evaporator (Stream 1011) are pumped to the wood boiler system, discussed in the offsite section. In addition to the concentrated organics, there are four other waste streams that are fed to the wood boiler to supplement the main wood feed (Stream 1019). These streams are: solids from enzyme hydrolysis (Stream 1002), yeast from fermentation (Stream 1003), lights from furfural production (Stream 1004) and lights from ethanol production (Stream 1020).

## 11. Waste Treatment (Section 1100) (Figure IV-A-11)

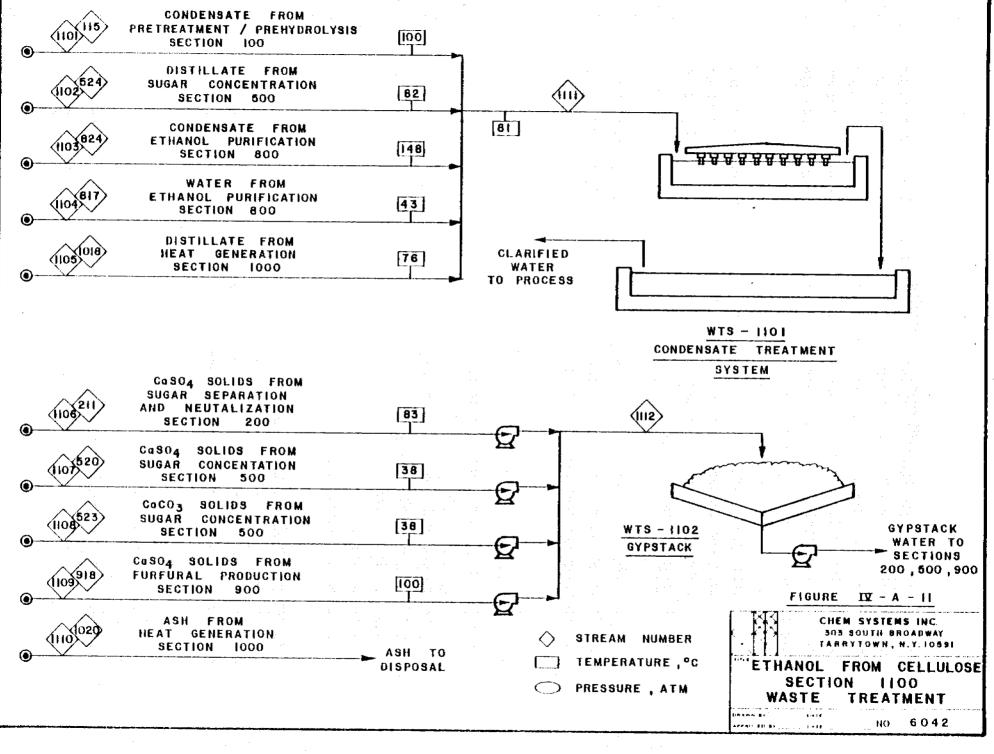
The waste treatment section can be subdivided into two subsections:

- Condensate treatment
- Solids handling

Five plant condensate streams are combined and sent to condensate treatment system WTS-1101. These streams are flash condensate from pretreatment/prehydrolysis in Section 100 (Stream 1101), multi-effect evaporator distillate from sugar concentration in Section 500 (Stream 1102), flash condensate from preheater HE-813 in ethanol purification (Stream 1103), stripper column bottoms from ethanol purification in Section 800 (Stream 1104), and multi-effect evaporator distillate from heat generation in Section 1000 (Stream 1105).

The combined condensate (Stream 1111) contains approximately 730 pounds per hour of waste organics with a BOD of 1,560 ppm. This stream is treated in an aerobic/anaerobic system to reduce the BOD level in the clarified effluent to 30-40 ppm. After treatment, this water is reused as process water in certain sections of the plant since the suspended solids will be approximately 20 ppm.

The gypstack, WTS-1102, receives waste solids from four process neutralization steps. These are calcium sulfate solids from sugar



separation and neutralization in Section 200 (Stream 1106), from sugar concentration in Section 500 (Stream 1107), and from furfural production in Section 900 (Stream 1109), and calcium carbonate solids from sugar concentration in Section 500 (Stream 1108). Each of these solids streams comes off a centrifuge and is slurried in circulating gypstack drainage water. The solids are pumped to the gypstack.

The gypstack is a lined solids storage area where the drainage liquor can be collected and pumped to the four process locations for transporting solids back to the gypstack. Approximately 6.5 tons per hour of dry solid waste is sent to the gypstack.

Ash from the heat generation steam boiler in Section 1000 (Stream 1110) can either be accumulated on the gypstack or sent directly to disposal. Approximately 1.5 tons per hour of ash are produced.

#### 12. Offsites

The offsites in this plant consist of:

- Raw material and product storage
- Wood feed and handling
- Utility systems
- Buildings
- Waste treatment

## Raw Material and Product Storage

The plant is provided with fourteen days storage for: ethanol, calcium hydroxide, sulfuric acid, caustic, corn steep liquor, furfural and sodium carbonate.

## Wood Feed and Handling

The total wood requirement to the plant (as a feedstock for the process and steam boiler) is 17,500 tons per week. The design is based on the

delivery of green wood chips to the plant. Based on a 5-day per week and 10-hour per day delivery schedule, the unloading facility is equpped with 2 hydraulic truck dumpers to unload 14 trucks total (25 tons per truck) per hour.

In this facility, the loaded trucks, after being weighed at the scale house, move onto the inclined dumper where the cabs are separated and the wood from the trailers is hydraulic dumped to the ground below the ramp. The chips are then moved into intermediate storage piles via front-end bulldozers. It is estimated that 20 acres are required to provide 60 days storage to allow for an inventory of wood during the bad weather months. From the piles, the wood chips are moved to a conveyor which continuously feeds a classifier where oversized chips are separated, crushed and recombined with the on-size chips. From here the chips are fed to three hold silos in parallel. From each silo the chips are metered to chip washers for the removal of dirt and sand. Finally, the washed chips are belt conveyed from each of the three washers to the refiner feed hoppers.

A separate conveyor from the classifier feeds wood to the steam boiler.

#### Utility Systems

#### Steam

The primary utility system in the plant is a steam boiler/power generator. The steam boiler is designed to produce 1,200 psia steam at 482°C which is let down to 600 and 250 psia through a steam turbine which in conjunction with a generator is used to generate electricity. The 600 and 250 psia steam is used in the process and the 250 psia steam is further let down to provide 65 psia steam where required.

The boiler system is a VU-40, CE-Power Systems type boiler fed with wood chips along with various organic waste streams as discussed in Section 1000. Based on a steam capacity of 269,000 pounds per hour of 250 psia steam and 46,100 pounds per hour of 600 psia steam, 12,400 kilowatts of power are generated. With a total plant demand of 13, 100 kilowatts, the net power import is 700 kilowatts.

## Cooling Water

The cooling water system includes a cooling tower, concrete basin, circulation pumps and field erection. The system is designed for a circulation rate of 27,000 gallons per minute which is an excess of 20 percent over known plant requirements, a supply temperature of  $30^{\circ}\text{C}$  and cooling water return temperature of  $44^{\circ}\text{C}$ .

#### Boiler Feedwater

The boiler feedwater system is to upgrade the quality of well water to the boiler. It includes water treatment (mixed bed quality deionizer) piping, pumps, tanks and installation. The capacity of 430 gallons per minute is an excess of 30 percent over total make-up of steam used directly in the process and 2 percent of recirculated process steam.

#### Electrical

Although 95 percent of the power requirements in the plant are provided by cogenerated power it is assumed that the plant will be tied into the main power grid to handle 25 percent of the plant power load. The electrical system provides for a tie-in to a main utility power line, transformer, substations and main breaker, metering systems, and main power cables.

## <u>Buildings</u>

This includes an office and administrative building, laboratory, change house and cafeteria, guard house, garage, maintenance shop and warehouse.

## General Utilities

This includes fire water pumps, instrument and plant air systems (compressors, dryers and surge drums), warehouse and shop cranes.

#### Site Development

The site development includes site clearing and assumes no unusual problems such as environmental problems or right of ways. Included in site development are fencing, curbing, parking lots, roads, wells, drainage, rail systems, borings and general paving.

#### Piping

This includes piping required for instrument and plant air headers, process water, fire water loop, flare system, process area tie-ins and interconnecting piping within the storage area.

#### Pollution Control

Pollution control is comprised mainly of the process waste treatment pond and the gypstack discussed in Section 1100. In addition there are sanitary waste treatment systems.

## B. <u>Design Basis</u>

A process design has been developed for a plant producing 25 million gallons per year of ethanol from mixed hardwoods. Eighty percent of the plant feedstock will be hardwoods from aspen forests and the remainder from maple/birch forests. As shown in Table IV-8-1, the average feedstock composition will be 57 percent aspen, 20 percent maple and 23 percent other hardwoods. On a moisture-free basis, the plant feed will contain approximately 47.2 percent cellulose, 31.3 percent hemicellulose and 13.5 percent lignin. The hemicellulose can be further broken down to 7.9 percent hexosans, 16.5 percent pentosans and 6.9 percent others. The moisture content is 50 percent.

The remainder of this section provides the design basis used to develop the material balances and major process equipment sizing.

TABLE IV-B-1

MIXED HARDWOOD COMPOSITION - MOISTURE FREE

	Trembling Aspen	Typical Hardw	ood Red	Maple	Average Plant Feedstock*	
Cellulose Hemicelluloses	51.5 29.6	43.0 33.1	40.0 33.9		47.2 31.3	
llexosans	7.2	8.2		9.5	7.9	
Glucan	1631 (141)		4.9	5.5		4.6
Mannan	National Property 2.2		2.3	3.4		2.5
Galactan	0.8		0.8	0.6		0.8
Pentosans	15.9	17.5		17.3	16.5	
Xylan	15.5 Line 1986		16.9	16.8		16.1
Arabinan	0.4		0.5	0.5		0.4
Other Hemis	6.5	7.5		1.1	6.9	
Uronic anhydride	3.2 3.2 3.2 3.2 3.2 3.2 3.2 3.2 3.2 3.2		3.9	3.4		3.4
Acetyl	######################################		3.6	3.7		3.5
Lignia	15.8 (1.1 )	[] <b>20.7</b>	23.4		18.5	
- Ash	0.2	0.3	0.2		0.2	
Extractives	2.9	2.9	2.5		2.8	
아이를 하면 되었다. 나라를 하고 못했다.	100.0	100.0	100.0		100.0	

\*Basis: 57% aspen, 20% maple, 23% other hardwoods.

## 1. Pretreatment/Prehydrolysis (Section 100)

#### Digester (CO-104)

Temperature 200°C

Pressure 16.0 atm
Residence time 1 minute

Maximum chip size 0.5" x 1" x 0.25"

#### Disk Refiner (CR-101)

Pressure 16.0 atm

Temperature 200°C

Power per dry ton per day 7 HP-day/ton

Power requirements are based on personal communications with C.E. Bauer.

## Prehydrolysis Reactor (R-101)

Temperature 200°C

Pressure 15.3 atm

Residence time 12 seconds

Solids concentration 15 weight percent

Acid conc. in solubles 0.75 weight percent

Total C<sub>6</sub> hydrolyzed 13.6 mole percent

Glucose yield on C<sub>6</sub> 13.4 mole percent

Total C5 hydrolyzed 70.8 mole percent

Xylose yield on C5 66.2 mole percent
Furfural yield on C5 4.5 mole percent

Furfural yield on C5 4.5 mole percent Lignin solublized 10.0 percent

actic sold wield an day and 0.075 and the

Acetic acid yield on dry wood 0.375 weight percent Methanol yield on dry wood 0.050 weight percent

Cellulose conversions and yields are based on kinetics developed by  $\operatorname{Dartmouth}(2,3)$ . Pentosan conversions and yields are based on kinetics developed by  $\operatorname{Dartmouth}(4)$ . Methanol production was measured by  $\operatorname{Dartmouth}(5)$ . Acetic acid production was assumed to be 25 percent of theoretical based on residence time in the digester and prehydrolysis reactor.

## First Flash Vessel (VT-102)

Temperature

148°C

Pressure

4.4 atm

## Second Flash Vessel (VT-103)

Temperature

118°C

Pressure

2.0 atm

## 2. Sugar Separation and Neutralization (Section 200)

## Centrifuge (CT-201)

Solids recovered

95 percent

Solids concentration

35 weight percent

## Rotary Drum Polishing Filter (F-201)

Solids recovered

95 percent

Solids concentration

35 weight percent

## Thickener (T-201)

Solids recovered

100 percent

Solids concentration

20 weight percent

## Calcium Sulfate Centrifuge (CT-202)

Solids recovered

95 percent

Solids concentration

83 weight percent

## 3. Enzyme Production (Section 300)

#### Enzyme Production Tanks (VT-301-313)

Type:

Fed batch

Temperature:

28°C

pΗ

5.0

Residence time

12 days

-- --, -

Cell density (initial)

7 grams/liter

Nutrient requirement

3 percent corn steep liquor

Cellulose concentration (initial)

1.7 percent

Total solids concentration (initial)

9.4 percent

Total solids concentration (final)

10.1 percent

Aeration rate

10.1 berceue

0.05 volume/volume/minute

Oxygen utilization

10 percent

Filter paper activity

15.0 IU/milliliter

Enzyme productivity

52 IU/liter/hour

Specific enzyme activity

0.64 IU/milligram enzyme

Carbon source requirement

0.17 grams enzyme/gram carbon source feed

109 IU/gram carbon source feed

Mycellium production

0.07 grams cells produced/gram cellulose

Cell recycle rate

55 percent

Carbon source is defined for the determination of enzyme production as the total prehydrolyzed wood feed which includes cellulose, lignin, ash, xylan, other insolubles and extractives. No credit is taken for any glucose or xylose present with respect to enzyme production and carbon source consumption. Only a fraction of the wood feed is actually utilized by the cells including cellulose, xylan and any sugars present. Complete consumption of these carbon source components is assumed. The carbon

source components consumed (cellulose, xylan, xylose and glucose) plus the oxygen utilized form enzymes, cells and carbon dioxide. Carbon source consumption is independent of type of enzyme production (fed batch vs. batch) and is only dependent on carbon source type (8,9). The carbon source, prehydrolyzed wood, produces enzymes at a carbon source consumption similar to steam exploded wood. Three percent corn steep liquor is used as the only nutrient source which satisfies all nutrient requirements and does not adversely effect enzyme production (3).

## Cell Centrifuge (CT-301)

Solids recovered 95 percent Solids concentration 15 percent

## Repulping Tank (AOT-301)

Solids concentration 10 percent

## Polishing Centrifuge (CT-302)

Solids recovered 95 percent
Solids concentration 15 percent
Wash rate on water and solubles 1.7 lbs/lb
Wash efficiency 95 percent

## 4. Enzyme Hydrolysis (Section 400)

## Enzyme Hydrolysis Tanks (VT-401-406)

Type: Continuous cascade

Temperature: 50°C

Residence time 24 hours

Solids concentration (initial) 10 percent

Cellulase loading 15 IU/gram solids
Cellulase loading 0.9 IU/gram solids

Cellobiase loading 0.9 IU/gram solids

Glucose yield 90 mole percent of theoretical

Enzyme hydrolysis yields are based on data published by Dartmouth (7). All acetic acid is neutralized with calcium hydroxide. Calcium sulfate formation via neutralization of sulfuric acid with calcium hydroxide is limited to an amount slightly below saturation at  $50^{\circ}$ C. Remaining sulfuric acid is neutralized with caustic. All xylan present is converted to xylose during enzyme hydrolysis. Cellulose not converted to glucose remains unconverted.

#### Centrifuge (CT-401)

Solids recovery 95 percent
Solids concentration 36 percent
Wash factor on water and solubles 0.8 lbs/lbs
Wash efficiency 88.5 percent

#### Polishing Filter (F-401)

Solids recovery 95 percent
Solids concentration 35 percent
Wash factor on water and solubles 0.8 lbs/lb
Wash efficiency 88.5 percent

## Dewatering Press (F-402)

Solids recovery 100 percent Solids concentration 55 percent

## 5. <u>Sugar Concentration (Section 500)</u>

## Multi-Effect Evaporator (E-501)

Type: Forward feed-forced circulation

Total evaporative capacity: 344,000 pounds H<sub>2</sub>0/hour

Feed glucose concentration: 6.8 weight percent

Product glucose concentration: 15 weight percent

#### Pressures

Steam chest first effect: 4.4 atm Steam chest second effect: 3.1 atm Steam chest third effect: 2.5 atm Steam chest fourth effect: 1.9 atm Steam chest fifth effect: 1.3 atm Steam chest sixth effect: 0.66 atm Vapor to condenser: 0.065 atm Vapor temperatures

First effect: 135°C Second effect: 12790 Third effect: 118°C Fourth effect: 107°C Fifth effect: 38°C Sixth effect: 38°C Steam: 148°C

### Hydroclone Bottoms (Part of E-501)

Solids recovered 100 percent Solids concentration 40 percent

## CaSO4 Centrifuge (CT-501)

Solids recovered 95 percent Solids concentration 77 percent

## CaCO3 Centrifuge (CT-502)

Solids recovered 95 percent Solids concentration 90 percent

## 5. Fermentation (Section 600)

## Fermenters (VT-601-606)

Total residence time

Temperature

Feed glucose concentration

Initial pH

Terminal ethanol concentration

Glucose conversion

Initial yeast concentration

Ethanol/CO2 selectivity

Glycerol/acetaldehyde/CO2 sel.

Fusel oil selectivity

Yeast growth/CO2 selectivity

Net ethanol per pound glucose

Net CO2 per pound glucose

Net glycerol per pound glucose

Net acetaldehyde per pound glucose

Net fusel oils per pound glucose

Net yeast growth per pound glucose

18 hours

30°C

13.2 weight percent

4.0

7.0 weight percent

100 percent

 $385 \times 10^6$  cells/milliliter

93 mole percent

4.9 mole percent

0.1 mole percent

2.0 mole percent

0.475 pound

0.477 pound

0.025 pound

0.012 pound

0.001 pound

0.010 pound

## Yeast Centrifuge (CT-601)

Solids recovered

Solids concentration

95 percent

15 weight percent

## Dewatering Press (F-603)

Solids recovered

100 percent

Solids concentration

55 weight percent

## 7. Carbon Dioxide Recovery (Section 700)

PACKAGE UNIT

## 8. Ethanol Purification (Section 800)

### Beer Column (TW-801)

Feed temperature 121°C

Overhead ethanol concentration 94 weight percent

Overhead temperature 110°C

Overhead pressure 3.4 atm

Bottoms temperature 138°C

Bottoms pressure 3.7 atm

### Dehydration Column (TW-802)

Entrainer Benzene Feed temperature 113°C Feed ethanol concentration 92.9 weight percent Overhead ethanol concentration 71 weight percent Overhead benzene concentration 12 weight percent Bottoms ethanol concentration 99.5 weight percent Bottoms temperature 93**0**C Bottoms pressure 1 atm

## Entrainer Stripper Column (TW-803)

Feed temperature 64°C Feed ethanol concentration 42 weight percent Feed benzene concentration 7 weight percent Overhead ethanol concentration 49.4 weight percent Overhead benzene concentration 8.4 weight percent Overhead temperature 85°C Bottoms benzene concentration 0.5 weight percent Bottoms temperature 96°C

## 9. Furfural Production (Section 900)

#### Furfural Reactor (R-901)

Temperature 220°C

Pressure 23.4 atm

Residence time 70 seconds

Acid concentration in solubles 1.0 weight percent

Xylose feed concentration 4.8 weight percent

Xylose converted 92.3 percent

Selectivity to furfural 51.6 mole weight

Xylose conversions are based on kinetics developed by  $\operatorname{Dartmouth}(4)$ .

#### First Flash Vessel (VT-902)

Temperature 120°C

Pressure 2.0 atm

Second Flash Vessel (VT-903)

Temperature 100°C

Pressure 1.1 atm

Thickener (T-901)

Solids recovered 100 percent

Solids concentration 20 percent

Calcium Sulfate Centrifuge (CT-901)

Solids recovered 95 percent

Solids concentration 61 weight percent

## Azeotrope Column (TW-901)

Feed furfural concentration 1.5 weight percent Feed temperature 100°C Reflux ratio 15.9 Actual trays 69 Overhead temperature 9800 Overhead pressure 1.0 atm Overhead furfural concentration 23 weight percent Bottoms temperature 104°C Furfural recovery in overheads 99 percent

Design parameters are based on column simulations developed by Smuk(10).

## Lights Column (TW-902)

Feed lights concentration	0.06 weight percent
Feed temperature	98°C
Actual trays	[15] 15] 차 (11] (12 H 의 [1

## Dehydration Column (TW-902)

Feed furfural concentration	84 weight percent
Feed temperature	98°C
Reflux ratio	0.34
Actual trays	10
Overhead temperature	98°C
Overhead pressure	1.0 atm
Overhead furfural concentration	35 weight percent
Bottoms temperature	162°C
Bottoms furfural concentration	99.5 weight percent

## 10. Heat Generation (Section 1000)

## Multi-Effect Evaporator (E-1001)

Type:	Forward feed-forced circulation			
Total evaporative capacity:	327,224 pound H <sub>2</sub> /hour			
Pressures				
Steam chest first effect:	4.4 atm			
Steam chest second effect:	3.1 atm			
Steam chest third effect:	2.5 atm			
Steam chest fourth effect:	1.9 atm			
Steam chest fifth effect:	1.3 atm			
Steam chest sixth effect:	0.66 atm			
Vapor to condenser:	0.065 atm			
Vapor temperature				
First effect:	135°C			
Second effect:	127°C			
Third effect:	118°C			
Fourth effect:	109°C			
Fifth effect:	88°C			
Sixth effect:	38°C			
Steam:	148°C			

## 11. Waste Treatment (Section 1100)

## Condensate Treatment System (WTS-1101)

Inlet suspended solids	1,000	PPM
BOD loading	1,560	PPM
Effluent suspended solids	20	PPM

Based on personal communications with EIMCO.

## C. Material Balance

A material balance for a plant producing 25 million gallons per year of ethanol from mixed hardwoods has been developed. The detailed material balance is presented in Appendix A by plant section. Overall raw material requirements are summarized in Table IV-C-1 and Table IV-C-2 summarizes plant product and waste effluent streams.

# TABLE IV-C-1 PROCESS RAW MATERIALS

	Pounds/Hour
Wood chips at 50 percent moisture To process - Stream 101 To boiler - Stream 1019	185,812 23,155 208,967
Sulfuric acid - anhydrous Stream 105 Stream 604 Stream 903	3,847 14 3,112
Calcium hydroxide - anhydrous Stream 206 Stream 403 Stream 911	6,973 2,477 350 2,366 5,193
Sodium hydroxide - anhydrous Stream 303 Stream 403	139 634 773
Sodium carbonate - anhydrous Stream 502	378
Corn steep liquor at 46 percent moisture Stream 305	1,473
Cellobiase at 65 percent moisture Stream 406	241
Benzene Stream 322	

The plant requires 2,508 tons per day of green hardwood chips. Approximately 89 percent of this feedstock is used in the process and the remainder is steam boiler fuel. Ethanol production is 75,000 gallons per day. This is a fuel grade product containing 0.5 weight percent water and 0.2 weight percent fusel oils. Two major by-products are also recovered, carbon dioxide and furfural. The carbon dioxide production rate is 244 tons per day. Approximately 55 tons per day of furfural are produced at a purity of 99.5 weight percent.

## TABLE IV-C-2 PRODUCT, BY-PRODUCTS AND WASTE STREAMS

		Pounds/Hour
Ethanol at 0.5 percent moisture Stream 820		20,576
Carbon dioxide - anhydrous Stream 703	en e	20,350
Furfural at 0.5 percent moisture Stream 928		4,562
Waste solids - anhydrous basis Stream 1112		12,580
Condensed organics - anhydrous basis Stream 1111		727
Ash Stream 1110		3,056

## D. Process Utility Summaries

## 1. Heat Balance Summary

The overall plant heat balance is summarized in Table IV-D-1 in terms of major process thermal interchanges. This table excludes heat exchangers that are located entirely within package units. Also, heat loads for sterlization, ejector operation and steam cleaning are not included in this table although they have been included in the utility summary.

## TABLE IV-D-1

## HEAT BALANCE SUMMARY

<u>Item</u>	Stream Heated	Stream Cooled M	M BTU/HR
 CO-104 R-101 HE-101 HE-102	Wood Chip Feed Prehydrolysis Feed Water to Prehydrolysis Water to Prehydrolysis	Steam Steam Flash Vapors Flash Vapors	40.7 53.5 17.3 26.2
 HE-201 HE-202	Neutralized Solids Cooling Water	Centrifuge Overflow Centrifuge Overflow	9.4 12.4
HE-301-314	Well Water	Enzyme Production	2.9
HE-407	Cooling Water	Enzyme Hydrol. Feed	24.4
E-501 E-501	Sugar Solution Cooling Water	Steam Evaporator Vapors	68.6 32.3
HE-607 HE-601-606	Cooling Water Well Water	Sugar Solution Fermentation	3.5 10.9
HE-801 HE-802 HE-813 HE-803 HE-804 HE-806 HE-807 HE-808 HE-809 HE-810 HE-811 HE-811	Beer Feed Beer Feed Beer Feed Beer Feed Beer Col. Reboiler Cooling Water Dehyd. Column Reboiler Cooling Water Stripper Reboiler Cooling Water Cooling Water Cooling Water	Dehydration Overheads Beer Col. Overheads Prehydrol. Flash Vapors Steam Steam Beer Column Vent Beer Col. Overheads Dehydration Overheads Beer Col. Overheads Fusel Oil Waste Water Ethanol Product	16.1 10.2 9.2 7.8 52.4 0.5 29.4 13.4 4.8 0.5 0.1
R-901 HE-901 HE-903 HE-908 HE-902 HE-904 HE-905 HE-906 HE-907 HE-909	Furfural Reactor Feed Azeotrope Reboiler Boiler Feedwater Cooling Water Cooling Water Lights Col. Reboiler Cooling Water Dehydration Reboiler Cooling Water Cooling Water	Steam Flash Vapors Azeotrope Col. Overheads Azeotrope Col. Overheads Furfural Product Flash Vapors Lights Col. Overheads Steam Dehydrated Overheads Lights Waste Stream	37.1 50.8 16.6 48.7 1.0 9.0 9.0 1.2 1.2 0.0
E-1001 E-1001	Stillage Cooling Water	Prehydrol. Flash Vapors Evaporator Vapors	44.3 67.8

#### 2. Steam Requirements

Process steam requirements are listed in Table IV-D-2. Three pressure levels (600 psia, 250 psia and 65 psia) are generated via letdown of 1,200 psia,  $482^{\circ}$ C steam through power generation turbines. The total steam required is 315,100 pounds per hour.

TABLE IV-D-2

UTILITY SUMMARY - STEAM
(Pounds per Hour)

<u>Item</u>	Name	65 psia	<u>250 psia</u>	600 psia
CO-104 R-101 Subtotal Se	Digester Prehydrolysis reactor ection 100		48,609 <sup>1</sup> 63,932 <sup>1</sup> 112,541	
F+302 Subtotal Se		2,400 <sup>1</sup> 2,400		
VT-401-406 Subtotal Se	Enzyme tanks ection 400	<u>500</u> 500	<u>.</u>	. <u> </u>
E-501 E-501 Subtotal Se		77,000 5001 77,500		<u>-</u>
F-602 Subtotal Se		 500 <sup>1</sup> 500		
HE-803 HE-804 Subtotal Se		3,590 - 8,590	63,500 63,500	
R-901 HE-906 Subtotal Se	Furfural reactor Dehydration reboiler ection 900	<u>.</u>	1,450 1,450	46,131 <sup>1</sup> 46,131
E-1001 Subtotal Se		500 <sup>1</sup> 500		<u></u> .
Steam cleanin Subtotal mi	scellaneous	1,500 <sup>1</sup> 1,500		
Total process Total exchang Plant total	ger steam	5,400 86,090 91,490	112,541 64,950 177,491	46,131
Steam gener	ated	91,490	177,491	46,131
<sup>1</sup> Process stea	ım			

## 3. Cooling Water Requirements

Cooling water requirements are summarized in Table IV-D-3. On the basis of a  $14^{\circ}\text{C}$  temperature rise, the total cooling water load is 22,700 gallons per minute. The cooling water supply is available at  $30^{\circ}\text{C}$ . Since approximately five percent of the cooling capacity is required at a temperature below  $30^{\circ}\text{C}$ , fresh well water is assumed for this supply which also acts as the make-up for the cooling water system.

TABLE IV-D-3

UTILITY SUMMARY - COOLING WATER

(14°C Temperature Rise)

<u>Item</u>	Name	<u>GPM</u>
	Filter Cooler Section 200	991 991
	Coolers Section 300	232 232
	Slurry coolers Section 400	1,950 1,950
 E-501 Subtotal	Evaporator condenser Section 500	6,579 6,579
 HE-607	Coolers Sugar cooler Section 600	871 280 1,151
	Trim condenser Vent condenser Dehydration overhead cooler Fusel oil cooler Waste water cooler Alcohol cooler Section 800	40 1,470 40 8 48 1,606
HE-902 HE-905 HE-907 HE-908 HE-909 Subtotal	Azeotrope condenser	80 719 96 3,893 2 4,790
E-1001 Subtotal	Evaporator condenser Section 1000	5,420 5,420
Plant to	ta) : : : : : : : : : : : : : : : : : : :	22,719

## 4. Process Water Requirements

Process water requirements are summarized in Table IV-D-4. The total process water load of 1,030 gallons per minute can be supplied by treated water from the consensate treatment system in Section 1100.

# TABLE IV-D-4 UTILITY SUMMARY - PROCESS WATER

<u>Item</u>	Name	GPN	1
	Prehydrolysis reactor Section 100	 <u>607</u>	
	Lime tank Section 200	39	3
VT-315	Enzyme production tanks Wash water tank Section 300	 45 156 201	5
VT-407	Slurry tank Wash water tank Section 400	53 77 130	7
	Scrubber Section 600	44	<del>1</del>
	Fusel oil washer/decanter Section 800		<u>2</u>
	Lime tank Section 900		<u>ā</u>
Plant to	tal process water	1,028	3
Water fr	om condensate treatment system	1,160	0
Net impo	rt process water		0

## 5. Boiler Feedwater Requirements

Boiler feedwater is required to make up for the steam injected into process as well as the two percent of circulating exchanger steam assumed as losses. The total boiler feedwater make-up is 334 gallons per minute as summarized in Table IV-D-5.

TABLE IV-D-5
UTILITY SUMMARY - BOILER FEEDWATER

<u>Item</u>	<u>Name</u>	in the second of			<u>GPM</u>
CO-104 R-101 Subtotal	Digester Prehydro Section 100		reactor		97 128 225
F-302 Subtotal	Filter Section 300				<u>5</u> 5
E-501 Subtotal	Ejector Section 500				$\frac{1}{1}$
F-602 Subtotal	Filter Section 600				$-\frac{1}{1}$
R-901 Subtotal	Furfural Section 900	react	or		<u>92</u> 92
	Ejector Section 1000				$\frac{1}{1}$
Miscellar BFW make- Subtotal					3 6 9
Plant tot	al boiler fee	edwate	۴		334

## 6. Power Requirements

Horsepower requirements are presented in Table IV-D-6. This tabulation is based on the design specifications for all plant equipment items and results in a total design horsepower load of 19,960. A number of major power users will actually be operating at less than design capacity. Thus, the steady-state power load is 13,100 kilowatts. Power cogeneration from high pressure steam (Section IV-A-12) yields 12,400 kilowatts. The net imported power from the utility grid is thus 700 kilowatts.

## TABLE IV-D-6

## UTILITY SUMMARY - POWER

<u>Item</u>	<u>Name</u>	Design HP
CR-101 CO-103 CO-104 R-101 CP-101 Subtotal	Disk Refiner Feeder (in CR-101) Digester (in CR-101) Prehydrolysis reactor Water feed pump Section 100	9,000 - 5 150 9,155
CT-201 CP-201 F-201 AOT-201 AOT-202 CP-202 CP-203 CO-201 CO-202 T-201 CT-202 CP-204 CP-205 Subtotal	Centrifuge Overflow pump Rotary drum polishing filter Lime tank Neutralization tank Lime pump Thickener feed pump Conveyor Conveyor Thickener Centrifuge Overflow pump Sugar transfer pump Section 200	50 30 3 30 40 2 25 40 7.5 10 25 3 25 290.5
CO-301 CP-301-314 FN-301 CP-315-328 CP-329 CT-301 AOT-301 CP-330 CT-302 CP-331 CP-332 CP-333 Subtota1	Conveyor Recirculation pumps Air blower Discharge pumps Cell centrifuge feed pump Cell centrifuge Repulping tank Wash water pump Polishing centrifuge Centrifuge feed pump Cell recylce pump Enzyme feed pump Section 300	7.5 260 10 260 20 25 5 10 150 5 7.5
AOT-401 CP-413 A-401-406 CP-407-412 CT-401 CP-415 F-401 CP-414 CO-401 F-402	Slurry tank Feed pump Agitators Discharge pumps Centrifuge Overflow pump Rotary drum polishing filter Wash water pump Conveyor Dewatering press	50 60 375 500 750 60 3 5

## TABLE IV-D-6 (Continued)

## UTILITY SUMMARY - POWER

<u>Item</u>	<u>Name</u>	Design HP
CP-416 CP-417 Subtotal	Press discharge pump Sugar transfer pump Section 400	5 50 2,098
E-501 CT-501 CP-501 CP-502 AOT-501 AOT-502 CP-503 CT-502 CP-504 CP-505 CP-506 Subtota1	Multi-effect evaporator Calcium sulfate centrifuge Sugar pump Overflow pump Precipitator Sodium carbonate solution tank Sodium sulfate pump Calcium carbonate centrifuge Overflow pump Filter feed pump Distillate pump Section 500	1,200 1 20 0.5 4.0 0.5 0.5 250 25 25 25
FN-601 FN-602 CP-613 CP-601-606 CP-607-612 CP-613 CT-601 CP-616 CP-615 CP-617 F-603 CP-618 AOT-601 CO-601 CP-619 Subtotal	Blower Air blower Fermentation feed pump Recirculation pumps Discharge pumps Scrubber return pump Yeast centrifuges Overflow pumps Underflow pumps Yeast pump Dewatering press Discharge pump Yeast hold tank Conveyor Purification feed pump Section 600	75 10 25 125 125 2 400 30 6 3 1 1 40 7.5 40 390.5
Subtotal	Section 700	2,100
CP-804 CP-805 CP-806 CP-801 CP-802 CP-807 CP-808 CP-803 Subtotal	Dehydration reflux pump Entrainer/stripper feed pump Entrainer column bottoms pump Beer column bottoms pump Reflux pump Wash water pump Fusel oil pump Dehydration column bottoms pump Section 800	5 2 2 25 20 0.5 0.5 2

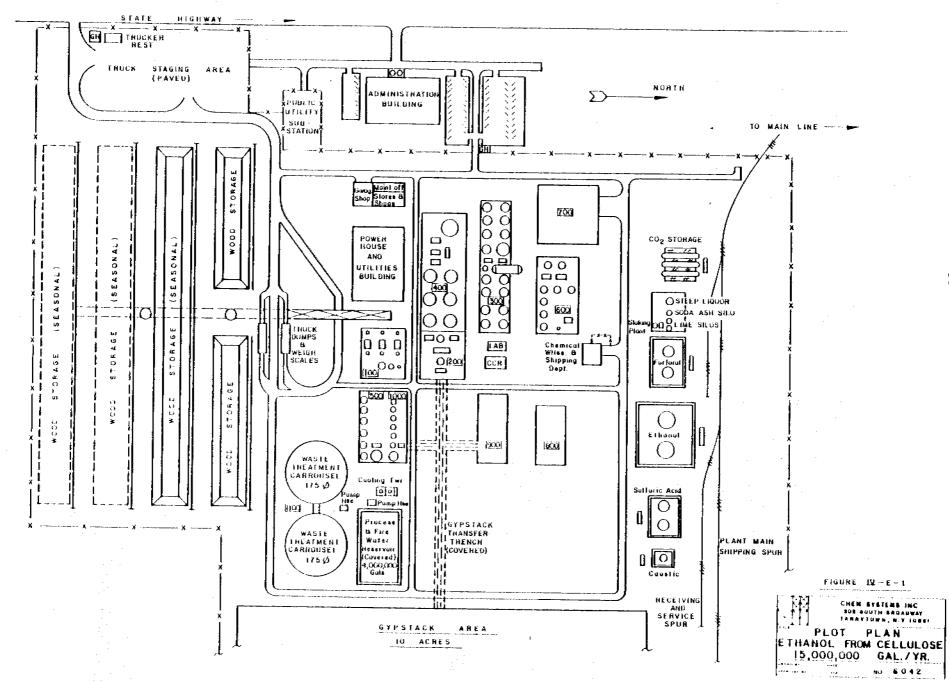
## TABLE IV-D-6 (Continued)

## UTILITY SUMMARY - POWER

<u>Item</u>	Name	Design HP
CP-901 A0T-902 CP-902 AOT-901 CP-903 T-901 CT-901 CP-904 CP-905 CP-910 CP-910 CP-912 CP-912 CP-911 CP-906 CP-913 CP-909 CP-908 Subtotal Sec	Furfural reactor feed pump Neutralization Tank Lime pump Lime tank Thickener feed pump Thickener Calcium sulfate centrifuge Overflow pump Azeotrope feed pump Lights column feed pump Dehydration feed pump Reflux pump Lights bottoms pump Azeotrope bottoms pump Stillage pump Reflux pump Dehydration bottoms pump	40 25 1 20 20 5 25 2 20 7.5 1 0.5 7.5 25 25 25 25 25 25 27
CP-1006 E-1001 Subtotal Sec	Distillate pump Multi-effect evaporator tion 1000	25 1,100 1,125
WTS-1101 WTS-1102 Subtotal Sec	Condensate treatment system Gypstack tion 1100	600 60 660
Subtotal off	Wood handling Steam boiler Cooling water system Miscellaneous Sites	1,000
Total plant	design horsepower	19,955.5
Actual runni	ng horsepower	17,580
Total plant	kilowatts	13,100
Generated po	wer, kilowatts	12,400
Net import p	power, kilowatts	700

### E. Plot Plan

The 25 million gallon per year ethanol facility is located in Gladwin County, Michigan. If centrally located within the county, the plant site would be midway (approximately 15 miles) from Interstate 75, U.S. 27 and U.S. 10, thus assuring reasonable highway access to all markets within the state and to the sources of wood feedstock. The facility itself requires a 130-acre site with a major portion of space reserved for wood storage and the gypstack (approximately 50 acres). The main facilities including the battery limits plant, powerhouse and waste treatment occupy 25 acres. Storage (other than wood) is sited on 10 acres and the administration facilities including a truck staging area occupy 20 acres. A preliminary plot plan for this facility is shown in Figure IV-E-1.



## V. SITING CONSIDERATIONS

## A. Selection of Michigan Site

#### 1. Introduction

The selection of a Midwest site for the ethanol facility and the selection of the wood feedstock based on available forest types are critical to the economic success of a project of this type. As illustrated in Table III-A-2, Aspen is an ideal hardwood for this facility based on potential reducing sugars. This results in high ethanol yields per unit of cost of wood feedstock. Thus, site selection was based on screening locations with abundant aspen resources. The selection process involved screening ethanol consumption patterns, gasoline demand, applicable tax laws, wood resource acreage and cost elements for delivered wood.

Based on these issues, Michigan was determined to be an ideal state in which to locate the proposed ethanol plant. Gladwin County, centrally located within Michigan's northern lower peninsula, was chosen as the plant site because of its access to abundant aspen forests as well as ethanol markets within the state.

In the following subsections, some of the major issues which affected the selection of the plant site and feedstock will be discussed in detail.

## 2. Feedstock Availability and Pricing

The state of Michigan contains 17.5 million acres of commercial forest land. As shown in Table V-A-1, twenty percent of the commercial forest land is aspen forest type. Another 35 percent is maple-birch type which also can be used in the ethanol facility although its ethanol yield will be somewhat lower. More than half of the state aspen forests can be found in the northern lower peninsula.

TABLE V-A-1

MICHIGAN HARDWOOD RESOURCES

(Thousand Acres)

	Northern Lower <u>Peninsula</u>	Southern Lower <u>Peninsula</u>	Upper <u>Peninsula</u>	<u>Total</u>
Total forest land	6,930	2,549	8,891	18,370
Commercial	6,695	2,463	8,332	17,490
Aspen	1,818	184	1,404	3,406
% aspen of comm'l	27	8	17	20
Maple-birch	1,662	903	3,532	6,097
% M.B. of comm'l	25	37	42	35

The aspen forest distribution by county within the northern lower peninsula is shown in Table V-A-2. It can be seen that Gladwin County contains 7 percent of the total state aspen forest land. When coupled with the four surrounding counties of Clare, Midland, Ogemaw and Roscommon, the aspen forest resource base constitutes 24 percent of the state total or 820,000 acres. Approximately 550,000 acres or 9 percent of the total state maple-birch forest land are also contained in these five counties. A map illustrating the aspen distribution throughout the Michigan lower peninsula is presented in Figure V-A-1. This map also shows the central location of Galdwin county.

The species of trees found in aspen and maple-birch forests in the northern lower peninsula are listed in Table V-A-3. Approximately 56 percent of the aspen forest is aspen and 45 percent of the maple-birch forest is maple-birch. With the exception of the softwoods, all the other species could be used as feedstock in the ethanol facility. If it is assumed that 30 percent of the plant feedstock will be hardwoods from aspen forests and the remainder from maple-birch forests, the average feedstock composition will be 57 percent aspen, 20 percent maple and 23 percent other hardwoods.

ASPEN DISTRIBUTION BY COUNTY (% of Total State Forest Type)

•	. Aspen	Maple-Birch	Percent of County  Total Forest Land
Alcona Alpena Cheboygan Clare Gladwin Midland Montmorency Ogemaw Oscoda Presque Isle Roscommon Total	6 6 6 4* 7 4* 5 5 5 4* 57	2 1 7 2 1 2 5 2 2 2 2 2 2	45 59 61 52 73 66 58 52 38 45 37

<sup>\*</sup>Within reasonable sourcing area of Gladwin County/Midland, Bay City site.

TABLE V-A-3

SPECIES OF TREES IN THE NORTHERN LOWER PENINSULA (Percent of Total)

	Aspen fores Type	it -	Maple-Birch Forest Type
Big Tooth Aspen Quaking Aspen	28 28		5 
Other hardwoods Maple-Birch Red Oaks Paper Birch Balsam Poplar Ash Beech Others	12 7 5 4 2 -		45 5 3 - 6 5
Softwoods	12	1	The state of the state of

## FIGURE V-A-1

MICHIGAN LOWER PENINSULA ASPEN RESOURCES



## ASPEN DISTRIBUTION

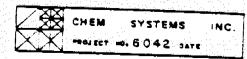
PROPOSED PLANT LOCATION

OVER 5% OF REGION TOTAL

3-5% OF REGION TOTAL

1-3% OF REGION TOTAL

LESS THAN 1% OF REGION TOTAL



The ownership of forest land in the northern lower peninsula is shown in Table V-A-4. This table also illustrates the average age and growth rate for the two forest types. It can be seen that the aspen forests are younger and have a higher growth rate. Based on feedstock requirements of 104 tons per hour or 33 million cubic feet per year of wood, Gladwin County could supply more than half these needs simply from the average annual growth rate. Gladwin and its four neighboring counties combined grow enough wood to keep two facilities of this size going.

TABLE V-A-4
NORTHERN LOWER PENINSULA

HARDWOOD RESOURCES

#### Maple-Birch Aspen Acreage, 000 1,818 1,662 Ownership, % National forest 9 29 24 State County, municipal 1 30 16 Farmer 8 6 Misc. pvt. corp. 35 Misc. pvt. indiv. 35 1 Forest ind. 1 Miscellaneous 100 100 Total

Average Growth/Ye (Cu ft/acre)	% of Aspen	% All Types	% of Maple-Birch	 % of All Types
85+ 20-84 0-21	44 56 0	47 21 32	24 76 0	24 25 51
Age Class				
1-2 yrs. 21-50 50+	44 24 32		5 27 58	

If all the wood came from these five counties, the average hauling distance would be 30-40 miles. A more conservative 50-mile hauling distance has been used in this study. Feedstock pricing is based on whole tree chipping performed in the forest by contract crews who supply their own equipment. Currently, hauling costs are two dollars per loaded mile for 25-ton loads. With an assumed average distance of 50 miles. The plant gate cost for hardwood chips becomes 18 dollars per wet ton as shown in Table V-A-5.

### TABLE V-A-5

MOOE	FEEDSTOCK	COST - I	MID 1984	
	(Dollars	per Wet T	on)	
Stumpage				3.0
Harvesting	and chipp	ing		11.0
Hauling				4.0
Total				18.0

Three different hardwood pricing scenarios were formulated for the plant operating years, 1987-2001:

- Escalation at the GNP deflator.
- Escalation to reach equivalency with softwood prices by the year
   2000 with softwoods escalating at the GNP deflator.
- Escalation according to Chem Systems projections for Midwest coal prices.

Chem Systems projections for the GNP deflator and Midwest coal price escalation are given in Table V-A-6. The base case ethanol facility economics assume that the hardwood chips will escalate with the GNP

deflator. The second scenario is based on the assumption that demand for hardwoods within the pulp and paper industry and the lumber industry will increase in Michigan as it finds increasing use in oriented strand board, pulp and paper, etc. The hardwood demand would equal that for softwoods by the year 2000 if these industries continue to increase their usage of hardwoods. Currently, hardwood is priced at approximately 88 percent of softwoods because of its lower demand by the pulp and paper industry. The third scenario assumes that a major use for hardwood in Michigan is as a fuel. If this were the dominant future use for hardwood, its competition would be coal, and therefore its value would escalate with Midwest coal prices. These pricing scenarios are summarized in Table V-A-7.

TABLE V-A-6
ESCALATION FACTORS

	GNP Deflator (%)	Midwest Price Esca (%)			
1984-1990 1991-2001	5.0 4.5		8.0 7.0	earl of the ar	

TABLE V-A-7

## PLANT GATE

## HARDWOOD PRICE PROJECTIONS Dollars per Wet Ton)

	Softwoods @ GNP Deflator	Hardwoods @ GNP Deflator	Hardwoods Via Coal Escalation	Hardwoods Approaching Softwoods
1984	20.5	18.0	18.0	18.0
1987	23.7	20.8	22.7	21.3
1990	27 <b>.</b> 5	24.1	28.6	25.4
1995	34.2	30.1	40.1	32.9
2000	42.7	37.5	56.2	42.7
2001	44.6	39.1	60.1	44.6

## 3. Gasoline Pool Analysis

The composition of demand for refined products in the United States has and will continue to be distinctly different from other industrialized regions of the world. Traditionally high in gasoline and low in residual fuel oil demand, consumption in the United States has resulted in a refining industry that is a highly complex structure designed to produce and market motor gasoline. The decline in gasoline demand, predicted to continue through the 1990s, is a major factor in the planning and actual operation of the petroleum refining industry. Similarly, demand for residual fuel oil will continue its recent decline and, again, refiners must adjust their operations accordingly. Kerosene/jet fuel and distillate fuel will experience a rise in demand level and will thus represent an increasing share of the composite refined product demand slate as shown in Table V-A-8, which presents the demand for the key products.

TABLE V-A-8

U.S. MAJOR REFINED PRODUCT DEMAND, 1980-1990
(MBPCD)

<u>Product</u>	1980	<u> 1982                                      </u>	<u>983</u> <u>1985</u>	<u>1990</u>	Annual Percent Change 1980-1990
Gasoline Kerosene/kerojet Distillate fuel Residual fuel Total	980 2,866 2 2,508 1	919 ,671 2,6 ,716 1,4	617 6,600 935 970 682 2,870 403 1,972 637 12,412	1,015 3,050 1,805	(0.9) 0.4 0.6 (3.2) (0.9)

## Gasoline Demand

Since fermentation ethanol is largely consumed for transport fuel applications both as an octane enhancer and volume extender, the emphasis in this section will be on current and anticipated demand for gasoline and

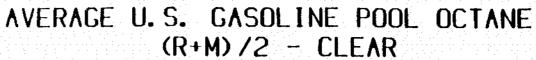
how the dynamics of this market will affect ethanol consumption during the 1980s. In the United States, transport fuels account for over 60 percent of refined product demand, with gasoline being the single largest volume product consumed. Although overall gasoline use is projected to decline, it will still account for about 40-45 percent of overall petroleum product demand in 1990, about the same share that occurred in 1978, the peak year for U.S. gasoline and refined product consumption.

#### Demand by Grade

Although total gasoline demand is expected to drop over the study period, the demand for unleaded grades will increase. The addition of lead compounds to improve the octane ratings of motor gasoline has been severely restricted by environmental regulations. Lead phasedown has caused the clear octane (octane prior to the addition of lead) of the U.S. gasoline pool to rise significantly and will cause the clear pool to increase further. Consistent with increasing demands for unleaded grades, in general, are increases in the demand for premium unleaded gasoline. Of the total gasoline demand, unleaded premium grades will increase from about 12 percent in 1983 to about 27 percent in 1990. The octane pool increase is displayed in graph form in Figure V-A-2. It is this change in the clear octane pool that provides the impetus for using octane enhancers, reflecting a growing requirement for ethanol and other octane enhancers.

Ethanol demand has grown from less than 100 million gallons in the 1979-1981 period to about 225 million gallons in 1982 and just over 400 million gallons in 1983. Examining the projected demand for gasoline by grade, U.S. refining capacity, the use of alternative octane enhancers, the status of state tax exemption laws, as well as current and anticipated ability to produce anhydrous fermentation grade product, it is estimated that overall U.S. ethanol consumption for transport fuel applications will equal about 600 million gallons in 1985 and grow to about 350 million gallons by 1990, as shown in Table V-A-9.

FIGURE V-A-2



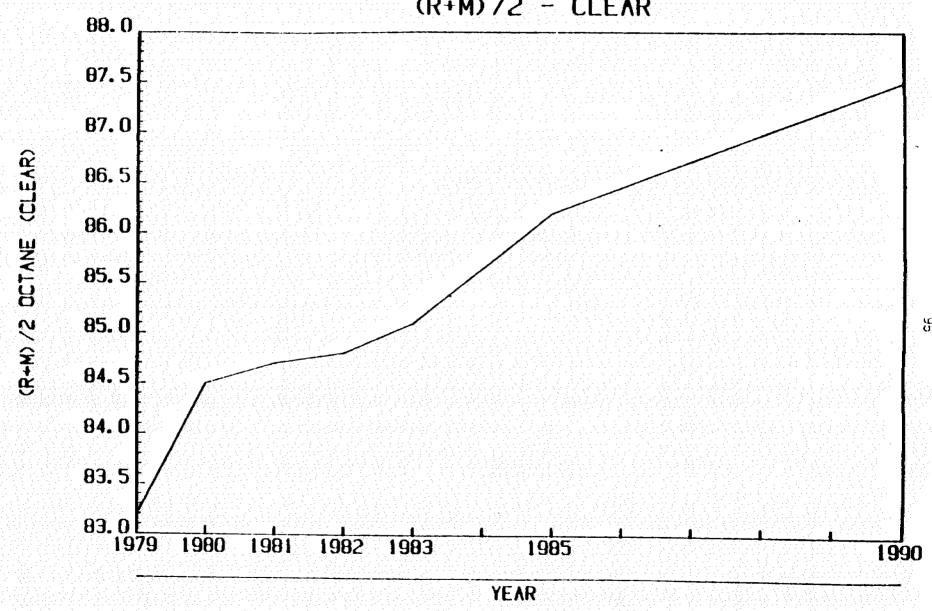


TABLE V-A-9

ESTIMATED U.S. DEMAND FOR TRANSPORT FUEL ETHANOL, 1981-1990
(Million Gallons)

1981	<u>1982</u>	<u>1983</u>	<u> 1985</u>	1990
77	225	407	600	850

Chem Systems' forecast indicates that ethanol will represent approximately one-quarter of the total consumption of octane enhancers used by U.S. refiners in the late 1980s and 1990. Total enhancer use, including MTBE, TBA, methanol, dimate and ethanol, is projected to grow from less than 50 MBPCD in 1980 and about 100 MBPCD in 1983 to 160 MBPCD in 1985 and just over 225 MBPCD in 1990. The ethanol share of this market will grow from about 10 percent in 1980 to an estimated 25 percent by 1990 as shown in Table V-A-10. In order to achieve this market share additional ethanol facilities will be required.

TABLE V-A-10

OCTANE ENHANCER USAGE FOR GASOLINE BLENDING, 1980-1990

(Percent of Total)

	1980	1983	1985	1990
Ethanol	10	23 👾	24	. 25
MTBE	15	18	15	15
Methanol and				
higher alcohols	40	24	30	33
Dimate	35	35	29	26
Total	100	100	100	100

## Regional and State Gasoline Demand

Regional gasoline demand in Michigan and surrounding states is summarized in Table V-A-11. In recent years, the growth rate in Michigan has outpaced that of the region as a whole and the national average.

TABLE V-A-11

GASOLINE DEMAND BY STATE

	<u>1982</u>	Growth Rates, %						
	(MBPCD)	<del>1960-1978</del>	1978-1982					
Michigan Minnesota	237 • 95	3.5 2.2	(7.2)					
Wisconsin Indiana	100 150	3.1 2.7	(6.4) (8.9) (5.3)					
Illinois Ohio	252 282	2.9 2.8	(6.6) (4.9)					
National to	tal 6,539	3.6	(4.4)					

Furthermore, there is a growing regional demand for unleaded fuel. This is exemplified in Table V-A-12 which summarizes Chem Systems' demand projections for the state of Michigan. It can be seen that while the total gasoline demand is projected to decrease throughout the study period unleaded gasoline will be growing rapidly until 1990. From that point, total unleaded will also begin to decline, but the premium grade will continue its growth. It is interesting to note that virtually the entire gasoline market in Michigan is within 200 miles of Gladwin County, the proposed plant site.

## 4. Tax Incentives

It has been a combination of federal and state legislative initiatives that have provided the impetus for the development of a domestic fermentation ethanol industry. Several of the federal programs have been specifically directed to benefit ethanol production and usage, while others have had a broader intent but nonetheless have directly supported the development of the domestic industry. Direct programs include the 5 cent per gallon federal exemption (increasing to 6 cents on January 1, 1985) for using blends of 10 percent ethanol and 90 percent gasoline as well as various loan guarantees for plant construction and so forth. Indirect programs include the phasedown of lead usage in gasoline as part of the Clean Air Act and various industrial tax incentives created as part

TABLE V-A-12
MICHIGAN GASOLINE DEMAND BY GRADE

	Unlea	ided	Leaded	
•	Regular	Premium	Regular	<u>Total</u>
1983	1,793	225	1,664	3,681
1985	2,130	355	1,065	3,550
1990	2,203	518	518	3,240
1995	2,085	556	139	2,780
2000	1,823	608	0	2,430

	Gì	ade Distr	(Percent)		
	Unlea	ided		Leaded	
	Regular	Premium		Regular	<u>Total</u>
1983	48.7	6.1		45.2	100.0
1985	60.0	10.0		30.0	100.0
1990	68.0	16.0		16.0	100.0
1995	75.0	20.0		5.0	100.0
2000	75.0	25.0		0.0	100.0

Market	Distribution from Gladwin County
Miles	Percent of Michigan Demand
0-50	9.6
50-100	38.6
100-200	50.2
200-300	1.6
	100.0

of President Reagan's tax reform initiatives. This section will review the federal programs that provide direct support, touch on those indirect issues that impact on ethanol use and plant construction, and discuss the state supports for fuel ethanol use.

#### Direct Federal Incentives

Perhaps the best known incentive, and certainly one of the most important, is the exemption from part of the federal excise tax for fuels containing ethanol produced from biomass sources - the law specifically excludes ethanol produced from petroleum and natural gas derivatives. Therefore,

"synthetic" ethanol produced from ethylene does not qualify for the federal credit (nor would it qualify for most state credits). original legislation exempted 10 percent ethanol blends from the full 4 cent per gallon federal excise tax on gasoline and diesel fuel - this exemption was originally scheduled to last until October 1984. As part of the Crude Oil Windfall Profits Tax Act of 1979, the exemption was extended until December 31, 1992. During the closing days of the 97th Congress, the Surface Transportation Assistance Act of 1982 was passed. This act increased the federal excise tax to 9 cents per gallon and raised the level of the exemption for 10 percent anhydrous alcohol blends by 1 cent per gallon to 5 cents per gallon and continued it in place until December 31, 1992 (the bill was effective on April 1, 1983). The increase in the exemption was the result of legislative compromise. The Senate wanted to provide ethanol with a full 9 cent per gallon exemption while many in the House wanted to continue the 4 cent per gallon exemption and terminate it in 1988. More recent efforts have resulted in an increase of the level of federal tax exemption once again, from 5 cents to 6 cents per gallon, effective on January 1, 1985 and continuing until 1992.

Coincident with the increase in the federal tax exemption, the blender tax credit was also increased. Effective April 1, 1983, and again applicable to December 31, 1992, biomass-derived alcohol used in or as a motor fuel was eligible for a credit of 50 cents per gallon for alcohol of at least 190 proof, and a 37.5 cent per gallon credit applies for alcohol from 150 to 190 proof. These benefits must be reduced by any gained from the overall federal excise tax exemption. This credit will increase to 60 cents on January 1, 1985.

In addition to the above credits, there is a full 9 cent per yallon tax exemption for neat alcohol transport fuels. This applies to any alcohol fuel containing at least 85 percent methanol, ethanol and any other alcohol derived from either biomass and coal and specifically excluding natural gas or petroleum-derived products.

#### Construction Assistance

To aid the construction of ethanol plants, both the Department of Energy and the Farmers Home Administration have been providing loan guarantees. With the budget cutting of the current administration, the level of funding available for guarantees has dropped considerably. Also, the uncertainties associated with independently financing these projects, given the upcoming end of the Energy Investment Tax Credit, higher feedstock costs, lower ethanol netbacks, and declining gasoline prices, have delayed many proposed projects.

The Department of Energy (DOE) has only limited funding available to support new plant construction, while the Synthetic Fuels Corporation (SFC) is specifically excluded from funding biomass ethanol plants. A DOE loan guarantee has been provided to New Energy of Indiana, whose plant construction is virtually completed. Conditional guarantees have also been provided to a number of potential large-scale ventures, although sources of both equity and debt financing must be arranged before a commitment can be obtained. In addition, companies must provide a detailed marketing plan and design and engineering details before a guarantee will be granted.

Having a conditional loan guarantee is no assurance that a project will in fact go ahead. Many details must be worked out, including securing the necessary equity and getting debt financing. Many project sponsors have also sought additional governmental arrangements (federal, state, and local) to secure their projects. Among the options are the following:

- Small Business Administration Loans
- Urban Development Action Grants (UDAG)
- Industrial Development Bonding

## Energy Investment Tax Credit (EITC)

The tax codes now provide a 10 percent energy investment tax credit, which is in addition to the regular 10 percent investment tax credit for alternative energy projects such as biomass ethanol facilities. This credit is due to expire on December 31, 1985, and recent attempts to extend the credit have been unsuccessful. Under current law a plant would have to be completed prior to January 1, 1986, in order to qualify for the EITC. The impending end of the credit has caused financing delays for several projects such as the proposed Minnesota Alcohol Fuel project. To qualify for the EITC the primary source of energy for the project must not be either oil or natural gas or derived from them.

Furthermore, if a project is funded with Industrial Development Bonds (IDB), the EITC will not be available. Also, if IDBs are used, the basis for depreciating property is changed.

#### Indirect Programs

The following are the most significant other federal programs that impact on ethanol:

- Farm support programs such as the Payments in Kind (PIK), which affects corn and other ethanol plant feedstock costs.
- The Clean Air Act and its lead phasedown requirements, which increases the demand for octane enhancers.
- Tax incentives that allow for accelerated cost recovery in the depreciation schedule and other similar issues.

Under the proposed regulations the level of lead would be reduced to 0.5 grams per gallon of leaded gasoline over an interim period such as from 1985-1987. After that time the lead level would drop to 0.1 grams per gallon. During the fourth quarter of 1984 final rules should be developed. The analysis that has been performed for this study is based

on existing law although it is recognized that such action by the EPA would significantly add to the demand for ethanol and other octane enhancers. Accordingly there should be reasonable upside potential to the base numbers presented in this report.

#### State Incentives

State tax incentives enable ethanol-gasoline blends to compete in the During 1982, thirty-one (31) states provided an incentive marketplace. that varied from a low of 1 cent per gallon to as much as 10 cents per gallon in New Mexico. These incentives in most cases apply to alcoholfrom any biomass source, but in several states there are restrictions that reflect feedstock type and source, ethanol production location, and reciprocal agreements between states. In this section a brief review has been developed of the various changes in state exemptions in the Midwest that have occurred during 1983 and the first half of 1984 and may occur thereafter. A summary of the existing state-by-state exemptions is also provided in Table V-A-13. It must be emphasized that there is no certainty that these tax laws will remain unchanged, given the exigencies of state budgets, political pressures, etc. However, as discussed in this section, more states have in fact revised their laws to provide greater incentives than have reduced them. The continuation of these exemptions will be based on a combination of revenue loss implications, political pressure and the level of industrial development and job creation that can be attributed to these exemptions. A brief description of the Michigan situation follows.

## <u>Michigan</u>

The state has maintained its phasedown ethanol exemption for local ethanol production. However, the state lowered its tax break, effective April 1, 1983, from 3 cents to 2 cents per gallon for blends containing ethanol produced out of state and in states that do not match Michigan's 4 cent per gallon tax exemption. The exemption for out-of-state material dropped to 1 cent per gallon in 1984 and expires on December 31, 1984.

TABLE V-A-13
SUMMARY OF STATE TAX EXEMPTIONS\*

	(All Units are Cents per Gallon Except as Noted)								In-state				
	1982	1983	1984	1985	1986	1987	1988	1989	1990	1991	1992	Production Only	Reciprocity
Alabama	3	3	3	3	ı	3	3	3	1	3	1		
Alaska	H H	В	В	8	8	8	В	š	B	8	8		
Ar i zona		₩.	-			31				그런 그를 그	- I		
Arkansas	6.5	6.5	5.5	5.5	5.5	5.5	5.5	5.5	5.5	5.5/0	. A	· · · · · · · · · · · · · · · · · · ·	10 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
Calilornia	4 1	3.	3/0			-	-				_		
Colorado	5	5	5	5/0	O .		¥					ľ	
Connecticut	1	1	1	i i	1	i	1	1	i i	1	1		
Delaware			-			n a'i 40 m							
f lor ida	5	51**/4	4	4	4	4/0	-						
Georgia													
Hawa i i		4	4	1/0									
ldahu	4	4	4	4	4/0		-					<b>1</b>	
- Illinoiy,	3%	32/22/41	41	41	4 %	41	4.6	4%	4%	4%	4%		<b>Y</b>
Indiana(1)	47	3%	34/2.5	2.5/0	-		-			트리스 네 걸.			
luwa	5	5/3	3/2	2/1	1/0								
Kansas	2	2/3/4	5	5	5	5	5	5	5	5	5		
Kentucky	1.5	3.5	3.5	3.5	3.5/0						•		X
iouisiana	ð	Ü	8/16	16	16	16	16	16	16	16	Ü	i i	
Maine			-		4	4	4	4	0	0	0		
Maryland		U/J	3	3	3/0								
Massachusett	5	÷ .						_					
Michigan -													
local prod'		4	4	2	Alexander		•						
out of stat	ს ს	4/3/2	1	U	Jan Very St	•							
Minnesota		0/2	2	2/4	4	4	4	4	4	4	4/0		
Mississippi				0/6%	6%	6#	6.5	6%	6%	64	61		
Missouri	위기에게 원기를				-	H							
Montana		V (1)		1/5	5/3	3	3.	3/0					
Helmaska	•	5	5		5	5	5 / 1	5 \	5	5	5	i i i i i i i i i i i i i i i i i i i	
Nevada				1.	1 J. 1	1	l l	1		$\mathbf{i}$	ì		
New Hampshir	e 5	5/0							_	rjana <u>I</u> rijan	1		
New Jersey				0/8	ક	IJ	6	6	4	4	Û		
New Mexico	LO	10/11	11	11	- 11	11/8	8/5	5/3	3/2	2/0	Ü	X	

<sup>\*</sup>As of August 1984.

<sup>\*\*</sup>Refers to percentage of wholesale price of gasoline.

A Applies to state unly.

<sup>(1)</sup> State now offers a special ethanol production subsidy. Effective 1/1/84 Indiana ethanol producers will receive 10 cents per gallon of production and after 7/1/85 subsidy will increase to 25 cents per gallon.

TABLE V-A-13 (Continued)

#### SUMMARY OF STATE TAX EXEMPTIONS\*

			(	All Dair	s are Cen	its per Ga	Hon Exec	oot as No	tedi			In-s	Lato		
	1982	1983	1984	1985	1986	1987	1988	1989	1990	1991	1992		ion Only	<u>kec i proc</u>	ity
New York	_	-	<b>-</b>	-		-	=	_	*	_	-				
North Carolina	ā 2	2/1/5	5	5	5	5	5	5	5	5	5/U				
North Dakota	4	4	5	б	4	4	4	4	4	4	4/0				
Ohio	3,5	3.5	3.5	3.5	3.5	3,5	3.5	3.5	3,5	3.5	3.5		•		
Ok La froma	6,5	-	-	. ~	· · ·		_	-		*	-				
Oregon	-	_	_	-	, <del>-</del>	-	-	_	-	-	-				
Pennsylvania	_	-	~	-	-	-	_	-	-	_	_			•	
Rhode Island	-		_	-	_	_	_	_	_	_	-				
South Carolina	a 7	Ű	<b></b>	-	-		-	~		-				•	
South Dakota	4	4	4	4/0	<del>.</del> .		. +		_	_	_			¥	
Tennessee		4	4	4	4	4	4	-	_	-	_			Ŷ	
Texas -													•	^	
local prod'u	5	5	5	5	5	4	3	2	1.	α	Ð				
out-of-state		5/1.2	1.2	1.2	1.2	1.0	0.7	0.5	0.25	ŭ	ñ				
Utah	5	5	5	5	. 5	5	5	5	5	Ğ	ñ				
Vermont	-	~		~		_	~		-	_	-				
Virginīa	ij.	8	8/6	6	ŭ/4	4	4/2	2	2/0	0	a	v			
Hashington	1.5	1.5/2.6	2.6/2.9	2.9	2.9	à	-,-	~		-	-	^			
- West Virginia			_	_		-		_	_		_				
Wisconsin 🕟		_		1					<u>.</u>	:				$x_{n} \in \mathcal{A}_{n}^{(n)}(x)$	
Wyoming 🔠	4	4	4 .	0 -	_	<b>-</b> .			· ·		-	. · X			

<sup>\*</sup>As of August 1984. \*\*Refers to percentage of wholesale price of gasoline. x-Applies to state only.

There are also two mutually exclusive efforts involving the state exemption. One would abolish the credit entirely, while another would double the credit and apply it only to alcohol produced from at least 50 percent Michigan-grown feedstocks. It is unclear what final action may occur on either of these measures, although it is considered likely that Michigan will extend its tax credit if industrial development occurs as a result.

#### B. Demand and Pricing Analysis

#### 1. Ethanol

#### Ethanol Demand in Michigan

Ethanol usage in Michign for gasoline blending was 56.6 million gallons in 1983 which made it the number one consuming state of fuel ethanol in the U.S. Meanwhile, Michigan ethanol production was only 0.3 million gallons in 1983 and will expand to 12.5 million gallons by 1987.

Chem Systems has examined the demand for gasoline in Michigan and has developed both volume projections as well as demand by grade. On the basis of these projections, ethanol demand assessments have been developed for various levels of market penetration and for varying distances from the plant site. These analyses were developed using existing lead use regulations. If an accelerated lead phasedown program is enacted, there will be even greater demand for ethanol. Overall gasoline volume demand projections by grade over the study period have been presented in Table V-A-12. Using these data as the basis the potential for ethanol was developed.

A profile of ethanol demand by year, radius from the Gladwin County plant site and penetration level is shown in Tables V-B-1 to V-B-5. In terms of unleaded premium, the proposed 25 million gallon per year ethanol plant would have to penetrate 48 percent of the potential 1990 market, 45 percent of the 1995 market and 41 percent of the 2000 market in Michigan.

However, in terms of total unleaded regular plus premium, the proposed plant would only have to penetrate 9 percent of the potential 1990 market and slightly more in future years. This should be an easy goal. In fact it should be reasonable for half the plant output to find its way into premium and the other half into regular.

TABLE V-B-1

ETHANOL MARKET POTENTIAL, 1983

(Million Gallons)

Distance from Gladwin site, miles	0-50	50-100 10	0-200	200-300	<u>Total</u>
Gasoline Unleaded regular Unleaded premium Leaded regular Total	172 22 160	692 87 642	900 113 835	29 4 27	1,793 225 1,664 3,681
Potential Ethanol Penetration 100% of ULR & ULP 100% of ULP 50% of ULR & ULP 25% of ULR & ULP	19.4 2.2 9.7 4.8	77.9 8.7 38.9 19.5	101.3 11.3 50.6 25.3	3.2 0.4 1.6 0.8	201.7 22.5 100.9 50.4

TABLE V-8-2

ETHANOL MARKET POTENTIAL, 1985

(Million Gallons)

Distance from Gladwin site, miles	0-50	50-100	100-200	200-300 <u>Total</u>
Gasoline Unleaded regular Unleaded premium Leaded regular Total	204 34 102	822 137 411	1,069 178 535	34 2,130 6 355 17 1,065 3,550
Potential Ethanol Penetration				
100% of ULR & ULP 100% of ULP 50% of ULR & ULP 25% of ULR & ULP	23.9 3.4 11.9 6.0	95.9 13.7 48.0 24.0	124.7 17.8 62.4 31.2	4.0 248.5 0.6 35.5 2.0 124.3 1.0 62.1

TABLE V-B-3

ETHANOL MARKET POTENTIAL, 1990

(Million Gallons)

Distance from Gladw site, miles	in <u>0-50</u>	<u>50-100</u>	100-200	200-300	<u>Total</u>
Gasoline Unleaded regular Unleaded premium Leaded regular Total	212 50 50	850 200 200	1,106 260 260	35 8 8	2,203 518 518 3,240
Potential Ethanol Penetration					
100% of ULR & ULP	26.1	105.1	136.6	4.4	272.2
100% of ULP 50% of ULR & ULP		20.0 52.5	26.0 68.3	0.8 2.2	51.8 136.1
25% of ULR & ULP	6.5	26.3	34.2	1.1	68.0

TABLE V-B-4

ETHANOL MARKET POTENTIAL, 1995

(Million Gallons)

Distance from Gladwi site, miles	n <u>0-50</u>	<u>50-100</u>	100-200	200-300	<u>Total</u>
Gasoline Unleaded regular	200	850	1,047	33	2,085
Unleaded premium Leaded regular	53 13	215 54	279	9	556
Total	13		70	2	139 2,780
Potential Ethanol					
Penetration 100% of ULR & ULP	25.4		132.6	4.2	264.1
100% of ULP 50% of ULR & ULP	5.3 12.7		27.9 66.3	0.9 2.1	
25% of ULR & ULP	6.3	25.5	33.1	1.1	66.0

TABLE V-B-5

ETHANOL MARKET POTENTIAL, 2000

(Million Gallons)

Distance from Gladwin site, miles	<u>0-50</u>	<u>50-100</u> <u>1</u>	00-200	200-300	<u>Total</u>
Gasoline Unleaded regular Unleaded premium Leaded regular Total	175 58 0	703 234 0	915 305 0	29 · 10 · 0 · 0 · 0 · 0 · 0 · 0 · 0 · 0 ·	1,823 608 0 2,430
Potential Ethanol Penetration 100% of ULR & ULP 100% of ULP 50% of ULR & ULP 25% of ULR & ULP	23.3 5.8 11.7 5.8	93.8 23.4 46.9 23.4	122.0 30.5 61.0 30.5	3.9 1.0 1.9 1.0	243.0 60.8 121.5 60.8

#### Ethanol Valuation

Ethanol has the highest value as an octane enhancer of the various alcohols and ethers now being utilized by refiners as well as gasoline blenders and marketers. Its use also exceeds that of any other single enhancer commonly utilized. However, because of its relatively high cost (\$1.55-\$1.70 per gallon), its use is largely a function of both federal and state tax credits granted for the use of 10 percent ethanol in gasoline blends.

The value of an octane enhancer to the refining industry is based on:

- The cost of crude oil
- The price of gasoline
- The value of butane
- The cost of incremental octane

For most octane enhancers, the traditional method of assessing their value is based on their physical properties compared to a gasoline reference. The properties of interest are octane level and vapor pressure. Both

blending octane number, (R+M)/2, and the blending Reid vapor pressure (RVP) are significantly influenced by the gasoline base stock composition (normally defined by the PONA analysis) and lead level. Furthermore, with the exception of MTBE, the blending RVPs of the other normally used oxygenates exhibit severe deviation from expected vapor pressures as predicted by Raoult's Law. These deviations are all on the up side which equates to a butane backout factor in order to maintain a specified gasoline RVP when oxygenates are blended with gasoline.

Ethanol has a blending octane of about 119 compared to the reference gasoline value of 88. Because of this difference of 31 octane numbers, ethanol has a significant octane credit associated with it. Also, its blending vapor pressure (RVP) is 21.8 psi in a 10 percent blend compared to a gasoline reference value of 11.

This gasoline reference vapor pressure represents an approximate national average. There are both seasonal and geographic variations that are due to local climatic conditions. The higher RVP of ethanol means that refiners would have to back out butane from their gasoline to bring the vapor pressure up to specification. Since butane is valued at less than gasoline, the refiner effectively loses value. In this instance the enhancer would receive a butane debit to reflect this downgrading were it not for special regulations.

In effect there is no butane debit since ethanol, at up to 10 volume percent, has a specific environmental exemption to allow a higher vapor pressure of the finished gasoline blend. This exemption does not currently apply to any other enhancer. Under most current practice, therefore, ethanol is not blended at the refinery but rather at the jobber/distributor level. Ethanol is generally added to regular-unleaded-grade gasoline to produce a premium-grade unleaded gasoline with a higher vapor pressure.

Accordingly, ethanol has two intrinsic values:

- One based on blending at the terminal which only takes into consideration the octane credit.
- The other applies if the blending occurs at the refinery and butane is backed out. Refiners have the option of meeting the vapor pressure specification, in which case they would back out butane. They have no legal requirement to do so, but some wish to maintain constant quality specifications. A refiner could also choose an alternative approach, which involves the preparation of a base stock at the refinery with a lower vapor pressure than normal. This base stock would then be used for blending purposes at the terminal.

The difference in ethanol value for these situations is about 5 cents per gallon currently and is expected to rise to about 9 cents per gallon by 1990. In a Midwest location offering a combined tax credit of 9 cents per gallon of blend (5 cents federal and 4 cents state), ethanol would be worth about \$1.69 per gallon in 1983 assuming that the ethanol is used in an unleaded-regular-grade gasoline. If unleaded regular is upgraded to premium gasoline by the addition of 10 percent ethanol, the ethanol value increases to approximately \$2.12 per gallon.

Chem Systems has developed several ethanol valuation cases for this study. The key factors in determining the netback to the ethanol plant are freight costs and the level of state tax credit which has been assumed as 3 cents per gallon from 1985 onward.

In developing the netback values Chem Systems also performed the following:

 Typical Midwestern terminal contract unleaded regular and premium prices were developed. While it is recognized that there will be some slight variation in terminal prices between various markets, it is deemed appropriate to use a typical value for feasibility study purposes. • There would be some discount required to give the jobber/blender/terminal operator an incentive to handle the additional product. The incentive used for calculation purposes was 1.1-2.0 cents per gallon of blend over the 1983-1995 period. If an ethanol producer also owned distribution outlets he would "capture" this incentive.

A typical calculation for a Michigan-only marketing case is illustrated for 1985 in Table V-B-6. Three cases were developed:

- Case I illustrates the value of ethanol in unleaded regular gasoline. While on a calculated basis, this represents the lowest value for ethanol, under current market conditions this tends to more closely reflect market clearing conditions.
- Case II reflects ethanol value when blended into unleaded premium gasoline.
- Case III represents the ethanol value when used to upgrade unleaded regular to unleaded premium gasoline. This represents the highest ethanol value, but market clearing transactions do not occur at this level. However, this value can be realized by an ethanol producer who also owns gasoline distribution outlets (either wholesale or retail).

A summary of the ethanol netback calculations for 1983-2000 appears in Table V-8-7. This table shows the three sets of values. The maximum value shown represents the ethanol netback value when used for upgrading regular unleaded gasoline to unleaded premium. The minimum case shows the ethanol value in unleaded regular gasoline and is more typical of market clearing transactions. The midrange case would reflect the ethanol value for boosting unleaded premium from 90 to 93 octane.

# TABLE V-B-6

# ETHANGL VALUATION CALCULATION, 1985 (Cents per Gallon)

I. Value of Ethanol in Unleaded Regular	
Unleaded regular gasoline (at terminal) Terminal incentive Net blend value	90.7 1.2 89.5
Federal incentive State incentive Total incentives	6.0 3.0 9.0
<pre>Implied ethanol value at terminal = 10((89.59(90.7)) + 9 Average freight Ethanol netback</pre>	$\begin{array}{r} .0) = 163.7 \\ \underline{3.6} \\ 165.1 \end{array}$
II. Value of Ethanol in Unleaded Premium	
Unleaded premium gasoline (at terminal) Terminal incentive Net blend value	96.5 1.2 95.3
Federal incentive State incentive Total incentives	6.0 <u>3.0</u> 9.0
<pre>Implied ethanol value at terminal = 10((95.39(96.5)) +9. Average freight Ethanol netback</pre>	0) = 174.9
III. Value of Ethanol in Upgrading Unleaded Regular to Unleaded Premium	
Unleaded regular gasoline (at terminal)	90.7
Unleaded premium gasoline (at terminal) Terminal incentive Net blend value	96.5 1.2 95.3
Federal incentive State incentive Total incentives	6.0 3.0 9.0
<pre>Implied ethanol value at terminal = 10((95.39(90.7)) +9. Average freight    Ethanol netback</pre>	$\begin{array}{rcl} .0) = & 225.7 \\ & & \frac{3.6}{223.1} \end{array}$

TABLE V-B-7

ETHANOL VALUATION WITH MICHIGAN ONLY MARKETING

(Cents per Gallon)

					Contract Contract of	
	<u>1983</u>	1985	<u>1986</u>	1990	<u>1995</u>	<u>20</u> 0
Unleaded Regular Midwest unleaded reg. MOGAS	90.3	90.7	92.0	118.3	156.0	202.0
Terminal incentive  Avg. ethanol transportation  Federal tax incention	3.3	1.2 3.6	3.7	1.5 4.8	6.5	
Federal tax incentive State tax incentive Delv'd ethanol value	4.0	3.0	3.0	6.0 3.0 193.3	6.0 3.0	6.0 3.0
Ethanol netback (minimum) nleaded Premium	166.0	165.1	165.3	188.5	219.5	262.0 253.6
Midwest unleaded prem. MOGAS	94.6	96.5	97.8	124.9 199.9	163.0	210.0
Ethanol netback (midrange) Ethanol upgrade value	- 170 R	170 a	171	100 1	000	270.0 261.6
Ethanol netback (maximum)	209.0	223.1	223.3	254.5	289.5	333.6

For this feasibility study, it has been assumed that half of the ethanol product would go into the upgrading of unleaded regular to unleaded premium and half would go into the unleaded regular market. Thus, as summarized in Table V-B-8, the ethanol netback values at the plant gate are the average of the minimum and maximum cases previously described.

TABLE V-B-8

ETHANOL NETBACKS, 1983-2001
(Cents per Gallon)

	<u>Minimum</u>	<u>Midrange</u>	<u>Maximum</u>	Plant <u>Average</u>
1983	166.0	170.3	209.0	**************************************
1985	165.1	170.9	223.1	187.5
1986	165.3	171.1	223.3	194.1
1987	170.8	176.8	230.8	194.3
1990	188.5	195.1	254.5	200.8 221.5
1995	219.5	226.5	289.5	254.5
2000	253.6	261.6	333.6	293.6
2001	261.0	269.2	343.2	302.1
				205+1

#### 2. Furfural

#### Current Furfural Demand

The principal use for furfural is as an intermediate for the manufacture of furfuryl alcohol (FA), and to a lesser degree, tetrahydrofuran (THF). Furfural is also used as a selective solvent for the separation of saturated from unsaturated compounds in petroleum lubricating oil, and in the extractive distillation of butadiene.

In 1983, total U.S. consumption of furfural was 71 million pounds as shown in Table V-B-9. The major applications for furfural and its derivatives are discussed in the paragraphs which follow.

TABLE V-8-9
U.S. FURFURAL CONSUMPTION, 1983
(Million Pounds)

Markets	Cons	umption
Furfural derivatives Furfuryl alcohol-foundry resins Tetrahydrofuran (THF)	28 23	51
Lube oil refining		10.2
Butadiene extraction		1.8
Miscellaneous uses		8
Total		71

## Furfuryl Alcohol

Furfuryl alcohol is used in resin form as a sand binder by the foundry industry, chemical resistant grouts and cements, and sand consolidation in oil well drilling. It is a mobile liquid which polymerizes readily under acid conditions. In addition to reacting with itself, FA is commercially co-reacted with aldehydes (formaldehyde), urea, and phenol.

The major market for FA resins is their use as binders for the production of molds and cores used for casting metals. Such resins were introduced as binders in the late 1950s and are used in two different metal casting systems - no-bake and hot-box.

In 1974, the consumption of FA resins amounted to about 70 million pounds with 30 million pounds being used in hot-box systems and 50 million pounds in no-bake applications. Based on an average furfuryl alcohol content of 75 percent for no-bake resins and 35 percent for hot-box resins, the consumption of furfuryl alcohol for foundry resins amounted to 48 million pounds in that year.

Although the prospects for growth looked good, the market for FA resins in the foundry industry actually decreased rather than increased in the period since 1974. By 1983, only 36 million pounds of FA resin were used by the foundry industry. Of this amount, 33 million pounds were used in no-bake systems and 3 million pounds in hot-box systems.

The reasons for this decline are as follows:

- The 6.7 percent per year decline in the shipments of foundry castings which occurred in the 1974-1983 period.
- The shortage of furfuryl alcohol that developed in the mid-1970s.
- Several price increases for furfuryl alcohol which effectively doubled the price of the material between 1974 and 1983. This undoubtedly had a dampening effect on the use of FA resin because the price of phenol, the key raw material in competing resins, only increased about 20 percent over the same period.
- The declining importance of hot-box systems as foundries switched from hot-box to cold-cure systems. This change was generally attributed to the energy intensive nature of the former rather than any advantageous technical advantage of the latter.

The future outlook for FA resins will depend on the growth of the foundry industry and the international competition between FA resins and other no-bake systems. Most industry representatives indicated that there is little opportunity for any recovery in the market for hot-box systems and, even if there were, FA resins have little change of regaining the market they lost to phenolic resins. Chem Systems estimates that FA resin consumption should increase to 50 million pounds by 1990 with no-bake systems using 47 million pounds and hot-box systems 3 million pounds. These projections are sensitive to price changes for the furfuryl alcohol and the effect of such changes are discussed later.

The demand for FA resins and furfuryl alcohol for 1983-2001 is presented in Table V-B-10.

TABLE V-B-10

DEMAND FOR FA RESINS IN FOUNDRY CASTINGS

(Million Pounds)

	1983	1984	<u>1985</u>	1990	<u> 1995</u>	2001
FA resins	36	40	43	47	51	56
No-bake	3	3	3	3	3	3
Hot-box	39	43	46	50	54	59
Furfuryl alcohol	28	31	33	36	40	44
Furfural	28	31	33	36	40	44

The major producers of foundry resins are Ashland Chemical, Acme Resin, Delta Resins, Kordell and CE Cast Products. Ashland is the largest consumer of furfuryl alcohol and will probably use 7-8 million pounds this year. Delta is next in importance and its use of furfuryl alcohol will probably amount to 5-6 million pounds in 1984. Table V-3-11 lists the producers of foundry resins and their plant locations.

#### TABLE V-B-11

#### U.S. PRODUCERS OF FOUNDRY RESINS

#### Company

Acme Resin (CPC International)
Ashland Chemical

Borden Chemical CE Cast Industrial Products Core-Lube Delta

Eronel
Georgia Pacific
Mar-Cam (Hill & Griffith)
Durex (Hooker Chemical)
Kordell Industries
Reichhold Chemical
Thiem
United Erie

#### Plant Location

Forest Park, Ill.
Cleveland, Ohio
Hammond, Ind.
Louisville, Ky.
Muse, Pa.
Danville, Ill.
Milwaukee, Wis.
Detroit, Mich.
Hawthorne, Ca.
Newark, Ohio
Hickory, N.C.
North Tonawanda, N.Y.
Mishawaka, Ind.

Milwaukee, Wis. Erie, Pa.

#### <u>Tetrahydrofuran</u>

The major derivative for tetrahydrofuran is polytetramethylene ether glycol (PTMEG), make commercially by DuPont and QO Chemicals. PTMEG is used in the production of a variety of thermoplastic and thermoset polymer systems. The most important are as follows:

- Thermoplastic polyurethane elastomers.
- Curable prepolymers for production of cast (thermoset) polyurethane elastomers.
- Spandex fibers.
- Copolyester-ether thermoplastic elastomers.

The THF demand for the 1983-2001 period is presented in Table V-B-12.

TABLE V-B-12

U.S. DEMAND FOR TETRAHYDROFURAN

(Million Pounds)

and the second second				and the second second	
·	<u>1983</u>	1985	1990	1995	2001
PTMEG Solvent Other	60 31 <u>1</u> 92	57 34 <u>1</u> 92	70 38 2 110	90 40 <u>2</u> 132	115 42 3 160
Exports Imports	15 107	19 111	20 130	25 157	30 190
Furfural	23	23	23	28	35

The current producers of THF are presented in Table V-B-13. Three of the four producers of THF use 1,4-butanediol. The fourth, QO Chemicals makes THF from furfural at its plant in Memphis. QO Chemicals also converts a portion of its THF output into PTMEG.

TABLE V-B-13

U.S. PRODUCERS OF TETRAHYDROFURAN
(Million Pounds per Year)

Location	Capacity
LaPorte, Tex. Geismar, La. Linden, N.J. Memphis, Tenn.	120 20 15 15
	LaPorte, Tex. Geismar, La. Linden, N.J.

<sup>\*</sup>THF derived from furfural.

#### Solvent Uses

One of the oldest applications for furfural is as a selective solvent for the refining of lube oil to increase its stability and improve viscosity index. The furfural refining process was developed by Texaco and is second in importance in the United States to phenol extraction. Furfural preferentially dissolves aromatic, unsaturated, sulfur and nitrogen compounds, thus reducing the oxidation sensitivity and changing the temperature/viscosity curve of the lube oil.

The quantity of furfural consumed by a refinery consists of filling the system initially, and the makeup to replace losses during recycling. In 1983, 10 million pounds of furfural were consumed in the United States for lube oil refining. The U.S. refiners with furfural extraction units are listed in Table V-B-14.

TABLE V-B-14

LUBE OIL REFINNING FACILITIES WITH FURFURAL EXTRACTION UNITS
(Barrels per Day)

Refinery	<u>Location</u>	Extraction Capacity
Amoco 011	Casper, Wyoming	2,300
Ashland Oil	Catlettsburg, Kentucky	15,000
Cit-Con Oil	Lake Charles, Louisiana	28,000*
Conoco	Ponca City, Oklahoma	4,300*
Farmland Industries	Coffeyville, Kansas	3,800
Mobil Oil	Beaumont, Texas	18,300*
	Paulsboro, New Jersey	15,500
Quaker State Oil	Farmers Valley, Pennsylvania	1,800
	St. Marys, West Virginia	1,000
Shell Oil	Martinez, California	6,800**
Sohio	Lima, Ohio	6,000
Sun english a balan da ka	Tulsa, Oklahoma	16,000
	Yabucoa, Puerto Rico	15,000
Texaco	Port Arthur, Texas	23,000**
Witco Chemical	Oildale, California	4,000
[ - 기급	사람은 얼마 아이들 아이를 가는 사람이 없었다.	160,800

<sup>\*</sup>Includes Duo-Sol extraction capacity.
\*\*Includes SO<sub>2</sub> extraction capacity.

The long-term prospects for furfural in lube oil refining are very modest, reflecting the maturity of this market. Chem Systems estimates that furfural consumption will probably increase 1-2 percent annually in the 1980s and 1990s. This is slightly lower than the projection for lube oil production in the 1980s because it anticipates the closing of 1-2 furfural extraction units in the 1985-1990 period and generally more efficient plant operation.

The furfural forecast for lube oil refining for 1983-2001 is presented in Table V-B-15.

TABLE V-B-15

U.S. FURFURAL CONSUMPTION IN LUBE OIL PRODUCTION (Million Pounds)

	1983	1984	<u>1985</u>	1990	<u>1995</u>	2001
Furfural	10.2	10.6	10.8	11.5	12.5	13.0

Furfural is also used for the extractive distillation of butadiene to effect separation from mixed C4 streams. This process was developed by Phillips Petroleum Company. In addition to furfural, other solvents are used commercially for butadiene extraction. Shell developed a process based on acetonitrile, while Exxon developed one based on cuprous ammonium acetate. In more recent years, newer processes based on other solvents have been developed that offer advantages of lower capital cost and utility requirements compared to the furfural and other systems.

The current consumption of furfural for butadiene extraction is estimated at 1.8 million pounds. Considering the long-term outlook for butadiene consumption in the United States, Chem Systems estimates that furfural used in solvent extraction will probably not exceed 2.0 million pounds by 1990. There is also the possibility that the existing extraction units could be switched from furfural to other solvents.

The demand for furfural in butadiene extraction is presented in Table V-B-16.

U.S. FURFURAL CONSUMPTION WITH BUTADIENE EXTRACTION
(Million Pounds)

	<u>1983</u>	<u>1984</u> <u>1985</u>	<u>1990</u>	<u> 1995                                     </u>
Furfural	the first of the second of the first of the	1.8 1.8		and the second of the second o

#### Miscellaneous Uses

The consumption of furfural in miscellaneous uses amounted to about 10 million pounds in 1983. In some of the outlets the application calls for the use of furfural itself, while in others FA is used. Some uses are complicated by the fact that both FA and furfural are used depending upon the specific formulation. In certain instances, the product is used as a solvent, in others as a wetting agent/adhestive aid, and in still others the final product is a resin.

## Future Furfural Demand

The markets for furfural and its derivatives consist of a number of mature applications for which no growth or relatively modest growth is projected for the study period. The demand for furfural is forecast to reach 83 million pounds in 1990 and 105 million pounds in 2001, representing an average annual growth of 2.0 percent. The consumption of furfural by end use for the 1983-2001 period is presented in Table V-B-17.

TABLE V-B-17
U.S. FURFURAL CONSUMPTION, 1983-2001
(Million Pounds)

<u>Markets</u> .	1983	1984	1985 1990	1995	Average Annual Growth, % 2001 1983-2001
Furfural derivatives Furfural alcohol-					
foundry resins	28	31	33 36	40	44 2.5
Tetrahydrofuran	23	23	23 23	28	35 2.4
Lube oil refining	10.2	10.5	11 11.5	12.5	13 1.5
Butadiene extraction	1.8	2	2	2	2 0.0
Miscellaneous uses	8	9	10.5	10.5	11 0.5
	71	75.5	<del>79</del> 83	93	105 2.0

#### Furfural Supply

QO Chemicals is the only producer of furfural in the United States. QO Chemicals is a privately-owned company that purchased the Quaker Oats Chemical Division from Quaker Oats during the first half of 1984. QO Chemicals is essentially being operated by the former management of Quaker Oats Chemical.

The current capacity for furfural is 132 million pounds per year. QO Chemicals is currently operating plants at Omaha, Nebraska; Cedar Rapids, Iowa; and Belle Glade, Florida. QO Chemicals was operating a fourth plant at Memphis, Tennessee, but this facility was shutdown this summer. Its capacity was 40 million pounds per year. All of these plants have been in operation for a number of years, although each has probably undergone considerable maintenance. This is because of the corrosion problems inherent with the conversion of agricultural wastes to furfural.

In addition to the above, Quaker Oats operated a fifth plant at Bayport, Texas between 1976 and 1983. This facility had a capacity of 44 million pounds per year and used rice hulls as raw material. According to several sources, the plant never operated at capacity and was plagued by severe corrosion problems. This facility was closed prior to the sale of the division to QO Chemicals.

QO Chemicals is also the only domestic producer of furfuryl alcohol. These facilities are located at Omaha, Nebraska and Memphis, Tennessee. Quaker Oats also had a furfuryl alcohol plant at its Bayport plant site. Chem Systems believes this facility has been moved to one of the other two sites to replace one or more older units. QO Chemicals' furfuryl alcohol capacity is estimated to be in excess of 50 million pounds per year.

QO Chemicals' U.S. plants and their capacities are listed in Table V-8-18.

TABLE V-B-18

QO CHEMICALS U.S. FURFURAL PLANTS
(Million Pounds per year)

<u>Location</u>	<u>Capacit</u>	<u>ጀ</u>	ricultural Feed	dstock
Omaha, Nebraska	40		orn cobs	
Cedar Rapids, Io			orn cobs, oat	nulls
Belle Glade, Flor	rida 72	DOMESTIC STATE OF STA	ugar cane bagas	sse
Total	$\overline{132}$			

Due to the fact that the sources of supply of furfural are dominated by countries with developed agricultural economies, there is considerable international trade between these sources and the industrialized nations which are the primary consumers of furfural and its derivatives. The United States stands out as the major supplier and user of furfural.

Even though there has been very little publicity given to the South Puerto Rico Sugar Company, a subsidiary of Gulf & Western Industries, its plant in the Dominican Republic is a major source for furfural. The company does not convert furfural to FA or other derivatives.

Table V-B-19 shows that furfural imports increased substantially in 1977-1979, then declined to the 5-10 million pound level in 1980-1983. However, imports during the first seven months of 1984 have already exceeded that level, amounting to 11.5 million pounds. For the year, imports could amount to as much as 18-20 million pounds with no change in the current rate of shipments.

TABLE V-B-19
U.S. IMPORTS OF FURFURAL, 1976-1983
(Thousand Pounds)

<u>Year</u>	Dominican Republic	South <u>Africa</u>	<u>Other</u>	<u>Total</u>
1976	552	- · · · · · · · · · · · · · · · · · · ·	526	1,078
1977	3,615		3,504	12,119
1978	16,668	378	5,441	22,987
1979	24,230	3,121	1,587	28,938
1980	2,899	416	1,722	5,037
1981	3,614	484	251	4,349
1982	5,489	2,141	38 <sup>1</sup>	7,668
1983	6,363	2,134	682	9,179

The importance of South Puerto Rico Sugar as a world supplier of furfural cannot be overlooked. Their nameplate capacity is 72 million pounds per year, which places them second to QO Chemicals as a free world producer. However, South Puerto Rico Sugar has been extremely reluctant to discuss any aspect of its furfural business. Since its basic business is sugar, it is highly likely that all of the furfural the company produces is sold through an intermediary.

The other important overseas source of furfural is SmithChem in South Africa. This company has been exporting furfural to the United States since 1978 with shipments in the last two years amounting to just over 2 million pounds. Several industry contacts noted that SmithChem has recently expanded its furfural plant and intends to increase its market position in the United States. However, this will more than likely result from an increase in furfuryl alcohol rather than furfural sales.

Data on U.S. exports of furfural are not available. However, the destination of U.S. exports in past years include Western Europe, Japan, Canada and Latin America. Chem Systems believes that U.S. exports may have amounted to 30-35 million pounds in the mid-1970s and 10-15 million pounds in the late 1970s. However, considering the global overcapacity in furfural and the increased level of competition in world markets, it is believed that U.S. exports are probably in the range of 5 million pounds in the last couple of years.

#### Furfural Pricing

Furfural is quite readily available in the United States and should remain so for all of the forecast period. QO Chemicals will probably not operate much above 50 percent of effective capacity in 1984 and, given the availability of furfural worldwide, can not be expected to improve this level of production in the next several years as shown in Table V-B-20.

TABLE V-B-20
U.S. FURFURAL SUPPLY/DEMAND BALANCE
(Million Pounds)

	<u>1983 1984 1985 1990 1995</u>	2001
Nameplate capacity	172 152 132 132 132	132
Effective capacity*	154 134 114 114	114
Domestic demand Exports	71 75.5 79 83 93	105
Imports	9 20 10 10 12	14

<sup>\*</sup>Based on Belle Galde plant operating 9 months a year.

This potential surplus supply of furfural in the United States should continue to exist unless one or more of the following changes occurs in the U.S. market.

- QO Chemicals closes one or more of the three furfural plants it is now operating.
- The market for furfural undergoes a significant change resulting in increased demand for either furfural or its derivatives.

Of the two, the former is likely to occur if no real change takes place in the outlook for furfural, or by-product furfural becomes available as a result of the cellulose-based manufacture of ethanol. The later is likely to occur if the market price for furfural is significantly reduced, opening up opportunities in existing or developing applications for furfural.

Two distinct opportunities exist to increase the demand for furfural beyond that forecast in Table V-B-20. Both are directly related to the sensitivity of the markets for foundry resins and THF to changes in the price of furfural.

The two key raw materials in the hot-box and no-bake resin segment of the foundry resin market are furfuryl alcohol and phenol. Currently, the market price for furfuryl alcohol is 62-64 cents a pound to large volume users compared to 30-31 cents a pound for phenol. This difference is directly related to the difference in the price of FA and phenolic resins. FA resins are now selling for about 75 cents a pound while phenolic resins sell for 30-40 cents a pound. Although some foundry resin producers are selling modified FA resins for 50-55 cents a pound in an attempt to offset this disadvantage, the modified resins lack the performance characteristics desired by most foundry operators.

All of the foundry resin producers agree that the foundry resin market is sensitive to raw material price changes. The most competitive segment of the market are the phenolic no-bake and the FA no-bake resins. Here,

changes in raw material prices could result in significant changes in future demand. Although FA hot-box resins have lost most of this market to phenolic hot-box resins, this is a segment of the market where pricing could also bring about changes.

Chem Systems believes that FA resin demand in 1990 could increase by 30 million pounds compared to 1984 demand if the price of furfuryl alcohol would be reduced to 45 cents a pound (1984 constant dollars). The demand could probably be increased by a further 15 million pounds if the price of furfuryl alcohol could be reduced to 40 cents a pound (1984 constant dollars). Such prices would be equivalent to a furfural price of 30 cents a pound in the former case and 25 cents a pound in the latter case, both expressed as 1984 constant dollars.

Table V-B-21 presents the demand for FA resins based on furfuryl alcohol being sold at 45 cents a pound (constant 1984 dollars). The revised demand forecast is based on the following assumptions:

- By-product furfural would not become available until 1987-1988.
   As a result, the impact of lower furfuryl alcohol prices would not be reflected in the forecast until 1990.
- The market adjustment to lower furfuryl alcohol prices would be completed by 1990 and the demand for FA resins in subsequent years would reflect the maturity of the foundry resin market with no further shift in demand patterns.
- The primary penetration of FA resins would be felt by phenolic no-bake and hot-box resins. However, FA resins would also replace about 10 percent of the phenolic urethane cold-box and no-bake resins.

TABLE V-B-21

DEMAND FOR FA RESINS IN FOUNDRY CASTING - REVISED

(Million Pounds)

	<u>1983</u>	1984	1985	1990	1995	2001
FA resins No-bake Hot-box Other	36 3 - 39	40 3 - 43	43 3 -	63 18 3	68 19 4	74 21 4
Furfural alcohol Furfural	28 28	31 31	33 33	56 56	61 61	66 66

A reduction in the price of furfural should also put furfural-based THF on an equivalent cash cost basis with that based on 1,4-butanediol. This should enable QO Chemicals to compete more effectively in the domestic and export segments of the THF market. If QO Chemicals could double its market share in 1990, the market for furfural would increase by 23 million pounds.

In summary, the net effect of reducing the price of furfural in 1990 to 30 cents a pound (1984 constant dollars) would be to increase the domestic demand from 83 million to 126 million pounds in 1990.

The current and future pricing of furfural is summarized in Table V-B-22, showing prices for both market situations, i.e., with and without the availability of by-product furfural from wood-based ethanol facilities.

TABLE V-B-22

U.S. FURFURAL PRICES

(Cents Per Pound)

	<u> 1984</u>	1985	<u>1990</u>	1995	2001
Without by-product Furfural available	47	47	55,		67
With by-product Furfural available	_		<b>4</b> 0	51	68.5

#### 3. <u>Carbon Dioxide</u>

There are many potential uses for the carbon dioxide by-product from a biomass-based ethanol plant. Figure V-B-1 shows historical carbon dioxide demand and how it is likely to change in the late 1980s.

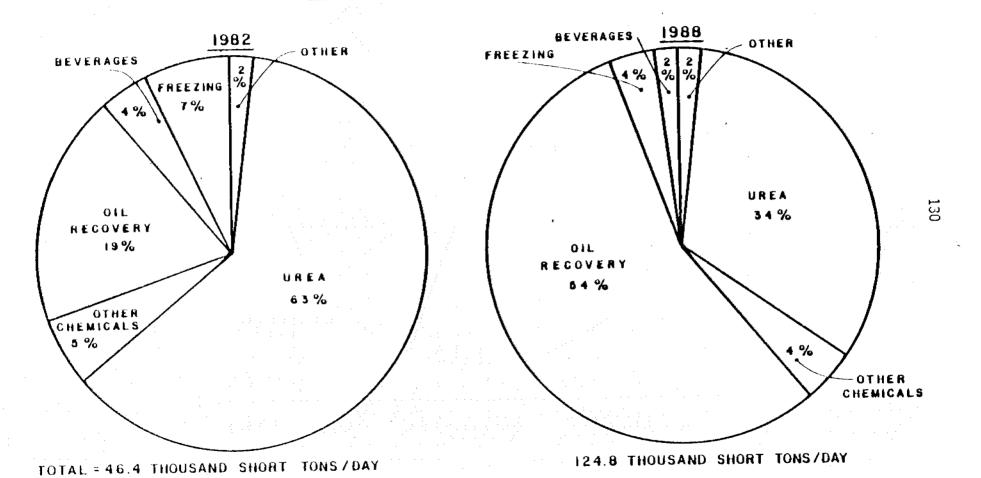
Growth in demand for traditional carbon dioxide end uses will be substantial. The largest end use historically has been in urea production, although carbon dioxide for this use is usually supplied captively from integrated ammonia plants.

Carbon dioxide is used in food freezing where it competes with nitrogen for markets. This application has grown sharply over the last decade and is expected to continue to do so, especially to supply frozen foods for the fast-food industry and for home microwave cooking. Consumption of carbon dioxide for beverage carbonation has also increased rapidly in the past. However, as a maturing market, demand will probably rise to close to levels for other consumer-related goods.

Other areas of demand in the chemical process industry include use as an inerting agent and as a raw material in the production of methanol, and sodium and other inorganic salt carbonates. In methanol production from natural gas, carbon dioxide is reacted with excess hydrogen to make up for carbon deficiencies in the synthesis reaction feed.

The predominant source of domestic carbon dioxide (Figure V-B-2) has historically been ammonia plants, which yield extremely high purity carbon dioxide. But high energy and feedstock costs, high interest rates and bumper grain crops have severely impacted the U.S. fertilizer and ammonia industry. Within the past three years, approximately 20 percent of domestic ammonia capacity has been shut down, which has had a major effect on carbon dioxide supplies.

# CO2 DEMAND\*

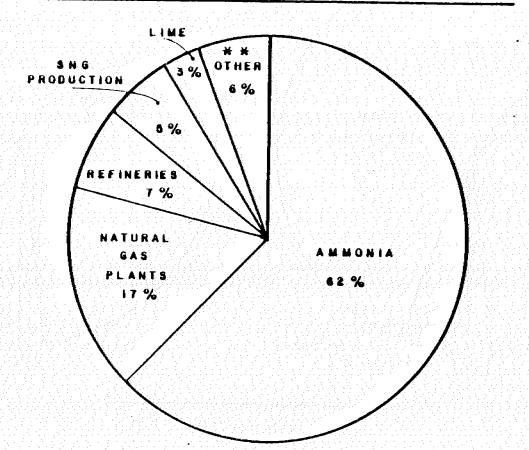


W PRELIMINARY ESTIMATES

INC.

CO2 RECOVERY

RECOVERY CAPACITY \*



1982 TOTAL = 53.6 THOUSAND SHORT TONS/DAY

H PHELIMINARY ESTIMATE

1982

H H NATURAL DEPOSITS, ETHYLENE OXIDE, FERMENTATION, COMBUSTION FLUE GASES, SODIUM PHOSPHATE



By the end of this decade, the largest single source of carbon dioxide will be high-purity natural deposits in the Rocky Mountains, which will be developed to supply new EOR projects in west Texas and New Mexico. This material will be delivered via pipelines now under construction or being planned. As these deposits are tapped, they may become important sources of carbon dioxide for alternate markets, as well. Furthermore, as the EOR projects are operated, they will produce perhaps one-quarter to one-half of the injected carbon dioxide in associated production. This CO2 can be reinjected in the same or nearby projects which will come on stream later.

Carbon dioxide off-gas from a fermentation ethanol plant can be a highly desirable product, depending on location. Many industrial gas companies such as Airco, Cardox, Liquid Carbonics, UGI actively seek out such sources of carbon dioxide and are prepared to enter into a long-term supply contract. These companies will typically install a liquefaction plant adjacent to the ethanol plant to minimize compression costs and then market the liquid carbon dioxide product usually for beverage carbonation.

For a Midwest location, the industrial gas companies have been paying 5-7 dollars per ton for the raw carbon dioxide stream. These are typically ten- to fifteen-year supply contracts with a built-in escalation tied in some manner to the ultimate price received for the liquid carbon dioxide product. A carbon dioxide liquefaction plant in the 300 ton per day capacity range will require an investment of about 6 million dollars. The cost of production, including a return on investment, will be in the 50-60 dollar per ton range, with the raw carbon dioxide gas priced in at 5-7 dollars per ton. Table V-B-23 summarizes the manufacturing cost for carbon dioxide liquefaction on a mid-1984 basis. Table V-B-24 shows a price forecast for raw carbon dioxide by-product assuming an arrangement is made with one of the companies mentioned above and wherein the price escalates with general inflation. Also shown in Table V-B-24 is a price forecast for liquefied carbon dioxide as projected for the Michigan ethanol facility.

TABLE V-B-23

# LIQUEFIED CARBON DIOXIDE MANUFACTURING COST SUMMARY (300 Tons per Day, mid-1984)

<u>C</u> ent:	s per	Pound
Raw materials (1)	0.36	
Utilities	0.30	The second secon
Operating cost	0.38	
Overhead expenses	0.39	3
Cash cost of production	1.56	1.14
Depreciation	$\frac{0.65}{2.22}$	
Net cost of production Carbon dioxide sales price at 10	2.44	
percent DCF return	2.8	

(1) Raw carbon dioxide at \$6 per ton.

TABLE V-B-24

CARBON DIOXIDE BY-PRODUCT PRICE FORECAST

(Dollars per Ton)

Year Raw CO2 Gas Liquefied CO2 1987 6.30 64.40 1988 67.60 6.60 1989 6.90 70.60 1990 7.30 74.80 1991 7.60 77.80 1992 7.95 81.40 1993 8.30 85.00 1994 8.70 89.20 1995 9.10 93.20 1996 9.50 97.40 1997 9.90 101.40 1998 10.35 106.00 1999 10.80 110.60 11.30 115.80 2000 11.80 2001 120.80

#### VI. ECONOMIC ANALYSIS

#### A. Base Case Economics

#### 1. Capital Cost Estimate

A capital cost has been estimated for a plant producing 25 million gallons per year of ethanol. The total inside battery limits (ISBL) cost is estimated at 66.26 million dollars based on mid-1984 costs for a Michigan site location. The cost estimate is based on Icarus cost data along with vendor equipment costs for major items.

A detailed equipment list for all major equipment items is presented in Appendix B. This list includes the design specifications for each item and its estimated purchase price on a mid-1984 basis. A breakdown of total ISBL capital cost by section is given in Table VI-A-1. Section 700, Carbon Dioxide Recovery System, is included in the ISBL as a complete turnkey facility.

The offsites include capital for storage, a steam/power cogeneration system, a wood handling facility, a cooling water system, waste treatment and other miscellaneous systems as described in Section IV-A-12. The offsites costs are summarized in Table VI-A-2.

## 2. Cost of Production Analysis

Table VI-A-3 is a production cost estimate for the base case 25 million gallon per year ethanol from wood plant. This estimate serves as the basis for the cash flow analysis presented in the following section. The costs are based on a mid-1984 time frame for the Michigan site. Most of the items in the cost of production estimate are self-explanatory. The by-product quantities, raw material and utility consumptions per gallon of ethanol produced have been discussed in Section IV-D.

TABLE VI-A-1

ISBL\* CAPITAL COST SUMMARY FOR BASE CASE, MID-1984

<u>Section</u> <u>Name</u>	Purchased Equipment, \$	Total Field Installed Cost, \$
100 Pretreatment/Prehyrolysi 200 Sugar Separation and	s 1,683,000	3,104,000
Neutralization	2,052,400	3,901,000
300 Enzyme Production	2,972,000	6,376,000
400 Enzyme Hydrolysis	4,245,000	7,069,000
500 Sugar Concentration	3,731,000	5,941,000
600 Fermentation	1,714,000	4,209,000
800 Ethanol Purification	1,153,000	4,111,000
900 Furfural Production 1000 Heat Generation	1,638,000	5,166,000
1000 Heat Generation	<u>2,100,000</u>	3,311,000
Total Purchased Equipment	21,238,400	
Total Field Cost	21,200,400	43,188,000
Engineering		4,180,000
Construction Overhead		4,350,000
Bare Plant Cost		52,218,000
Contingency		5,222,000
Contractors Fee		1,509,000
Special Charges on the first enterties at		1,550,000
Total Facility Cost		60,490,000
CO2 Recovery System		5,800,000
Total ISBL		66,290,000

<sup>\*</sup>Section 1100, Waste Treatment, is included in offsites costs.

# TABLE VI-A-2 OFFSITES CAPITAL COSTS

	Total Installed Cost, \$Mid-1984
Storage Ethanol (14 days, 1,020,000 gals) Calcium hydroxide (14 days, 160,000 gals) Sulfuric acid (14 days, 85,000 gals) Caustic (14 days, 40,500 gals) Corn steep liquor (14 days, 40,500 gal) Furfural (14 days, 160,000 gal) Sodium carbonate (14 days, 23,500 gal)	676,800 560,000 195,000 82,600 82,600 171,000
Total storage	1,928,000
Steam boiler (320,000 lb/hr, 1,200 psig steam, wood-fired)	11,700,000
Generator/turbine/switch gear (12,400 kw)	6,600,000
Wood handling system (104 tons per hour)	8,100,000
Cooling water (27,000 GPM)	2,760,000
Electrical system from main power grid (3,275 kw)	982,000
Boiler feedwater system (440 GPM)	490,000
Pollution control Pond and gypstack (Section 1100) Sanitary waste (0.5% ISBL less CO2 system)	2,560,000 300,000
Buildings (3% ISBL less CO <sub>2</sub> system)	1,815,000
General utilities (5% ISBL less CO <sub>2</sub> system)	3,024,000
Site development (6% ISBL less CO2 system)	3,630,000
Piping (3% ISBL less CO <sub>2</sub> system)	1,815,000 45,704,000

NET COST OF PRODUCTION

REQUIRED SALES PRICE AT 10% DCF

39,312 157,28

206.5

#MILLION 36.3

112.0

# TABLE VI-A-3

# COST OF PRODUCTION ESTIMATE FOR ETHANOL PROCESS- ENZYME HYDROLYSIS

# CAPITAL SUMMARY

Harrier - Carliage	numeki					
RASIS Location: Michigan Mid-1984 Capacity: 25.00 million gallo	ns/vr		AL COST y Limits tes			
Str. Time: 8000 hours per year		and the first of the control of the	Fixed Inv ng Capital			
PRODUCTION_COST_SUMMARY						
UNITS  RAW MATERIALS PER GAL  Mixed Hardwood, Lb 66.88212  Sulfuric Acid, Lb 2.23180  Calcium Hydroxide, Lb 1.66251  Sodium Hydroxide, Lb .24741  Catalyst & Chemicals	¢/ <u>UNIT</u> .9 2.8 1.6	COST, \$M 15,046 1,562 665	CENTS PER GAL			
TOTAL RAW MATERIALS  UTILITIES Purchased Power, kWH .22404  Generated Power, kWH 3.96878  Bfw, M Gal .00641  Cooling Water, M Gal .43609  Steam, 250 psia, M Lb .08612  Steam, 600 psia, M Lb .01480	139.0 9.8 80.0	223 1,048	30.97			
TOTAL UTILITIES  OPERATING COSTS Labor, 46 Men @ \$ 27,900 10 Foremen, 9 Men @ \$ 31,700 1 Supervision, 3 Man @ \$ 38,200 3 Maint., Material & Labor 6% of	m/s Man	4,972 1,283 285 115 3,978	19.39			
TOTAL OPERATING COST  OVERHEAD EXPENSES  Direct Overhead 45% Lab. & S  Gen. Plant Overhead 65% Oper. Co Insurance, Prop. Tax 1.5% Tot. Fi	sts	5,661 757 3,680 1,680	22.65			
TOTAL OVERHEAD EXPENSES BY-PRODUCT CREDIT Carbon Dioxide, Lb 6.51328 Furfural, Lb 1.46013	2.9 30.0	6,117 -4,558 -10,949	24.47			
TOTAL BY-PRODUCT CREDIT  CASH COST OF PRODUCTION		15,507 21,482	The state of the s			
DEPRECIATION 20% ISBL + 1						
		and the second s				

Raw material costs total 80.97 cents per gallon of ethanol produced hardwood priced on at 18 dollars per wet ton, sulfuric acid at 56 dollars per ton and calcium hydroxide and sodium hydroxide at 32 dollars and 150 dollars per ton, respectively. A breakdown of catalyst and chemicals costs is presented in Table VI-A-4.

TABLE VI-A-4

CATALYST AND CHEMICALS SUMMARY

	Unit Cost (\$/1b)	Lbs/Yr	Annual Cost, \$
Cellobiase (65% moisture) Sodium carbonate Corn steep liquor (46% moistur Benzene	0.700 0.062 e) 0.081 0.210	1,928,000 3,024,000 11,784,000 56,000	1,349,600 187,500 954,500 11,200 2,502,300

Total utility costs amount to 19.89 cents per gallon of product. The operating and overhead expenses, both largely capital related, contribute 22.65 and 24.47 cents per gallon to the cost of ethanol.

In this facility the commercial by-products are furfural and carbon dioxide with by-product prices of 2.8 and 30.0 cents per pound for carbon dioxide and furfural respectively, there results a credit of 62.04 cents per gallon of ethanol.

The cash cost of production is 85.94 cents per gallon. The net cost of production, which includes depreciation, is 157.28 cents per gallon of product. Depreciation is calculated on a 5 and 10-year straight line basis for ISBL and offsites capital investments, respectively, accounting for 71.33 cents per gallon. In order to achieve a 10 percent discounted cash flow rate of return, an ethanol sales price of 206 cents per gallon is required.

## 3. Cash Flow Analysis

#### a. 100 Percent Equity Financing

A discounted cash flow (DCF) financial model has been developed for the ethanol facility project economic analysis. Table VI-A-5 presents the model summary for the base case design. The capital and production cost estimates presented above serve as the basis for the model.

Parameters used in the cash flow analysis are defined below:

- Project life the project has a 15-year operating life following a 3-year construction period starting in 1984 and zero salvage value at the end of project life.
- Land cost for this plant, it is estimated that the Michigan site consisting of 130 acres would cost 0.1 million dollars in 1984.
- <u>Fixed investment cost</u> the total capital investment cost in 1984 dollars was presented earlier. A 3-year construction period, 1984 through 1986, is assumed.
- Start-up cost these include provision for the expenses of a technical start-up team, consumption of feedstock and utilities, and other operating costs during this period of little or no real production. These costs are estimated at 10 percent of ISBL capital cost and are expended in the year of plant start-up (i.e., the last year of plant construction). These costs are amortized on a 5-year straight line basis and deducted from operating profit prior to the calculation of income taxes. These are included in the depreciation expense.
- Working capital working capital requirements for the project are calculated using the procedure presented below:

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GENERAL PLACE OVID.	. 0	.0	.0		71.7	75.0	75.3	~5.6	1,1 5,9	71.2 6.2
THSURABLE	, 6	. 0	. 0	2.0	2,1	72.3	12,4	2.5	2.7	2.0
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#### TADLE - VI A-5 ENZYME HYDROLYSIS 1002 FOULLY ETHANOL FROM CELLULOSE

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DCF\_TRR

- Feedstock inventory 2 months supply of delivered chips plus 1/2 month's supply of the other raw materials valued at delivered prices.
- Finished product inventory = 1/2 month's supply of principal products and by-products valued at gross cost of production.
- Accounts receivable 1 month's gross cost of production.
- Cash 1 week's out-of-pocket expenses estimated at gross cost of production less depreciation.
- Warehouse/spare parts inventory 3 percent ISBL capital cost.

#### LESS

 Accounts payable - 1 month's supply of raw materials at delivered prices.

Because working capital requirements increase in each year of the project life due to the impact of inflation and production levels, the annual increases are added as a negative cash flow item in the analyses. The total working capital expended is recouped in the last year of the project as a positive cash flow.

- Interest during construction the project is based on 100 percent equity financing. Therefore, there is no interest or loan repayment required in the cash flow model.
- Operating capacity buildup new process plants do not immediately reach nameplate production at start up. To allow for this, Chem Systems has assumed a capacity buildup to nameplate levels of three years, with plant operating capacities at 60 percent of nameplate in the year of start up, 80 percent in the second year of operation and 100 percent in the third year of plant operation. Operating levels remain at 100 percent of nameplate for the remaining years of plant operation.

Escalation - the discounted cash flow analysis is performed on a current dollar basis. It is therefore necessary to escalate the elements in the financial analysis to reflect escalation in capital costs, product revenues and operating costs. Table VI-A-6 presents Chem Systems' estimate of the escalation rates for general inflation, capital costs and utility prices for the period 1984 through 2001.

## TABLE VI-A-6

#### ESCALATION RATES, 1984-2001 (Percent per Year)

<u>Period</u> <u>G</u>	NP Deflator	Capital	Cost	Utility Costs
1984-1990	5.0	6.	5	7.5
1990-2001	4.5	5.	5	6.9

- Revenues the revenues are received from sale of the ethanol product as well as carbon dioxide and furfural by-products. Projected ethanol prices are based on a combined federal/state tax incentive of nine cents per gallon for 10 percent ethanol blends in gasoline. The price forecasts over the life of the project are presented in Section V.
- Expenses the expenses correspond to those presented in the cost of production estimate for 1984, Table VI-A-3. For the years in the project life the expense elements in the financial analysis are escalated as presented in Table VI-A-7.

# TABLE VI-A-7

# ESCALATION RATES FOR EXPENSES

		and the second of the	
Wood		GN	IP .
Chemicals		GN	IP .
Utilities			ilities
Labor		GA	IP .
Maintenance		Ca	pital
Direct overhead		GN	IP .
General plant overh	ead	Ca	pital
Insurance		Ca	pital

- Operating margin this is the revenues less operating expenses.
- MSARD Expense this is the cost of management, sales, administration, research and development for the operation of the plant, taken at six percent of the total revenues.
- <u>Depreciation</u> this is a non-cash cost and is based on straight line methods over five years for ISBL capital and ten years for offsite capital. The amortization of start-up expenses over five years is also included.
- IDC Amortization interest during construction (IDC) does not apply.
- Gross profit this is the operating margin less MSARD and depreciation expenses. It is the same as the adjusted profit since there is no interest expense.
- Loss carryforward the losses for each operating year are carried forward and credited against tax liabilities in future years. In this project, the loses would be taken against the tax liabilities of 1992-1993.
- Taxable profit this is the adjusted profit before tax less losses.
- Corporate tax corporate tax is taken at 45 percent of profit.
   Losses are allowed to be carried forward at 100 percent toward the next years taxable income. Unused tax credits are carried forward until depletion.
- Investment tax credit the investment tax credit is 10 percent of the total capital expenditure for the plant and is credited against current or future tax liabilities. In this case the credit is used in 1993 through the year 1995.

- Net profit this is the after-tax profit, but is a non-cash profit since depreciation is included as an expense.
- Cash flow this is the net profit plus depreciation.
- IRR the internal rate of return (IRR) for this project (based on 100 percent equity financing) is 5.9 percent.

## b. Debt Financing

An alternative to 100 percent equity financing in the DCF cash flow analysis is financing of the investment capital including land, fixed investment, start up and initial working capital based on 30 percent equity and 70 percent debt. This analysis is presented in Table VI-A-8. All the data are the same as the 100 percent equity case presented above with the exception of financing which involves interest and loan repayments. In this project the average interest rate is taken at 13 percent per annum and the loan repayment commences in the first year of operations. In the construction years the interest is capitalized (financed with debt and equity) and amortized in the first five years of plant operations.

With debt/equity financing the IRR on equity is negative for this project, indicating the high cost of debt financing relative to the present worth of the potential profits.

# c. SERI's Comparative Case

In order to obtain a cash flow analysis which can be compared with economic analyses presented by other contractors, SERI has provided a set of economic parameters (Table VI-A-9) to be used for a standard cash flow analysis. Using these values, a base year 1984 cost of production estimate was calculated (Table VI-A-10). The cash flow analysis for 1984 through 2001 is presented in Table VI-A-11. In this analysis the ethanol selling price was calculated such that a 15 percent DCF return on equity

PATE RUN: 10/3/84 FILE NUMBER: 67			ENZYNI ETI	TABLE - 1 E HYDROLYSI IANOL FROM	/1- A-8 (8-30% EQU) CELLULOSE	ŢΥ			DISKETT	E - 26
PARITAL EVENTURES	1904	1985	1986	1987	1988	1909	1990	1991	1992	1993
APITAL EXPENDITURES	1	. 0	0	.0	. 0	. 0	. 0	. 6	, 6	
TXED INVESTMENT	39.6	39.6	39.6	. 0	Û	Ö	. 0	ō	0	,
TART-UP COST	, Ø	. 0	7.0	Ō	Ö	0	Ö	ű	Ö	
ORKING CAPITAL	, 0	, 0	. 0	"8.9	2.4	2.7	6	ē	. 6	··· , i
HITERESE DURING CONS	71.9	5.9	10.5	, 0	. 0	. 0	, Ó	. 0	. 0	
OTAL CAPETAL EXPEND	"H1.6	45.5	57.1	-8.9	2,4	~2.7	~.0		".8	
EBT	29.1	~31,B	"10.0	76.2	. 0	. 0	. 0	. 0	. 0	
QUITY	12.5	13.6	~17.1	72.7	~2.4	2.7	0	8	0	
eerating rate, %	, 9 ,		. O	(*************************************	**************************************	100.0	**************************************	100.0	**************************************	****** .00
EVERUE	. ö	. 0	Ö	40.8	54.5	73.3	76.1	78.6	91.3	84,
XPENSES ***I		******	******	******	******	****	******	*****	****	******
000	. 0	. 0	.0	10,4	~14 · 6	19.2	720.1	721.0	122.0	· 23.
IENTCALS	. 0	, 0	. 0	-3.6	75.0	~6.6	~6.9	7.3	*7.6	~7.
TILITIES	. 0	. 0	. 0	~3.7	5.3	77.2	~7. <b>7</b>	70.2	~8.8	~9.
ABOR	. 0	, 0	. 6	~1.1	71.6	~2.1	2.2	72.3	" <u>2</u> . lj	2.
AINTENANCE	, 0	.0	. 0	~4.7 .	~5.0	~5.3	5.7	~გ.0	~6.3	~გ.
LRECT OVERHEAD	0	. 0	, Ø	- 9	. 9	-1.0	1.0	$^{-1}.1$	~i.i	1.
ENERAL PLANT OVHO,	, 0	. 0	, 0	~կ , կ	4.7.	<sup>-</sup> 5,0	<b>~5.3</b> .	. "5. <b>6</b>	~5.9	~6.
NSURANCE	ű	٨	. 6	72.0	~2. <b>1</b>	2.3	72.4	2.5	12,7	~2.

TOTAL EXPENSES ~30.<del>9</del> 139.4 151.4 48.6 54.1 ~56.9 59.8 PROFIT, TAX AND CASH FLOUR \*\*\*\* OPERATING MARGIN . 0 9.9 17.1 24.7 24.4 24.5 24.4 24.2 MSARD EXPENSES . 0 . 0 . 0 72.4 73.4 74,4 ~4.6 14.7 74.9 5.0 DEPRECIATION . 0 . 0 120.3 **~20.3** ~9.9 . 0 ~20.3 720.3 20.3 74.9. 3.7 IDC AMORTIZATION . 0 . 0 ~3.7 ~3.7 \_\_\_3.7 73.7 GROSS PROFIT . 0 14.5 ~3.7 ~3.9 . 0 10.2 ~4.2 . 0 14.7 14.3 INTEREST PAYMENT . 0 . 0 , 0 ~13.9 "13.6 13,2 112.9 12.3 11.7 ~11.1 ADJUSTED PROFIT . 0 . 0 . 0 730.4 723.B 116.9 16.7 \*16.4 3.0 3.3 LOSS CARRYFORUM . 0 . 0 . 0 . . . 0 . . 0 . . . . 0, . . . 0 0 3,0 ~3.3 TAXABLE PROFIT . 0 . 0 730.4 ~23.A 7.16.7 . 0 16.9 16,4 . 0 . 0 CORPORATE TAX @ 95 % , Ü . 0 . 0 . 0. , Ø . 0 . 0 . 0 . 0 . 0 INVESTMENT TAX CRED. , Ü . 0 . , 0 .....0 . 0 . 0 , 0, . 0 NOT PROFIT 723.8 16.9 716.4 . 0 . 0 30.4 ~16.7 . (1 . 0 LOAN REPAYMENT . 0 . 0 . 0 72.7 73.0 ~3, IF 73.0 74.3 "4.9 5.5 DEPRECIALTON. . 0 20.3 20.3 20.3 20.3 20.3 4,9 4.9 FDC AMORTIZATION , 0 3.7 3.7 3.7 3.7 3.7 . 0 , Ű LOSS CARRYFORUD . 0 . 0 3.0 3,3

30 % FQUITY 12.5 13.6 17. E 711.8 5.4 2.7 2.5 2.2 1.7 **XXXXXXXXXXXXXXXX** DCF 1RR 6.0%

DISKETIE 📑

# TABLE - VI A-8 ENZYME HYDROLYSIS-302 EQUITY ETHANOL FROM CELLULOSE

CAPITAL EXPENDITURES	1994	1995	1996	1997	1996	1999	2000	200
LAND FIXED INVESTMENT	0	. U	0	0	, O	.0	. 0	
START-UP COST	Ö	.0		0	. 0	0	. 0	
WORKING CAPITAL	9	ÿ	1.6	1.0	1.1	1.1	0	***
INTEREST MIRTHG CONS	.0	. 0	0	.0	. 0	. 0	1.2	24.0
TOTAL CAPITAL EXPEND	- 9		1.0	1.0	1.1	1.1	1.2	/ F175
DE DT	.00	. 0	. 0	0	Ô	. û	.0	24.t
инклининининининининин ЕППГІЛ	· .9	.9	1.0	-1.0	1.1.	1.1	1.2	24.4
OPERATING RATE, X	100.0	100.0	100.0	106.0		100.0	100.0	. 100 .
REVENUE NA	86.9 ********	89.8	92.9	96.1	99.5	103.0	106.6	110.3
uoon	24.0	25.1	26.2	27.4		**************************************	**************************************	HKHHKMI
CHENICALS	8.3	0.7	9.0	9.5	9.9	10.3	10.0	32.7 11.3
UIII.ITIES	10.1	710.8	11.5	<sup>-</sup> 12.3	13.1	14.0	15.6	16.0
LABOR HAINTENANCE	2.6		72.9	T3.0%	73.1	3.3	3,4	3.6
DIRECT OVERHEAD	7.0 1.2	7.4	7.9	8.3	B. 7	9.2	7.7	10.3
DENERAL PLANT OVID.	6.6	1.3 7.0	71.4 77.3	1.4	1.5	1.5	1.6	1.7
INSURANCE	73.0	3.2	3.3	7.0	8.2 3.7	8.6	9.1	79.6
					3. f	<b>-3.9</b>	4.1	`
TOTAL EXPENSES PROFIT, TAX AND CASH FLO	62.9	66.1	69.5	73.1	76.9	80.9	05.1	-89.6
OPERATING MARGIN	1619 8 9 9 8 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9	23.7	*****	*****	******	********	*****	******
HSART EXPENSES	5.2	23.1 5.4	23.4 5.6	23.0	22.6	22.0	21.5	20.8
DEPRECIATION	4.9	4.9	4.9	5.9 .0	6.0	6.2	6.9	6.6
TOC AMORTIZATION	. 0	Ō			. 0 . 0	.0	.0	. 0
GROSS PROFET						A HASIR		
INTEREST PAYMENT	13.9	13.5	13.0	17.2	16.6	15.9	15.1	. 14 I
	10.3	2.5	6.6	7.6	6.4	5.1	3.6	1.9
ADJUSTED PROFTT	3.6	3.9	4.3	9.7	10.2	10.8		
LOSS CARRYFORUM	3.6	"3.9	4.3	~9.7	10.2	10.8	11.5	12.2
						nya <b>tu</b> k <b>u</b> kuya. Nadrodet	II.U	12.2
TAXABLE PROFIT	. 0	0	. 0	.0	. 0	. 0	.0	. 0
CORPORATE TAX @ 45 %	0		0	. 0	. 0	e e	0	. 0
THA LEET	. 0	<u></u>	. 0	. 0	. 0	. 0	. 0	. 0
VET PROFIT	. 0	<b>C</b>	•	0				
JOAN REPAYMENT	6.2	7.0	0.0	9.0	10.2		.0	0
EPRECIATION	4.9	4,9	4,9	Ü	10	11.5	13.6	74.7
LIC AMORTIZATION	. 0	.0	.0	0	ů.	Ü		.0
OSS CARRYFORUM Chripping and ankananae	3.6 *********	3.9	4.3	9.7	10.2	10.8	11.5	12.2
		77 7 7 7 7 7 7 7 8 8 8 8 8 8 8 8 8 8 8	**************************************	· 表示背景高级的 ( )	********	**********	<b>ENKNENKN</b> NN	****
30 % FQ01TY	girak di statu. Mga <b>1. 3</b> gan	, <b>(</b> (1)		en vien en e	Alegai ediga Sale <b>I, I</b> Sele	1.9		3. 3. 4. 1. 5. 199. <b>6</b> 0
эмникинынкикиныныны Ж. IRR	иниянияныя) -6.0%	(	*********		*****	*******		. И. В.
- 12:陈紫水水等 30~00~00~10 克油	0,00				机成分类化 电扩展机			

escalation of ethanol, the year-by-year ethanol selling prices required to achieve the 15 percent return are summarized in Table VI-A-12. Since the data used in the analysis do not coincide with Chem Systems forecast, specific details of the analysis are not discussed further.

#### TABLE VI-A-9

# PROVIDED BY SERI

Capacity factor (hours/year) Energy costs (per KWh) Electricity buyback rate (per kWh) Inflation rate (%) Base year for constant dollars Year of cost information	8000 .06 .05 7 1984 1984
Year of first commercial operation (3-yr. const	r. period) 1987
Linear distribution of costs during construction	
Facility lifetime (years)	15
Depreciation tax life (years)	
	5%,22%,21%,21%,21% per year
Salvage value	1% of invocations
Taxes, insurance (each)	1% of investment
Income tax rate (%)	45
Income tax rate (%) Investment tax credit (%)	45 10 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
Income tax rate (%) Investment tax credit (%) Ratio of debt to capitalization (%)	45 10 30
Income tax rate (%) Investment tax credit (%) Ratio of debt to capitalization (%) Ratio of equity to capitalization (%)	45 10 30 70
Income tax rate (%) Investment tax credit (%) Ratio of debt to capitalization (%) Ratio of equity to capitalization (%) Annual rate of return on debt (%)	45 10 30 70 13
Income tax rate (%) Investment tax credit (%) Ratio of debt to capitalization (%) Ratio of equity to capitalization (%) Annual rate of return on debt (%) DCF rate of return on equity (%)	45 10 30 70
Income tax rate (%) Investment tax credit (%) Ratio of debt to capitalization (%) Ratio of equity to capitalization (%) Annual rate of return on debt (%) DCF rate of return on equity (%) Real escalation rate on capital costs (%)	45 10 30 70 13
Income tax rate (%) Investment tax credit (%) Ratio of debt to capitalization (%) Ratio of equity to capitalization (%) Annual rate of return on debt (%) DCF rate of return on equity (%) Real escalation rate on capital costs (%) Real escalation rate on operating costs (%)	45 10 30 70 13 15 1
Income tax rate (%) Investment tax credit (%) Ratio of debt to capitalization (%) Ratio of equity to capitalization (%) Annual rate of return on debt (%) DCF rate of return on equity (%) Real escalation rate on capital costs (%)	45 10 30 70 13

# B. Optimization Alternatives

In order to study variations of the base case design, a version of the enzyme hydrolysis simulation developed for SERI by Chem Systems  $^{(13)}$  was modified to closely approximate this design. This resulted in an ethanol sales price on a mid-1984 basis of 206 cents per gallon using standard economic parameters.  $^{(13)}$ 

#MILLION 36.3

> 112.0 9.6

# TABLE VI-A-10

# COST OF PRODUCTION ESTIMATE FOR ETHANOL - SERI PROCESS- ENZYME HYDROLYSIS

# CAPITAL SUMMARY

BASIS		nder State of the state o	(2) - 보이라는 생각 회사 사람 1985년 - 1985년	20 12 12 13 13 13 13 13 13 13 13 13 13 13 13 13
Location: Michigan			F9711	AL_COST ry Limits
Mid-1984			Offsi	'
Capacity: 25.00 mi		ons/yr		
Str.Time: 8000 hours p	er year		Total Work ii	Fixed Inv. ng Capital
<u>PR</u>	ODUCIION C	OST_SUMMA	RY	
DAIT WATER ON A	UNITS	PRICE,	ANNUAL	CENTS
RAW MATERIALS Mixed Hardwood, Lb	PERLEAL	¢/UNIT	<u>COST</u> , \$M	PER GAL
Sulfuric Acid, Lb	44.88212 2.23180	1.0	77,553	
Calcium Hydroxide, Lb	1.66251	2.8	1,562	
Sodium Hydroxide, Lh.	.24741	1.6 7.5	665	
Catalyst & Chemicals			464 2,503	
TOTAL RAW MATERIALS				
UTILITIES			22,747	91.00
Purchased Power, kWH	.22404	6.0	336	
- Generated Power, kum	3,96878	1.3	1,290	
Sfw, M Gal	.00641	139.0	223	
Cooling Water, M Gal	. 43609	9.8	1,068	
Steam, 250 psia, M Lb Steam, 300 psia, M Lb	.08612	80.0	1,722	
	01480	90.0	333	
TOTAL UTILITIES		·	4,972	
PPERATING_COSTS			7,714	19.89
Labor, 46 Men @ \$ 27,90			1,283	
Foremen, 9 Men @ \$ 31,	700 1	M/S	285	
Supervision, 3 Man @ \$ Maint, Material & Labo	38,200 <u>3</u>	Man	115	
		128F	3,978	
TOTAL OPERATING COS	au		5,661	22.65
RYERHEAR EXCENSES				
Direct Overhead 45 Gen. Plant Overhead 65	% Lab. & Si	up.	757	
Insurance, Prop. Tax 2.	A. SPETA COS	375 - T-"	3,680	
		· · · · · · · · · · · · · · · · · · ·	Z,Z4U	
TOTAL OVERHEAD EXPEN			6,677	26.71
BY-PRODUCT CREDIT Carbon Dioxide, Lb Furfural, Lb	2 53.788			
Furfural, Lb	1.46013	2.8	_ 74,558	
	20013		<sup>-</sup> 10,949	
TOTAL BY-PRODUCT CRE	DIT		15,507	-62.04
CASH COST OF PRODUCTION			======	======
		egation of the second of the second	24,550	
DEPRECIATION 20%	: ISBL + 20	Z OSBL	22.400	80 47
	el el el beb		======	======
NET COST OF PRODUCTION			46,950	187,84
REQUIRED SALES PRICE AT	10% DCF			
				215.5

TABLE - VI-A-11 ENZYME HYDROLYSIS ETHAROL FROM CELLAROSE - SERI ECONOMIC BATA

				****************	OEAR EASI	amanic anith				
	1984	1905	1988	1987	1788	1989	1220	1991.	1992	1993
CAPITAL EXPENDITURES		** ** ** ** **	*** *** ** ***	40 m, arr .gc		To the company				
LAUT		, (1	. 0	. 0	. 0	, θ	. 0	. 0	, ŧr	, (1
FIXEO THVESTHENT START-UP COST	~90.3°	40.3	~40.3	. 0	. 0	. 0	. 0	. 0	. 0	. 0
MORKING CAPITAL	. ()	. 11	7.2	. 0	, 0	. 0.	. 0	, ()	. 0	. 0
INTERFET DURING COMS	. U O	.0 12,41	. 0	1)t ' it	1.2	71.3	7.6.19	~1.5	71.7	14.18
THE PERSON NAMED IN COMPANIES	. []	40 7 14	74.3	. 0	. 0	. 0	. 0	, <b>Q</b> .	, ()	: 0
EDTAL CAPITAL EXPEND	791,2	742.7	51.0	*** <u>1</u> 4 , 14	11.2	on group				
DETCT	12.4.	12.8	15.5	**************************************	. 9	~1.3 .0	‴ Ł . ⅓	1.5	~1.7	. 1.8
ERNITY	726,8	29.9	736.2	"10,I	1.2	" t.3	, () ™1 , 1£	.0 1.5	. 0	, (t
жиндрыянчичичимияния. 	*******	*********	*********	*****	 		**************************************	 	1.7 (нижимжик)	- В.1." ххиянина
OPERACING RATE, %	. 0	. 0	. 0	100.0	100.0	100.0	100,0	100.0	100.0	100.0
REVENOU	. 0	. 0	. 0	76.5	83.1	90.1	97.7	105.0	115.1	179. O
	*********	******	*****	<b>ИНРИНКИКЖЖ</b>	********	**********	*********			(HENSEN NE
0000	. 0	. 0	. 0	~22.4	124.3	726.4	120.6	"31.1	~33.7	36.4
CIENICALS	. 0	, 0	. 0	6.6	77.2	"7. <b>8</b> ]	~O.4;	"♀ <b>,</b> [	9.9	210.8
UTTL###ES	. 0	. ()	. 0	76.Jr	<u>"6.9"</u>	7.5	^ <b>8</b> ,2	~8.9	9.6	" LO . h
LABOR :	. 0	, 0	. ()	2.1	2.3	72.5	2.7	12.9	3.2	13.4
HATRITENANCE UTRECT OVERHEAD	, 0	. 0	. 0	~5.0	~5.4	~5.8	~6.3	"ል.ፀ	7.3	77.9
GUNERAL PLANT OVOR.	. 0	, 0	. 0	"1,0	"1 1	74.2	1.3	1.4	1.5	1.6
INSURANCE	. 0	. 0	. 0	<u> "</u> ፋ . ፋ	75.0	~5.4	15.8	~6.3	"6.8°	7.3
DROWNING C	. 0	.0	. 0		12.3	12.4	2.6	12.9	3.1	3.3
TOTAL EXPENSES	ű .	. 0	. 0	50.2	<sup></sup> 54 , կ					
TROFIL JOX OND CASH FLE	ЭЦкинкинаны			OUTE.	і і тісі СКЕННЕН МЭННЕ	"59.8 Жананыны	"63.9"	769.3 *********	75.1	81.9
OPFRATTNG MARGIN	0	. 0	, 0	26.3	28.6	31.1	33.0	36.7	чининиких чо.о	икккин: 41,5,0
MSARIC EXPLIBES	. 0	Ü	. 0	11.6	5.0	5.4	75.9		40.0 6.9	77.5 S
DEPRECIATION	. 0	. 0	Ö	19.2	~20.2	26.9	726.9	726.9	. 0	
TOC PHOBITS VITOR	. 0	. 0	. 0	1.5	1.5	1.5	71.5	1,5	. 0	0 (
21 PASS PA PA - RA PASS A LA PASS A	de ea de en se	ter on the art or	record on party .	all was not use and	to the second					
GROSS PROFIT	. 0	, 0	. 0	1.0	1.67	12.7	. 5	1.9	33.4	35.9
MIEDERE LAURENT			. 0	5,9	5.7	15.5	15.4	75.2	*4.9	4.6
AUJUSTEIC PROFIT					* ****	ar early on the		100 1 to 1 to 1 to 1 to 1	· de la	
LOGS CARRYFORMS	.0	.0 :	. 0	4.9	11.0	0.3	5.9	3.1	20 , t	31.3
1975 SA PERIOR OF THE PROPERTY		. 0	.0	. 0		. 0	. 0	, 0	199 (	156, 9
JAXADLE PROFIT		. 0	. 0	" I <sub>E</sub> , 9. :		nin nr				
CORPORATE TAX @ 95 %	Û	. 0	, O	0	11.0 0	"0.3 .0	5.9	737 P.	. ()	250.3
DIVESTMENT TAX CRED.	. 0	, 6	ů.	. 0	0	, 0	, <b>0</b> , , <b>0</b> .	. N . O	, (t ()	14.4
				***		• • • • • • • • • • • • • • • • • • •		. U	·	11.4
HET PROFIT	. 9	. 0	, 0. 3.	~4.9·	11.0	"8.3 .	7157 19	3.3	·	256.3
, LDAR REPAYMENT, ·	0	. 0	.0	1.1	1.3	1.4	1.5	1.8	12,4	2.3
REPRECIATION	. 0	. 0	.0	19.2	28.2	26.9	26.9	28.0	, û	. (i
CHE AMORTIZATION	. 0	. 0	. (1 - 1)	1.5	1.5	1.5	1.5	1.57	O.	
EOSS CAPRIFOROD	. , (1	. 0	, 0	. 0	. 0	, it	, 0	. 0	29 . ĭ	5,9
— <u>Баниянияния и и и и и и и и и и и и и и и </u>		*****	**********	******	<b>инчиний</b>	иниямяныя	<b>чүкнүки</b>	*****	<b>чинининия</b>	инжини
										•
70/ % FOOTEY		1.00								
	i li Marie le le Mire a la Mire. La Marie le le Mire a la Mire.	19.9 XXXXXXXXX		9.7	15.5	17.4	19,5	21.0	"मा, प	27,3
DCT THRE		оенергаца.	- пивиня и и в	IK SEKHANA	екинини	M W W W W W W W W W W W	ниоринымы	***************************************	4 M W 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	ининини
EAST LIMIT	15.0%						•			

# TADLE VEA 11 ENZYME HYDROLYSTS FINANDA FROM CELLULUSE - SERE ELONOMIC DATA

	1.999	1995	1996	1997	1998	[999	2000	2001
CAPITAL EXPENDITURES		The second section of the section of the second section of the section of the second section of the se					* * * * * * * * * * * * * * * * * * *	2 ** If \$
- PVIII	. 0	. 0	. 6	0	. 0		. 0	. 0
Flaeb Thvesthehr	0	0	Ð	. 0	. 0	0	Ö	. 0
START OF COST	, 0	. 0	0	. 0	0	io	o o	Ö
WORKING CAPTION	2.0	2.1	2.3	72.5	2.7	2.9	3.7	41 . 0.
INTEREST DORTHO COMS	0	. 0	. 0	. 0	ø	. 0	, p	Ú.
North Anna Canada and Anna Can Anna Canada and Anna Canada an			The state of the s	er er er ça 🖦 📜	The second of	And the second		
TOTAL CAPITAL EXPEND	"2.0	2.1	2.3	72.5	2.7	2.9	3.20	41.9
	. 0	. 0	. 0	. 0	. 0	. 0	0	n
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0000	39.7	43.1	46.7	50.7	55.0	*****	******	*****
CHERTCALS	11.7	12.7	13.8	14.9		59.7	64 - 7	70.3
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Lating	3.7	4.0	" <b>4</b> , <b>4</b>	4.7	5.2	17.0 5.6	18.5	70.1
MATRIEPARKE	0.3	9.3	10.0	10.8	11.7	12.6	6.1 13.6	6.6
OURLET OVERHEAD	1.9	1.9	2.1	2.3	2.5	2.7	13.a 2,4	14.7
GENERAL PLANT OVIII.	7.9	8.5	9.2	9 9	10.7	11.6	12.5	3.1 13.5
INSURANCE	3.6	3.9	4.2	4.5	4 9	5,3	5.7	6.2
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OPERATING HARGIN	97.1	51.3	55.8	60.7	65.9	71 . 7	17.9	89.7
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		4.0	1.6	3.2	2.7	2.1	1.5	0
ADJUSTED PROFIT	44.2	30.4	42.6	47.1	52.0			
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					, R	.0	.0	0
TAYABLE PROFILE	39.7	381.4	92.3	97.1	52.0	57.3		
CORPORATE TOX & 45 %	15.6	17.3	19.2	21.2	23.4	25.0	63. f. 20. q	60.5
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		16.10	47,000	17.6	11.6	g : 23 <b>,</b> (t. j. )	in Pay ford	73.1
DCF LRR	15.0%		*********	********		* N el S s N = a M p )		HEND BEEN

TABLE VI-A-12

REQUIRED PRICE OF ETHANOL TO ACHIEVE A 15 PERCENT DCF

RETURN ON EQUITY

<u>Year</u>	•	Price,	\$/Gal
1984		1.78	
1987		2.27	
1988		2.46	e francisco
1989	e .	2.67	
1990		2.90	
1991		3.14	
1992		 3.41	
1993		3.70	
1994		4.01	
1995		4.36	
1996		4.73	
1997		5.13	
1998		5.56	
1999		6.04	
2000		6.55	
2001		7.10	

Two types of optimization studies were performed:

- Alternatives involving different equipment or processing steps using known and tested engineering techniques.
- Process improvements involving unproven but feasible advances in enzyme hydrolysis technology.

While the process design chosen as the basis for this study is only one of many that could have been chosen, it represents the current state of research in enzyme hydrolysis coupled with reasonable engineering assumptions. For the most part, optimization of processing steps involving standard commercially proven unit operations was carried out during the base case design effort and incorporated into the final process design. Some of these efforts are briefly discussed below.

Numerous alternatives to the commercially proven ethanol purification scheme used in this design have been proposed. Two such schemes were evaluate in detail:

- Multi-effect distillation (by Hoechst)
- Distillation with vapor recompression followed by potassium acetate extraction (by Dartmouth)

Both of these alternatives involved substantially less steam but more capital than the base case design. They were ruled out primarily to avoid substantial innovation in an area where conventional technology is adequate. It is reasonable to assume, however, that both alternatives would work and may offer some overall production cost savings if properly integrated into the process.

Multi-effect evaporation as employed in both sugar concentration and waste solids concentration involves a heavy capital investment and substantial energy usage. Mechanical recompression involving steam-driven compressors was investigated as an alternative for both six-effect evaporation systems. Recompression generally was found to be slightly more capital intensive, but required less steam. When considering the overall plant steam and power balances including power cogeneration from wood via letdown of high pressure steam, the conventional multi-effect evaporators were found to be the more economic alternative in both cases.

The option of eliminating sugar concentration entirely was investigated with the process simulator. This results in a reduction of the glucose concentration entering fermentation from 15 to 6.85 weight percent. While capital and utilities in this section of the plant decrease, there are substantial capital and utility cost increases in the fermentation, purification and heat generation sections associated with the greater water content present. This more than offsets the larger furfural by-product credit associated with a more dilute xylose stream entering furfural production. The net result is an increase of 17 cents per gallon in ethanol sales price as compared to the base case.

Another alternative investigated with the computer simulator was elimination of the waste treatment system. Since the condensate streams contain only 1,000 ppm of organic matter, it may be possible in certain circumstances to simply discharge this waste. Make-up process water would, however, be required. This alternative reduces the ethanol sales price by 2 cents per gallon, due to a slight decrease in capital which is somewhat offset by the addition of make-up process water requirements.

The waste streams currently routed to heat generation could be sent to an anaerobic digester where the organic content is reduced producing methane and carbon dioxide. Without recovery of the carbon dioxide from the anaerobic digester, this option is 6 cents per gallon more expensive than the base case. This is due to added capital for waste water treatment as well as additional raw materials to overcome the lower carbon utilization efficiency for steam generation. However, if the carbon dioxide is recovered along with the fermenter off-gas, the resultant ethanol sales price is 4 cents per gallon lower than the base case. This alternative includes a substantial increase in the carbon dioxide recovery system and its power requirements which partially offset the added by-product credit.

In summary, it seems that optimization of conventional technology areas offers little economic incentive as compared to the base case design. However, it is very important to determine which process parameters in the more innovative sections of the plant could be most effective in improving process economics. In this way the areas for future research and development can be established, which can ultimately result in an optimized enzyme hydrolysis process design.

Several different process improvements in pretreatment, enzyme production and enzyme hydrolysis were investigated using the simulation model. For each case a change was made in the design of one of the process subsections which is either based on a real research option or a hypothetical research goal. These cases are summarized below:

- 1. An enzyme recovery section was added in enzyme hydrolysis. It was assumed that by contacting the enzyme-sugar stream from enzyme hydrolysis with the solids feed to enzyme hydrolysis, that 60 percent of the original enzyme activity can be recovered. This is a conceptualized design, however, research at Berkeley<sup>(14)</sup> has shown that enzyme recovery may be feasible.
- 2. A steam digestor pretreatment replaced the prehydrolysis pretreatment of the base case. Recent research at  $Dartmouth^{(12)}$  has indicated that a steam pretreatment at 350 psig and five-minute residence time may be as effective a pretreatment as mild acid prehydrolysis.
- 3. Enzyme productivity was increased from 52 IU/ml/hr (FPA 15 IU/ml) to 106 IU/ml/hr (FPA 30.4) which represents the best results obtained for solka floc in a fed-batch system. Also enzyme production per gram of carbon source consumed was increased from 0.17 to 0.31 grams enzyme per gram of carbon source. This is also the best result obtained for solka floc in a fed-batch system.
- 4. The solids concentration in enzyme hydrolysis was increased from 10 to 15 weight percent with no corresponding reduction in yield as predicted by experimentation, or increase in enzyme loading.

The first case, which incorporated enzyme recovery, reduced ethanol sales price by about 10 cents per gallon. This was due primarily to the reduction in raw materials cost by the 60 percent reduction in wood being diverted to enzyme production.

There was also a slight reduction in capital caused by the decrease in enzyme production tank size which was offset somewhat by the enzyme recovery equipment.

The second case, use of steam explosion in place of prehydrolysis for pretreatment, resulted in a four-cent per gallon ethanol cost reduction. This was primarily caused by the raw material cost reduction associated

with the elimination of the acid and neutralizing chemicals added at the front end. There was a slight increase in capital due to the extra cost of disk refining at higher pressure and the increased solids (less xylan is solubilized during steam explosion compared to prehydrolysis) in the streams going to separation and enzyme hydrolysis.

The third case, increased enzyme productivity, resulted in a 17 cent per gallon ethanol sales price decrease. This is due to less feed diverted to enzyme production, resulting in decreases in wood, acid, calcium hydroxide and nutrient raw material requirements. Also capital in the front end is decreased due to the reduced wood feed, and enzyme production capital is reduced significantly resulting from the increased productivity.

The fourth case, increasing enzyme hydrolysis solids concentration to 15 weight percent, reduced ethanol cost 27 cents per gallon, the largest single decrease. Increasing the solids concentration in enzyme hydrolysis significantly reduces capital in enzyme hydrolysis and fermentation, (reduces size of sugar concentrator), as well as reducing the steam boiler cost in the offsites. The reduction in steam load also decreases the wood feed requirement and utilities cost.

A summary of the results for the alternative process schemes as compared to the base case at 10 percent DCF return are given in Table VI-B-1. If the results are assumed nearly cumulative and these are combined, the reasonable potential for reducing ethanol production costs via further process development can be analyzed. By combining these four cases a net decrease of approximately 50-55 cents per gallon could be realized. Since the base case ethanol sales price was 206 cents per gallon with a 10 percent DCF return, the optimized case might lie in the 150-160 cent per gallon range in mid-1984 dollars.

It must be emphasized that these optimization results represent attainable targets for future research and development. At the current state of enzyme technology development, even the 206 cent per gallon ethanol price would require experimental verification of design parameters.

# Table VI-B-1

# Summary of Process Alternatives

1.1	Alternative			Deviati Base Case,	on from Cents/Gal
1.	Enzyme recover				4:: <b>-10</b> - [[]]
2.	Steam explosio	)n			-4
3.	Increased enzy	me productivity			-17
4.	15% enzyme hyd	irolysis solids	concentrat	ion	-27

#### VII. DISCUSSION

## A. Sources and Limitations of Data

The bases for the design of the enzyme hydrolysis process described in this study are derived from many different sources. These include research efforts in biomass conversion technology, equipment manufacturers serving related areas such as the paper and pulp industry and Chem Systems experience in the biotechnology field. The heart of the process represents the fruits of the efforts of several researchers and design parameters are based on experimental results. Manufacturers of equipment for the paper and pulp and other industries have provided equipment specifications and costs for some of the larger package items such as the disk refiner, multi-effect evaporators and the wood boiler. Where no data exist or there is no analogous commercially available equipment, Chem Systems has extrapolated on existing data or made assumptions based on experience in biomass conversion processes. A brief discussion by plant section of the sources and limitations of the design and equipment bases ensues.

# Section 100-Pretreatment/Prehydrolysis

The wood handling system equipment requirements and cost are based on information supplied by Charles T. Maine of Boston, Massachusetts, an engineering contractor who has designed and constructed a wood-based power plant for Burlington Electric in Vermont. A major concern with the wood handling system is whether a debarker is required. The main problem with bark is that it may carry dirt through the disk refiner, potentially damaging the disk grinders or hampering enzyme production and/or hydrolysis. Therefore the chips are thoroughly washed before entry into the disk refiner eliminating the need for the debarker.

The disk refiner is a commercially available item used in the paper and pulp industry for thermomechanical pulping. The equipment specifications and cost for a disk refiner producing particle fibers 0.5-1 millimeter in

diameter were provided by C.E. Bauer of Springfield, Illinois, and Sprout Waldron of Munsey, Pennsylvania, both manufacturers of this type of equipment.

The prehydrolysis plug flow reactor (PFR) presented a unique problem since there is no commercially available equipment that is comparable. The design parameters and material balance around the reactor are based on Dartmouth experimental data as described in Section IV-B-1. However, there are some questions concerning minor components in the prehydrolysis product. The amount of acetic acid, other organic acids and phenolics formed during prehydrolysis has not been quantified. The presence of other organic acids could significantly affect the processing scheme in so much as these acids may have to be removed due to their potential toxic effects on the microbial systems downstream. In the very least they may have to be reneutralized. Chem Systems has made an assumption concerning the formation of actic acid based on the acetyl groups present as discussed in Section IV-B-1, however other byproducts have not been quantified.

The equipment conceptualized for the PFR has been based on the bench scale Dartmouth PFR and discussions with the disk refiner manufacturer, C.E. Bauer. The disk refiner is envisioned to serve as the feeder to the prehydrolysis reactor when operated at the PFR pressure of 230-240 psia. The PFR itself is similar to the Dartmouth pipe reactor. However, it has not been demonstrated that instantaneous heatup, uniform acid mixing and residence time control can be achieved for a scaled up reactor.

# Section 200-Sugar Separation and Neutralization

The major concern in this section, and one of the principal concerns of the feasibility of the design in general is the solid-liquid separation characteristics of the materials being processed. Laboratory studies at Dartmouth have indicated that the filter rate of the prehydrolyzed ligno-cellulosic material is very low due to clogging of the filter. However, the material did settle out giving an indication that centrifuging would be possible. Experiments by Dartmouth using Bird

centrifuges yielded filter cakes of 30-40 weight percent solids and demonstrated the feasibility of solid-liquid separations. However, these experiments have not been well documented, actual performance data are lacking, and certainly this separation has not been demonstrated on commercial sized equipment. Therefore a number of assumptions have been made concerning the performance of the solid-liquid separation equipment which are summarized in Section IV-B-2. Rotary drum filters are only used for cool (less than 70°C), low solids loading separations, such as polishing.

#### Section 300-Enzyme Production

Perhaps the largest question exists concerning the design parameters of the enzyme production system, since this section contributes significantly to the cost of production (approximately 30-40 cents per gallon). A great amount of research effort has been expended in this area, especially at However, design criteria for an optimized system are still Berkeley. Berkeley has individually demonstrated the feasibility of high enzyme concentrations and productivities for fed batch systems, and the use of low cost substrates (steam exploded wood) and nutrients (corn steep liquor) for enzyme growth. However, a single design incorporating all these advances into an optimized system has not been attempted. Therefore a number of assumptions have been made based on extrapolating what data are available. These assumptions have been summarized in Section IV-B-3. The major assumption made is that the carbon source used in this design, prehydrolyzed wood, will produce RUT C-30 enzymes at a productivity, concentration and carbon source consumption similar to steam exploded wood. In an attempt to verify this assumption, Chem Systems had Dartmouth conduct experiments on growing RUT C-30 on prehydrolyzed wood. The results of these experiments indicate that RUT 6-30 will grow on the prehydrolyzed wood if it has been thoroughly washed, which apparently removes inhibitory effects of some prehydrolysis products. similar to results obtained by Berkeley. (8) However, the enzyme concentration and productivity are approximately half of what Berkeley obtained for steam exploded wood. (13) More research is needed in this area to verify these results.

Another area of concern in the enzyme production section (as well as the enzyme hydrolysis section) is the feasibility of agitation, pumpability and reaction of the 9-10 weight percent slurries in the enzyme production tanks. Goulds, a manufacturer of pumps, was contacted. The Goulds people felt that although testing of the material would have to be performed, based on their experience in pumping wood slurries, a recessed impeller type centrifugal pump would provide adequate agitation and pumpability of the material.

# Section 400-Enzyme Hydrolysis

The yield in enzyme hydrolysis is based on experimental results obtained by Dartmouth (see Section IV-B-4) and is correlated to a number of design parameters including prehydrolysis conditions, enzyme loading, residence time and solids concentration. The yield assumed (i.e., 90 percent of theoretical) is considered to be somewhat conservative, since Dartmouth has obtained quantitative yields at the conditions used.

The cellobiase added to the RUT C-30 cellulase is a very small amount and is assumed to be purchased. Current price quotes from NOVO indicate that very small amounts of cellulase used for research purposes cost 5-7 dollars per pound. However, it is unlikely that bulk quantities of enzyme would be priced that high. In addition NOVO probably uses an expensive substrate like Solka Floc to grow their enzyme. In breaking out the cost of producing enzyme from wood in the current design a cost including return of 60-70 cents per pound of cellulase is obtained. Therefore a conservative value of 2 dollars per pound is used as a purchase price for cellobiase. Even if this value is too low, recent work at Dartmouth has indicated that the cellobiase addition level assumed in this study may be too high.

The equipment specifications and cost associated with the dewatering presses were obtained from Black Clawson of Middletown, Ohio. Based on discussions with Black Clawson, it was decided that it is reasonable to assume the dewatering presses could concentrate the ligno-cellulosic materials from 35 to 55 weight percent solids.

# Section 500-Sugar Concentration

The six-effect forced circulation evaporator used for sugar concentration is a conventional item used in the chemical industry. The minimization of calcium sulfate plating is achieved by using an evaporator system with slurry recycle, which encourages calcium sulfate crystal growth as opposed to surface plating in the evaporator. The design, cost and equipment specification for the evaporator system were provided by the Unitech Division of Ecodyne, a manufacturer of various types of evaporator systems.

## Section 600-Fermentation

Continuous sugar fermentation to ethanol is a well documented and commercially available technology. Ethanol yield and by-product formation are based on commercial fermentation ethanol data. By-products formed are limited for the sake of simplicity to glycerol, acetaldehyde, fusel oils, carbon dioxide and yeast. Fermentation residence time requirement is correlated with cell density for a continuous system with cell recycle and is based on industrial practice.

A major problem in fermentation, especially with yeast recycle is contamination. In order to minimize the possibility of contamination, the yeast being recycled is acid washed at low pH for several hours. In addition, in order to avoid mutations, complete yeast population change out is achieved within 1-2 weeks by purging the yeast grown during fermentation. Also a select yeast strain is chosen.

The design specifications of the yeast centrifuges were obtained from information from Alfa-Laval, of Ft. Lee, New Jersey. Alfa-Laval manufacters a wide range of equipment for the fermentation industry.

# Section 700-Carbon Dioxide Recovery

The carbon dioxide recovery unit is a standard sized (300 ton per day) package unit. The design specifications and cost were provided by a major industrial gas supplier.

#### Section 800-Ethanol Purification

The design of the ethanol purification system is based on the cascade distillation system developed by Katzen, (14) and used in commercial fermentation ethanol facilities. One design alternative which warrants careful evaluation is the distillation/heat pump system followed by potassium acetate extraction of ethanol under development at Dartmouth. This scheme is potentially highly energy efficient at low ethanol concentrations. When coupled with the elimination of multi-effect evaporation prior to fermentation and the replacement of concentration in the heat generation section with anaerobic digestion, the net result may be more economic than the standard systems proposed in this study. Due to second-order reaction kinetics, the furfural yield would be greater from the more dilute solution encountered in this scheme. These interactions require investigation and, if sufficient economic potential exists, this ethanol purification system should be experimentally verified.

#### Section 900-Furfural Production

The material balance and design of the furfural production reactor are based on experimental data provided by Dartmouth, which is summarized in Section IV-B-9. The reactor itself is similar to the prehydrolysis PFR except that it is a liquid phase reactor, which eliminates many of the problems associated with the slurry reactor. The reactor is envisioned to be a column with baffle plates to simulate plug flow (i.e., minimize backmixing). As with the prehydrolysis PFR, materials of construction present a problem in the high temperature acidic environment of the system. Zirconium has proved satisfactory in the Dartmouth PFR. Also, as with the prehydrolysis PFR, an exact material balance quantifying all by products is not available. The design described herein does not quantify any light components which may be formed (i.e., methanol). However, a column is included in the design which removes lights (ethanol) present in the feed.

## Section 1000-Heat Generation

The heat generation section is a relatively straightforward design using commercially available equipment. The stillage evaporator is a six-effect forced circulation type. Design specifications and cost for the stillage evaporator were provided by the Unitech Division of Ecodyne. The wood boiler/cogeneration system which handles a combination wood/organic liquid feed is a standard package unit which is used in commercial wood-based electric generating stations. Design specifications and cost for this system were provided by CE Power Systems of New York City.

## Section 1100-Waste Treatment

The waste treatment system handles all the waste water from the process and decreases the organic content from 1000 to 20 ppm by aerobic digestion. The design specifications and cost were provided by EIMCO of Salt Lake City, Utah. The equipment specified is of the type commonly used to reduce the organic content of the water effluent from fermentation ethanol plants.

## <u>Ge</u>neral

A general comment concerning the equipment specified is that varying degrees of technical applicability exist. That is, some of the equipment specified is used commercially for very closely related applications. This includes the concentrators, waste ponds, wood boiler cogeneration, carbon dioxide recovery and the distillation and fermentation equipment. However, other equipment specified would have to be extensively tested with respect to their specific application in this design. This would include the centrifuges, disk refiner/prehydrolysis reactor, slurry pumps, dewatering presses and filters.

## B. Critical Technology Issues

The major critical technology issue concerning the process described in this study can be succinctly summarized as the demonstration of several individual unit operations and the technical feasibility of several process steps in a scaled-up integrated facility. Although many of the process steps have been individually demonstrated in the laboratory, it has not been shown that similar results can be obtained for a feedstock undergoing several process steps in succession. This of course would require a pilot plant-sized demonstration facility including, at a minimum, pretreatment, separation, enzyme production, enzyme hydrolysis and fermentation processing steps. The fermentation of wood hydrolyzate via pretreatment and enzymatic hydrolysis has to be studied in a continuous system to determine the acceptable level of inhibitors or toxins from pretreatment as well as to demonstrate that hydrolysis enzymes and fermentation yeasts can adapt to certain toxin levels.

The individual unit operations which are questionable for this specific application have been discussed with regard to the basis and justification of their selection in the previous section. These include the centrifuges/filters/dewatering processes for solid-liquid separations, slurry pumps for high-solids transfer and agitation, and the disk refiner/prehydrolysis reactor for pretreatment. The feasibility of these items in performing the desired unit operations has yet to be demonstrated on commercial sized equipment. Only the prehydrolysis reactor (using a slurry pump instead of the disk refiner as the feeder) and some centrifuge experiments have been demonstrated on a bench scale. Testing of the equipment to verify suitability for these applications is required in collaboration with the various manufacturers. This involves supplying the vendors with suitable quantities of material for testing, which unfortunately is not readily available.

It has not yet been demonstrated that disk refined prehydrolyzed wood can actually be used for enzyme production, hydrolyzed to sugar using such enzymes and then fermented to ethanol in integrated operation. In addition, because of the incompleteness of the material balance and the

speculation on the effect of by-product and salt formation it is unknown what processing steps may be required upstream and downstream in order to purify the stream of all toxic components prior to entry into the microbial systems. Enzyme production probably requires washing of the unneutralized solids to be used as the carbon source. Recent experiments at  $Dartmouth^{(13)}$  have already indicated that some toxic materials are apparently present in prehydrolyzed wood inhibiting cell growth and enzyme Although it has been shown by these experiments that the presence of poplar bank apparently does not inhibit enzyme hydrolysis of prehydrolyzed wood, its effect on enzyme production is unknown. Also the effect of bark from other wood species and the by-products formed during prehydrolysis are unknown. Solutions to these problems involve the possibility of using some kind of steaming as a pretreatment, thus minimizing the by-product formation associated with mild acid hydrolysis. Overneutralization of the prehydrolyzate to precipitate some of the phenolic by-products formed during prehydrolysis may also facilitate their removal. These options are discussed in more detail in Section VII-D.

# C. Commercial Potential

The commercial potential of the enzyme hydrolysis process described in this study, or, for that matter, any cellulose hydrolysis process, will be dependent on the demonstration of the technical feasibility of several of the process steps discussed in Section VII-B. If technical feasibility is established, then the economic competitiveness of the process, vis a vis corn and acid hydrolysis based ethanol, will determine its commercial potential.

The base case design for the enzyme hydrolysis process produced ethanol for 206 cents per gallon at 10 percent DCF. This facility produces 25 million gallons per year of ethanol in Michigan (in mid-1984 assuming instantaneous plant construction and start-up. A 50 million gallon per year whole kernal milling corn ethanol plant at a U.S. Midwest location produces ethanol for 219 cents per gallon at 10 percent DCF in mid-1984. Costs of production from these two plants are compared in Table VII-C-1.

The wood enzyme hydrolysis case produces ethanol for about 13 cents per gallon less than the corn ethanol case, based on a new plant with instantaneous construction and start-up in 1984. However, these estimates are based on a current corn price of 3.10 dollars per bushel, which is expected to be one of the highest corn prices for the years 1984-2001. Therefore, ethanol cost of production from the corn plant can be expected to be somewhat lower over the plant lifetime.

It is presumed that a wood based acid hydrolysis facility would produce ethanol for an even lower cost of production, perhaps 180 cents pergallon, than the enzyme or corn cases on a 1984 basis.

TABLE VII-C-1
COMPARISON OF ETHANOL PROCESS ECONOMICS

Wood	Corn
Enzyme	Whole Kernel
Hydrolysis	Milling
<u>(25 MM Gal/Yr)</u>	(50 MM Gal/Yr)
66.3	80.6
45.7	62.8
112.0	143.4
80.97 <sup>(1)</sup>	130.89(2)
19.89	43.90
22.65	12.89
24.47	14.13
(62.04)	(67.86)
85.95	133.96
71.33	44.83
157.28	178.79
206.0	213.5
	Enzyme Hydrolysis (25 MM Gal/Yr)  66.3 45.7 112.0  80.97(1) 19.89 22.65 24.47 (62.04)  85.95 71.33

<sup>(1)</sup> Mixed hardwoods at \$18/wet ton.

<sup>(2)</sup> Corn kernels at \$3.10/bushel.

The commercial potential for the enzyme hydrolysis process is probably dependent on whether the potential improvements discussed in Section VI-B can be realized. The base enzyme case is certainly competitive with a corn ethanol facility on a current basis, however, significant process improvements are still necessary for the enzyme hydrolysis process to become competitive with a wood-based acid hydrolysis process. Furthermore, the technical feasibility of enzyme hydrolysis must be demonstrated in an integrated facility before it will gain acceptance in the commercial marketplace as a viable alternative to corn-based routes.

#### D. Recommendations for Future R&D

The most important goal of future research and development work for a wood-based enzyme hydrolysis ethanol process is the demonstration of the technical viability of several unit operations as well as the satisfactory performance of an integrated process. This has been discussed in terms of process sensitivities in Section VI-B. This would likely require the construction and operation of a pilot plant sized unit demonstrating at a minimum the pretreatment, separation, enzyme production, enzyme hydrolysis and fermentation process steps.

Bench scale research should concentrate on those areas where process economics can be improved most significantly. These areas include, as discussed in Section VI-B, optimization of pretreatment, enzyme hydrolysis and enzyme production. Additionally, bench scale research is necessary to characterized the, as yet, incomplete material balance with regard to tars and minor organics such as acetic acid, methanol, levulinic acid, formic acid and soluble phenolics from lignin breakdown. Understanding these minor constituents is a prerequisite to finalizing this process design.

The optimized pretreatment serves two important functions. The first and most important is to give high enzyme hydrolysis yields in the most cost effective manner, (low capital, low enzyme loading and low hydrolysis residence time). However, it is also important that the pretreated wood feedstock serve as an inexpensive, effective carbon source for enzyme

production. So far it has not been demonstrated that prehydrolyzed wood can be used as a carbon source for enzyme production and yield enzyme productivities as high as steam exploded wood. However, although steam exploded wood yields satisfactory enzyme productivities, it has not been shown to give enzyme hydrolysis yields as high as prehydrolysis when supplemented by cellobiase. More research is needed to verify these preliminary results and to optimize the pretreatment system.

Enzyme production is another area where significant cost reductions can be achieved by research improvements. The ultimate research goal should be to be able to utilize an inexpensive substrate like steam exploded or prehydrolyzed wood and achieve similar enzyme yields and productivities to those achieved with solka floc. Also optimization of the enzyme production system including nutrient type and requirement, and system type (batch, fed-batch or continuous) is required. Partial recovery of enzymes following hydrolysis coupled with the above improvements would reduce enzyme production costs even further. This is another area where more research needs to be done.

As evidenced by the discussion in Section VI-B, the key to significantly improved process economics may be in the amount of water required in enzyme hydrolysis. An increase in solids concentration of 5 percent (increasing from 10 to 15 percent) in enzyme hydrolysis reduces costs of production by 27 cents per gallon if the same hydrolysis yield is assumed. At present, as solids concentration increases, hydroysis yields correspondingly decrease. It is not known exactly what causes this decrease in yield. However, it is presumed that mixing difficulties and product inhibition (glucose) are the major factors. This problem should be researched for possible improved mixing techniques and physical or chemical removal of the glucose as it is formed. Also, minimum enzyme loading for both RUT C-30 cellulase and cellobiase should be achieved.

#### VIII. REFERENCES

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### X. APPENDIX

## A. Detailed Material Balance

This sub-section contains the detailed material balance for the eleven plant sections based on 25 million gallons per year of fuel-grade ethanol.

STREAM NU.			ler	10	<b>15</b>	1	<b>4</b> 3		184		i <b>0</b> 5		<b>e</b> 6		187		105		H <b>4</b> 9
DESMETTE 420	MOLECUL AR	LBS	MOLES	195	MOLES	LBS	MULES	LBS	NULES	LDS	MILES	LBS	MULES	LUS	POLES	LBS	HO ES	Flei	MOLES
FILE SECI	METGHTS		10	i in	HA	<b>H</b>	HR	HØ	HŘ	HA	HIA	HA	HA	H	H	*#	<b>HR</b>	H#(	40
WATER	16	32905.9	5161.4	45640.7	2790.5	141514.5	7661.9	303675.0	16870.4	0.0	1.1	63331.9	3551.8	503121.5	28284.5	506823.6	28156.9	58720.6	3202, 1
CELLULUSE Insoluble Lighth	165	43851.7 17894.6	279.7	0.0 0.6	0.6	43851.7 17094.6	270.7	1. f 1. f	0.0	1.0 4.4	1.1	0.0 0.0	0.0	43851.7 17834.6	270.7	41601.6 15384.0	256.8	0.6 0.0	0.0
ASH		185.6		6.0		185.4		1.1		0.0		0.0		185.6		185.6	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	0.0	
AVCELL TUP	1.1 1.4 1.4	0.4		0.0		1.4		1.1		0.6		0.0		\$.0		0.0			
OTHER INSOLUBLES		9848.2	·	0.0		9848.2		1.1		9.0		0.0		9848.2		9854. 8		0.0	
YEAST		0.0		1.1		1.1		1.4		4.6		0.0		4.0		4.1		1.1	
CALCIUM SIA FATE	136	1.1	0.1	6.0	1.6	1.1	1.1	4.0	8.6	0.0	0.0	8.0	8.6	0.0	6.0	0.6	0.0	0.0	0.0
HEME. KYLAN	132	15050.7	114.0	1.0	1.1	15050.7	114.6	4.6	4.0	8.0	1.0	8.9	1.1	15050.7	114.0	4391.6	11.3	i.J	. 1.1
HENT. GLUCAN	165	4273.4	26.4	1.1	1.1	4273.4	24.4	0.6	1.6	0.0	0.0	8.0	4.4	4271.4	26.4	0.0	9.0	0.0	0.0
CALCIUM CARBONATE	100	0.0	1.0	0.0	8.0	1.0	1.0	1.6	1.1	8.6	1.0	6.0	1.1	4.4	1.1	6.6	1.0	0.0	1.0
ONYGEN	32	6.0	8.6	0.0	1.1	1.1	0.0	0.0	0.0	0.0	8.6	0.0	1.1	6.6	0.6	0.0	9.6	1.0	1. \$
NA TROGEN	28	0.0	1.0	6.0	0.0	1.1	1.0	1.0	0.0	1.0	6.6	6.0	0.0	1.1	0.0	0.0	6.6	· · • • • • • • • • • • • • • • • • • •	· 4. £
CARBON DIGITOE	- 44	8.6	1.1	0.0	1.1	0.67	0.6	0.0	0.0	4.6	1.1	0.0	0.6	1.1	4.4	1.1	0.0	1.6	0.1
COM STEEP LIGURE		1.0		1.0		1.0		1.0		1.1		0.0		8.6		8.8		8.0	
MUTATERIS	. ( ) ( ) ()	0.6		0.0		6.0		0.6		1.0		6.0		0.0		1.0		. 6.6	
CELLUM ASE		4.1		1.1		0.0		6.6		1.1		6.6		4.0		1.6		1.1	
CELL OBLASE		1.1		0.0		8.0		6.6		1.1		0.0		6.0		0.0	·	1.1	
EXTRACT (VES		2681.6		6.6		2601.6		1.6		1.1		0.0		2661.4		2641.5	***	0.0	
SOLUBLE LISHIN	••	1.1		0.6	: . <del></del> -	6.0		1.1		1.1		8.8		6.0		1709.8		8.0	
FUSEL OIL		0.0		8.8		6.0		1.4		1.6		0.6		1.1		8.6		6.0	
FURFURAL	96	0.0	0.0	8.6	0.0	1.0	1.1	1.1	1.1	1.1	8.8	8.6	1.1	1.1	1.1	484.6	5.0	210.9	2.2
H₩F	158	0.0	1.1	4.4	4.0	0.0	6.0	1.1	4.6	6.1	1.6	0.0	0.0	0.4	6.0	55.7	8.4	6.0	0.0
JIYLOSE	150	0.0	0.0	0.0	1.1	1.1	6.0	1.0	1.1	1.0	1.1	6.0	0.0	8.0	1.0	11333.2	15.6	1.1	1.1
ALICOSE	180	9.0	1.1	1.0	1.0	1.0	6.0	1.1	1.1	1.1	1.6	1.1		4.0	1.1	7168.2	39. A	0.0	0.4
SULFURIL ACID	96	0.6	1.1	1.1	1.1	1.1	0.0	0.0	0.0	3817.4	39. 3	1.1	0.0	3847.4	39.1	3847.4	39. 3	1.1	0.6
SODILY HYDROLIDS	W.	6.6	8.0	0.0	1.0	6.8	4.6	4.0	6.6	1.1	1.1	1.1	1.0	8.0	1.1	6.0	1.4	0.0	0.6
CALCIUM HYDROXIDE	74	1.0	0.0	9. 5	0.0	0.0	0.0	0.0	6.6	1.1	1.0	1.1	1.1	1.0	1.1	0.0	0.0	1.1	1.0
BENZENE	74	1.6		1.0	1.0	0.0	1.1	1.1	1.1	6.0	1.0	0.6	1.1	1.1	1.0	1.0	0.0	1.1	0.0
<b>GLYCEAGL</b>	92	<b>8.8</b>	1.1	8.8	0.0	6.0	1.1	1.0		1.1		0.0	LI	1.0	1.8	8.6	4.6	9.4	0.0
£THA'-Q:	46	1.1	1.1	9.6	0.6	0.6	0.0	1.1	1.0	1.0	6.6	0.0	8.6	1.0	0.0	1.1	0.0	1.6	1.1
SOUTUM CARBONATE	105	0.0	1.1	1.1	8.6	6.0	1.1	1.1	1.1	0.0	1.0	0.0	1.1	1.0	0.1	1.1	1.1	1.1	1.1
SCOTUM SULFATE	115	ar e teta	1.0	0.0	i. 6.6	1.1	0.1	1.1	3.0	4.0	1.0	8.6	0.0	1.1	9.0	1.1	0.0	1.1	1.1
SCOTUM ACETATE	82	1.1	1.5	0.0	1.1	1.1	1.1	4.1	0.0	8.0	1.6	6.0	1.1	1.0	L. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.	1.1	0.0	1.0	1.1
CALCIUM ACETATE	158	1.0	1.0	1.1	1.1	0.0	0.0	1.1	0.0	1.0	1.1	6.6	8.0	1.1	€.€	0.4	0.0	1.1	1.1
ACETIC ACID	€0	9.6	6.4	1.1	1.1	1.1	8.0	1.1	0.0	6.0	0.0	8.6	8.4	1.0	8.0	\$161.7	19.4	67.4	1.1
<b>EHM</b>	15	50 m (0.00)	1.1	1.6	1.4	0.0	1.1	4.0	0.6	1.1	• 0.€	0.0	0.0	0.0	0.6	67.4	1.5	55.7	1.7
ACETAL DENVOE	• • • • • • • • • • • • • • • • • • •	0.5	6.6	0.0	1.1	0.0	0.0	0.0	6.8	6.6	1.0	1.1	6.0	1.0	0.0	0,0	0.0	1.1	0.0
TOTAL		185811.7		48698.7	Mais.	234120.4		# .C1 31 6t.		3847.4		63931.9		645874.4		685873.8		59054. /	
IEMFAATURE, C		18.0		284.0		264.6		27.0		21.0		284.8		200.0		200.0			
PRESSURE, AIM		1.0		17.0		16.6		1.0		1.0								118.6	
				***		10.0		1.		1.0	3 37 4 4	17.0		15.3		15. 3	er artis	4.6	4 4 1 1

	DATE 19-Aug-04				11		ETH	MUL VIA EN	ZYME HYDROLI	YSIS SECT	10N 1 <b>NO</b> : PR	ETRENTMENT/F	REHYDROLYS!	IS				
٠.									5.54									
	STREAM NO.			tiē .		ni j		115	1	113		116	1	115	· : 1	116	 :	117
	DISKETTE 420 FILE SECT	MOLECULAR Welghiis	LBS HA	MOLES IIA	LBS HA	MOLES IM	LBS IBR	HOLES	LBS: Hit	HOLES HR	LBS HR	MILES HÆ	LBS IM	HOLES Hir	LBS HR	MOLES HA	LBS HAR	MULES HR
			~	*****	:													
	HOTE	10	1/4/47 #	niani e	20511.0	I CAE A	419556. 0	23308.7	15840.1	244 6	(837)5.9	22128.7	44307.1	2465.9	9886. 3	544. 6	48714.3	2717.5
	MATER	10	446193.0	24894.6	28547.0	1585.9				B84. 0	41601.6	256.8	11307.1	£103.3	3000. 1	9,0	0.0	0.0
	CELLULOSE	162	41601.6	256.6	0.6	0.0	41691.6 15384.8	256.8	0. <b>6</b>	1.1	15381.6	£70.8	0.0		1.1		1.1	
	INSOLUBLE LIGHTN		15381.6		1.0										0. U		0.5	
	RSH		185.6		0.0		185.6		1.1		185.6		1.1 1.5		0.0		0.0	
	AYCELLIUM Dinka insulubles		0.0		0. <del>0</del> 6. 0		9054.8		0. G 0. G		0. 0 9854. £		1. U	****	0.0		0.0	
			9654.6											*	0.0		0.0	
	YEAST		0.0		0.0		1.6		8.8		9. 0	0.0	4. # 0. 0	1.8	0. Q	9.8	0. 0	0.0
	CALCIUM SULFATE PENI, KYLAN	136 132	0. 0 4393. 6	9. 0 33. 3	0.0 0.0	0.1 1.1	0. <b>8</b> 4391. 6	0. 6 31. J	9. 6 9. 6	9. 6 9. 6	9. <b>6</b> 4393. 6	33. 3	8. 8	8. E	0.0	1. 6	0. t	8. 8
		162	9,373, 0	4.0	0.0	1.1		0.0	1.1	9. 9	1,333,0	34. 3 8. 6	1. 6	9.0	1. 9	0.0	1.1	0.0
	HENT, GLUCAN	100	0.0	1.0	6.0		6.4 8.6	1.1	1.1	₩. #	1.1	0. T	9. 8	6.6	t. 1	B. 0	0.0	8. 5
	CALCIUM CARBONATE OXYGEN	32	9.0	0.0	0.0	0.8 0.8		0.0	1.1	9. F	0. E	1.1	8. 8	6.0	0.0	1. f	1.0	8.6
	NETROGEN	26	1.0	8.0	1.0	1.0	8.6	1.4	0. F	8. 0	1.6	1.1	1.4	1.1	1.1	0.0	0.8	8.0
	CARBON BLOXIDE	- 44	1.1	8. B	0.0	1.1	8. 6	1.4	1.1	6. Ó	6.6	6.6	1.1	3.8	0.0	1.1	1.1	0.0
	CORN STEEP LIGHT	11	1.8	***	0.0	7.1	5.5	7.1	1.6		1.1		0.0		0.0	***	9.0	
	NUTRIENTS	****	9. 8		0. E		5.0		1.1	4	1.6		1.1		0.0		0.0	
	CSLLULASE		6.6		6.6		8.0		1.1		1.1		1.1		0.0		0.6	***
	CELLODIASE		1.1		0.0		8.0		1.1		1. 6		1.1	****	8.6		8.8	
	EXTRACTIVES		2601.8	***	6.0	~-*	2601.5		1.6		2601.8		0.0		9.4		9.6	
	SOLUBLE LIGHTN		1789. 6		0.0		1709. 8		1.1		1709. 8		0. 6		0.0		1.6	
	rusel all		£. 6		1.0		1103.G		1. 6		1.4		1.1		1.4		0.0	
	FURFURAL	96	273. [	2.9	90.2	0.9	183.5	1.9	45.7	0.5	137.8	1.4	135.9	1.4	35. 2	8.4	175.7	1.8
	IPF	125	55.7	5. 7 0. 4	1.1	1.1	55.7	0.5	1.0	0.0	55.7		1.1	8.0	9.6	4.0	6.0	8.0
	XYLOSE	158	11333.5	75.6	0.0	4.4	11333.2	75.6	1.0	B. B	11333.2	75.6	1.1	0.0	0. 6	1.1	0.6	8. 6
	GLUCGSE	184	7168.2	39. <b>a</b>	0.0	6.0	7168.2	39. <b>a</b>		0.0	7168.2	39. 4	0.0	0.0	8.0	1.4	0. 0	8.8
	SULFURIC ACTO	98	3847.4	39. J	0.1	4.0	3847.4	37. a 39. 1	1.1	0.0	3847.4	39. J	6.6	0.0	1.1	8.0	1.1	1.1
	SOUTHWE HYDROXIDE	70 16	3017.1	31.3	0.6	8.6	3017,1	32: J · 0. 0:	1.1	6.6	3877.7	9.6	1.1	9.1	6.0	4.1	1.0	4.4
	CALCIUM BYOROXIDE	74	0.0	1.6	0. S	1.1	6.0	1.0	1.1	0.0	1.6	1.1	1.0	1.1	1.5	1.6	0.6	1.1
	BENZENE BENZENE	18	1.8	1.0	0. T	0. E	4.0	1.0	1.1	6. 8	1.1	6.1	0.0	1.0	1.1	0.1	0.0	8.6
	GLYCEROL	92	9. B	0.1	1.1	1.1	1.1	8.0	0.1	6. 8	0. E	0.0	0.0	0.0	1.1	0.0	0.0	0.6
	ETHANGE	46	1.5	1.0	8.6	6.8	1.1	3. 6	0.6	6.6	0.0	1.1	1.0	1.6	1.0	0.0	0.0	0. F
	SODIUM CARBONATE	106	1.1	0.0	1.6	9.6	4.4	1.4	2.4	B. 6	4.0	6.6	0.0	0.0	8.0	8. 8	0.0	9.6
	SUBTUM SULFATE	142	1.6	8.0	1.4	1.8	4.6	1.6	0.0	1.8	9. 0	1.1	1.1	9.6	1.1	1.1	0.0	0.0
	SODIUM ACETATE	82	6.1	0.0	1.5	9. 0	4.0	0.0	1.1	0.6	1.1	0.6	0.0	1.0	0.0	4.4	0.0	8.0
	CALCIUM ACCIATE	158	1.6	0.0	4.0	6.6	1.0	1.1	0.0	6.4	1.1	0.0	0.0	1.1	8.0	1.6	1.0	9.0
	ACETIC ACID	E3	1094.2	18.2	36.1	0.6	105 <b>0.</b> t	17.6	18.6	6.3	1839.5	17.3	54. 7	6.9	11.3	8.8	56.2	. 6.9
	METHANO.	12	11.7	9.4		9, 4	1639. E	17.0	10. G	1.0	9. 0	4.0	11.7	0. 5	9.3	0.3	46.4	1.1
	ACETALDENYDE	36 44	8.0	0.6	11.7	1.0	0.0	0.0	6.4	6.0	1.4	0.0	. 11.7	0. ¥ 0. 6	9. 0	8.4	0.8	6.6
	PICE INCOLUNIE			7. ₹	U. <b>4</b>		w. U	7.0	V. W		T. T.			•. •	7.0	7.7	V. P	7.4
	TOTAL	* 4.4	546819.2		an⊾85. <b>€</b>		518134.2		15994.4	et e	502229.8		11589.4		9862. 1		49192.5	
	TEPT PIL		J10013.5		CHOOP.		albin.c		17.52.2		Jerre 1. 0		1130317		avuer I		1217610	
	TEMPERATURE, C		115. ĕ		116.6		118.8		189.0		189.6		104.4		148.0		146.0	
	PRESSURE, AIR		4.4		2. 6	3 To 10 To 10	2.0		1.1		1.1		1.1		4.4		1.4	
	THE STATE OF THE PARTY OF THE P	•	1. 1		Ę. <b>V</b>		E. 0	State of the second	Be I	s galeja.		tieficiele e	11.5		. 1.1		1, 1	

STREAM NO.			201	2	<b>9</b> 2		201		184		205	2	<b>16</b>		<b>20</b> 7		206	2	<b>8</b> 9
BIGHETTE 420	HOLECUL AA	LDS	MOLEG	LDS	MOLES	LUS	MOLEG	LPS	MOLEG	LIS	MOLES	LDS	MLES	i DS		186			
FILE SECS	ME LONTS	111	HA	H <b>it</b>	H	HA	HI	HR	H	М	141	на	H	H	HQLES HQ	LDS HAT	HOLES	1 BS Hef	HOLES HAR
MATER	16	183715.9	22128.7	116542.4	6474.6	207173.5	15954. 1	5827. &	222 8	ARISIR *	15010 1			-					
CELLULOSE	165	41601.6	256. 8	39521.6	244.0	2888.0	12.8	1975.5	323.8 12.8	201315.7	15638.3	1.1	1.8	19488.4	. 1002. 2	302030.8	16779.5	12918.9	117.7
INSOLUBLE LIGHTH	No. of the second	15384, 6		14615.9	2 (	768.9		729.0	15.5	104.5 39. i	4.6	9.4	1.1		<b></b>	184.5	8.6	g 104.5 c	0.6
ASH		185.6		175.9		9.8		7.8		37. <u>1</u>	Arrive V	1.1		1.0		19. [		19.1	
NACETT (TIN		4.6		6.6		8.0		8.6	***	8.4		1.\$		8.0		1.1	****	4.5	
OTICA INSOLUBLES		9054.0		8602.5		452.4		429. 9		22.5		8.6		1.1	••••	1.1	43 <del>75</del> 4	1.6	بنب
YEAGT	3 1 1 1 - <del></del>	1.0		8.6		4.0		1.1		1.1		1.0		0.0		22.5	**	22.5	
CALCIUM SIAFATE	136	1.6	1.1	1.1	1.1	1.1	3.8	6.8	8.6	8.4	8.0	1.1	17	1.0		1.1	A Section	0.0	
HEAL. IYLAN	135	1391, 6	33.1	4173.7	31.6	219.8	1.7	209.1	1.6	18.7	0.1	8,0 8,0	1.4	1.1	1.1	3729.2	27.4	3261.2	24.
HENT, GLUCAN	165	1.1	1.1	6.6	0.0	0.6	6.8	8.6	6.4	0.0	1.1	0.0	0.0	1.4	4.6	14.7	1.1	19.7	0. E
CALCIUM CARBONATE	196	1.1	1.0	6.6	1.4	1.5	LI	6.6	1.0	1.1	4.1	4.0	0.0	6.8	0.0	1.1	#. N.	1.1	0.6
DIVGEN	35	1.6	0.8		1.1	1.1	8.6	1.6	0.5	1.0	1.0	0.0	1.0	6.6	1.1	1.1	1.6	4.6	6.6
NI TROGEN	24	0.0	6.0	1.1		4.0	0.6	6.0	6.6	6.6	Li	1.1	1.1 1.0	1.1	1.1	0.0	0.1	A	. 0.0
CARBON DIUXIDE		1.1	1.1	8.0	8.0	1.1	1.6	1.1	6.6	6.6	1.1	1.1	6.0	1 7 70	1.6		4.1	9.8	<b>ĕ.</b> ₽
CORN STEEP LIQUOR		4.0		1.0		1.1		4.0		8.6		8.0		0.0 0.0	6.0	1.1	9,0	9.0	1.0
MUTATENTS		0.0		1.1		0.0		6.6	4	1.1		0.0		6.0		1.1	•	1.1	
CELLULASE		1.1		1.0		0.0		6.6		8.6		1.1		1.1	<u> </u>	0.0		1.0	
CELLOGIASE	77	1.1		8.6		0.6		1.4		1.1		4.4		8.6		1.1		0.0	
ENTRACTIVES		2601.8		751.3		1858.4		37.1		1413.3		1.0		\$. <b>6</b>		0.0	•••	8.0	
SOLUBLE LIGNIN		1789.6		493.4		1216.4		24.4		1191.9		0.0		0.0		1813.3		17.2	*
FUSEL DIL		1.1	***	4.6		1.1		1.8		1.1				1.4		1191.9		50. 8	•
FLIAF LIAGE	<b>%</b> .	137.4	1.4	10.1	1.1	97.7	1.0	2.4		95.7	L.D	0.6	0.0	1.1	0.0			0.0	****
HMF Tylose	126	55.7	4.1	15.6	1.1	40.1	4.3	1.0		39.1	6.1	6.6	0.0	1.0	1.6	93.7	1,0	1.9	
	150	11333.2	75.6	1215.	21.6	8461.2	53.7	163.2	1.1	7896.1	52.7	6.6			1.1	39. 1 789 <b>8.</b> 1	<b>1</b> .3	2.0	
SULCOSE	164	7168.2	.39. 4	2069.3	16.5	5099.#	28.3	163.6	1.6	4995.4	27.8	0.1	0.6	4.6	6.1	and the second of the second	52.7	J38. 0	5.3
SULFURIC ACID	98	3817.1	39. 3	1118.4	11.3	2736.6	27.9	49.8	1.5	2686.0	27.4	4.4	0.4		1.0	1995. 1	21.6	514.4	1.2
SOOTUM HYDAOXIDE	4	0.6	6.6	1.6		1.1	1.0	6.0	0.0	1.0	1.1	1.1	6.6	1.0	1.1	0.1	1.1	1.1	1.1
CALCIUM HADBOALDE	74	9.0	1.1	1.1	1.0	8.6	4.6	0.6	6.6	8.6	1.1	2476.7	33.5	1.0 1.0	1. 0.0	1.6	1.0	1.0	L.
BENZENE 31. Veerde	/4		1.4	1.1	8.0	8.8	4.0	1.0	1.6	1.6	0.0	0.0	8.8	1.0	6.6	1.1	6.0	4.0	1.1
THRNOL	95	1.1	1.1	1.1	1.6	1.1	1.0	8.6	1.1	0.0	1.1	6.6	4.4	1.0	1.0	0.0 6.0	1.1	1.1	0.0
ODILM EARBONATE	46	8.6	1.1	1.0	1.1	0.0	1.1	4.6	6.6	1.1	1.1	1.1	0.0	1.1	0.0		0.0	0.0	9.6
OCIUM SILFATE	196	4.6	8.8	1.1	1.0	6.0	1.1	1.1	1.1	1.1	0.6	0.0	0.0	1.1	0.0	1.1	1.1	6.8	1.1
ONIUM ACETATE	145	1.6	1.6	6.0	1.1	0.0	0.6	0.0	1.0	1.1	4.6	6.0	0.6	4.6	1.1	9.0	0.0	1.1	1.1
ALCHA ACETATE	82	1.1	1.1	1.1	1.1	4.0	1.0	1.1	6.0	1.1	0.0	8.6	6.1	1.1	1.1	0.0	1.1	1.0	1.1
ETIC ACID	i 58	1.0	1.1	4.0	0.0	0.6	1.1	1.1	1.6	0.0	0.0	8.8	8.0	1.4	1.6	954.5	8.6	0.0	1.1
ETHANCE	60 12	1039.5	17.3	299.9	5.0	739.6	12. 3	14.7	0.2	724.9	12.1	0.1	8.0	8.0	6.1	\$.4.3	6.0 6.0	11.0	1.3
CETALDEHYDE		1.0	6.0	0.8	1.1	1.1	4.6	0.6	0.0	1.1	0.0	1.1	1.1	0.1	0.1	1.1		1.1	4.6
TE LIST RESULTED	49	9.0	0.0	0.0	0.0	1.6	0.0	1.1	1.1	0.6	1.1	8.8	1.1	0.0	8.0	8.6	0.0	0.4	6. f. 8. f
DTAL		502229.6		191684.5		318545.3		9517.5		300967.4		2476.7		19480. 4					
ENFERATUNE, C	MARKE	100.0		100.6												322924. 9		17063.6	
RESSURE, AIM		1.6				180.6		160.6		68.0		27.6		27.4	1,575.5	60.4		63.0	
		1.0		1.0		L.		1.0	11111	1.6		1.6	es a Milit	1.0		1.0		1.0	

STREAM NO.		n=n=+d+c	210	2:	11		\$1 <b>\$</b>		213	í	214	21	Ś	i	516
DISKETTE 420 File Sec2	MOLECULA Ne 16415		MOLES Ha	LBS :	MOLES	L DS HA	MOLES HA	LBS	MOLES Hø	LBS Jai	MOLES HA	LBS HR	MOLES HR	LOS HR	MOLES HA
										J. 184		- 100	. THE	. na	HA
MOTER	, i	8 289111.9	16061.0	616.5	. 34,2	12302, 4	683.5	301414.3	16713.2	122374 4	/70e =				:
CELLOLOSE	16		0, 0	99.7	0.5	4.5	£.,00	1.9		\$22374.2	6798.3	14684.3	815.8	107685, 9	5982.6
INSULUBLE LIGHTN		and the second s		37. Ł		2.6		2.0	.0	41497. 1 15345. 7	256.2	4979.6	36. 7	36517.3	225.4
ASH				0.0		. 0.0		£. 0				1641.6		13504. [	
MYCELLILM				8.6		6.0		1.6		185.6		22.5		163.2	
OTHER INSULURLES		6.6		21.5		l. e		1.6	***	90 32. 4		1.0		1,4	
YEAST		1.4		4.0		1.6		0.0		•		1003.5		7948.9	
CALCIUM SINFATE	138		3.4	3079.5	22.6	101.7	1.3	65 <b>6</b> , 7		1.6		6.6		0.0	
HENT. KYLON	136	9.0	1.0	9.8	0.1	1.1	.0	1.0	1.1	0. 0	9.0	1.1	8.8	9. 0	0.4
HENT, GLUCON	166		1.6	0.0	0.0	4.1	6.0	6.6	.1	4382. 6	33, 2	525.6	4.0	3857.2	29.2
CALCIUM CARBONATE	104		0. 6	1.6	£.0	8.8	1.1	1.0	0.0	4.4	4.4	1.1	0.0	1.1	0.4
DXYGEN	38		0.0	1.1	0.6	1.1	1.0	0. Ú	8.0	6.0	0.6	0.0	1.1	1.1	1.1
NETROGEN	28		6.0	6.0	6.6	1.1	1.1		0,0	0.0	1.1	1.1	1.0	8. 0	. 0.0
CARBON DIDITIDE	44		4.4	0.0	1.4	1.1	0.0	0.5	1.6	0.4	1.1	8. 6	0.0	0.0	ē. <b>8</b>
CORN STEEP LIQUOR				6.6	7.4	1.4		J. (	4.4	0, 0	6.6	1.4	₽. ₽	0, 0	1.6
MUTRIENTS		1.4		0.0				1.0		1.1		1.1		8. f	
CELLULASE				1.1		8.6		1.6		1.0		0.0		1, 6	
CELLOBIASE		0.0	~	1.1		1.6		1.1		1.1		1.5		1.1	
EXTRACTIVES				1.9		0.0		0.0		0.0		0.4		1.1	
SOLUBLE LIGHIN		***************************************		2.9		73.3		1809. 4		788.4		94.8		693.7	
FUSEL DIL		6.0		4.6		17.5	· · ·	1189.		317.6		62.5		455, 3	w v
FURFURAL	. 96		i, e	1.1		4.6		0.8		1.6		1.1		8.0	
HML	156		0.3		8.6	3.9	.1	95.7	1.0	42.●	0.5	4.9	9. L	37, 1	4, 4
XYLUSE	156		50. §	<b>0.6</b>	1.1	2.0		39. 1	4.1	16. 6	0. 1	2, 8		14.7	0. <u>t</u>
GLUCOSE	1881	4781.4		16.6	9.1	321.4	2.1	7881.5	52.5	3135.1	22.9	412.3	2.7	3922.0	20.8
SULFURIC ACID	98	4.6	26.6	18.7	0.1	203, 2	1.4	4984.7	27.7	2172.6	12. 1	268. 9	1.4	1912. 0	10.6
SCOTUM HYDROXIDE	16	1.4	0.5	6.6	0.0	8.6	1,6	1.1	1.1	1160.7	11.6	139.7	1.4	102[.0	10.4
CALCIUM HYDRUKIDE	74		4.6	6.6	1.6	1.1	0.0	0.0	6. E	1.1	1.4	6.0	8.6	8. D	1.6
BENJENE	78		0.0	1.1	1.1	1.0	1.1	e, e	6.8	1.6	0.0	0.0	6.9	8. 9	0.0
GL YCEROL	92	1.6	1.0	1.0	6. E	1.4	1. <del>f</del>	1.4	8.0	6. 🛍	0.0	8. 6	0.0	1.1	0.0
ETHORIOL		1.4	0. 8	1.6	1.1	8.6	1.1	0.0	0.0	9. 5	6. 6	0.8	0. 0	0.0	0.0
SUDILUM CARBONATE	46	9.0	6.0	1.1	1.1	1. 1	. 0.0	0. 6	0. 0	1.6	1.4	1.4	e, e	8.8	1.0
SOUTH SHEATE	186	0.0	1.1	1.1	4.4	1.1	1.6	0.0	0.0	4.5	6.5	4.4	0.0	1.4	8. 8
	145	0.0	0.0	0.6	0.0	0. 6	· 6.1	£. #	1.5	0.0	1.1	0. 8	1.6	0.6	0.6
SOPIUM ACETATE	82	0.8	9.4	4.6	B. D	£. 0	4.4	1.4	0.0	0.0	0.0	0.0	8.6	0.6	0.0
CALCIUM ACETATE	158	913.5	5. 8	₹. 0		39. 1	1. Ž	952.6	6.0	6. 6	8. 0	1.4	0.0	ð. <b>0</b>	A. 0
ACETIC ACID	64	8.6	1.4	<b>0.0</b> 3	1.4	1.0	· 1, €	1.0	. i - 1, i -	314.6	5.2	38.1	0.6	276.5	1.6
METIGNAL DESTAIL DESIGNA	32	4.0	6.6	6.0	1.6	0.4	8.0	0. 0	9. 8	1.6	9. 6	Ø. Ø	6. £	1.1	0.0
ACETAL DELIYOE	44	1,8	₽, ₽	1.1	0, 6	9. 6	5 · 0,0	· 6.6	4.4	0.0	●, 6	0.0	8.0	0.0	0.0
101AL		305842.0		3989.2		13183.6	•	319025.7	•	201262.0		24152.4		177103.E	
TENPERATURE, C		83.6		43.6		83, 6		63.8		98. ●		CA &		ne e	
PRESSURE, ATM		1.0		1.0		1.8	100	1.8				91.1		96, #	
• 1				•••		1.0		1.4		1.0		l. 0		1.6	

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0:

STREAM NO.		3	41		342	3	03		<b>34</b>		<b>e</b> 5		186		07		160		109
DISKETTE 420 FILE BEC3	MOLECULAR MEIGHIS	LOS Ha	MOLES	LBS	MOLER	LPS	MOLES	1.95	MOLES	LOS	HOLES	LDS	MALES	LPS	PULES	L DS	MULES	LBS	<b>#0.8</b> 9
rice octa	METON 13			HE	HE	HÆ		HØ.	HA	HR (1.1.1	<b>H</b>	HA	H	HI	HÆ	HA	HÆ	НЯ	HA
WATER	10	14684.3	815.4	22329. 3	1214 6														
CELLULUSE	165	4979.8	30.7	6.0	1240.5	6.6	1.0	1.6	1.1	670.0	37.1	18649.6	1036.1	56103.2	3133.5	1.1	0.0	33865.4	1881.
INSOLUBLE LIGHTM		1841.6		0.4	6.6	9. 0 9. 0	1 . A.	1.1	1.1	1.1	0.0	1.1	6.0	0.0	0.0	0.0	0.0	1.1	<b>)</b> ,
<b>19</b> 1		22.5		8.6		0.0		4.0		1.1	•	IBLE.		3661.1		1. <b>1.</b> 6.		1439.6	
NYCELL LUM	en e	1.0	***	6.4		1.0		6.6 0.0		6.0		22.5	·	11.9		0.4		43.0	
THER INSOLUBLES		1003.5		4.6	=			1.1		6.6 6.6		340.0		697.6		1.1		662.4	**
reast		1.1		Li		4.4		1.4				1063.5	•••	2167.4	- <del></del> -	1.6		2058.5	<del></del>
ALCIUM SILFATE	136	0.0	0.4	6.0	0.0	0.0	1.0	1.1	0.0	1.1	1 11 1 1 1 1	1.1	1.5 · •	1.1		1.1	, s s	1.1	· · · · · · ·
ENI. SYLAN	112	525.6	4.0	6.6	A.D	8.0	1.0	4.4	1.1	1.1	1.0	1.1	1.6	1.1	0.0	1.1	6.0	1.0	0.0
HENT. GLUCAN	162	0.0	0.6	0.0	1.1	6.0	1.6	1.1	4.4	1.4	1.1	8.0	1.6	1.1	1.1	1.1	1.1	9.0	4.4
CALCIUM CARBONATE	100	4.4	8.0	0.0	1.4	LE	1.1	4.0	1.1	0.0	9.8 0.4	0.0	1.1		1.1	1.0	1.1	6.0	• •
DI YGEN	32	8.8	0.0	0.0	1.0	1.0	2 2 <b>0.0</b> 1	14107.5	114.5	8.0	6.0	0.6	4.0	1.1	1.0	6.0	1.1	1.1	<b>(.</b> )
II I ROGEN	28	4.1	0.6	4.0	8.4	1.0	0.6	53072.6	1895.4	6.6	0.0	0.0		8.6	4.0	12697.1	3%.6	1.1	4.6
CAABON DIGITIE	- 44	0.0	0.6	0.8	6.4	1.0	0.0	1.1	4.0	6.6	8.0	1.5	6.6	0.0	5.0	53072.6	1895. 4	0.0	<b>1.</b>
CORN STEEP LIGUR	V (j. 1. 444)	1.0		0.0		1.0		1.4	***	795.3		11.7	6.0	0.0	0.0	5774.1	131.2	1.0	<b>0.</b> 0
NTRLENTS		6.6		0.4		0.6		8.6		1.1		0.0		867.4		1.1	•	101.6	
ELLULASE	***	6.8		8.0		8.8		Li		4.6		11.7		1480.2		1.1		6.6	· · · ·
ELLOBIASE		1.1		1.1		0.0		0.4		4.6		13.7		0.0				889.1	
LITRACTIVES		94.8		1.0		0.0		0.0		8.8		1.0		95.1		6.6		1.4	
OLUME LIGHTH		62.3		0.0		0.0		6.6		2.0		1.0		61.5		1.1		57.6	
USEL OIL		8.6		1.0		4.0		6.0		1.1		4.0		8.6		8.6 8.6		36. 1	· · ·
UNFURAL	96	4.9	0.1	0.0	8.6	0.0	8.8	1.0	1.1	6.0	0.4	1.0	0.0	1.9	0. 1			0, 6	
<b>6</b>	126	2.\$		1.1	0.0	4.0	4.1	4.0	0.0	8.8	1.0	6.6	1.0	2.0		0.0	1.1	2.9	
IYL OSE	150	412. 3	2.1	1.1	1.1	1.1	1.1	9.0	0.0	1.0	1.0	0.0	1.0	0.1	0.0	8.8	8.8	1.0	(
A LICOSE	186	260.9	1.4	1.0	1.0	6.0	A.F	0.0	0.0	2.6	4.6	1.1				, Alexandro Tellino	•	1.1	9.1
BULFURIC ACID	9	139, 7	1.4	0.0	0.0	0.4	6.0	0.6		0.0	1.1	1.4				6.0 0.0	1.6	9.0	6.1
OPTION HYDROLLOE	49	5.6	0.8	1.0	0.0	138.7	1.5	6.8	0.4	1.1	1.1	8.6	0.6	0.0	0.0	1.1	0.6	1.1	0.0
ALCIUM HYDROXIDE	74	0.0	1.1	0.0	0.0	8.8	1.1	1.4	1.1	1.1	9.1	0.0		6.6	0.1	1.0	1.1	1.6	6.0
ENZENE	78	1.0	8.6	6.0	6.0	1.5	0.0	1.1	0.0	1.0	8.0	0.0	8.0	1.1	1.6	6.6	9.0 1.1	1.1	0.1
L YCE ROL	35	1.6	1.1	0.0	1.0	1.1	1.1	8.6	1.1	8.8	6.4	8.6	4.6	9.0	0.0	6.0	6.0	1.8	0.1
THANGE	46	1.1	1.1	6.0	1.1	1.1	1.0	6.6	6.6	6.6	4.0	1.4	0.4	8.8	1.1	8.8	1.1	6.6	0.0
COTUM CARBONATE	186	1.1	0.0	0.0	1.1	1.1	6.0	1.1	6.0	4.6	1.0	1.1	1.1	8.6	6.0	1.1	1.1		6.1
OOLUM SILFATE	142	1.0	1.6	4.0	0.0	0.0	6.0	0.4	1.1	1.1	1.1	2.8		294.2	1.1	9.4	8. F	123, (	3 1.0
ODIUM ACETATE	82	1.1	8.0	0.6	0.0	6.6	0.0	0.6	0.0	0.0	1.1	1.0	0.6	52.1	0.6	6.6	6.0	123,1	. 6.9
ALCIUM ACETATE	158	1.6	8.5	6.0	4.0	6.0	0.0	0.0	0.0	4.6	0.0	0.0	1.6	4.6	0.1	1.1	1.0	0.4	8.4
CETIC ACID	60	34.1	0.6	1.1	0.1	1.1	0.0	9.0	0.0	1.0		6.6	1.5	4.6	6.0	1.1		4.4	4.0
ETHIRMOL	75	1.1	0.6	0.0	0.0	0.0	1.6	4.0	1.1	0.0	1.1	0.6	9.4	1.1	0.0	1.1	9.0		
CETAL DEHVOE	46	0.0	0.0	0.0	6.0	0.0	6.1	1.1	0.0	1.1	1.1	0.0	8.8	0.0	0.4	0.0	0.0	6.6 0.0	0. 0 0. 6
DIAL		24152.4		£5153° I		138.7	ad page Markan	67180.5		1473.3		21975.7		65766.6		71513.6		41757.	
EMPENATURE, C		100.0		27.6		27.0		27.6		27.0		21.0		20.0					
RESSUME, ATM		1.8		1.0		1.0		1.0		1.0		11.		20.0		20.0		28. 0	
	Albert School			• • •	44 1 4 24	1.7				1.0	are discount			1.0	1 1 1 1 1 1	I.€.		1.0	

SIREAN NO.		3	114	3	11 .	: 3	15	3	13		314		315		16	31	17
DISHETTE 428	MOLECULAR	LDS	HOLES	LBS	MOLES	LBS	NOLES	LPS	MOLES	LBS	MOLES	LOS	NOLES	189	MOLES	LBS	MILES
FILE SECI	Retoils	HA	I 📽	HR	HR	HII ·	HA	HA	開	HA	RH	na ·	H	HÆ	HÆ	<b>岩</b> 欄	Hi
		*****							·····								
WATER	I B	22537.4	1252.1	28678.5	1159.9	54744.2	3841. 3	33673. J	1876.7	78398.4	4355.5	57327.4	3:44.9	100935.8	5607.5	15024.3	834.7
CELLULOSE	165	1.6	0.0	0.0	0.0	8.0	. 4.0	0.0	8.8	6.6	0.0	0.0	8.0	0.0	0.0	0.0	0.6
INSOLUBLE LIGHTN		163.7		1.1		3199.6		3324.7		174.9		9.8		358.6		1483.1	
ash		2.0		0.6		43.6	~~*	41.0		2.0		0.4		3.9		18.6	
NYCELLIUM		35. 2		0.5		662.4		629. 2		33.2		1.4		68.4		280, 4	
OTHER INSULUBIES		108.4		0.4		2956.\$		1956. 0		182.6		4.6		211.€		872.5	
YERST		0.0		0.0		1.6		6.6		0.0	~~~	1.6		1.1		0.0	
CALCILIA SIX FATE	136	0.0	0. F	0.0	8.0	1.6	9. 0	8.9	0. 0	0.0	9.9	1.1	0.0	0.6	4.6	0.0	6.6
HENT, NYCHA	132	1.4	1. <b>£</b>	6.6	4.4	1.6	1. 0	0.6	1.1	Đ, Š	1.1	0.0	0.0	8.4	6.0	6.0	4.4
HENT, BLUCOM	165	1.1	\$. £.	0. 🕈	9. 1	1.1	8.6	6.4	1.6	0. 0	4.4	1.1	1.1	0.0	4.0	0. Ŭ.	<b>₽.</b> ⊈ ,
CALCEUM CARBONATE	198	4. 6	1.4	0. 6	4.6	0.0	1. \$	6.6	0. T	0.0	4. 5	1.4	£.	9. \$	4. 6	0. 0	₿. <b>₽</b>
DXYCEN	32	0.0	₽.€	<b>0.0</b>	4. ŧ	0,6	4.4	8. 6	1.1	8.8	0.4	1.1	1.1	1.4	0.0	. 1. 1	4.4
NT 1 ROGEN	26	4.0	0.6	0.0	4.0	0.6	1.1	6.0	0. 0	0.4	8.4	1. 0	8. 6	1.1	0. 0	0. 0	0.0
3011018 MORRA	44	6.4	9. 0	1.0	9.4	1.1	9. 4	0.6	1.6	0.0	9. 0	0,0	9. 0	1.1	0.6	8. 0	6. B
CORN STEEP LIQUOR		322, 4		6.6		484.6		14.7		469. 9		<b>1.</b> F		792.3		2.9	
MUTRIENIS		8.6		0.6		1.5		0.6		0.0		1.0		0.0		0.0	
CELLULASE		591.1		1.1:		889. L		26. 4	*	862.7	4- p.	1.1	*~~	1453. B		12.7	
CELLODIASE EXTRACTIVES		8.6 26.4		4.1		1.1		1.1		0.0		0.0		0.0		0. 0	
SOLUBLE LIGNIN		38. j. 25. š		1.1		57.6		6.0		55.7		1.1		93.6		1.0	
FUSEL DIL		23, 1 1. 1		8.0 0.0		36. L		1.0		37.1		1.1		62.5		0.0	
FURFUROL	96	2,6		0.0	1.1	6.0. 2.9		1.9:		1.6		1.1		0.0		0.0	
HOF	126	1.6	.0	4.8	0.0	1.6	.ŧ .t	0. 0 0. 0	8. S 8. S	2,9: 1. <b>6</b> :	.4	0.0	0. 6 6. 6	4.9 2. <b>8</b>	0. i	0.0 0.0	0. 0 0. 6
XYL OSE	150	0.6	0.0	1.1	1.1	1.V	1. 1	6.6	8.4	4.0	.6 8.6	1.1	0.8	0.0	1.1	4.0	0. 6
SLUCOSE	180	6.0	4.0	0.0	6.1	4.0	1.1	1.4	8.8	6.6	4.6	0.0	5.0	0.0	0.0	1.6	0.0
SULFUNIC ACID	98	0.0	0.0	1.1	1.1	6.4	0.0	0.0	6.6	1.1	6.6	1.1	0.0	1.1	0.0	1.1	1.1
SODELM HYDROXIDE	48	0.4	0.0	0.6	0.4	8.0	0. 6	0.4	8.6	1.1	8.6	0.6	8.0	0.6	1.4	1.1	0.0
CALCIUM INVORONTOE	74	1.1	1.1	0.0	1.1	1.4	0.6	8.0	1.4	4.4	1.6	1.1	0.0	0.0	0.0	1.6	0.0
BENZENE	78	0.0	1.8	0.0	1.8	1.1:	6.0	1.0	0.0	1.4	4.4	1.1	6.0	0.0	0.4	0.6	0.0
GL. YCEROL	92	0.0.	4.9	0.0	1.1	0.4	1.1	1.6	1.1	0.6	1.1	1.1	0.6	1.6	1.1	1.1	0.6
ETHORIOL	46	6. \$	1.1	1.6	1.0	0.4	1. 1	1.1	1.1	0.0	4.4	0.0	8.0	6,0	0.0	1.0	0.0
SODIUM CARBOMATE	1 <del>96</del>	0. 0	4.1	4.4	8. 8	0.0	0.0	8.8	0.0	1.0	1.1	1.1	0.0	1.1	0.0	6.0	0.0
SODIUM SULFATE	142	A1. 1	4.6	1. 0	1.1	123. f	8.9	1.9	. 0	119.2	4.8	8.6	4.9	200. 3	1.4	2. 0	. 0
SUDIUM ACETATE	85	21.5	1.3	1.1	1.1	. 31.3	1.4	1.0		30. 3	0.4	0. t	0. 8	51.6	0.6	0.0	8.6
CALCIUM ACETATE	158	6.0	9. 0	1.6	1.6	0.,0	1.1	4.1	8. 0	0.0	0.0	1.4	8.0	8. 0	0, 0	0.0	0.0
ACETIC ACID	€0,	1.0	0, 0	1.0	0.0	0. 6	B.E.	1.1	9.0	0,0	0.0	£. 0	9.0	. 1.1	e, õ	0.0	0.0
METHANOL :	15	1. 6	6. 8	4. 8	0. 0	0.0	I.f.	1.5	e. 6	e. e	0.0	B. #	8. €	6.6	€. €	1.4	0.0
ACE TAL DELIYDE	441	1.0	ē. ē	0.0	0.0	0,0	1. (	1.6	9, 6	0.0	0.0	6.6	0.0	1, 6	0.0	9. 0	0. 6
TOTAL		23919.2		20076, 5		62635.5		39673. 0		80289.9		57327.4	1 1,1	101239.1	5+	17697.4	
TEMPERATURE, C		20.0		27.0		28. 6		27.		27.0		27.0		27.0		21.6	
PRESSURE, AIM		1, 0		1.0		L€		1.0	.*	1. 🕡	٠.	1. 🕯		1.0		1.0	

### ETHANOL VIA ENZYME HYDROLYSIS - SECTION 1001 ENZYME HYDROLYSIS

STREAM NO.		1	đị		102	January 4	03	•	O.		495	9 1 1 V	<b>6</b> 6		487		198		489
DISKETIE 450 File seca	MULECIAL RAE ME 1 GHTS	189 HR	MOLES HR	1.99 H#	MOLES Hill	LBS HA	MOLES HR	LDS HR	MOLEG	LDS H#	HOLES HR	LBS HB	MILES Hil	LDS HR	NOLES HR	LAS HAR	MILES	LDS	MÜLE HA
											1.5 3 3 3							*****	
WATER	i <b>i</b>	107605.9	5982.6	301414.3	16745.2	1.6	8.6	26435, 7	1464. 6	100935.6	5607.5	156.3	8.7	53/083.3	27030.0	532904.7	23685.8	38484.9	3133
CELLILOSE	162	36517.3	225. I	4.9		8.6	0.0	4.6	4.1	0.4	1.1	8.6	8.8	36522.2	225. 4	3652.0	22.5	32191, 3: 8. 8	2133.
ingoluble lighth Ash	1. 1. • • • • • • • • • • • • • • • • •	13544.1	•	2.0	***	0.0		0.0		354.6		8.8		13864.6		13864.6		1.1	•
REP) Rycellium	***	163.2		1.1		4.4		8.0		1.9	, . <del></del> .	0.6		167. 1		167.1		0.0	
HILELLIUM HIHCA INSOLUBLES		1014.0	: · · · · · · · · · · · · · · · · · · ·	1.0		8.0		8.0	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	64. 4		0.0		68.4		64.4		1.1	
EAST		1918.9	1 177	1.0		0.0	****		a" = a <del>"-</del> "	211.4		0,0%		816 <del>0,</del> 9		8160.9		0.6	· · · · · · · · · · · · · · · · · · ·
3 F 5 4 1 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2		8.0		1.6	****			1.0		0.0		1.0		1.0		8.8		1.1	
ALCIUM SULFAIE	136	0.6	1.6	650. 7	<b>1.1</b>	1.1	4.4	4.0	0.0	1.1	0.0	LO	8.0	979.0	7.2	979.0	1.2	1.4	
ENI. IYLAN	132	3657.2	29.2	L.D.	. •	4.0	0.0	1.0	6.6	1.1	6.6	0.0	0.0	3858.2	29.2	1.1	0.0	1.6	I.
ENI. GLUCAN	165	1.1	4.4	0.0	1.0	6.6	8.0	8.8	1.1	0.0	1.1	0.0	8.0	8.0	1.1	4.6	8.0	6.4	
LCIUM CARBONATE	100	4.4	4.0	6.8	6.0	1.5		LI	0.0	1.0	1.0	1.1		0.6	4.0	6.0	0.0	0.0	
XYEN .	35	4.0	0.0	1.0	1.6	1.1	4.1	0.4	0.0	8.0	9.0	1.1	6.1	LI	8.0	L.O	6.6	1.0	
TAGGEN	28	1.1	1.1		LI	1.0	1.1	6.0	0.0	0.0	8.0	1.0	0.0	0.1	0.0	1.1	J. 1	1.1	•
ARBON DIOXIDE	44	1.1	4.0		1.1	0.0	0.0	0.0	0.6	1.1	0.6	0.6	0.0	8.0	0.0	1.1	1.1		1.
NAM SIEEP LIANDA		1.8				1.0		0.4		192.1		1.1		792.1	•	792.1		9. 8 6. 8	•
IRLENIS		1.1		4.6		8.6		1.1		4.4		4.1		1.1		6.4			
LULASE		4.0		1.0		1.6	: :	4.0		1453. 8		6.0		1453.6	•	1453. 8		1.1	
LODIASE		0.0		1.0		0.0		6.0		1.6		85.6		65.0	***	85.0		0.0 0.0	· · · · · ·
MACTIVES	· · · · · · · · · · · · · · · · · · ·	693.7		1609. 4		8.6		8.6		93.6		0.0		25%. 5		2596. 9		7.5	4.4.57
THE FIBRIN		155. 3		1189.0		1.6		1.4		62.5		1.0		1706.0		1706.0	N=7	0.6	4 / 5
EL OIL		L.I	***	1.1		0.5		0.0		1.1		6.0		0.0		0.0	***	0.6	
NTURAL .	96	37.1	8.4	95.7	1.0	1.1	0.0	0.0	0.0	4.9	0.1	8.0	0.6	137.8	1.4	137.8		1.1	[14] [15]
	156	14.7	8.1	<b>19.</b> l	4.3	1.0	1.0	1.0	1.0	2.0		1.6	1.1	55.7	0.4	55.7	0.4	1.1	0.(
. D&E	150	3022.6	24.2	7641.5	52.5	6.6	0.0	1.0	1.1	1.1	0.0	1.1	1.1	10944.3	72.7	15285.1	161.9	0. 8 0. 6	J. (
LICOSE	160	1912.0	18.6	4984.7	27.1	4.0	1.1	8.6	0.0	0.0	8.0	0.6	9.0	6896.6	38.3	43418.5	27 7 4		8.0
LFURIC ACIA	***	1021.0	16.4	1.0	1.1	0.6	1.1	0.0	0.0	0.0	6.6	8.6	0.0	7.6	0.1		241.2	1.1	
DIAM HABBOTIDE		1.0	1.8	8.0	1.0	634.1	15.9	6.6	0.0	0.0	1.0	0.1	1.1	1.1	7.4	7.4	1.1	1.1	1.0
CIUM HYDAGIIDE	74	0.0	0.4	0.6	1.1	349.8	4.7	1.4	L.O	9.0	4.1	1.1	1.1	1.0	1.1	6.6	0.6	1.1	- <b>⊕.</b> ।
NTENE	74	0.0	1.1	4.0	1.1	6.6	1.0	0.0	6.0	1.1	4.0	0.6	6.0	0.0	8.0	1.1	1.1	1.1	
(CE NOL	22	1.6	1.1	1.1	1.1	0.0	8.0	0.0	6.6	1.0	0.0	8.8	0.6	6.6	0.5	0.0	1.0	1.1	1.1
ano.		0.0	8.8	6.0	1.0	1.0	1.0	1.0	1.1	0.0	0.0	1.6	1.1	7.5	1.1		1.1	8.6	0.4
ILIM CARBONATE	106	<b></b>	. 1.1	8.0	1.4	0.6	0.0	4.1	1.1	1.0	9. 0	1.1	1.1	1.1	0.0		1.6	0.5	9. (
ILM SULFATE	115	6.0	1.1	4.6	0.4	1.8	1.0	0.0	6.6	200.3	1.6	1.0	1.0	1325.4	9.3	1.1	1.1	1.1	0.0
ILIM ACETATE	82	0.0	4.1	6.0	6.6	1.1	9.8	0.0	6.6	51.8	0.6	0.0	1.1	31.6		1,325. 8	9. 1	4.6	- 4.0
CIUM ACETATE	158	1.1	4.0	952.6	6.0	1.1	6.0	1.4	0.4	0.8	8.4	4.1	1.4	1317.4	1.6	51.0	0.6	6.6	· •.6
TIC REID	60	276.5	1.6	1.4	1.0	6.6	1.1	1.1	4.4	4.0	1.0	4.6	1.1	0.0	6.3	1317.0	0.3	1.6	6.0
HANDL	22	1.1	0.0	1.0	1.0	6.6	1.1	1.0	4.4	0.0	1.0	6.0			0.0	0.0	4.0	0.0	
TALDEHYDE	**	0.0	0.0	6.6	0.0	1.1	8.8	0.0	4.6	6.6	1.1	0.6	1.6	6.0 6.0	0.8 0.8	1.0	8.6 8.0	0.8 0.0	0.0
<b>RL</b>		177109.6		319025.7		963.6	•• ••	26135.7	•	101239. [	-	241.3		628035, (		628435, j			• •
PERATURE, C		98.0		13.0		27.6		97.4					선임님님	Alberra,		~ 04 11, 1		38464.9	
SSURE, ATM		1.8		1.0				27.6		21.0		27.0		50.0	医毛囊溶液	50.0		27.0	
		• • •	机多数电路			1.4	Section 1994	1. I.O. 1		. i. 1.0	artik Alber	L	31 P. S. S. S.	1.0	交通 有效	1.0		1.0	打造 特定

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STREAM NO.		3 to 1	10:		11	}	115	` 4	13	4	14	1	115		116	•	17		418
DISKETTE 450 File seca	MOLECULAR Melohis	l DS	MOLES HJ	LÖS IA	MOLES HA	LØS HØ	MOLES	LDS	MOLES	LBS	MOLES	LDS	MOLES	LBS	MOLES	1.85	HOLES	LAS	MOL
					na .		HA	106	紙	HR	Ř	116	H <b>A</b>	HA	Hit	RA	HÄ	服	HA
MATER	18	36573. 6	2031.8	1831.9	Let. a	524363.7	29131.3	45114.6	2586.3	2258.8	125, 5	523936.8	29107.6	15024, 3	834.7	62397, 1	3466.5	23079, 7	1282
CELLULOSE	162	8, €	0.0	1.4	4.4	182.7	· 61	3169.3	21.4	173.9	1. f	8.8	0.1	0.6	0.0	3643.2	22.5	3643.2	22
INSOLUBLE LEGNIN		I.F.		0. E		693, 7		13170.9		659.5		34.2		1483. 1		1531 1.5		15313.5	-
ish ·		0.0		4. 0		8.8		158.3		6.6	~~~	1.1	***	18.6		185.6		185.6	
YCELL IUN		4.0		6. F		2.9		65.5		2.9		1. i		290.4		346.6		348.8	
THEA INSOLUBLES		4.1		0, 8		488, 4		7752.5		387. 9		20.5		872.5		9612.5		340.0	
east	of to set	6. \$		1. (		0.0		6.4		4.0		0.4		6, 8		8.8			_
ALCIUM SULFATE	136	6.4	8.0	6.0	6.0	970.2	7.1	6.4	0.1	0.0	6.0	970.2	7.1	5.4	2.5	8.8		8. 8	
EKE. XYLAN	132	0. 6	0.0	0.0	0.0	1.0	Ĥ	0.0	1.1	1.1	1.1	0.0	9.0	1.6	6.6	0.0	0. L 0. 0	2.9 8.#	
EMI, GLUCAN	162	0.1	4.0	0.0	0.0	8.6	1.1	1.1	1.6	0.0	0.6	0.0	1.1	6.6	6.6	1. 1	1.1	1.1	
ALCIUM CARBONATE	100	6. 6	6.6	0.0	0.0	1.6	6.1	0.0	0.0	8.4	1.1	1.4	1.1	1.1	1.1	0. C			
XYGEN	35	0.0	8.6	1.1	1.0	9.4	4.4	0.0	1.1	8.8	0.0	6.6	8.0	1.1	- 1.1	0. O	0.0	6.0	•
HI 180GEN	26	1.\$	8.0	0.4	4.0	1.4	0.0	8.6	1.1	8.8	0.0	6.4	1.1				1.0	0.0	
CARBON DIOLIDE	44	0.0	4.4	0, 6	6.0	0.0	1.1	5.6	0.0	1.1	8.0	6.4	0. S	0. S 0. S	1.1	9. 0	<b>8.</b> €	9.0	
CORN STEEP LIQUOR		0.4		1.6		785.5		6.6	-	4.4	***	765.5			0.0	0.6	0.0	1,0	0
UTAJENTS		4.6		8.0		0.0		4.4		4.4				2.9		9. 6	~~=	1.9	
ELLULASE		0.0	•••	0.6		1441.1		12.7		1.0		0.0		8.0		8.9		1.0	-
ELLODIASE		8.6		0. 6		84. 6		1.0		1.0		1410. [		12.7		26. \$		9.6	-
ITTAACTIVES		0.8		6.0		2574. 4		22.5				84.0	7.75	1.0		į, I		4. 0	_
OLAIBLE LIGHTN		0.4	***	0.6		1692.2		14.7		4.6		2573.4		1.0	~	24.4		0.6	-
USEL OIL		1.4		1.0		0.0				1.0		1691.2		1.1		15.6		5. 9	-
URFURNL	96	B. <b>8</b>	4.6	8.6				8.6		1.1		4.6		<b>(.</b> )		0. 0		1.1	-
PF	126	1.6	1. 6	4.6	6.0	136. č	1.4	1.0	.1	6.1	1.4	136. 8	1.4	9. 0	4.1	1.0	6	1.4	€.
VI.OSE	150	4.4	0.6	1.0	0, 0 0, 0	55.7	1.4	1.1	5.0	1.1	1.4	55.7	0. (	0. 0	8.8	9. 🛊	0, 8	0.0	8
ALUCOSE .	18#	0.8	1.5			15155.2	101.0	133.8	4, 9	6.4		15148.4	iol. O	0. 6	0.0	140.7	0.9	51.8	•
ULFURIC ACID	98	1.6	9. 0	0,6	1.1	43037.8	239. 1	379. L	2.1	19.5	6. 1	43020.2	239.	1.1	8. 0	398. 6	2.2	147.5	. · · · •
ODIUM HYDROXIDE	10	8.0	1.0	0. 6 0. 6	1.1	7.8	6.1	0.0	1.1	1.1	1. \$	7. 6	0. [	0.0	₽, €	4.4	1.0	0.0	●.
ALCIUM HYDROLIDE	74	0.0	1.0		8.6	0.0	1. E	1.6	1.1	1.1	0. 0	0. 0	₽. ₿	4.0	4.0	1.6	0.0	1.0	6.
ENZENE	75			6.0	1.4	1.4	1.1	1.1	6.0	0.0	0. 0	0. 0	E. S	4.4	1.4	1.4	1.0	0.6	4.
LYCEROL	92	1.0	1.6	6.0	1.6	0.0.	1.1	0.6	8.1	₽. ₽	1.1	4.4	6.6	1.1	1.0	1.0	0.0	0.0	1.
THRANCE	7E 46	4.4	1.1	0.0	1.1	0.0	. I.4 :-	1.1	₽. \$	4. 6	4. #	Ð. <b>Ú</b>	9.0	€. €	1.1	1.1	6.0	6.4	8.
ODIUM CARBONATE	186	4.0	9.0	0.0.	1.1	1.0	9. 6	0.0	1.6	0.0	J. T	4.41	1.6	9, 5	ı, j	1.1	0.6	9. 6	0.
DOTUM SULFATE	112	0.0	8.6	1.1	1.1	8.4	1.6	0.0	<b>0.</b> €	1.6	1.4	0.0	0.0	9. 6	Ð. Ø	0.0	0.0	1.1	
DOLLA ACETATE		0.0	4.0	0.0	8.0	1314.1	9.3	$\Pi_i T_j$	6.1	1. ₽	.1	1313.1	9.2	2.4	.4	14.7	<b>8.</b> 1	5. 9	
	88	0.0	6.6	9.6	8.6	51.4	. 6.1	1.1	1.1	0. 6	Ø. Ø	51.6	O. 6	8,4 -	8.8	4.4	<b>1.6</b>	0.6	8.
ALCIUM ACETATE	158	4.0	<b>8</b> , 6	6.0	4. 6	1305. 3	1.3	11.7	f. l	1.0	. 0	1304. 3	0.3	0. 0	8.8	12.7	0. j	4, 9	
CETIC OCID	66	4.4	0.0	8.0	1, 0	0.0	1.1	1,1	R, Ø	h.t.	1.1	9. 6	9.4	1.4	1.6	0.0	0.0	0.0	1.
ETHANOL	12	6.0	0.0	0.6	6.0	. 6.6	4.0	1. 1	8. 5	• 1,€	0.0	1.6	4.6	6.6	9. 6	0.0	0.0	0.4	ā.
CETALDEHYDE	44,	0.0	0. 6	0, 0	1. f	1, 0	1, 9 .	0.0	8. €	0.6	4. ₽	0.6	4.4	0. 0	. 0.0	O, Ű	0.8	4.6	0.
TAL :		36573. 0		1831,9		591273.9		78334.2	 	3523. £	•	592582. /		17697.4		91554, 7		51825.	
INPERATURE, C		27.		27.0		19.8		48.0		47.# ∞		49. 8	•	28.0		55.#		44. 8	
RESSUME, ATM		1. €		1.6		1.1								F-0.4		77.10		77.0	

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STREAM NO.		41	19		120
DISKETTE 458	MOLECULAR	LBS	MOLES	L89	HOLES
FILE SEC4	WEIGHIS	HE	. /順 : : :	<b>Hil</b>	1 <b>6</b>
WIE C	10	39317.4	7184 1	56,3251, 2	11201 0
CELLULOSE	165	173111.4 0.8	2184, 3	363631. E	31291.9
DESOLUBLE LIGHTN		Li		31.2	W- 1
ASIL		1.1		6.6	·
MACEFFIRM		1.0	<del></del> .	1.1	
OTHER INSOLUBLES		0.5		21.5	
YEAST	LA BERT	1.0		4.0	
CALCIUM SULFATE	136	5.9		976.0	7.2
HEN]. IVLAN	132	1.1	6.0	1.5	, <b></b>
HENT. BLUCAN	162	0.0	1.0	1.1	4.0
CALCIUM CARBONATE DIYGEN	35 100	8. 6 0. 6	6.0 6.0	6.8 6.8	1.1
NITROGEN	28	1.0	1.0	8.0	6.6
CARBON DIGITAL	44	A.O	8.6		Li
CORN STEEP LIGHOR		5.9	31, 44	/91.4	
MURIENTS		0.0		1.1	
CELLULASE		16.6		1456.7	
CELLODIASE		1.6		85.4	<del></del> .
ELTAACTIVES		15.6		2589. [	
SOLUBLE LIGHTM		9, 8		1701.0	
FUSEL OIL		1.1		1.1	
FURFURAL	<b>%</b>	1.0	.1	137.0	1.4
IN OSE	126 15 <b>0</b>	6. <b>6</b>	1.6	<b>55.</b> 7 15237. 3	8.4 101.6
ALUCOSE	180	251.1	1.4		240. 6
SLAFIANC ACID	94	0.0	0.0	1.8	4.1
SOUTUM HYDROTTOE	, i	1.0	0.0	4.0	
CALCIUM HYDROXIDE	74	6.0	1.6	0.0	0.0
BEKZENE	71	LI	1.6	1.0	6.0
GL YCEROL	92	0.0	1.1	0.1	0.8
ETHANOL	46	1.0	2.0	4.0	0.5
SODIUM CARBONITE	106	1.1	1.1	6.6	6.1
SIGNIM SAFATE	142	4.8	0.1	1321.9	1.3
SODIUM ACETATE	85	1.1	1.1	51.6	0.6
CALCIUM ACETATE ACETIC ACID	15 <b>8</b> 68	7.8 8.4	1.1	1312.1	4.1
METHANDE	12 12		1.0	4.6	8.0 8.0
ACE IALBEHYDE	32 11	1.4	A.4	4.1	1.0
TROOP I FILL BRANCH I MALE		V. W	•••	<b>4. 4</b>	
101AL		19729.7		632312.4	
		STATE OF	V 10 V	VIII :	
TEMPERATURE, C	A PAGE	44.0		18.6	
PRESSURE, AIR		1.0		3.5	

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STREAM NO.			501	5	<b>6</b> 2		503		584		505		546		507		548		509
DISKETTE 450	MOLECUI AN	LBS	MOLES	LBS · ·	MULES	186	HOLES	LBS	MOLES	1.05	MOLES	LOS	MOLES	I.ÐS	MOLES	LBS	MULES	1.60	war to
FILE SECS	WE LOHIS	II.	HR	HAT	HUR	HR	HÆ	HA	HE	HA	HR	HŘ	HA	HA	HI	Hill	HIL	l.&S Hr	MOLES HR
1,212						~~~~~				*********	*******					166	1/11	138	TUR
WITER		******																: :	
CELLULGSE	10	56,3254. 2	31291.9	8.0	1.0	38136. 2	2118.7	562803.8	31266.9	514666.#	28592.6	455283.0	25293.5	397052, 8	22058.5	336982.0	16721.2	257691.6	14316.2
INSOLUBLE LIGHTN	168	6.4.	0. j	0.6	1.4	585.1	3.1	513.9	3.2	513.9	1.5	513.9	3.2	513.9	3.2	513.9	3.2	513.9	3, 2
ASH	***	34.2		6.0		1963, 🛭		1998.0	·	1998.0		1998.	***	1998, 6		1978, Ø		1938.0	
MYCELL RUM		1.6 1.1		1.1		1.1		1.1		1.0		9. 8		1.6		1.6		1.6	
OTHER INSOLUBIES		20.5		1.6	***	1.6		0.0		Ø. <b>\$</b>		1.1		1.1	·	1.4		0.0	
YEAST		1.0		<b>0.6</b>		1177.3		1197.8	••-	1197. 6		1197.8		1197.0		1197. 8		1197.6	•
CALCIUM SULFATE	136	976.4		0. D		0.0		0.0		8. Ø		0. j		8.0		1.1		9.4	
HENT. TYLAN	135	2/Q.W.	7.2	0.0	0.0	29694.9	210.3	39671.0	225.5	30671.0	225.5	30671.0	223.5	30571.0	225. 5	30671.0	225.5	30671.0	225.5
HENT. GLUCAN	132	8. G	0. S 0. O	1.6	1.0	0.0	0.0	4.4	1.1	. 6.0	1.6	1.6	1.1	0.4	4. #	1.4	1.6	0. 1	1.1
CALCILM CARBONATE	166	8.6	6. 6	6.6	0.0	1.0	1.6	1.1	0.0	1.1	1. (	1.1	0.0	0. 0	€. €	0.0	9.8	0.0	6.8
DIVEEN	75	0.0	8.6	0. 0 0. 6	1.1	8.0	0.0	1.1	1.1	1.1	6. 6	9. 0	0.0	1.1	· 0.0	0.0	0.0	1. 8	8.0
NITROGEN	28	1.8	9.0		0,0	8.8	1.1	1.1	4.1	1.1	0, 61	9.8	6.9	1.6	0. 0	0.0	8. 6	9.4	4.0
CARBON DIGKIDE	44	0.0	1.6	0.0 1.0	1.4	8.6	1.6	1.6	1.1	0.0	1.6	1.6	1.1	· 1. f	1.1	1. 1	1.0	9.6	. 4.0
CORN STEEP LICKION	***	791.4		4.0	1.4	0.0	0.4	1.6	1.1	1.1	0.0	1.1	0.0	0. 5	8. 6	1. 1	4.4	8.6.	5.0
MUTRIENTS		8.6		6.0		133.6	***	925.2	****	925.2		925, 2		925. <b>2</b>	~	925.2		925.2	· ·
CELLULASE	***	1456, 7		1.6		4.4		1.1		1.6		0. 2		9.4		0.0	~ •~	8. 6	
CELLONIASE		85.0	***	6.1		245.2		1701.9	~~~	1701.9		1701.9		1761.9		1781.9		1741.9	
EXTRACTIVES	4	2589. f		1.1		13.7		94.7		98. 7		98.7		98.7		98.7		98.7	
SOLUBLE LIGHTN		1701.6		1.0		436.7		3825. 8		3025. 8		1025.0		1025. B		3025.4		3025.0	
FUSEL OIL		9.8		6.6		287.2		1988.2	****	1988.2	. :	1988. 2		1988.2		1988.2		1988.2	•
FURFURAL	96	137.8	1,4	0.0	0.0	1.4		1.0		0.6		0. F		1.1	· ••••	4.4		0. O;	
HAF	126	55.7	0,4	1.6	6. E	8. 0 9. 6	0.0	123. 1	1.3	104.5	. 1.1	8 <b>9</b> . L	6. B	56.7	0.6	32.₽	4.3	0.0	0.0
TYLUSE	154	15237.3	101.6	1.6	1.6	257 <b>6</b> , 5	€.1 17.1	65. 5 17697. 8	0.5	65. 5	4.5	65. 5	<b>#.</b> 5	65.5	0.5	65.5	0.5	65.5	0.5
GL UCOSE	188	13271.1	240.4	0.0	0.4	7571.8			118.7	17847. 6	118.7	17667.4	116.7	178 <b>9</b> 7. <b>8</b>	118.7	17807.8	116.7	17687.8	116.7
SULFURIC ACID	98	7.5	<b>5.</b> (	0.0	0.0	1.0	42. I	50043, 1	262.5	50013.1	262.5	50843.1	262.5	58843, 1	282.5	50843, (	202.5	50043.1	282.5
SUBTUM HYDRUKTOE	40	0.6	. 0.0	1.6	6.6		1.4	8. A 8. B	8.1	B. B	<b>0.</b> [	8.6	0.1	ā. 6	1.1	6. 6	0.1	8.6	<b>₽.</b> i
COLCIUM HYDRATIDE	74	0.6	1.1	4.4	1.#	0.0	1.4	1.1	1.5	0.0	. 6.0	8.8	1.0	8.6	1.1	₹. ₽	1.8	1.1	1.0
BENZENE	78	1.0	0.0	9.8	1.1	0.6	1. 2	1.4	0.6	0, 6	4.6	6.6	1.1	1.4	0,0	4.6	9.4	6. 8	0.0
OLYCEROL	92	1.4	1.1	0.6	1.1	6.6	1.6	1.1	0.0	6.6 6.6	1.4	1.4	1.6	0, 0	8, 8	0.0	0.0	0. 0	0.0
THANCE.	46	0.6	6.4	1.4	8. 6	8.0	1.6	0.6	1.1	0. C	0. 0	1.1	4.0	4.0	8. 8	0.0	4.6	€, €	6.4
SODIUM CARBONATE	186	8.6	4.4	378. L	3, 6	1.6	0.0	0. 0	4. F	6.4	6.0	0.0	0.0	1.1	1. 6	9. 0	6. ø	8. 8	1.1
SODIUM SULFATE	112	1321.9	9.3	1.0	4.6	. 222, 8	1.6	1511.5	10.9	1544.6	<b>0.0</b>	0.0	0.6	0.0	1.5	ŷ, ø	0. <del>ŭ</del>	1.4	0.0
CODIUM ACETATE	SA	51.8	0.6	0.0	0.1	0.6	4.1	64.6	0.7		18.9	1544.6	10.9	1544.6	18.9	1541, 6	10,9	1544.6	10.9
CALCIUM ACETATE	158	1312.1	4.3	4.6	0.6	226.6	1.4	1532.9	9, 7	60. 6 1532, 9	1.7	60.6	0.7	60.6	1.7	60.6	8.7	60. 6	0.7
CETIC ACID	68	4.6	0.6	6.0	0.0	1. 1	0.0	1335.3	2. / 0. 6	1336.9	3.7	1532, 9	9.7	1532.9	9.7	15.12.9	9.7	1532.9	9.7
ETHUNOL	32	0.9	4.6	0.6	4. 0	4.4	6.8				0.0	8.6	4.0	4.0	1.1	0. 0	9, 8	6.6	8.0
CETALDENYDE	44	0, 6	0.0	1.1	8.8	1.0	6.0	9. B 9. 8	1. # 1. #	9.9	6. 0	0.0	0.0	0.0	1.6	0. 6	€. €	8.8	1.0
				****	417			V. E	1.0	9, 9	6.0	0.0	1.1	9.0	9.0	0.0	, <b>(). ()</b>	1.1	0.0
OTAL.		632312.4		378.1		83199.4		676910.5		628754.2	-	569316.6		511093.1	-	450997.9		371675.2	
EMPERATURE, C		18.8		35, €		38.0		135.		101 #					13.5				
RESSURE, AIM	_	3.5		1.0		3.1				127. 6		116.0		107.0		88. €	•	J <b>a.</b> 4	
	5	414		1+▼		3.1		3. i		2.5		1.9		1.1		0.64		0, 955	

STREAM NO.			it <b>0</b>		51 6		118		HJE TO THE		116	5	i15		516		517		ie.
DIGHETTE 450	NUL ECIL AR	LBS	NOLES	LDS	MOLES	LBS	MOLES	LPS	MOLES	LOS	MGLEG	LDS	MOLES	LBS	MILES	1.05	MOLES	UKS	
FILE SECS	MEIGHTS	HAR	H#	HA			HII	HR	HIE	HR	HA	H <b>A</b>	н	HA	н	HØ	HI	HR	HALES
NA (EJ)	18	J8586. 6	2143.7	48137, 4	2674. 3	59383, 0	3299. 1	54230, 2	3235.4	5 <b>86</b> 76, 6	1771 1	70004							
CELLUL DSE	165	0.0	8.0	4.0	0.0	1.4	0.0	1.1	1.1	9. 8	3337.3	79290. V	4485, 8	261106. 1	14689.4	218891.8	12160.6	664.4	36. 9
INSOLUBLE LIGHTN	31 3 4 <del>-</del>	L. C.		1.1		4.0	A 41 113	0.6				4.4		0.0 0.0	8.0	9.6	9.4	6.4	j. (. 1)
ASH		1.1		1.1		1.6		6.0	1 1 1 - 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	0.0	***	8.8	ing and the second	1.1		0.0 0.0		34.2	er in terminal
NYCELL IUM Othen insolubles		0.8	· · · · · · · · · · · · · · · · · · ·	6.0	15.75 <b>***</b>	1.6		1,1		0.0		1.1	'	6.0		1.1		1.1	
YEAST		4.6		1.1		1.0		8.0		6.4		1.1		0.0		0.8		28.5	
CALCIUM SULFAIE	136	1.0 1.0		4.6		4.6		4.6		8.0		6.6		1.1		0.8		1.1	
HEME. IYLAN	132	6.6	6.0 6.0	6.0 8.0	1.1 1.4	1.1	1.0	1.1	1.1	6.6	1.1	6.0	0.0	8.4	6.0	459.2	1.6	516.4	3.0
HEMI. BLUCAN	165	0.1	0.8	6.8	6.0	1.0	4.0	0.0	LI	10	1.1	4.1	0.0	1.6	0.6	8.6	1.1	0.0	
CALCIUM CARBOHATE	198	4.6	4.0	6.6	8.0	4.6	0.0 0.0	4.0	0.0	1.1	6.0	4.0	1.0	9.6	6.0	0.0	1.0	9.6	8.0
OXYGEN	32	0.0	0.0	1.1	6.0	0.6		6.0	1.1	0.0	1.0	0.0	1.0	1.4	1.1	4.6	0.0	0.0	1.0
NITROGEN	25	0.0	0.0	8.0	8.0	1.6	4.0	Le	4.4	1.1	8.8 8.4			1.0	, A. 440	1.0	1.1	0.0	1.8
CARBON DIGITOE	44	1.1	A.D.	8.0	1.1	4.0	4.0	0.6	4.0	0.0	0.1	8.8	6.8	1.1	LI	0.6	1.1	0.0	8.0
COM STEEP LIGHUR		1.1		1.0		4.0		0.0		6.0		6.6			rt	0.0	9.8	6.6	8.0
MULATENTA	1. 1 1. 1 <del>1. 1</del> 1.	1.1		1.6		6.6	: : : <del></del> :	1.1		1.0		1.0		1.1 1.1		789. 4		2.0	
CELLULASE CELLODIASE	***	1.0		1.1		6.0		LI		8.0		0.0		1.1		4. 6 1451. 8		0.6	÷ ====
ENTRACTIVES		0.0		9. 9		1.0		8.0		0.0		0.0		0.0		H. 6		1.0	
SOLUBLE LIGHTH		0.0		4.0		0.0		1.5		0.0		1.6		6.0		2581.2		1.4	
FLISEL OIL		0.1 0.6		6.6		0.0	<del></del>	1.1		1.1		1.1		0.6		1696.1			
FURFURAL	**	14.7	8.2	18.6		1.0		1.1	-	8.0		0.6		1.1		1.1		0.0	
186	156	6.6	1.0	8.8	0.2	24.4	1.3	23.4	9.8	24.4	0.3	12.2	6.3	105.5	1.1	1.1	0.1	0.0	0.0
XYLOSE	158	0.0	0.1	6.6	8. O	0.6 0.0	1.1	LI		1.6	in the	1.1	3 to 6.00	6.6	1.6	55.7	1,4	1.6	0.6
GLUCOSE	150	1.1	0.0	6.8	1.1	1.0	1.1	1.1	1.1	1.1	1.1	0.0	1.5	1.0		15193.3	101.3	44.6	0.3
BLE FURIC ACID	94	0.1	6.0	1.1	1.1	0.0	0.0	6.0 6.6		4.0	6.0	4.0		1.1	4.4	43139.4	219.7	131.9	6.7
SODIUM HYDROLIDE	40	0.8	8.6	6.0	1.0	0.0	i i		6.0 6.0	0.6	1.6	6.0	0.0	1.1		7.4	6.1	1.1	0.0
CALCIEM HYDROTIDE	76	0.6	1.0	8.6	0.0	6.0	1.1	0.0	8.0	0.0	1.0	1.1	6.6	4.0	0.0	0.4	1.1	0.6	8.8
BENZENE	74	1.1	LO	0.0	0.0	6.0	4.4	1.4	4.0	0.6	0.5 8.0	• • •			6.0	0.0	1.1	1.1	0.0
EL YCE AOL	92	1.1	0.8	1.1	1.1	1.1	0.0	6.6	0.0	0.0	1.1	0.0 0.0	0.0	0.0	1.1	1.1	6.6	1.6	1.1
ETHANDE		1.0	6.0	1.0	0.0	4.0	6.0	0.6	1.1	0.1	1.1	1.1	8.0 8.6	1.6	4.0	1.1	1.1	1.1	0.0
SODIUM CARBONATE	166	0.0	1.6	8.8	1.1	6.6	1.1	1.6	1.1	0.0	1.1	1.1	0.8	1.6	1.0	0.6	0.6	<b>J.</b> 0	6.6
SOOTUM SULFATE	145	LO	1.0	1.1	4.6	1.1	1.0	1.0	0.4	0.0	8.6	0.4	0.0	1.1	0.4	0.0	4.9	6.0	0.0
SOBIUM ACETATE	12	1.1	8.0	1.1	6.6	4.1	0.0	1.1	0.0	0.0	4.0	0.0	0.0	9.4	4.4	131 <b>0.0</b> 51.0	9. 3 0. 6	3.9	
CALCIUM ACETATE ACETIC ACID	158		6.0	6.0	8.6	l.t	1.6	1.1	1.4	1.1	0.0	0.0	0.0	6.0		1304.2	6.3	1.9	9.4
METHANOL	6 <b>6</b>	0.0	1.1	1.1	1.1	8.0	1.1	1.1	8.0	1.1	1.1	0.0	6.1	1.1	0.4	8. E	8.0	1.7 0.0	1.0
ACETAL DEHYDE	32 11	0.0	1.0	6.6	4.6	1.1	1.0	1.1	6.0	9.0	4.0	8.6	0.0	1.0	8.6	0.1	0.0	6.4	0.0
The second second second	77	1.1	4.4	0.0	6.6	1.1	4.6	8.6	4.6	4.4	1.1	8.6	4.0	0.6	6.6	0.0	0.4	4.4	8.6
TOTAL		38601.1		48156. J		59407.5		58253.6	**	60095. 3	-	79322.6	<del>-</del> •	264514.0		20/027.0		******	
TEMPERATURE, C			ANNE													-u ( PL ( , )		1146.9	
PRESSURE, ATM		135.6		151.4		114.0	ALC: NO	107.0		85. 0		34.0		86.6	23 72	38. 6		38. 0	
THE SWITTE   MIN		1.1		2.5		1.9		1.3		8.66		0.065		6.66		1.8		1. 6	

	STREAM NO.		51	1 <b>9</b> - 1 - 2 - 1	5	20	:	251		525	56	23		124
	DISKETTE 45# FILE SEC5	MOLECULAR NE LOUIS	lbs Hii	MOLES	LBS HA	NOLES HR	105 IM	MOLES HR	LBS HR	MOLES HR	L DS HR	MOLES IIR	LBS HA	MOLES HR
					:									
	MATER	18	536.4	29. 8	128. 🕯	7.1	219427.4	12190. 4	219466.0	12186.9	27.4	1,5	313698.6	19694.4
	CELLULUSE	162	1.6		7.8	. 1	1.0	.8	0.0	0. 6	1.0		4.4	0.0
	INSOLUBLE LIGHIN		1. #		33.2		1.0		0.0	****	1.8		1.1	
	ASII		0.6		1.1		6.0		0. 0		6.6		4.6	
4.4	MYCELL LLM		0. 8		0. 6		1.4		8.0		0.0		1. 0	
÷	OTHER INSULIBLES		1.0		19.5	-	1.0		Ð. <b>Ú</b>		1.0		8. ♦	~~~ .
	YEAST		6. 8		<b>4.</b> #		1.6		4.4		8.8		0.0	***
	CALCIUM SULFATE	136	26.4	9.2	494.5	3. 6	0.0	0.0	Ð. Ò	9. 0	0.0	1.0	1.6	1.0
	HENT, ILYLAN	132	6.4	0.0	8. 6	4.6	6. 6	8.4	₽. @	4.4	8. 0	1.4	<b>1. t</b>	0. 0
	HEMI, BLUCAN	531	0.0	0. 0	4.4	8.8	1.1	0.0	4. 6	4.4	0. E	1.1	0.0	0.0
100	CALCIUM CARBONATE	166	4.0	0.8	8. 8	4.8	356. 6	3. 6	28, 3	4.3	328.3	1. 1	Q. G	0. 0
	OXYGEN	32	0.0	0.0	1.1	0.0	1.1	0. 0	9. 0	8.0	0.0	6.0	. 6,0	0.0 ,
	NI I ROGEN	26	1.8	0.0	4.0	8.6	1.1	0.0	0. 0	4.6	0.6	1.0	4.0	0. Ú
	CARBON DIOIRDE	44	1. t	. 0.0	0.0	8.0	0. ť	1.1	€. \$	0.0	1.6	8.6	8.6	0.5
:	CORN STEEP LIQUIN		2. 🛊		6. B		791.4		791.4		1.1		1.1	
	MUTAJENIS		0.0		0.0		1.1		9. 0		0.0		1.4	
	CELLULASE	. ***	3.9		1.6		1455.7		1455.7		6.0		0.1	
	CELLOBIASE		1.0		0.0		85. 0		85. 🛡		1.1		4.0	
100	EKTRACTIVES		6. 6		1. 6		2588. 1		25 <b>8</b> 7. (		1. 0		. 0.6	
	SOLUBLE LIGHTN		3.9		1.0		1700.0		1796.0		1. (		8. 6	
	FISCL OIL		6. 6		6.8		4. 0		9.6		6. 5		1.6	
	FURFURIAL.	96	9.0	0. 6	0.0	4.4	0, \$	0. <b>(</b>	8. 8	0.4	1.6	1.6	137.6	1,5
	HMF	126	6. 6	1.0	0.0	4.4	55. 7	0.4	55. <u>?</u>	1.4	0. E	4. 6	1. 6	<b>2.0</b>
100	AYLOSE :	156	35.2	9.2	8. 6	<b>₹.</b> I.	15228.5	191.5	15226. 5	101.5	2. 👣	, į	8. 0	Ð. Ø
	GLUCTISE	160	196.5	4.6	25. 4	1. i	13215, 9	240. 3	43240.1	240.2	5. 9	.#	1.4	4.0
	SULFURIC ACID	98	1.1	1.1	1.1	1.1	7.4	0.1	7.6	0. (	9. û	0.0	8.4	a. e
	SOUTUM HYDROXIDE	40	6. 0	0.0	8. 0	8.0	1.4	<b>0.</b> E	0. 0	0, <b>E</b>	0. 6	1.1	6.0	4.0
. :	CALCIUM HYDROXIDE	74	9. 0	0.0	6.6	₽. <b>Ē</b>	1.\$	1.1	0. 0	O. Ú	0.0	1.1	0.0	0. 0
	BENZENE	78	9. 0	8.8	1.4	8.6	1. F	6.0	6.6	0. 0	0.6	9. 0	8. 8	8.#
- F	GLY LERDL	25	1.1	0, 0	4. 6	1.1	9.0	1.0	0.6	0.0	1.1	0.0	0.0	<b>1. 2</b>
	e (nandi.	46	8. 6	0.0	6. 6	l. f	€, €	1.1	0.6	6. 0	1. (	1.1	1.1	1.1
	SODIUM CARDONATE	106	1.1	0. \$	1.0	1.1	4.4	1.0	0.0	1.1	1.6	1.1	0.0	1.1
100	SODEUM SIX FATE	115	2.9		l. 🗓		1828.0	12.9	1828. 6	12.9	1.1	1.6	1.1	1.6
	SODIUM ACETATE	82	0.4	0. 0	0.0	1.8	51.8	1.6	51.6	0.6	1.6	9.8	1.1	0.0
	CALCIUM ACETATE	15&	2.9		1. 🛊		1111.1	4.3	1311.1	6.3	1.0	4.4	1.0	0.0
	ACETTO ACTO	60	1.6	0, 0	1.4	0. 6	4.5	6.6	6. 6	0.0	1.1	4.6	1.6	0.0
	HE THENOL	32	1.9	0.0	1.6	6.6	4.6	1.1	1.1	0.0	1.4	1.6	1.1	8.6
	ACE FALDEHYDE	- 44	0. 6	0, 0	8.8	0.6	4.4	8.6	0.0	1.8	9. 0	1.5	0.6	9.0
	101AL 1 1 1 1 1 1 1 1	- 	730.6	•	710.1		268135.9		287768.5		361.4		343036.6	
	TEMPERATURE, C		30.0		38.0		J&. 8	4,4,4,3	38.0		36. 0		62.0	
	PRESSURE, AIM		1.0		1.0	•	1.0	1.5	1.5		1.0		1.0	
	· · · · · · · · · · · · · · · · · · ·													

ISKETTE 450	MOLECULINA	LBS	MOLES	LBS	MOLEO	LDS	MOLES	LOS	MOLES	LOS	MOLES	LDS	MOLES	BC:	ww.co	1.00			
LE SECE	ME IGHTS	Hå	HØ	Hel	HA	Hill	HR	Hif	H	Hy	HQ WATER	198	Hill	LBS	MALES	LBS HE	MOLES	LBS HB	MOLES Hæ
en .	18	219100.0	12168. 9	£2521. T	1251.5	241927.7	13440.4	0.0	0.0	12164.5	676. €	253606.7	14649, 3	489.5	27.2	22076.4	1226, 1	12.2	
LUL OSC	162	4.4	5.0	0.0	9.0	6.4	1.0	0.0	0.0	8.4	0.0	8.0	R.O	8.0	1.1	6.6	12.0.1	JE. E	1.8
SOLUBLE LIGHTH	64 feb <b>5</b> 73	8.6	***	. <b>1.0</b>		0.0		0.0		1.6		6.1		1.0		1.6		1.5	
		8.6		0.0		6.6		0.6		0.0		6.0		1.0		0.0		6.6	
ELLIM		1.4		0.0		1.0		8.8		1.1		1.1		1. 6		0.0		4.4	•••
ER INSULUBLES	**************************************	8.0	• •	1.1		1.6		1.1	:	1.0		1.0		0.0		8.0		1.4	
ST .		1.0		0.0		4.0		1.1		2396.6		2765. 4		0.1		8.0		0.0	
EILM SULFATE	136	8.0	6.6	1.1	1.1	1.1	0.0	1.1	0.0	1.1	0.0	0.0	0.0	0.0	1.1	0.0	1.0	0.6	8.6
11. IYLAN	132	1.6	0.0	0.6	4.0	0.0	1.6	\$. <b>\$</b> .	1.1	0.0	6.6	9.8	4.4	6.0	1.1	4.1	1.0	1.1	0.0
N1. GLUCAN CHUN CARBONATE	162	0.0	1.1	8.8	8.0	0.0	0.0	1.1	8.0	9.0	0.0	6.6	1.5	0.0	₽.0	1.0	4.8	0.0	1.1
LLEUM EMMBEMMIE YBEN	1 <b>00</b> 32	28. ]	4.3	1.1		24. 3	1.3	0.6	1.1	174.\$	1.7	203.2	2.0	1.1	6.0	0.8	4.0	1.0	9.8
TROGEN	3.2 28	1.0	0.6	6.6	8.0	6.6	1.0	1.1	1.1	0.0	1.4	0.0	0.0	U. U	0.0	6.0	6.0	1.0	8.0
ROOM BIGXIDE	44	1.6	6.1	<b>8.1</b>	1.0	1.1	1.1	1.1	1.0	6.6	J. 1. 1.	319.5	11.4	1.1	1.0	1.0	6.6	0.0	0.0
RN STEEP LIQUOR		791.4	4.0	6.0 0.0	6.6	<b>1.1</b>	1.1	1.1	0.0	15.6	0.6	831.4	18.9	20319. 9	162.5	0.6	0.0	20349.9	462.5
RIENIS		4.6		1.0		791.4		0.0		44.1		1.1		1.6		1.1	• • •	1.1	
LULASE	.4.777 ( <u>* 1.1</u> 2.1	1455. F		6.0		0. 0 1455, 7		0.0		1.4		0.0		1.1		4.0		1.1	
LONIASE		85.4		0.0		65.6		6.0 0.1		73.1		1529. 6		4.1	***	1.1	•	0.0	
NACTIVES		2587.1		0.0		2587.1		1.1		4.9	<del></del> -	89.9	-	6.6		- 6.6		₽.#	· · · · · ·
UBLE LIGHTH		1704.4				1700.0		6.6		136.9 86.0		2710.0	***	1.6	***	1.1		1.1	:: . <del></del>
EL OIL		0.0		1.0		4.6		1.1		2. B		1786. B		1.1		1.0		1.1	3 : <b>***</b> :
Fural	96	0.0	0.0	1.6	0.4	4.4	6.0	6.6	1.6	1.1		4.5		0.0		1.0		1.1	
	126	55.7	8.4	1.0	6.0	55.7	0.4	i.i	1.1	2.9	1.1	9. 0 58. 6	8.8 8.5	1.1	0.0	1.1	1.0	1.1	0.0
05£	150	15226.5	101.3	0.0	6.0	15226.5	101.5	1.1	1.1	766.9	5.1	15993.9	186.6	0.0	6.6 8.6	1. t 1. t	1.1	1.1	4.4
COSE	186	13240.1	248.2	0.0	1.6	13240.1	248.2	8.6	0.0	0.6	0.6	1.1	1.1	1.0	6.6		1.0	1.5	0.0
FURIC ACID	98	7.6	6.1	0.0	0.0	7.8	0.1	13.7	0.1	1.0	7.7	22.5	0.2	1.1	1.1	9.0 8.0	6.6	1.1	1.4
LUM HYDROILDE	(0.1	LO	0.0	8.0	1.6	8.0	6.6	6.0	1.1	8.0	1.1	1.1	8.6	1.1	1.1	1.4	6.0	0.0 0.9	1.1
CILM HYDADIIDE	74	1.1	0.0	8.0	1.1	1.1	4.0	0.5	1.1	1.1	0.0	0.0	0.6	1.1		0.6	1.1	9 4 2	3.75
lene	V <sub>1</sub> , 1, 1, 14, 1		1.0	8.0	0.0	0.5	4.0	1.1	4.4	1.0	1.1	8.6	0.6	6.6	1.6	1.1	0.0	9. <b>0</b> . 0	0. F 8. E
CE ACI.	92	9.0	1.1	4.6	8.8	6.0	1.1	1.1	1.0	51.7	0.6	1130.2	12.4	6.6	8.0		0.0	1.4	0.0
₩€L	46	1.6	8.8	210.3	5. 2	240.3	5.2	1.4	8.0	1035.6	22.5	21563.9	469.2	245.2	5.3	4.6	0.0	4.9	0.1
IUM CARBONATE	106	8.6	1.1	1.1	1.1	0.0	1.0	0.0	6.4	0.0	6.0	0.0	4.1	0.0	0.1	0.6	0.0	0.0	0.0
IUM SULFATE	145	1826.0	12.9	4.4	1.4	1626. 6	12. 9	1.6	1.6	95.7	0.7	1923.7	13.5	1.1	0.0	1.0	1.1	0.6	1.4
ILM ACETATE	85	51.8	0.6	0.1	1.1	51.0	0.6	4.4	1.1	2.9	0.0	54.7	8.0	1.1	0.6	1.0	0.1	4.0	0.0
ILM ACETATE	156	1311.1	4.3	9.0	6.0	1311.1	4.1	1.6	1.4	66. 1	8.4	1177.6	4.7	8.6	1.1	6.0	0.0	1.1	6.0
IC ACID	£0	5.0	1.6	8.8	0.0	1.6	6.0	1.1	Ø. B	0.0	0.0	6.0	6.0	0.6	6.0	6.0	6.0	0.0	
IANOL	75	1.6	1.0	0.0	0.6	6.0	1.1	0.0	1.4	4.6	1.1	1.1	1.1	1.1	0.0	1.0	6.6	1.4	0.6
IAL DEHYDE		6.0	8.8	6.0	0.0	0.0	0.0	0.0	4.4	26.4	8.6	544.2	12.4	0.0	0.0	1.1	6.6	1.1	8.6
		287766.5		22765.0		110536.5		13.7	y de la S	17145.4		J06610. 9		21 <b>884.</b> 6		22070.4		24387.1	
PERATURE, C		38.0		jø. e		30.0		27.		30.6		30.0	NAM:						
		~~~	4 4 5 5 4 5 6	44.4		38. B		£ ( )		10 to		48 8	and the second of the second	37.0		30.4		34.4	

### ETHANOL VIA ENZYME HYDRALYSES - SECTION 600; FERMENTATION

	. 19 Aug 84								ELIMNOT AL	ENZYNE HYD	AULYS (5	SECTION 600;	FERMENTATI	LON			
					•											•	
	EAN NO.	200		6	10		61E		LL2	. 6	13	6	l <b>4</b>	6	15		616
	METTE ASO E SECG		ECULAR 1691s	LBS IIR	MOLES HAR	LBS HR	MOLES HA	L BS	MOLES HÆ	LBS	MOLES 接	LDS HA	MOLES HR	LBS	MOLES IIR -	LBS HA	MOLES
													11/4			na 	718
	ITER			1343E T													
	ILI ULOSE		18	13435, 7	745.4	210171.0	° 13342.6	12160.5	676. 0	1267. 2	78.14	182.7	18.1	1084.5	68.2	241255.5	13403. E
	VSOLUBLE LIBNIN		162	0.0	. I,#	1.0	. 0.0	4.4	0. 6	Ø. <b>S</b>	1. 1	€. #	4.4	0. 0	1.4	1.1	0.0
	SII			1.4	. ~~~	8.0	:	0. 6		9. 6		1. f	***	4.4		0.0	***
	YCELLIUM			6.6		0.0		1.1	*	₩. ₩		0.0		₽. ₩		1.1	~~~
	ITHER INSOLUBLES		~	4.6		1.0		0. 6		0. 0		8.0		0.0	-4-	4.4	
	(EAST			1.1		0.6		- <b>0</b> . 0	-	9. 0		4.6		0.0	·	4.4	
	Calcium Sulfate			2645.7		139.7		2396.6		249. 1		249.1		8. 6		139.1	
	HENT, KYLAN		136 132	1.0	0.0	1.1	8.0	1.6	0.0	8. 6	0.6	4. 6	0. 0	0. 0	8.0	0.0	0.0
	HEMI. GLUCTN		162	6.0	0.0	4.6	4.6	1.4	t. 4	0. 6	0. 6	<b>0.</b> ₽	8. 8	9. 0	9.4	8.0	8. 6
	CALCIUM CARBUNATE			0.0	0.0	4.6	0.0	1.1	₽, ₽	1.6	0.4	0. 6	8.4	6.6	0.0	8.4	0.0
	OXYGEN		1 <b>00</b> 32	193. 4	1.9	9.6	A. L	174.9	1.7	18.6	1.2	18.6	4.2	6. 6	8.0	9. 8	4.1
	NETROGEN			0,8	1.4	1.1	0.0	Đ. Đ	1.1	1.1	4. 1	4. €	4.4	0. 0	9.6	8.6	8.8
			58	1.0	4.0	0.0	0.0	6.8	1. 6	8.6	4. 🛊	8.0	1.1	8.6	0.8	1.1	1.1
	CARBON DIGITIDE		44	16.6	0. 4	302.9	6.9	15.6	0.4	1.6	. 6	0. G	0.0	1.1	.0	383.8	6.9
	COAN SIEEP LIQUOR NUIRIENTS			44.0		767.5		40. (		3. 9	H =	1.0		2.9		790.4	
				0.0		6. 6		0.0		0.6		5. 0		0.0		1.1	
	CELLULASE			81.1		1447. 9		71. 3		7. 8		1.0		6. 6		1454.6	
	CELL DBIASE			4.9	*	85. 4		4.9		9. 6		1. (	***	8.6		85.€	
	EXTRACTIVES			144.6		2513.4	***	139.9		13.7		2. 🛊		11.7		2585. (	
	SOLUBLE LIGHTN			94.6	-:-	1691.2	<del></del>	86. 0		6.8.		1.0		7.8		1699. 6	
	FUSEL OIL			2.0		43. <b>#</b>		2.6		8.0		1.6		0.0		43.0	
	FURTURAL.		96	€. 6	0, 6	0. 0	8.6	9. 0	1.1	1, 8	0.4	4.4	€, €	0. 6	0.6	0.0	8.5
	INF		156	2. 9	. 0	55.7	0.5	2.9		6.6	0.0	0. 0	0.0	1.4	0.0	55.7	0.4
	TYLOSE		156	847.1	5. 6	15146. 4	101.6	766.9	5.1	60. 1	0.5	11.7	0.1	68. 4	0.5	15214.8	181.4
	6LUCOSE		180	4.0	1.0	. 0.0	0.0	8.0	1.1	€. 8	1. 9	1.1	1.0	9.0	6.6	0.0	0.0
	SULFUAIC ACID		98	1.0	.0	21.5	4.2	1.0	.6	0.6	1.1	1. 0	0.6	0.0	6.6	21.5	4.2
	SODIUM HYDROXIDE		46	0.0	0.0	8, 6	0.0	0.0	1.6	1.6	0.0	0.0	0.0	1.1	0.6	9.6	0.0
	COLCIUM HYDROXIDE	100	74	0. 6	4.4	0.0	0. 6	1.1	5.8	0.6	0.0	0.0	0.0	8.8	0.6	1.0	0.6
	BENTENE		78	4.6	9. 8	6.6	1.6	8. €	1.6	6. 6	1.0	1.4	1.1	0.0	0.6	8.8	0. Ú
	EL YCE ROL		92	60.6	0.7	1077.6	11.7	54.7	0.6	5. 9	1.0	1.#	.4	4.9	8.1	1982.5	11.8
	ETHOHOL		46	1143. f	24. 6	20149. 8	144.4	1035.6	22.5	107.5	2.3	15.6	0.3	91.4	2.0	20532.6	
	SODIUM CARBONATE		196	0.0	0.0	0.0	0.0	0.0	0.6	8.6	0.0	0.0	0.8	8.0	1.0	8033214 1.6	446.4
	SODILM SLATATE		142	105. 5	1.7	1818.2	12.4	95.7	0.7	9.6	0. t	1.4		6.6	0.1	1827.0	8. <b>€</b> 12.9
	SUDIUM NCETATE		62	2.9	1.0	51.6	6.6	2.9	0.0	6.6	6.6	1.0	9. 8	0. Ú	0.0	51.6	8.6
	CALCEIM NCETATE		156	13, 3	0.5	t 304. J	8.3	66. 4	6.4	1.0	9.6	1.0	1.4	5, 9	. 0	1310.2	e. e
	ACETTC ACTO		60-	0, 6	9. 6	Ø. Ŭ	0.0	0.0	0.0	1.1	4.6	8.4	4.4	0.0	0.0	0. E	0. S
•	ME ENGENIAL	1.1	12	1.6	ě. 6	0. 0	0. 2	0. 0	1.4	0.0	0.0	0.0	0.0	6.9	4.6	4. 0	
	ACETALDEHYDE		44	29. 3	0.7	5(4, 9	0.7	26. 4	0.6	2. 9	Ø. I	0.0	0.0	2.9	8. 1	517.6 .	0.0 . 11.8
	TOTAL		•	18928.4	. **	287682.5	·· . <del></del>	17145.4		1783. 0		185.6		1297.5		288980. 0	
1.						1 1 1 m						COLUMN ME	1.0	(Ca) and		LOD 700. B	
	TEMPERATURE, C			30.0		30. 6		30.0		73.0							
	PRESSURE, AIM					27		. J.E. G.		30. @		.14. 💗		30. €		30. 0	

SIREAN NO.		700		102			763
DISKETTE 450 FILE SECT	MOLECULAR METRITS	LBS HB	MOLES	LBS Hif	HOLES Hil	LOS Ha	MON ES Hai
WATER		32.2					
CE: LULDSE	162	1.6	1, 8 8, 6	0.8	4.0		
INSOLUBLE LIGHTH		6.4		0.0	0.0	0. E	
ASH		0.0		6.6		4.4	
NYCELLIUM		6.0		1.0		6.0	
OTHER INSOLUBLES		0.0	( ) <del>  -</del> -( )	1.1	•	6.0	
<b>YEAST</b>	· · · · · · · · · · · · · · · · · · ·	4.0		4.0		0.6	
CALCIUM GULFATE	136	6.6	6.0	1,1	8.0	4.1	1.0
HENT, LYLAN	155		LE	1.1	1.6	9.0	0.0
HEAT, GLUCAN	165	1.1	1.1	1.1	1.1	8.0	0.0
CALCIUM CARBONATE STYBEN	i <b>M</b>	1.0	1.1	4.6	4.1	0.0	
NITROGEN	32 28	0.0 0.0	0.6	0.0	8.0	1.1	
CARBON DIOXIDE	44	26349.5	1.6 462.5	6. 6 20349. 9	6.6	L.C.	8.0
CURN STEEP LIQUOR		4.0	102.3	\$. <b>6</b>	162.5	20349. 9	
MUTRIENTS		4.6		0.0		1.0 6.6	
CELLUL ASE		0.0		1.6		1.1	
CELLOBIASE		6.0		1.1		1.1	
ENTRACTIVES		0.0		6.1		0.0	
SOLUBLE LIBNIN		0.0	777	1.1		0.0	
FIISEL DIL		0.0		1.1			•••
Flanfural	96	0.0	0.0	1.6	1.6	8.8	1.0
HMF	126	4.0	6.8	1.0	1.0		4.0
TYLOSE	150	0.0	1.1	1.1	4.4		0.0
GLUCOSE SULFUATO ACTO	184 94	6.0	4.0	4.0	0.0	8.0	1.1
SOUTH HYDAILIDE	77 U	1.1	6.0	4.0	. 8.6	1.1	
CALCIUM HYDROXIDE	$\tilde{n}$	4.0	0.0 0.0	1.6	1.0	1.0	
BENTENE	78	8.6	4.0	9. 0 6. 0	1.1 L.1	6.6 6.6	5.6
FIL YEE ROL	92	0.0	6.0		1.0	4.1	1.0 1.0
ETHANOL.	46	4.9	0.1	8.8	4.4	4.4	
STIBLUM CARBONATE	196	8.0	1.1	0.0	8.4	6.6	0.0
SODIUM SULFATE	142	0.0	0.0	4.0	4.0	6.0	4.4
SUDIUM ACETATE	₽2	6.0	6.0	1.1	0.0	8.8	4.4
CALCIUM ACETATE	158	0.0	1.1	1.0	8.0	6.0	1.6
CETIC ACID	60	1.0	1.1	1.1	5.0	0.0	0.0
E IHANDE	ĸ	6.6	1.0	4.0	1.6	6.0	0.0
CETAL DEHYDE		6.6	a <b>60</b> 00	1.1	1.0	8.8	0.0
(CIA)							
OTAL		20307. [		20349.9		28.149, 9	
EMPEANTURE, C		30.0		34.0			
RESSURE, ATM		1.0		1.6		30.0	

DISMETIE 420   MALEGIAR   LES   MALES   LES   LES   LES   LES   LES   LES   MALES   LES   LE			
FILE SECS MEIGHTS IN NA	866	807	669 669
MATER 18 241255.5 13403.1 1457.2 51.0 240573.0 13301.6 7868.0 437.2 1818.4 189.6 CELLULOSE 162 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	S LBS HOLE	ES LBS MOLES	1.85 MOLES LBS MOLE
MATER  18 241255.5 13491.1 1457.2 61.0 240471.0 13181.6 7266.0 437.2 1818.4 189.5  CELLULOSE  162 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	1號 1號	TOR HA	HA HA HA HA
CELLULOSE  162 8.0 8.0 8.0 8.0 8.0 8.0 8.0 8.0 8.0 8.0			was proposed and an analysis of the same
CELLUSSE 162	6 6056.4 336.4		5
THISTURGE LIGHT			
ASII  MOCELLIM  B. C.			
MYCHILIN	***		5 <b>6.6</b> 6.6
DITER INSTAURCES			1.0 mg 1.0 mg
TRAST  CALCIUM SAFATE  136  0.8  0.0  0.0  0.0  0.0  0.0  0.0  0.	4.9	W. W	1.0 1.0
CRICIN SAFATE  136		•.•	#.# #.#
HEAL, STUCKN 152 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.			1.1
FEN. GLUCAN  162	I II		
DAYLON CARRINITE  188 9.4 0.1 0.8 0.8 0.8 0.8 0.8 0.8 0.8 0.8 0.8 0.8			6.6 0.6 8.4 A.
DAYSEN 32			1.0 1.6 1.6 1.
NITROGEN CARBON DICKIDE CARBON DICKI			9.0 0.0 0.0 0.
CARBON DIGITIDE  44 1841.8 6.3 6.6 6.6 199.4 6.6 1801.8 6.9 18.3 1.6  DIRN SIEEP LIGIDAR			0.0 0.0 0.0
TIME STEEP LIQUID 1794.4			0.0 0.6 8.6 6.
NUTRIENTS		***	200.3 4.6 233.5 5.
CELLORIOSE	***	₩1 ₩ 1	1.0 0.0
EXTRACTIVES - 2585.1 - 4.6 - 2585.1 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6 - 6.6	9.9	9.8	
EXTRACTIVES	***	9. 4	0.0 I,0
SOLIDAE ELISAIN	4.4		
FIRE OIL  FURTURAL  96  6.8  6.8  6.8  6.8  6.8  6.8  6.8	W. D.	• •	8.6 · · · · · · · · · · · · · · · · · · ·
FURTURAL 96 8.8 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	4.6	4.4	
PMF		T- W	1.0 · · · · · · · · · · · · · · · · · · ·
TYCSE 150 15214.6 101.4 0.6 0.0 13214.6 101.4 0.6 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0			ne ule elect d'
GLUCOSE  180  180  180  180  180  180  180  18	****		4.0 0.0 0.0 0.0
SULTURITY RCTB 96 21.5 0.2 0.0 0.0 21.3 0.2 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	***		9. 0 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0.
SODIUM HYDROXIDE  46	7.7		0.0 0.0 0.0 0.0
CALCIUM HYDROLIDE 76 0.8 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0		****	0.6 f.6 f.9 f.9
BENIENE 78 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.			
SETTORRIC. 92 1982.5 11.6 0.0 0.0 1882.5 11.8 0.0 0.6 0.6 0.0 0.0 0.0 0.0 0.0 0.0 0.0		717	note that the thin
ETHORICE  46 20532.6 446.4 20429.9 444.1 102.7 2.2 110321.9 2398.3 25374.6 551.6  SODILIN CURREDHATE  106 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0			B. B
SOBTIM CARBONATE  186			4.1 6.8 6.6 6.6
SODILM SALFATE  142 1885.6 13.3 0.0 0.0 1885.6 13.3 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0			73854.2: 1598.1 84947.2 1846.3
SODIUM ACETATE 82 51.8 0.6 0.0 0.6 51.8 0.0 0.0 0.6 0.0 0.0 0.0 0.0 0.0 0.0 0.0			0.0 0.0 0.0
CRECIUM ACETATE 156 1310.2 8.3 0.0 0.0 1310.2 6.3 0.0 0.0 0.0 ACETIC ACID 60 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0		****	6.0 6.6 0.0 0.0
ACETIC ACID  60 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0			8.6 0.0 0,4 0,6
METHYMEN. 12 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.			
ACETAL DELIYDE 44 517.6 LL.B 0.0 0.0 0.0 517.6 LL.B 119.2 2.7  TOTAL 283030.6 21607.1 267361.5 LL9012.3 27374.6		***	0.0 0.0 0.0 0.6
TOTAL 283030.6 21007.1 267361.5 119012.3 27374.6			9.0 6.6 6.6
Thereas are	91637.7		342.6 7.4 398.6 9.1
TEMERATURE, C 38 & 112 & 120 C	11931.1	12031, 9	78895.8 91637.7
DRECCIDE OTH 113.8 113.0	110.0	119.6	ile. 6 tja, 6
PRESSURE, ATM 3.7 3.4 3.7	1.4	3.4	3.4

DISKETTE 420	MOLECULAR	LDS	MOLES	LBS	MOJ ES	t DS	MOLES	1 16	MOLES	LDS	MCIL ES	LDS	MOLES	LDS	NOI ES	1.85	PULES,	185	POLES
FILE SECS	METOTIE	Hill	H#		HØ	HÆ	HI	HA	横	Н	Hill	HA	HI	<b>Н</b> Й	HA	HA	H <b>II</b>	168	4#
<b>K</b> ATE#		THE A	433.0																
ELLIA OSE	162	7868, 6	437.2	1.0	8.0	7667.6	426.6	611.6	354.7	4373.6	243.	3019. 6	167.7	10686.6	593.7	1354.6	75.3	1074.7	59.7
ZELULUSE Insoluble lighth	106	. 0.0 0.0	0.0	6.6	8.0	0.0	8.0	1.1	1.0	6.0		6.0	0.6	1.1	1.0	: 1 - 1 <b>0 0</b> 1	1.0	<b>(. (</b> .)	1.0
SH SIA LIBERT OF THE SHARE				0.0		6.6		1.1		1.1	****	1.1		. 0.0		0.0		1.1	14 St. 55
YCELL IUN		1.4	·	1.1	***	0.0		l. I		1.1		1.1		1.1	****	1.1		8.6	
THEO INSOLUBLES		0.0		1.0	***	0.0		1.1	: : : : : : : : : : : : : : : : : : :	1.1		. <b>1.0</b>		4.0		0.0	11 je 🕶 e	0.0	
EAST				1.1		0,0		1.4		1.0	10 ( <del>777</del> )	1.4		0.0				1.0	
URI ALCIUM SULFATE	and the state of t	8.0		LI		1.1		1.0		8.0		6.0		1.1	· · · · · · · · · · · · · · · · · · ·	1.6		6.6	
MILIUM SALPHIE EMI. KYLAN	135 176	0.0	6.0	8.6	1.1	1.1	1.0	r.	1.1	1.0	1.0	1.0	0.0	1.1	1.1	1.1	1.1	1.6	6.6
ENI. GLUCAN	162	0.0	4.0	1.0	1.0	LI	4.6	1.6	6.6	L.S	0.0	1.1	0.0	4.4	1.1	1.1	6.6	1.1	8.6
ALCIUM CAMBONATE	100	1.4	0.0	0.0	1.0	1.1	1.1	1.1	6.0	4.0	1.6	1.1	6.0	1.1	4.4	0.0	0.0	4.1	( ) B.O.
NAGEN Saftran Challeanis	32		1.6 1.1	4.6	0.0		6.6	1.1	1.1	1.1	1.1	1.1	1.1	8.0	0.1	1.1	0.0	9.6	1.1
LTROGEN	26	1.1	L.S		6.0	1.0	8.6	1.4	1.1	4.0	8.8	l.f	L.S	1.0	0.0	4.0	6.0	0.0	1.6
ARBON BIOXIDE	20 11	8.8	4.6	6.6	LI	4.0	LI	1.6	4.0	4.0	1.1	4.0	0.0	1.1	6.6	1.1	6.6	0.0	0.0
DAN STEEP LIBUOR		6.4		1.4	6.0 	6.6	6.6	1.1	1.1	1.1	4.4	1.0	0.0	4.1	6.0	1.1	0.0	0.0	8.6
LIRIENIS	- 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1	1.1				1.0	***	4.4		1.0	•••	0.0		1.1	• • • • • • • • • • • • • • • • • • • •	6.0		1.6	
LLULASE		1.0		1.1		1.1		1.1	~~~	0.0		0.0		1.1	·			1.0	
LLONIASE		1.1				1.1		0.0		6.6	· · · · · · · · · · · · · · · · · · ·	1.1		9.4		1.1		1.1	
TRACTIVES		****		1.0		1.1	1.5	6.0		6.0		1.1	**************************************	1.0		0.0		6.8	
TERE LIENIN		4.0	· · · · · · · · · · · · · · · · · · ·	0.0		6.0		6.0	€ 1778 1	1.1		1.1		A.F		9.0		0.0	
SEL DIL		4.0		0.6		0.0	· · · · · ·	1.1		1.1		1.1	****	1.1		1.6			<del></del> .
AFURAL	%	1.1	V 2000 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	41.6		1.0		r.		1.1		1.1		Li	•	1.1		8.0	***
	126	4.0	0.0	0.0	3.8	6.0	6.0	r.	1.1	1.1	1.4	1.1	<b></b>	L.F	6.0	0.0	6.0	0.0	0.0
YL OSE	150	4.0	8.8 6.8	0.6	1.6	0.0	1.1	1.0	1.1	11	1.1	F.8	4.0	1.1	1.0	1.1	1.1	0.0	1.1
LUCOSE	184	4.0		0.0	8.0	8.0	8.8	1.1	4.1	1.1	4.1	4.0	4.0	1.4	1.1	0.0	1.1	1.0	1.1
LLFLAIC ACID	98	1.1	6.6 8.6	0.6	0.0		L.F	4.0	4.0	4.0	1.1	8.0		1.0	1.1	1.1	4.1	1.1	1.1
ODIUM INVORDATIDE					4.6	1.6	L.	1.1	1.1	1.1		1.1	6.0	1.1	1.0	1.1	1.0	1.1	1.1
ALCIUM HYDAGIIDE	74	6.0	0.0	0.0	1.1	1.0	1.1	4.0		1.6	1.1	1.1	6.0	3.4	1.1	1.1	<b>0.0</b> , □	1.4	0.4
ENZENE	78	8.0	6.0	0.0	1.1	1.4	1.1	4.0	0.0		0.0	6.0	0.4	0.0		1.0	1.0	8.0	1.4
YCE BOL	35 70	L.F		1.0	0,4	76257.6	977.7	76257.6	977.7	616.2	7.9	616.2	7.9	83676. 1	1475.4	6.8	<b></b>	0.0	1.1
INDICE.		6.6	1104 7	1.8	1.1	1.1	4.0	0.0	6.0	0.6	1.1	1,1	0.6	0.0	8.0	8.0	0.6	0.0	3.0
MAMAL GATUM CARBONATE		116121.9	2391, 3	0.0	0.0	28648.4	444.7	20610.4	448.7	3643. 9	74. 3	3681.9	74. 3	24244.3	527.0	1.1	6.0	8.6	6.0
COLUM CHROUNNIE	148	1.1	1.5	0.0	0.0	1.1	8.6	1.0	1.1	1.1	1.1	1.1	1.1	1.0	0.0	6.0	0.0	. 1.1	0.0
ABILIN SALTHIE Dâilm aceiate	115	0.6	0.6	1.0	1.0	4.0	0.0	1.1	LI	1.1	1.1	0.0	1.0	0.0	1.1	6.0	1.1	6.6	1.1
ALCIUM ACETATE	86 158	4.0	9.6	0.6	0.0	4.6	1.1	8.6			4.0	4.1	8.8	8.0	1.1	L1	1.1	6.1	1.1
ELLUM HEEMIE Elic acid	130	L.	8.8	8.0	0.0	8.6	1.1	0.0	1.1	1.1	1.1	4.0	8.6	1.1	4.1	1.1	0.0	1.1	1.1
	6# 12	1.1	11	1.0	4.0	1.1	1.1	1.1	6.0	1.6	4.4	4.4	9.0	1.0	1.1	0.0	0.0	0.0	8.0
HANDL TETM DELIVE		1.6	1.1	6.0	0.0		1.1	5.0	4.0	1.0	1.6	<b>(.)</b>	1.1	0.0	1.1	4.6	1.1	0.0	4.1
CETAL DEHYDE	<b>44</b> -	1.1	• • •	0.0		0.6	0.0	0.0	4.6	6.0	1.6	0.0	1.1	6.6	0.6	0.0	6.0	0.0	0,0
OTAL		116190.6		43.0		104565. &		193211.0		<b>8</b> 593. 7		7239. 1		118869. 0		1361.4		1074.7	
EMPERATURE, C		110.6		21.0		71.0		61.0		64.0		85, 8		64.0		43.6		21.6	
RESSURE, AIN		1.4	and the second	1.0											the second second second	4.9. 6		E1.9	1 5 4 5 4 5 A

BIREAM NO.				813		<b>820</b>		151	A	22	8	<b>21</b>	. 6	24
DISKETTE F\$LE	420 SECS	MOLECULAR MEJGHTS	LBS LBS	MOLES HR	LBS HA	HOLES HA	LOS HA	MILES HR	LBS HR	MGLES 排尿	LBS HR	MOLES IIA	L BSI HA	MOLES HAR
MATER		. :	193.7		440.5			**********						*****
CELLULOSE		16	182.6	5.7	162.6	5.7	1.1	1.1	0. 6	<b>f.</b> #	9896. 3	544. <b>a</b>	9866. 3	544.8
INSOLUBLE	a leater	165	1.1	4.0	1.1	8.0	1.1	0.0	8. 0	9. <b>0</b>	9. 0	· 1. f	0. (	€. €
ASH	LIVILIA		1.1		6.6	***	6.4		l. f		. 6.0		1.6	
NACETT ION			6.0		1.1		4.8		1.4		8.0		O: O	
OTHER INSO	A HEN CE:		1.1		1. 6		8.0		6. <b>f</b>	.===	1.4		<b>8.8</b>	
YEAST	A UOS ES		1.1		1.1	***	0,0		4.6		. 6.0		9. 6	
CALCIUM SU			1.0		0.0		6. 8		£. 0		0. 0		0.0	
HEMI, TYLA		136	1.1	1.1	0.0	1.1	₽. ₽	₽. 0	4. 5	0.0	1.4	6.0	0, 0	0. D
		132	1.6	8.0	4. 6	0.0	₩. €	1. E	8. 6	0.0	4. 6	0. 0	2, 6	8. 6
HERI. BLUC		162	4.4	4.4	9. 0	8, 8	8. 6	0. Ē	0.6	8.8	€. €	0. 0	. 1, 0	0.0
CALCIUM CA	REPORTE	106	0.0	4.6	0.0	8. 6	l. f	f. f	8.0	1. 6	8.8	<b>₽. ₽</b>	1.5	0. 0
OXYGEN:		35	0.0	1.1	0. 0	0. <b>G</b>	0. 6	4. 6	6. 0	0.0	8.6	0. 6	4. 0	0.0
MITAGGEN		58	1.8	1. 8	ŧ. o	6. 8	8. 6	6. 6	1.6	8.0	8.6	0.4	0.0	0. 6
CARBON DIO		44	0, 6	4. 8	6.6	0. 0	303. 8	6.9	€. €	ē. Q	9. 0	0.0	0.0	0.6
CORN STEEP	LIGUON		0.0		8. 6		6. 6		6. 1		9. 8		0.6	
NUTRIENTS			4.4		0.0		0.0		0.0	***	6.0		0.0	
CELLUL ASE			8. 8		4.4		. 4.0		6.0		1.6		1.6	
CELLOBIASE			0. 0	^ <del></del>	1.0		8.0		0.6		0.6		1.4	
EXIRACTIVE:			1.1		6, 8		0.0		0.0		0. f		0.0	
SOLUBLE LIK	CHIN		0. i.		1.1		8.6		1.1		1.1		0.0	pr se .
FUSEL OIL			6.6		43. 🛊		1.1		6.0		1.1		1.6	
FUNFUHAL.		96	1. 1	8.8	6.6	6.8	4.4	6.6	0. E	1.1	35. 3	8.4	35.3	0.4
HE	4.0	156	1. 6	I. 6	6.6	4.0	0. E	0. Ú	1.1	6.6	1.4	0.0	0.0	0.0
TYLOSE		158	₹. €	1.1	0.0	4. \$	1.1	0.0	0. 0	8.8	1.6	1.1	6.0	4.4
BLUCOSE	١	la•	4. 0	6.8	8.0	0.4	0.0	4.4	6.0	1.6	0.0	0.0	1.0	6.0
SULFURIC AC		98	0. 0	1.1	€. €	0. 6	4. 0	8.6	6.0	9. 0	8.0	8.8	0. 6	1.4
SOCIUM HYDA		- 49	€.	4.6	9.4	1.0	1.1	8.6	9. 6	0.8	1.6	1.6	1.6	0.6
CALCIUM HYD	DROX LOE	74	6.6	0.0	0.0	4.4	0.6	0.0	0.0	1.1	1.0	1.1	1.6	0.0
BENTENE	15.00	74	Ø. @	4. 8	8.8	9. 🛊	1.1	6.0	6.8	0.1	0.0	8.6	6.0	0.0
GL Y CE ROL		92	0. 6	₿, ₿	6. 6	0.6	8.8	1.0	9. 9	6.6	0.0	0.8	0.0	0.0
ETHOME.		46	20129.9	444. 1	20129.9	144. E	1.1	1.1	1.6	0.0	8.6	8.0	1.6	
SODIUM CARB	OWATE	196	0. 0	0.4	1.1	0. 6	1.1	0.4	0.0	0.0	0.0	1.0	1.1	1.8
SCORTUM SULF	AJE	311	4. 8	0, 6	0.6	4. 6	0.6	0.6	0.0	1.0	0.0	0.0	1.1	0.0
SODIUM ACET	IAJE	82	0. 0	1.0	0. 6	0.0	0.0	0.0	0.0	8. 6	0.0	0.0	0.0	1.6
CALCIUM ACE	TATE	158	6.6	4.0	8.6	6.0	0.0	1.0	4.4	4.4	6.0	6. 0		0.6
RETTE ACTO	)	68	8.8	0.6	1.6	9.8	1.0	8.4	0.6	0.0	11.2	4.2	<b>6.6</b>	₩. ₩
E THENDE	100	32	4.0	0.0	0.0	1.4	0.0	0.0	0.0	0.0	9.3		11.2	1.2
CETAL DEHYD	E	44	0.0	9, €	1.1	0,0	517.0	11.6	0.0	0.6	1.#	0. 3 0. 0	9. 1 4. #	0. 3 0. #
TO TAK			20532.5		20575.5	· · · · ·	821.7	·	6.8		9862, 1		1.5388	
IENPERA!URE	r		61.5		12.5									
	-		91.		43,#		30. 0		30.0		148.4		140. #	
MESSUME, A	חיי		1, 0		1, 👫		1.0	100	1.0		4. 4		4.4	

SIREAM MO.			<b>X</b> I	9	₽?	98	3		904		905		986		907		986	9	999
DISKETTE 453 FILE SEC9	MOLECULAR VE1644S	LDS HR	MOLES HE	LDS	MOLES	LBS HR	MOLES	LIFS HAR	HOLES HA	LOS HAT	MOLES	t BS	MALES Ha	LDS	MOLES HR	LDS HR	MULES HA	LBS HA	MOLES HR
														********					
MATER	18	246873.6	13361.4	46131.0	2568. 0	6.6	6.6	287064.0	15944.7	292657.4	16225.4	231295.9	12849.8	60761.6	3375.6	221118.5	12284.4	18177.4	565. 4
CELLULOSE	165	8.0	8.6	6.0	1.6	8.0	1.1	1.1	1.6	0.0	6.0	6.0	6.6	0.0	0.0	1.1	0.0	4.6	0.0
INSCLUDLE LIBRAN		4.0		0.0		1.6		1.1		1.4		1.1		8.0		1.0		1.1	
ASH		0.6		1.1		0.6		1.0		6.0		1.1		0.6		0.0		0, 8	1
MYCELL TUM		1.1	: · · · · · · · · · · · · · · · · · · ·	0,0	: : : <del> :</del> :	1.0		4.0		0.0		4.6		1.5		8.6		0.0	
OTHER INSOLUBLES		0.0		L.O		1.0		0.0		4356.6		4356.6	<del></del> -	1.1		4350.6		0.0	
YEAST		139.7		4.6		4.0		139.7		139.7		139.7	***	1.6		139.7		1.6	
CALCIUM SULFATE	136	1.1	9.6	6.6	6.0	9. 0	4.0	1.1	1.1	1.5	1.1	1.6	0.0	1.1	1.6	8.0	0.0	0.1	0.6
HEMI. XYLAN	132	4.0	6.0	6.0	<b>₽.</b>	1.0	1.1	1.1	1.6	8.8	1.5	1.1	1.6	1.1	0.0	0.0	6.0	1.1	8.6
HENT, BLUCAN	165	6.8	6.0	1.1	8.6	6.0	6.6	8.1	0.8	0.0	0.0	6.6	0.0	8,0	8.8	0.0	8.8	0.0	4.4
COLCIUM CARBONATE		9.6	6.1	LI	1.1	1.6	1.1	9.1	1.1	9. 8	0.1	9.8	1.1	6.0	0.0	9. 8	6.1	9.6	1.1
OXYGEN	32	0.0	6.1	1.4	1.1	8.6	1.1	L1	0.1	8.6	1.1	1.1	1.1	1.1	1.1	0.0	1.0	1.0	1.1
NITROGEN	28	1.1	6.0	F.0	1.0	6.6	0.0	4.0	4.6	0.0	1.0	0.0	1.1	1.1	6.0	1.0	0.0	0.0	0.0
CAABON DIOXIDE	**	1.0	1.1	0.0	1.1	8.0	1.1	1.1	4.0	0.0	1.1	0.0	1.1	1.1	0.0	0.0	6.6	1.0	1.1
CORN STEEP LEALING	•	796.4		0.6		0.8		190.4		790.4		796.4		1.1		790.4		0.6	
MUIALENTS	<del></del> -	1.1		4.0		1.1		4.0		6.6		1.0		0.0		9.0		1.0	
CELLUL ASE		1454.4		0.6		8.0		1454. 4		1454.4		1454.8		. 1.0		1454.4		1.0	
CELLOBIASE		85.0		0.6		0.0		85.0		85.0		85.0		1.6		85.0		1.6	3 1 112
EXTRACTIVES		2585. 1	****	0.0		1.0		2585.1	<del></del>	2585. L		2585.1	•	1. 6.6		2585. j		1.1	
SOLUBLE LIGHTN		1699.4		0.0		0.6		1699. 6		1699.6		1699. 6		1.0		1699.0		0.0	
FUSEL OIL		L.O	•••	6.6		1.1		1.6		1.1		1.1		1.1		1.1		1.4	
FURFURAL	96	8.8	1.0	8.0	6.0	4.0	6.6	1.1	1.0	4633.1	48.3	1604.6	16.7	3428.5	31.5	1159.1	12.1	445.5	4.6
l <b>ef</b>	156	55.7	8.4	1.6	1.1	1.5	6.6	55.7	4.4	55.1	1.4	55.7	1.4	1.1	0.0	55.1	9.4	0.1	1.6
I'M OSE	150	15214.8	101.4	1.1	1.1	1.0	4.0	15214.4	181.4	107.7	7.9	1177.7	7.9	1.6	1.0	1177.7	7.9	0.0	1.1
<b>GLIJI'USE</b>	168	<b>(4.0</b> )		1. <b>1. (</b>	L1	6.0	1.1	4.4	1.0	1.0	1.1	1.0	0.0	1.1	1.1	6.0		0.6	1.1
SULFUAIC ACID	98	21.5	6.8	1.1	1.1	3111.7	31.6	3133.2	12.6	3133.2	32.0	3133.2	12.0	1.1	8.0	3133.2	32.6	6.4	0.0
SOOTUM HYDROTTOE	•	7 - A 4 E	1.6	0.0	1.0	4.0	0.0	1.1	(i 60)	1.1	1.1	1.0	6.0	1.0	0.4	8.0	0.1	6.0	1.1
CALCEUM HYDAOXIDE	74	4.0		8.0	8.0	1.0	9.6	4.0	1.6	6.0	6.0	0.0	1.1	1.0	8.8	1.1	1.1	0.0	0.0
BENJEHE	74	( ) ( <b>)</b> ( )	L.C	1.1	1.0	0.0	1.6	2.6	LI	1.1	1.1	1.0	1.1	1.1	0.0	1.6	8.0	0.6	9.4
&L YCE ROL	92	1002.5	11.0	6.6	8.4	0.0	1.1	1082.5	11.4	1082.5	11.6	1062.5	11.0	1.1	9.0	1882.5	11.0	4.4	0.0
ETHANOL	46	192.7	2.2	6.4	6.6	0.0	1.1	102.7	3.8	102.7	3.5	24.4	8.5	73.1	1.7	6.6	6.6	24.4	0.5
SODIUM CARBONATE	196	1.0	9.6	1.6	1.1	0.8	1.1	1.1	8.6	1.1	1.1	1.1	0.0	1.1	0.6	1.1	0.0	0.6	1.1
SOOTUM SULFATE	142	1827.6	12.9	0.8	0.0	1.1	1.1	1827.0	12.9	1827.4	12.9	1827.6	12.9	8.6	4.1	1827.	12.3	1.1	0.0
SOULM ACETATE	95	51.6	1.1	8.8	0.6	0.0	1.1	51.4	1.1	51.8	1.1	51.8	0.0	0.0	0.0	51.8	0.0	8.0	0.6
CALCIUM ACETATE	158	1310.2	1.3	1.6	1.1		1.0	1316.2	1.3	1310.2	4.1	1310.2	1.3	1.0	0.6	1316.2	8.1	6.0	0.6
ACETIC ACID	60	1.1	1.1	6.0	1.0	6.6	8.6	0.0	1.1	1.1	0.0	6.0	1.1	6.0	6.4	1.1	1.0	0.0	0.0
METHANIA	12	1.6	4.0	1.1	0.0	1.6	1.1	0.0	1.6	8.6	1.1	0.0	0.6	1.1	0.0	1.1	0.0	0.0	0.0
ACE TAL DEHYDE	**	1.1	8.8	0.0	0.1	1.1	1.1	9.4	0.0	6.0	0.0	1.4	4.0	0.0	1.1	9.0	0.0	0.0	0.0
TOTAL		267302.9		46131.6		3111.7		316545, 7		316545.7		252677.1		63868. 3		242430.0		10647. 1	
TENPERATURE, C		138.6		253.6		21.0		220.8		220, 8		120. 0		150.0		100.0		100.0	
PRESSURE, AIM		3.7		10.6		23.4		23.4		23.4	2352	2.		2.0		1.0		177.0	

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DA!E 19-Aug-84						ETH	ANCIL VEA ENI	EYNE HYDROLYS	T <b>B ≥</b> €Cì	ION 980: FUR	FURAL PROD	<b>UCTION</b>	·				•		
STREAM NO.		91	•		911		912	91	3	9	16		15		916	9	u U	9	116
DISKETTE 453 FILE SECT	MOLECULAR NETGHTS	LBS HR	MOLES	195 188	MOLES Hir	L BS Hil	MOLES HR	t 8S I∰	MOLES HA	i BS IIR	NOLES HR	L BS HR	MULES HR	1 DS HR	#OLES HÆ	L BS	MOLES HR	LBS HR	MOLI HR
							*******			********									
MAJER	18	2366.3	171 €			000143.8	10110 6	A#11 f .		44784	100 0							1	
CELLULOSE	165	6. f	131.5 0.6	0.0 6.6	4.4	222111.0	12339.6	2523.6	140.2	12701. 6	785. 6	31983, 9	1777.2	190123.1	10562.4	27132.3	1507.3	4857.6	269.
INSOLUBLE LIGHTA	100	6.6		1.1	8.0	0.0 0.6	1.1	6.6 4.6	0.0	0.0	6.0	8.6	1.0	0.#	0.0	1.1	1,6	0.6	6.
ASI		1.1		0.4		1.1				1.6		1.1		1.0		1.1	· .	1.1	
MYCELLIUM	***	1.4		9.4				6.6		6.6		6.0	+	1.0		1.1		6.0	
OTHER INSOLUBLES		6.6		4.0		0. 0 4350, 6		6.6 6.6		1.0		1.1		1.6		1.1		1.8	
YERST		1.0		4. t						1.1		4354.6		4.0	• •	217.9		4132.7	
CALCIUM SULFATE	136					139.7	70.0	6.1		8.0		139.7		0.0		6.0		132.9	
HENT, TYLAN	132	0.0 0.0	1.6	1.6	1.0	4347.7	32.0	1.1	6.6	1.1	1.1	4039.9	29.7	307.6	2.3	24J. J	1.6	3796.6	27.
HERE, GLUCAN	195	0.0	4. 0	1.6	1.0	0.0	1.1	0.0	4.6	6.0	0.0	0.8	8.6	0.0	1.1	0.6	8.4	0.0	₫.
CALCIUM CARBONATE			0.0	0.6	0.0	8.8	1.6	1.1	6.6	1.6	0.0	0.0	6.6	1.1	1.1	1.1	6.0	9.0	6.
DIYGEN	1 <b>81</b> 32	0.6	0.0	1.6	1.6	9.8	0. 8	1.1	1.1	1.1	1.1	9. 8	0.1	1.1	1.1	8. 8	8.8	9.6	4.
NI 1ROGEN		1.1	0.0	0.8	4.0	4.4	1.6	0.6	0.6	0.0	0, 0	9.6	4.0	0, 6	0.0	1.4	8.8	. 1.0	
	28 -	1.6	0.8	1.1	1.6	1.6	1.6	0.0	1. 6	1.0	6.1	1.1	0.0	0. 0	ę. s	₽. ₽	8. 6	1.1	
CARBON DIGITOR	41	1.0	9. \$	1.1	4.4	0,0	' 4.6	0. 6	6. 6	8.6	4. 6	0.4	. 0.0	0.0	0.0	4.4	₽, €	1,0	: •
CORN STEEP LIQUOR		0.6		1.0		790.4		1.4		1.1		114.3		67 <b>6.</b> [		96.7		17.6	-
MUTRIENTS .		1.1		0.0		1.1		1.1		1.1		1.1		0. 0		0.0		0,0	-
CELLULASE		1.6		1.1		1454.8		1.6		1.0		209.1		1245.7		177.0		31.3	-
CELLOBIASE		4.6		0.0		85, 0		1.1		1.1		12.7		72.3		i0.7	*****	2.0	-
EXTRACTIVES		0, 6		6.0		2585. 1		1.1		€. € _		372.2		5515.3	:	315.6	***	56.7	
SOLUBLE LIGHTN		0.0		1.4		1699. 0		0.0		0. 0		244.3		1454. 8		207.1		37.1	
FUSEL DIL		0.0		8. 0		4.0		1.4		6.0		8.6		1. (		0. Ű		9. #	
FURFUNAL.	96	0.6	8. 6	8. 6	0.0	1050.7	11.4	68. 4	4.7,	513, 9	5. 4	157. 3	1.6	933. 4	9.7	133.6	1.4	23.4	
柳	126	6.0	ø. ø	9. 6	1.0	55.7	0.4	0.0	1. 1	0.0	8. Ú	7.8	6.1	47.9	0.4	6. 🐔	9. (	1.0	
XYL OSE	150	₽. ₽	8, 8	8.0	6. \$	1177.7	7.9	6. 6	6.0	0. 0	1. 1	170.0	1.1	1267.7	6.7	144.6	1.0	25. 1	
<b>G</b> LUCOSE	186	1. 9	8. 6	8.0	1. 6	1.1	6.0	ð. <b>4</b>	1.1	6.8	6. 6	8. €	4.4	0.0	1.1	0. 8	6. 6	1.1	•
SULFURIC ACID	98	4. 0	4.4	1.1	1.1	1.6	1.1	8. 0	1.1	6. 1	₽. 0	8. 6	4.4	1.1	1.1	0. 0	0.0	1.4	1
SODIUM HYDROXIDE	48	0. 0	0, 0	4.4	8.6	1. Ú	6.0	1.4	1. 1	6.6	8. 6	1.6	1.1	9. 8	0.0	0.0	0.0	6.6	
CALCIUM HYDROLIDE	74	€. €	0.0	2366. 3	32. 🖡	6. 6	0.0	4.1	1.0	1.0	0. 0	4.0	8.0	0.0	0.6	0.6	0.6	1,4	
BEHLENE	76	0. Ø <sub>2</sub>	4. 0	8.4	1.1	0.8	0.6	1.6	6. E	1.6	8.0	1.6	0.0	0.0	4.0	9.0	0.6	4.0	i
CAL YCE ROL	98	6. Ú.	0. 0	0.0	1.1	1062.5	11.5	1.4	1. 0	1.0	1.4	156.3	1.7	926.2	10.1	132.9	1. 1	23.4	•
ETHANOL	46	Ø. Ø.	6.4	0.0	8.4	0.6	4.6	9. 0	Ø. B	24.4	4.5	0.0	0.0	0. 5	0.6	6.0	0. 8	0.0	•
SUBJUM CORBUNATE	105	0.0	8. 6	ē. š	1.1	0.0	1.0	0.6	4.4	4.6	6.0	0.0	0.0	0.0	8.0	0.0	0. E	0.4	Ī
SCOTUM SULFATE	145	0.0	8. 6	8. 0	1.0	1027.	12. 9	0.6	1.6	0.0	1.1	271.6	1.9	1555, 4	11.6	234.6	1.6	41.6	
SODIUM ACETATE	82	0.1	4.4	9.4	1.4	51.6	1.0	0.0	1.1	4.4	0.0	7.8	0.0	44.0	1.1	6.4	8.0	1.0	
CALCIUM ACETATE	156 .	4.6	8. 6	6.0	1.4	1310.2	1.3	0.6	9.0	0.0	0.0	180.6	1.2	1121.6	1.1	160.2	1.0	28. J	i
ACETIC ACID	£ø	1.0	0.0	1.6	8.0	1.1	0.0	0.0	1.4	0.0	0.0	1.1	0. 6	0.8	1.0	9.0	6.0	0.0	
ME THIRNOL	32	4. 6	0.0	8.0	1.6	0.0	0.0	0.6	0.0	0.0	0.0	1.6	6.6	1.0	1.4	9.6	9.0	8.0	·
ACETAL DEHYINE	44	3.8	0.6	9.0	4.4	0.0	1.0	1.1	0.0	0.0	0.6	8.8	0.0	0.4	9.4	0.0	4.0	8. 6	
		*****	-			. <u>साम्</u>							#. # ·	7. V		V. V	-,-	B.T	: 10
TOTAL:	1.1	2366. 3	: . :	2364, 3		244170.6		2592.		13239.3		42441.9		201720.7		23224. #		13217.8	
TEMPERATURE, C		27.		27.6		188.0		188.0		100.0		105 #		100 =		102.0		(4) 6	
PRESSURE, ATM					100					190.4		189.0		199.0		199. 6		100.6	
Lursonur' Wiu		1.0	5	1.4		1.4		1.0		1.4		1.0		1.0		1.0		1.0	

#### ETIMANOL VIA ENZYME HYDROLYSIS SECTION 900; FURFURAL PRODUCTION

STREAM NO.			919	***********	320		321		922		<b>21</b>	9	24	9	25		)2 <b>6</b>		<b>%</b> )
DIGNETIE 453 FILE SECO	NOLECULAR METOHTS	LBS Hat	HOLES HA	LBS	MOLEG HA	LBS HR	NOLES HR	LOS	MOLES	LOS IOR	MONES Hif	LOS	MILES HA	LBS HM	MOLES HA	LBS HR	MAI ES	LBS	MOLES HIR
uatea	10	217255.4	12069.7	73462.6	4481.3	£2399, <b>0</b>	3466.6	290616.9	leise s	£4007 Ø	***								: V
CELLIA OSE	195	0.0	1.1	8.8	0.0	1.1	3,0	5.6	16145. 4 0. 6	62297.9	1461.	957.9	53.2	935.	51.9	62375.2	3465.3	78.3	4.3
INSOLUBLE LIGHTH		8.0		0.4		4.6		6.6		6.6	6.0	6.0	0.0	4.0	1.1	1.1	1.4	0.0	1.1
ash		0.0	- C	0.4		0.0		0.0		6.8		1.6	***	6.0		1.1	***	4.4	•
MYCELL ILM		6.6		0.0		1.1	***	4.4		8.6		1.1	-	1.0		6.0		8.4	
OTHER INSOLUBLES		217.9		0.6		8.6		217.5		0.0		0.0		1.0		0.0		4.6	
YEAGI		6.5		0.0		4.4		6.6				8.0		1.0		1.1			1.1.1.1 <b>11.</b>
CALCIUM SULFATE	136	551.4	4.1	0.0	0.0	0.4	0.0	551.0	4.1	6.6 1.0		1.1		1.1		8.1		1.1	
HEME, BYLAN	132	0.0	0.0	1.0	6.0	0.0	4.4	1.1	8.6	1.1	1.6	8.6 9.8	6.0		1.0	3.0	1.1	1.1	1.1
HENS. GLUCAN	162	0.0	8.9	1.0	0.0	0.0	0.0	1.6	1.0	0.0	6.4	0.1	1.1	1.5	1.0	1.1	1.6	1.6	1.1
CALCIUM CARBONATE	100	1.0	6.0		0.0	L.S	1.4	1.4	6.6	0.6	6.6	1.1	L.	f. j	0.0	1.1	0.0	1.6	0.0
DIYGEN	32	0.0	0.6	8.6	1.1	1.1	0.0	0.4	0.0	1.0	8.0	0.0	0.6 0.0		1.0	1.0	1.1	1.1	6.0
ML I RUGEN	28	0.6	1.6	8.8	8.0	0.0	8.0	0.0	0.4		1.1			0.0	0.0	0.6	0.1	1.0	8.0
CARBON BIDILDE	44	1.1	4.0	1.1	6.0	0.6	0.0	8.6	1.1	1.1	1.1	1.4	1.0	1.1	11	<b></b> .	0.0	4.6	
COM STEEP LIDION	4 da 444	112.8		0.0		6.6		712.4		9.0		1.0	6.6	1.1	0.0	1.1		1.1	1,0
MITATENTS		4.6		0.0		1.1		0.0		1.0		0.0		1.1		1.1	• • • • • • • • • • • • • • • • • • • •	1.1	
CELLULASE		1423.5		6.6		4.0		1123.5		1.6		1.1		9.0		0.0		0.0	
CELLODIASE		81.6		0.6		0.0		83.0		4.4		1.4		0.6 0.0		0.0		8.6	•••
EXTRACTIVES		2524.5		0.0		1.1		2524.5		8.6		0.0				0.0		1.1	
SOLUBLE LIGHTN		1661.9		8.4		1.1		1661.9		0.0		0.0		1.1		1.1		0.0	
FUSEL DIL		0.6		0.0		1.1		6.0		0.0		8.6		9.8 0.8		1.1	•	1.1	77
FURFURAL	96	1667. 3	11.1	3542.4	36.9	18614.0	193.9	46.1	0.5	14051.2	146.4	5842.3	52.5			1.1		4.6	· · · · · · · · · · · · · · · · · · ·
HMF	126	54.1	8.4	8.6	1.0	4.0	6.0	56.1	0.4	4.4	1.1			503.2	5.2	14075.6	146.6	24. 4	<b>4.</b> J
LYLOSE	150	1152.3	7.7	1.0	1.0	0.6	0.0	1152.3	1.1	0.0	1.0	1.0	0.0 0.0	8.8 8.8	1.1	1.0	4.4	1.1	5.1
BLUC09E	180	1.1	L.	0.0	1.0	6.6	1.1	9.4	8.0	6.0	0.0				8.6	<u> </u>	• •	10 Mg	1.1
SLEFURIC ACID	98	1.0	0.0	1.1	6.0	1.1	1.1	6.0		0.6			•	1.1	1.1		1.0	1.1	(N. S. S. B. B.)
SOCILIM HYDROLIDE	14	0.0	6.8	1.1	4.0	1.1	1.0	6.0	1.1	0.0	Li	0.0	0.6	0.6	1.1	1.1	1.0	1.0	0.0
CALCIUM HYBROTIDE	24	0.0	6.6	6.6	0.0	0.4	6.6	6.0	0.0	0.1	6.6			8.6	1.1	6.0	1.0	1.6	4.4
BENTEHE	76	1.1	1.8	1.1	6.0	0.6	0.0	0.4	6.0	1.1	1.1	6.0	1.1	1.1	1.6	1.6	1.1	1.1	1.1
GL YCE ROL	92	1059.1	11.5	1.1	4.60	8.8	0.0	1859, 1	11.5		0.0	0.0	1.1 1.1	1.0	1.1	ostiljs	- 1.1 J.1 /	8.0	55 2 N.F
ETISMOL	46	8.8	0.0	102.7	2.2	142.7	2.2	0.0	8.6	8.6	0.0	8.8		1 7 7 7 7	4.4	9.8	1.1	1.0	8,6
SODILM CARBONATE	186	6.6	0.0	1.1	1.1	1.1	4.1	4.1	6.0	1.1			1.1	1.1	1.1	192.7	2.2	102.7	2.2
SOOTUM SULFATE	112	1786.4	12.6	8.8	8.0	8.0	4.0	1786.0	12.6	1.1	6.0	1.1	1.1	1.1	1.1	Lt	0.0	1.1	0.0
SOULM ACETATE	82	50.0	1.1	0.1	0.6	6.6	6.8	50.4	12.6	1.1		1.0	1.1	1.1	1.1	4.1	1.6	1.1	6.1
CALCIUM ACETATE	158	1201.0	4.1	0.5	1.1	0.0		1201.4	A.i	4.6	1.1	6.6	6.6	4.4	0.0	1.0	0.0	4.1	1.0
ACETIC ACID	60	1.1	1.1	0.1	0.0	0.0	1.1	1.1	1.0	1.4	8.6	1.4	1,44 PM TO TO 1	1.6	1.0	1.8	6.6	6.6	1.1
RETHANDL	15		0.0	0.6	8.0		6.4		1.1	0.1	0.6	0.4	8.6	0.0	0.0	1.1	1.6	, <b>6.0</b> ]	1.0
ACE IAL DEHYDE	44	0.4	0.0	1.6	0.0	0.0	8.0	0.0	1.1	0.0	0.0	0.0	0.0 0.0	1.1 1.1	1.1 1.1	1. f	0.6 0.6	1.1	0.0 0.0
(O)AL		230954.7		77(47.1		81116.5		303295.0		76349.1		6000.1		1436, 1		76554.5		(95.4	
TEMPERATURE, C		[88.8		150.0		98. 6		104.0		96. 🛊		68 A							
PRESSURE, AIM		1.0		1.0		1.0		and the state of t				94.0		96.4		96.0		69. ¢	
						1.0		1.1		1.0		i. L∎.		1.0		1.0		1.0	

Y .:

STREAM NO.		96	28
DISKETTE 453 FILE SEC9	MOLECULAR Weights	LBS	MOLES
WATER	. 18	22.9	1. 3
CELLULOSE	162	0. 0	8.6
INSULUBLE LIGHTN		8.6	
AS11		4.4	
MYCELL CLIM	****	1.4	
OTHER INSOLUBLES		8. 6	
YEAST		8. 6	
CALCIUM SULFATE	136	0.0	8.0
HENT, XYLAN	132	8.6	6.8
HENT. GELICAN	162	4.6	0.8
CALCIUM CARBONATE	180	4.4	1.0
DLYGEN	32	1.6	0.6
NETROGEN	28	4.6	0.0
CARBON DIUKIDE	44	1.1	8.6
CORN STEEP LIQUIA		4. 5	
MUTRIENTS		4.6	
CELLULASE		0.0	
CELLDAIRSE		0.8	
EXTRACTIVES		8.6	:
SOLUBLE LIGHTH		0.0	
FUSEL DIL		4.4	
FUPFURAL	96	4539. 1	47.3
HMF	156	0.0	6.0
XYLOSE	150	0.4	4.8
GLUCOSE	188	1.6	0.0
SULFURIC ACID	98	4. 6	0.0
SOUTH HYDROXIDE	18	6.6	4.4
CALCIUM INORGAZOE	74	0.0	0.0
BENJENE	78	4.0	6.6
GL. YCEROL	92	0. 0	4.6
ETHRNOE	46	1.8	4.8
SODIUM CARBONATE	196	1.4	8.8
SCOTIM SULFATE	142	0.6	8.9
SODIUM INCETATE	62	0.0	8.0
CALCIUM ACETATE	158	0.0	8.8
ACETIC ACID	60	0.0	6.8
METHANOL	15	1. 8	8.8
ACETAL DENYDE	44	0.0	0.6
			D. V
TOTAL		4562.0	
TEMFERATURE, C		43. 6	
PRESSURE, AIN		1.8	
		1	

STREAM MO.			1001		<b>16</b> 2		003		984	i. i	1803		1966		1667	1	PAB .		1019
DISKETTE 453 FILE SECTO	NOLECULAR NEIGHTS	LBS HA	MOLES HR	LOS HA	MOLES HÆ	LOS Ha	HOLES HA	LDS HO	MOLES HA	LDS HR	MOLES 119	LDS HA	MOLES IN	LDS HA	MOLES Hil	LOS HÆ	MD ES HR	LBS HA	MOLE HÆ
WATER	10	290616.9	16145.4	23079.7	1282.2	182.7	10.1	14.3	4.3	48914. 3	2717.5	257 <b>668</b> . (	14278.2	218789.7	12150.5	176736.4	3822. 0	131170.5	7267.
CELLUN OSE	. I&	0.0	4.0	3643.2	22.5	0.6	0.0	1.0	0.0	\$.0	8.0	4.4	1.1	1.6	1.6	6.0	: 0.4:	0.0	9.
INSOLUBLE LIGHTN		1.0		15313.5	***	6.0		4.0		8.8		1.4		6.6		1.1		0.0	•
ASH		0.0		185.6		8.6		6.0		6.6		6.6		1.0		1.0		4.4	
NYCELLIUM		0.6		348.8		1.1		1.0	***	1.4		1.6		4.0		1.1		6.0	
OTHER INSOLUBLES		217.9		9012.8		9.6		1.1		6.0		217.9		217.9		217.9		217.9	
YEAST		6.0		0.0		249.1		0.0		8.4		6.0		6.4	• • • • • • • • • • • • • • • • • • •	6.8		6.8	
CALCIUM SULFAIE	1.36	551.4	4.1	2.9		0.6	1.1	1.4	1.4	1.0	0.0	551.0	4.1	551.0	9,1	551.0	4. 6	551.0	4.
HENT, TYLAN	132	L.C	4.6	1.0	8.0	1.6	1.8	1.0	1.0	0.0	8.1	8.6	0.0	. 0.0	6.0	4.4	4.6	9.0	8.1
HEMI. GLUCAN	185	0.0	0.0	0.6	1.1	1.1	0.0	8.8	8.0	1.1	0.0	1.1	0.6	0.0	0.0	1.6	0.0	1.1	• •
CALCIUM CARBONATE	106	1.0	0.0	1.1	1.1	10.6	6.2	0.6	8.0	6.0	6.0	0.6	0.0	0.0	0.1	1.6	0.0	0.5	•.0
OTREEN	32	0.0	0.0	1.1	1.1	1.1	6.6	4.4	0.0	0.6	1.1	9.8	0.0	0.0	0.1	1.1	9.6	4.4	6.1
NLTROGEN	28	1.0	0.0	1.1	1.1	1.0	1.1	6.0	6.6	0.6	6.6	0.0	6.0	0.1	8.0	1.1	6.6	8.0	0.
CARBON DIDLIDE	11	0.0	0.0	1.5	1.1	0.0	0.0	6.6	1.6	4.0	6.0	6.6	8.0	1.1	0.0	0.0	6.0	0.6	8.6
CORM STEEP LIQUID		772.0		3.9		1.6		1.1		0.6		112.8		712.8		772.4		112.0	• •
MITALENTS		0.6		1.1		0.0		1.1		0.6		8.6		0.0		0.0		0.0	
CELLULASE	1	1423.5		9. a		1.0		1.0		8.4		1423.5		1423.5		1423.5		1423.5	
CELLOBIASE		61.6		0.0		6.6		1.1		0.8		83.8		63.0		63.6		81.0	
EXTRACTIVES		2528.5		1.6		2.1		0.0		0.0		2528.5		2528.5		2528.5	+	2528.5	
SOLUBLE LIBNIN		1661.9		5.9		1.0		6.4		1.1		1661.9		1661.9		1661.9		1661.9	***
FUSEL OIL		0.0		1.1		1.1		0.0		0.0		8.0		9.0		0.1		#. <b>.</b>	
FURFURAL	96	46.1	0.5	0.4	0.1	8.8	6.6	21.4	6.3	175.7	1.6	8.8	6.0	1.1	0.0	1.0	0.0	0.0	
INE	126	56.7	6.4	1.0	0.0	1.0	0.0	6.0	1.1	0.1	0.1	56.1	6.4	56.1	6.4	56.1	0.4	56.7	0.1
IYLOSE	150	1152.3	1.1	51.6	8.3	11.7	0, 1	0.0	0.6			1152.3	1.1	1152.3	1.7	1152, 3	1.7	1152.3	0.4
<b>BLUCOSE</b>	188	4.0	6.6	147.5	4.4	5.6	0.4	6.1	l.e	1.1	1.1	1.1	1.0	9.6	1.1	6.6	9.8	11.2.3	1.1
SILFURIC ACID	98	8.0	0.0	1.1	1.4	1.6	0.0	0.0	1.6	0.8	1.1		1.6	1.6	1.1	1.6			<b>9.</b> (
SOCIUM HYDROTTEE	48	1.0	1.0	0.0	6.0	1.1	4.6	1.1	0.0	6.6	1.0	1.1	1.4	1.1	1.4	1.4	8.0 8.0	1.0	1.9
CALCIUM HYDROXISE	16	0.0	0.8	0.0	0.0	0.0	1.1	0.1	0.6		1.6	0.0	5.0	0.0	1.4	1.0			6.6
BENJENE	76	0.0	5.0	1.1	0.1	1.1	0.6	6.0	6.0	1.0	1.1	8.0	1.6	1.1	0.6	4.4	0.0	8.0	0.0
GL YCEROL	92	1059.1	11.5	8.6	0.0	1.0		6.0	1.1	0.6	1.1	1059.1	11.5	1859.1	11.5	( <b>85</b> 3. j	11.5	0. 0 1059. i	0.6
ETHANOL	46	1.0	6.0	0.4	0.4	15.6	0.1	192.7	2.2	0.1	0.0	0. 6	0.0	0.0	1.1	0.0			11.5
SCOULM CARBONATE	106	0.0	1.0	0.0	0.0	1.1	1.1	1.1	1.0			1.1	1.1	8.6	0.6	0.0		7.8	1.1
SOUTUM SULFATE	112	1786.0	12.6	5.9		1.0		1.1	1.4	0.0	6.6	1786.0	12.6	1786.0	12.6	-	0.4	₽. \$	8.4
SODIUM ACETATE	82	54.8	1.1	0.4	0.0	1.0	0.1	0.6	8.6	1.1	1.1	50.0	1.1	54.4	1.0	1786.4	12.6	1786.	12.6
CALCIUM ACETATE	158	1281.0	6.1	1,9		0.6	0.5	1.6	1.1	1.0	0.4	1281.4	<b>5</b> L	1281.8		50. å	4.0	50.8	0.0
ACETIC ACID	68	8.0	0.0	9.6	9.8	6.6	0.0	1.1	0.0	56.2	0.3	0.0	0.0	1ca1.#	8.1 9.0	1281.6	6.1	1281.8	8.1
HETHANIL	15	6.0	0.0	1.6	1.0	0.0	1.6	1.6	8.6	16. 4	1.5	0.0	1.0	0.0		0.0	8.0	0.0	0.0
ACE I AL DEHYDE	**	8.6	4.4	4.4	0.0	0.0	1.1	1.1	9.0	1.1	8.6	0.0	4.4	1.5	9. 6 9. 6	8.4 8.6	0.0	0. 0 0. 0	1.1
TOTAL		343295.0		51825.0		185.6		205. 4		43192.5		26%40.1		8.11.W1.7		189428.4		143842.5	
TEMPERATURE, C		154.0		14.#		30.0		80. D		148.0		l 15. €		127.0		110.0		107.4	
PRESSURE, ATM		3. t	化氯化氯 医皮肤	1.0		1.4		1.0	and a start	1.4		3, L	Seattle for the first	2.5	2014/03/05	1.9		1.1	2000

STREAM NO.		1916		1011															
2.11.11.11.11		·	410			1012		1913		1914	1015		1816		1017		1918		
DISKETTE 453 FILE SECIO	MOLECULAR METGHTS	LOS :	MOLES	LBS	MOLES	1 BS	MOLES	LBS	HOLES	1.09	MULES	LBS	MOLES	LBS	MOLES	LPS	MOLES	LAS	MUI E
TILL SEED	MC10/HQ	ns	118 -	IIA.	HR	HR	HA	HR	HIR	HR	HR	IIŘ	HR	HA	IIA	HR	HR	HR	HA
														***********				*********	
WATER	là .	76653.9	4258. 5	12631.	781.8	33688. 6	1867. 2	38298, 4	2127.7	41913.3	2324.5	45625.9	2534.8	54516.6	3028, 7	64022.0	JEET A	106 800 X	
CELLULOSE	162	0. 0	0.0	1.4	0.0	0.0	1.0	8.0	6.4	0.0	P. B	8.0	B. #	9.9	JVCG, 1		3556.0	356899.7	18161.
INSOLUBLE LIGHTN		ű. S		1.1		0.0		1.6		1.1	***	1.1		1. ē		9. 5	9. 0	0.0	●.
ASIT		0.0		6.6		0.0		0.6		0.0		0.0				6. 6	***	9. 0	
MYCELLIUM		1.0		0.0		1.1		8.0		9.6		1.1		1.1		1.1	*	8. 0	
OTHER INSOLUBLES	<b></b> .	217, 9	~~~	217.9		0.6		0.0	<del></del> -	1.1		0. E		1.0		0.0		0. 8	
YEAST	***	6.8		6. 6	~~~	6.0		0.0						8.0		0. ć	M. 1-1-	€. €	
CALCIUM SULFATE	136	551.0	4.1	551.0	4.1	4.6	8.0	9. 0		6.0		1.0		1.1		1.1		Ð. Ø	
HENT, KYLAN	132	1.0	1.0	4.6	6.6	9. 6	1.6	1. T	1.1	9. 0	1.1	1.0	1.1	1.4	8.4	<b>1. (</b>	1.4	4. 0	€.
HEME, BLUCIN	165	4.0	0.0	4.4	0.0	0.0	1.6		8.8	1.1	6.6	6.0	0.0	4.0	0. 0	0. 8	9. 0	0.0	€,-
CALCIUM CARDONATE	:00	0.0	0.1	0.0	0.0	6.6		0.0	4.1	1.1	1.1	₽. 0	9. 8	4.0	0. E	0.0	0. B	9. 0	0.1
CHYCEH	32	6.0	0.4	0.0	0.0		1.1	8.6	1.1	0.0	1. (	1.1	2.6	0.0	1.1	₽. ₽	P. #	a. a	0,1
NITROGEN	28	4, 8	4.0	1. f		1.1	0.4	4.4	1. f	9. 0	8.6	€. €	0. f	0.0	6. S	a. a	0.6	8.0	9.1
CARBON DIOXIDE	44	0.0	0.0		8.6	1.4	1.4	0. 0	₽. ♦	1.4	6. 0	0. 6	0. 8	1.4	€. €	0. 0	9.0	2.6	€. 1
CORN STEEP LIQUOR		112.6		1.6	4.0	9. 0	1.1	1. 0	8.0	1.1	ŧ. •	0. 0	0. 0	1. 0	9. 0	0. 0	0.0	0.0	8.4
NUTRIENTS		0.8		772.6		1.1		4. #	~~~	6.0		1.6		6.6		8. 6		€. €	
CELLULASE		1123.5	***	0.0		4.6		1.0		e. é	7.7-	0. €		9. 0		6. 0		0.6	
CELLODIASE			***	1423.5		0.8		6. 6		4.6		8. 8		1.1	***	0, ₽		1.6	
EXTRACTIVES		83.4		83.0		0. 0		0.0		9. 6		0.6		1.1		0.6	***	8.0	
SOLUBLE LIGHTN		2528.5		2528.5	1	9.0	·	5. 8	· :	. B. D.		0.6		6.6		0.0		0.0	
FUSEL OIL		1661.9		1661.9	·,	3. F	·	1.1		0.0	• • •	0.6		4.6		0.0		1.6	
		1.0	***	1.1		8.0		0.0		1.4		0.0		0.6		f. t		6, 9	
FURFURAL	96	4.5	0.0	8. 6	1.6	46, 1	1.5	1.1	Ð. <b>D</b>	0.6	0.0	0.0	0. 0	0.0	4.4	0.0	0.0	221.8	
HIME	126	56.7	0.4	56. 7	8.4	1.4	f. f.	1.4	1.6	9.0	1.4	1.4	8.8	1.0	0.0	6.6	0.0		2. 3
TYLDSE	150	1152.3	7.7	1152.3	7.7	1.6	0.0	0.0	0.0	0.6	1.1	0.0	6.0	4.4	6.6	1. 0	1.1	. 6.6; 6.6	0.0
6LUCOSE	180	1.6	₽. ₩	6. 6	1.6	0.0	1.4	0.0	0.0	1.6	1.1	9.6	0.6	0.0	7				₽. 6
SULFURIC ACID	98	8.4	6.0	8, ∰ .	1.4	1. 6	4.6	0.0	0.6	1.8	6. 6	0.0	1.0	8.6	f. 0 0, 1	<b>0. 0</b>	<b>9. 0</b>	6.0	8.4
SODILM HYDROLODE	40	Ø. Ø.	£. 8	0.6	0.0	4.6	0. 6	6.6	1.4	0.0	1. 6	1.6	e. 6	0. B		0. 0	0. 0	0.0	. €.6
CALCIUM INDROXIDE	74.	●. ●	1.1	6.4	0.0	1.0	1.0	8.4	1.0	0.6	1.1	1.6			1.6	6.0	2. 9	3.0	9.0
BENZENE	78	0. 6	4. 0	6. 6	1.1	0.6	1.4	9.6	6.6	4.4	8. 6		₽. Đ	1.1	1.1	8.6	0.0	0.0	₽. 2
GL.YCEAOL	98	1059. 1	11.5	1059, [	11.5	0.0	1.6	1.1	1.6	1.2	0.0	1.1	9. 6	1.0	0.0	0.0	0. 🛊	ě. 🛊	1.1
ETHIRMOL	46	0. 6	8.8	6.8	6.0	0.0	6.0	0.0	0.6			1.0	1.6	1.0	0. 6	e, e	0. 6	2.6	9.0
SODIUM CARBONATE	105	0. 6	0.0	0.6	6.0	0.0	9.6	6.0		0.0	1.4	1.1	0. B	8.8	4. 0	₽. €	€. €	2.0	0.0
SOUTH SULFATE	142	1786.	12.6	1786. 6	12.6	0.8	1.6		1.1	0.0	1.0	0.0	0.6	ē. š	4.4	1.1	1. š	€. 🛊	0.4
SODIUM ACETATE	82	50.6	0.0	50. a	1.8	0.0		9.0	8.0	8.6	0.4	0.6	0. 1	0.0	4. 6	₽, #	0.0	0.0	8.0
CALCIUM ACETATE	150	1281.6	6.1	1281.6	8.1	1. 1	9. 6	1.1.	0.6	8.0	6, 6	1.1	0.0	1.4	0, 0	0, 6	4.6	0. 🕅	0.0
ACETIC ACTO	68	6.0	6. 0	0.0	1.0	8.4	0. 0	0.0	4.41	1.0	8.4	6.6	0. 0	0. 0	8.6	8. 0	€. ₿	8. 9	4.0
ME THANGE	32	1.1	1.0	0.6			0.0	0.6	1.1	4. 6	₽. ●	0, 6	0.0	0.0	6. 0	8. €	6.4	56.∂	<b>e.</b> 9
CETAL DEHYDE	44	1.4	0.0	0, 8	1. 0 1. 4	0, 0 - 0, 6 -	1. E	1.6 - 4.4	0.0	0.0	1.0	1. 6	9. 0	4.4	€. €	<b>e</b> . 4	8.8	46, 4	1.5
									4, 0	0.0	9. 4	4.4	1.6	6.8	0, 8	6.6	€. €	0, 0	₽. €
OTAL		69285, <b>9</b>	**	25263.9		33654.9		38298.4		41913.3		15625, 9		51516.6		64022.0		321223.7	
ENCERATURE, C		88.0		36, 6	3.5	135. ₽		137.4						•					
RESSURE, AIN		6, 66		0. 665		3.1		127. <b>0</b> 2.5		118.8		103.6		88. #		36. 0		76. 8	
		V. L/L		o. cl.i		• •						1.1		0.66		8. 865			

STREAM NO.			419		1020		1051		1855
DISKETTE 453	MOLECIA AR	LDS	MOLES	LBS	MU	: .	MOLES	L PS	POLES
FILE SECIO	METERIES	網	HR.S	· Hall	. , Н⊈.	HAR	HE .	HR	H
									**********
MATER	18	11577.5	643.2	6.0		. 8 47554. 0	6 2641.7	1.1	1.6
CELLULOSE	165	5464.6	33.1	4.6		9107.4		4.0	4.0
INSOLUBLE LIGHTN		2130.2		9.0		17443.7		LA	
ASH		23.1		0.0		264. (		200.0	
MYCELL IUM		0.0		0.1	-	348.1		9.6	
OFFER INSOLUBLES		1227.3	, <del></del> .	5.2 <b>0.</b> 0	₽ jubiu-	18457.5	) <b></b> .	500.2	
YEAST		8.0				256.0		0.6	
CALCIUM SULFAIE	136	1.1	1.0		1	. 554. 0	4.1	554.0	4.1
HERI. IYLAN	132	1875.6	14.2	4.0	•	. 6 1875.6	5.41	1.1	0.0
HEMI, GLUCOM	162	532.5	1.1	0.4	1), 1 <sub>), 1</sub> , 1	532.5	3.3	0.0	0.0
CALCEUM CARBONATE	100	1.6	1.0	. e. (		.4 18.6	1.2	0.0	1.1
CHYGEN	32		0.0	<b>8.</b> (		4.0	1.0	1.0	0.6
VLTADGEN	58	1.6	1.1	4.6	•	.6 6.0	L i Li	1.1	1.8
CANBON BLOKTUE	44	0.0	1.1	303.1	6	. 9 303. (	6.9	1.1	0.0
CORM STEEP LIQUOR		1.1		1.1	•	111.1		- 1. <b>(</b>	
NUTRIENTS		1.1		: <b>6</b> (			•	0.0	
CELLULASE	· · · · · · · · · · · · · · · · · · ·	1.1		22.0		1434.8		4.0	
CELLGBIASE		, es <b>1.1</b> s		9.1		- 17.0		6.1	
EXTRACTIVES		324.2				2863.4		1.1	
SOLUBLE LIGHTA		1.0	***	1.0	1	1668. 7		1.1	
FUSE OIL		0.0		. A.		41			
FURFURAL.	%	1.6	1.1	0.0		. 24. (		1.1	0.0
HPF XYI USE	126	1.1				\$ 55.7	5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	1.0	1.1
	150	8.6	<b>. ! !</b> .			. 1215. 0		1.1	6.8
GLUCOSE SUCFURIC ACID	1600 90		• • •			6 147.5		4.0	1.1
SOUTH HYDROLIDE	70 40	0.6 0.4	, Val. <b>6.6</b> 0	•	- 1. ·	4		0.0	1.0
CALCIUM HYGROLIDE	74	7	1.0	0.0		0.0		1.6	0.0
BENZENE	78		6.6 6.6	14 B	T	1 0.0		1.1	0.0
GL YCENOL	92	0.5	1.1	L(		. 0 . 0 . 0 . 0 . 0 . 0 . 0 . 0 . 0 . 0		l.t	1.1
ETHANOL	46	4.6	1.6	9.1	1 1 1			1.0	0.0
SUBTUM CARBONATE	106	4.4	0.0	8.0		. 116.3 . 0.0		1.1	1.0
SCOTON SIAFATE	142	4.1	1.0			. 1792. 6		0.0 1792.6	1.1
SODIUM ACETATE	82	1.1	0.4	0.0		51.6		0.6	12.6
CHECIUM ACETATE	154	1.1	4.6	1.0		\$ 1266.7	1 1 1 1 1 1 1 1 1	6.0	0.0
ACETTE ACID	64		4.4	1.1		# 100. F		4.4	1.6
METHANOL	32	1.0	1.1	0.0	1.3.7	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		1.1	1.0
ACE TAL DEHYDE	41	1.1	0.0	517.1	· · · · · ·			0.1	0.0
JOIAE		23155.6	edan sir <del>a</del> Anglasia	B21. Z		101756.4		3055.8	
temprature, c		30.6		30. 0		10.0		na ~	
PRESSURE, AIN					the second of	10.0		84. 9	
Fut 22/38 F HIM	Service to a	1.0		1.0	1 3	1.0		1.0	

STREAM NAI.			1110		810: ·		1815
DISHETTE 453 FILE SECTI	NULECULAR NEIGHIS	1.9S H41	MULES Har	LDS Hal	HOLES	L DES	MA ES
WATER	18	1.0	6.4	726146.2	10311.5	5629. 5	312.7
CELLO, OSE	162	0.0	<b>8.</b> (	0.6	1.1	100.4	0.7
INSULUBLE LIGHTH		0.1		1.6		71.3	
ASI!		208.1	·	1.1		0.0	
MARCHINE		6.1					: , , , <del>,</del>
OHER INSULUBLIS		584.2				4174.7	
YEATT						132.9	
CALCIUM SELFATE	136	554.0	·		6.6	7366.6	54.2
HEMI. NYLAN MENI. GEUCAN	132	1.0			L.	5.1	6.1
CALCIUM CAMBONATE		1.0			1.1	1.0	1.6
GXYEEN	IN 32	8.0				334.0	1.4
VI TRUGEN	26	4.6			0.0	9.0	8.8
CARBON DIDLIDE	44	8.4			0.0	4.0	8.0
CORN STEEP LIQUOR		4.6		1.0		0.0 17.6	0.0
N. TRIENTS		6.6		1.1		8.6	
COLLUCASE		0.1		4.6		12.2	
CEI LODIASE		1.1		6.8	<u></u>	2.4	
EXTRACTIVES		1.0	<del></del>	6.0	· · · · · · · · · · · · · · · · · · ·	62.5	
SOLUBLE LIGHTN		0.0		6.0		41.6	
FUSE: OIL		9.0		1.6		6.0	
FUSFURAL	%	6.6	0.0	530.7	5.5	23.4	6.2
<b>0</b> €	156	1.0	6.0		6.0	1.0	
TYLUSE	150	0.0	1.1	1.0	8.0	2.1	0.1
6LUCUSE	188	6.0	4.0	1.0	8.0	12.0	1.2
SULFURIC ACID	98	1.0	1.1	1.1	1.1	9.0	0.6
STOREM HYDRINIDE		8.0	6.0	1.0	4.0	4.0	l.t
CA. CILIN HYDROXIDE	74	0.0	1.1	1.6	1.1	4.4	1.4
85MIEVE	18	I. I	L.O	6.8	0.1	6.6	1.1
GL VCF ROL	₩.	1.1	4.0	1.1	0.0	23. 4	1.3
£10/9/ <b>10</b> /2	46	4.0		1.4	8.0	1.0	4.1
SUDILIN CARBONATE	186	8.8		1.0	6.6	1.0	0.0
SUDJUM SA FAJE	115	1792.	12.6	1.0	1.5	12.5	1.1
SODIUM ACETATE	95	1.1		4.6		1.0	1.6
CRI CHUM ACETATE	158	1.0	1.0	1.0	1.0	11.3	1.2
ACETIC ACID	60	0.1	0.0	122.1	P. D	1.1	1.4
- 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1	12	4.6	1.0	67.4	2.1	6.0	1.1
PCETAL DEHYDE	44	0.4			1.1	0.6	1.0
fo'at		3055.0		126813.2		16216.3	
" " SPATURE, C		80.0		61.6			
PHESSUFF, ATM		1.0		1.0			

## B. Major Equipment Specifications and Costs

This sub-section contains the major equipment specifications for the eleven plant sections based on a capacity of 25 million gallons per year of fuel grade ethanol with an on-stream factor of 8,000 hours per year. The equipment sparing philosophy was designed to achieve this on-stream factor. Bare equipment costs were developed in part from vendor quotations and in part from an Icarus Cost computer run.

# EQUIPMENT SPECIFICATIONS AND COSTS SECTION 100 - PRETREATMENT/PREHYDROLYSIS

Conveyor (CO-101)

Part of offsite wood handling

Chip Washer (VS-101) - 3 Items

Part of offsite wood handling

Conveyor (CO-102) -3 Items

Part of offsite wood handling

Chip Bin (VT-101) - 3 Items

\$69,300

Type: Cone bottom

Capacity: 1,600 cubic feet Size: 9' dia. x 23' t/t

Material: CS

Temperature: 100°F Pressure: 0.05 psig

## Feeder (CO-103) - 3 Items

Part of refiner (CR-101) package

# <u> Digester (CO-104) - 3 Items</u>

Part of refiner (CR-101) package

Disk Refiner (CR-101) - 3 Items

\$1,215,000

Type: Single disk

Capacity: 380 dry tons per day

Drive: 3,000 HP Temperature: 392°F Pressure: 220 psig

Prehydrolysis Reactor (R-101) - 3 Items

\$55,300

Capacity: 80 gallons Size: 1' dia. x 13.5' t/t Material: Zirconium clad Internal: Discharge wiper

Drive: 1.5 HP Temperature: 392°F Pressure: 210 psig

First Stage Water Preheater (HE-101)

\$12,200

Area; 840 square feet Type: Fixed T-S

Duty: 17.3 MM Btu/hr

Material: CS/CS

Temperature: 250/150°F Pressure: 20/150 psig

Second Stage Water Preheater (HE-102)

\$26,100

Area: 2,100 square feet

Type: Fixed T-S
Duty: 26.2 MM Btu/hr
Material: CS/CS

Temperature: 250/150°F Pressure: 30/150 psig

Flash Vessel (VT-102)

\$124,600

Type: Vertical

Capacity: 6,100 gallons Size: 7' dia. x 21' t/t Material: Zirconium Internal: Demister Temperature: 300°F Pressure: 50 psig

Flash Vessel (VT-103)

\$93,700

Type: Vertical

Capacity: 3,800 gallons Size: 6' dia. x 18' t/t Material: Zirconium Temperature: 250°F Pressure: 15 psig

## Flash Vessel (VT-104)

Type: Vertical

Capacity: 12,900 gallons Size: 9' dia. x 27' t/t

Material: 317 SS Temperature: 212°F Pressure: 2 psig

# Water Feed Pump (CP-101) - 2 Items

\$15,800

Type: Centrifugal, AVS Medium Capacity: 750 GPM

SG: 1.0 Material: CS Head: 575' TDH Drive: 150 HP Temperature: 30°F

\$1,652,400 30,600 \$1,683,000 Subtotal Section 100 Undefined miscellaneous Total Section 100

# SECTION 200 - SUGAR SEPARATION AND NEUTRALIZATION

Cost Centrifuge (CT-201) - 2 Items \$911,000 Type: Solid bowl Size: 44" dia. Material: 317 SS Drive: 25 HP Overflow: 311 GPM Prehydrolyzate Bin (VT-201) \$143,200 Type: Cone bottom Capacity: 90,000 gallons Material: 317 SS Temperature: 2120F Pressure: 0.05 psig Holdup: 4 hours Overflow Pump (CP-201) - 4 Items \$37,600 Type: Centrifugal, AVS medium Capacity: 310 GPM Material: 317 SS Head: 100' TDH Drive: 15 HP Temperature: 2120F Rotary Drum Polishing Filter (F-201) \$50,200 Area: 100 square feet Capacity: 622 GPM Material: EPLCS Drive: 3 HP Thickener Exchanger (HE-201) \$55,800 Area: 1,175 square feet Type: Fixed T-S Duty: 9.4 MM Btu/hr Material: 316 SS/321 S Temperature: 135/2120F Pressure: 15/15 psig Filter Cooler (HE-202) \$34,100 Area: 880 square feet Type: Fixed T-S Duty: 12.4 MM Btu/hr Material: CS/321 S Temperature: 120/135°F Pressure: 15/15 psig

# Lime Tank (AOT-201) \$87,000 Type: Mixer Capacity: 11,000 gallons Material: 304 SS Drive: 50 HP Holdup: 4 hours Neutralization Tank (AOT-202) \$66,100 Type: Mixer Capacity: 7,000 gallons Material: 317 SS Holdup: 10 minutes Drive: 40 HP Lime Pump (CP-202) - 2 Items \$5,400 Type: Centrifugal, AVS medium Capacity: 50 GPM Material: 304 SS Head: 100' TDH Drive: 2 HP Temperature: 80°F Thickener Feed Pump (CP-203) - 2 Items \$8,800 Type: Centrifugal, AVS medium Capacity: 700 GPM Material: 304 SS Head: 100' TDH Drive: 25 HP Temperature: 140°F Conveyor (CO-201) \$174,100 Type: Screw Material: 317 SS Drive: 40 HP Length: 100' Screw Diameter: 24" Conveyor (CO+202) \$50,700 Type: Screw Material: 317 SS Drive: 7.5 HP Length: 100' Screw Diameter: 9"

## Thickener (T-201)

\$41,000

Type: Gravity w/rake Capacity: 650 GPM Material: CS Drive: 10 HP Diameter: 30'

## Centrifuge (CT-202)

\$71,600

Type: Solid bowl Size: 18" dia.. Material: 304 SS Drive: 25 HP Overflow: 27 GPM

#### Overflow Pump (CP-204) - 2 Items

\$5,400

Type: Centrifugal, AVS medium

Capacity: 30 GPM Material: 304 SS Head: 100' TDH Drive: 3 HP

Temperature: 180°F

## Sugar Receiver (VT-202)

\$167,200

Type: Cone roof

Capacity: 325,000 gallons Material: 304 SS

Material: 304 SS Temperature: 180°F Pressure: 0.05 psig Holdup: 8 hours

## Sugar Transfer Pump (CP-205) - Items

\$8,800

Type: Centrifugal, AVS medium

Capacity: 725 GPM Material: 304 SS Head: 100' TDH Drive: 25 HP Temperature: 180°F

Subtotal Section 200
Undefined miscellaneous
Total Section 200

\$1,918,000 134,400 \$2,052,400

#### SECTION 300 - ENZYME PRODUCTION

Cost Solids Bin (VT-318) \$13,500 Type: Cone bottom Capacity: 12,000 gallons Material: 317 SS Temperature: 210°F Pressure: 0.05 psig Holdup: 4 hours Conveyor (CO-301) Type: Screw Capacity: 13 tons/hour Material: 316 SS Drive: 7.5 HP Length: 100' Screw Diameter: 9" Enzyme Production Tanks (VT-301-314) - 14 Items Type: Cone roof Capacity: 185,000 gallons Material: 304 SS Temperature: 85°F Pressure: 0.05 psig Coolers (HE-301-314) - 14 Items Area: 40 square feet Type: Spiral plate Duty: 0.225 MM Btu/hr Material: 304 SS Pressure: 15 psiq Recirculation Pumps (CP-301-314) - 21 Items \$98,700 Type: Centrifugal, recessed impeller Capacity: 50 GPM Material: 304 SS Head: 100' TDH Drive: 20 HP Temperature: 850F Filter (F-301) \$2,000 Area: 50 square feet Type: Cartridge

Material: CS

Filter (F-302)

\$2,000

Area: 50 square feet Type: Cartridge

Material: CS

Air Blower (FN-301)

\$10,000

Type: ROT-Blower Capacity: 100 GPM

Material: CS Drive: 10 HP

Discharge Pumps (CP-315-328) - 21 Items

98,700

Type: Centrifugal, recessed impeller

Capacity: 50 GPM Material: 304 SS Head: 100' TDH Drive: 20 HP Temperature: 85°F

Hold Tank (HT-301)

\$152,600

Type: Horizontal drum Capacity: 290,000 gallons Material: 304 SS

Temperature: 85°F Pressure: 0.05 psig Holdup: 36 hours

Cell Centrifuge Feed Pump (CP-329) - 2 Items

9,400

Type: Centrifugal, recessed impeller

Capacity: 150 GPM Material: 304 SS Head: 100' TDH Drive: 20 HP Temperature: 35°F

Cell Centrifuge (CT-301)

56,400

Type: Disk Size: 10" diameter

Material: CS Drive: 25 HP Overflow: 50 GPM

Repulping Tank (AOT-301)

\$19,500

Type: Mixer

Capacity: 1,500 gallons Material: 304 SS-clad

Drive: 5 HP Temperature: 85°F Pressure: 0.05 psiq Holdup: 10 minutes Wash Water Tank (VT-315) \$17,600 Type: Cylinder Capacity: 10,000 gallons Material: CS Temperature: 80°F Pressure: 0.05 psig Holdup: 1 hour Wash Water Pump (CP-330) - 2 Items \$7,100 Type: Centrifugal, AVS medium Capacity: 160 GPM Material: CS Head: 100' TDH Drive: 10 HP Temperature: 30°F Polishing Centrifuge (CT-302) \$143,200 Type: Disk Size: 20" dia. Material: CS Drive: 150 HP Overflow: 160 GPM Centrifuge Feed Pump (CP-331) - 2 Items \$6,800 Type: Centrifugal, AVS medium Capacity: 125 GPM Material: 304 SS Head: 100' TDH Drive: 5 HP Temperature: 85°F Cell Recycle Hold Tank (VT-317) \$115,800 Type: Cone roof Capacity: 175,000 gallons Material: 304 SS Temperature: 35°F Pressure: 0.05 psig Holdup: 36 hours

Type: Centrifugal, AVS medium Capacity: 90 GPM

Cell Recycle Pump (CP-332) - 2 Items

Material: 304 SS Head: 100' TDH Drive: 5 HP

Temperature: 85°F

Enzyme Receiver (VT-316)

\$192,500

Type: Cone roof Capacity: 410,000 gallons Material: 304 SS

Temperature: 35°F Pressure: 0.05 psig Holdup: 36 hours

Enzyme Feed Pump (CP-333) - 2 Items

\$7,800

Type: Centrifugal, AVS medium

Capacity: 220 GPM Material: 304 SS Head: 100' TDH Drive: 7.5 HP Temperature: 85°F

> \$2,777,300 Subtotal Section 300 194,700 Undefined Miscellaneous \$2,972,000 Total Section 300:

# SECTION 400 - ENZYME HYDROLYSIS

Cost \$99,400 Sturry Tank (AOT-401) Type: Mixer Capacity: 15,000 gallons Material: 317 S\$ Drive: 50 HP Temperature: 190°F Pressure: 0.05 psig Holdup: 10 minutes \$305,400 Slurry Cooler (HE-407) Area: 11,000 square feet Type: Fixed T-S Duty: 24.4 MM Btu/hr Material: CS/304 SS Temperature: 115/165°F Pressure: 15/15 psig \$13,800 Feed Pump (CP-413) - 2 Items Type: Centrifugal, recessed impeller Capacity: 1,250 GPM Material: 304 SS Head: 100' TDH Drive: 100 HP Enzyme Hydrolysis Tanks (VT-401-406) - 6 Items \$1,077,600 Type: Cone roof Capacity: 370,000 gallons Material: 304 SS Temperature: 122°F Pressure: 0.05 psig \$120,000 Agitators (A-401-406) - 6 Items Type: Propeller Capacity: 40,000 GPM Material: 304 SS Size: 30" Drive: 75 HP Temperature: 122°F \$62,100 Discharge Pumps (CP-407-412) - 9 Items Type: Centrifugal, recessed impeller Capacity: 1,250 GPM

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Material: 304 SS Head: 100' TDH Drive: 100 HP Temperature: 122°F

Steam Coils (HE-401-406) - 6 Items

\$25,200

Area: 100 square feet Type: Heater steam Material: 304 SS Pressure: 15 psig

Centrifuge (CT-401) - 3 Items

\$1,239,600

Type: Solid bowl Size: 54" dia. Material: 304 SS Drive: 250 HP Overflow: 400 GPM

Overflow Pump (CP-415) - 6 Items

\$23,400

Type: Centrifugal, AVS medium Capacity: 450 GPM Material: 304 SS Head: 100' TDH Drive: 20 HP

Temperature: 120°F

Rotary Drum Polishing Filter (F-401)

\$40,200

Area: 100 square feet Type: Rotary drum Capacity: 1,200 GPM Material: 304 SS Drive: 3 HP

Wash Water Tank (VT-407)

\$12,000

Type: Cylinder
Capacity: 5,000 gallons
Material: CS
Temperature: 8005

Temperature: 80°F Pressure: 0.05 psig

Wash Water Pump (CP-414) - 2 Items

\$6,000

Type: Centrifugal, AVS medium Capacity: 80 GPM Material: CS

Head: 100' TDH Drive: 5 HP

Temperature: 800F

\$12,400 Conveyor (CO-401) Type: Screw Capacity: 30 tons/hour Material: CS Drive: 15 HP Length: 100' Screw diameter: 12" Dewatering Press (F-402) - 3 Items \$618,000 Type: Fibercone press Capacity: 60 GPM Material: 304 SS Drive: 75 HP \$6,400 Press Discharge Pump (CP-415) - 2 Items Type: Centrifugal, AVS medium Capacity: 100 GPM Material: 304 SS Head: 100' TDH Drive: 5 HP Temperature: 115°F \$292,000 Sugar Hold Tank (VT-408) Type: Come roof Capacity: 630,000 gallons Material: 304 SS Temperature: 120°F Pressure: 0.05 psig Holdup: 8 hours Sugar Transfer Pump (CP-417) - 2 Items Type: Centrifugal, AVS medium Capacity: 1,300 GPM Material: 304 SS Head: 150' TDH Drive: 50 HP Temperature: 120°F Subtotal Section 400 \$3,967,300 Undefined Miscellaneous 277,700 \$4,245,000

Total Section 400

## SECTION 500 - SUGAR CONCENTRATION

Cost Multi-Effect-Evaporator (E-501) \$3,260,000 Type: Six effect-forced circulation Duty: Heating - 68.6 MM Btu/hr Cooling - 82.3 MM Btu/hr Capacity: 344,000 lb/hr distillate Material: 304 SS Drive: 1,200 HP CaSO<sub>4</sub> Centrifuge (CT-501) \$10,000 Type: Solid bowl Size: 5" diameter Material: 304 SS Drive: 1 HP Overflow: 2 GPM Sugar Pump (CP-501) - 2 Items \$8,200 Type: Centrifuge, AVS medium Capacity: 550 GPM Material: 304 SS Head: 100' TDH Drive: 20 HP Temperature: 100°F Overflow Pump (CP-502) - 2 Items \$4,000 Type: Centrifugal, AVS medium Capacity: 2 GPM Material: 304 SS Head: 50' TDH Drive: 0.5 HP Temperature: 1000F Precipitator (AOT-501) \$45,300 Type: Mixer Capacity: 6,500 gallons Material: 304 SS Drive: 40 HP Temperature: 100°F Pressure: 0.05 psig Holdup: 10 minutes Na2CO3 Solution Tank (AOT-502) \$7,000 Type: Mixer

Capacity: 50 gallons

Material: 304 SS Drive: 0.5 HP Temperature: 100°F Pressure: 0.05 psig

### Na<sub>2</sub>CO<sub>3</sub> Feed Pump (CP-503) - 2 Items

\$4,000

Type: Centrifugal, AVS medium

Capacity: 5 GPM Material: 304 SS Head: 20' TDH Drive: 0.5 HP

Temperature: 100°F

### CaCO3 Centrifuge (CT-502)

\$336,000

Type: Solid bowl Size: 44" dia. Material: 304 SS Drive: 250 hp Overflow: 642 GPM

#### Overflow Pump (CP-504) - 2 Items

\$8,400

Type: Centrifugal, AVS medium

Capacity: 600 GPM Material: 304 SS Head: 100' TDH Drive: 25 HP

Temperature: 100°F

### Filter Feed Pump (CP-505) + 2 Items

\$8,400

Type: Centrifugal, AVS medium

Capacity: 600 GPM Material: 304 SS Head: 50' TDH Drive: 25 HP

Temperature: 100°F

### Distillate Pump (CP-506) - 2 Items

\$8,400

Type: Centrifugal, AVS medium

Capacity: 690 GPM Material: CS Head: 100' TDH Drive: 25 HP

Temperature: 180°F

Subtotal Section 500
Undefined Miscellaneous
Total Section 500

\$3,699,700 31,300 \$3,731,000

#### SECTION 600 - FERMENTATION

Cost Concentrated Sugar Receiver (VT-608) \$25,500 Type: Cylinder Capacity: 7,000 gallons Material: 304 SS Temperature: 100°F Pressure: 0.05 psig Holdup: 10 minutes Blower (FN-601) \$37,500 Type: ROT blower Capacity: 3,040 CFM Material: CS Drive: 75 HP Pressure: 5 psig Air Blower (FN-602) \$10,000 Type: ROT blower Capacity: 100 CFM Material: CS Drive: 10 HP Pressure: 5 psig Filter (F-601) \$2,000 Area: 50 square feet Type: Cartridge Material: CS Filter (F-602) \$2,000 Area: 50 square feet Type: Cartridge Material: CS Fermentation Feed Pump (CP-613) - 2 Items \$8,400 Type: Centrifugal, AVS medium Capacity: 600 GPM Material: 304 SS Head: 100' TDH Drive: 75 HP Temperature: 100°F

# Sugar Cooler (HE-607) \$27,300 Area: 805 square feet Type: Spiral plate Duty: 3.5 MM Btu/hr Material: CS/304 SS Temperature: 80/100°F Pressure: 15/15 psig Fermentation Tanks (VT-601-606) - 6 Items \$614,400 Type: Cone roof Capacity: 150,000 GPM Material: 304 SS Temperature: 86°F Pressure: 0.05 psiq Recirculation Pumps (CP-601-606) - 12 Items Type: Centrifugal, AVS medium Capacity: 600 GPM Material: 304 SS Head: 100' TDH Drive: 25 HP Temperature: 36°F Discharge Pumps (CP-607-612) - 12 Items \$50,400 Type: Centrifugal, AVS medium Capacity: 600 GPM Material: 304 SS Head: 100' TDH Drive: 25 HP Temperature: 36°F Coolers (HE-601-606) - 6 Items \$146,400 Area: 710 square feet Type: Spiral plate Duty: 2.2 MM Btu/hr Material: CS/304 SS Temperature: 30/86°F Pressure: 15/15 psig Scrubber (TW-601) \$20,000 Type: Packed Size: 4' diameter x 10' t/t Material: 304 SS

### Scrubber Return Pump (CP-613) - 2 Items

\$5,400

Type: Centrifugal, AVS medium

Capacity: 50 GPM Material: 304 SS Head: 100' TDH Drive: 2 HP

Temperature: 85°F

#### Yeast Centrifuge (CT-601) - 2 Items

\$229,600

Type: Disk

Size: 30" diameter Material: 304 SS Drive: 200 HP Overflow: 300 GPM

#### Overflow Pump (CP-616) - 4 Items

\$16,000

Type: Centrifugal, AVS medium

Capacity: 300 GPM Material: 304 SS Head: 100' TDH Drive: 15 HP

Temperature: 36°F

#### Underflow Pump (CP-615) - 4 Items

\$10,400

Type: Centrifugal, AVS medium

Capacity: 25 GPM Material: 304 SS Head: 100' TDH Drive: 3 HP

Temperature: 86°F

#### Yeast Hold Tank (AOT-601)

\$68,200

Type: Mixer

Volume: 10,000 gallons

Material: 304 SS Drive: 40 HP Temperature: 86°F Pressure: 0.05 psig Holdup: 4 hours

#### Yeast Pump (CP-617) - 2 Items

\$5,200

Type: Centrifugal, AVS medium Capacity: 40 GPM Material: 304 SS Head: 100' TDH Drive: 3 HP

Temperature: 86°F

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Dewatering Press (F-603) \$45,000 Type: Fibercone Press Capacity: 3 GPM Material: CS Drive: 1 HP Discharge Pump (CP-618) - 2 Items \$4,000 Type: Centrifugal, AVS medium Capacity: 5 GPM Material: 304 SS Head: 50' TDH Drive: 1 HP Temperature: 36°F Conveyor (CO-601) \$10,000 Type: Screw Capacity: 0.5 tons/hr Material: CS Drive: 7.5 HP Length: 100' Beer Well (VT-607) - 2 Items \$204,800 Type: Come roof Capacity: 150,000 gallons Material: 304 SS Temperature: 86°F Pressure: 0.05 psig Holdup: 8 hours Purification Feed Pump (CP-619) - 2 Items \$8,400 Type: Centrifugal, AVS medium Capacity: 600 GPM Material: 304 SS Head: 180' TDH Drive: 40 HP Temperature: 86°F Subtotal Section 600 \$1,501,300 Undefined Miscellaneous 112,700 Total Section 600 \$1,714,000

# SECTION 700 - CARBON DIOXIDE RECOVERY

Cost

Package Unit

\$5,800,000

Capacity: 300 tons/day Drives: 2,100 HP

Total Section 700

\$5,800,000

# SECTION 800 - ETHANOL PURIFICATION

	Cost
	<u>COSL</u>
Beer Column (TW-801)	\$324,000
Type: Distillation Column Size: 9.5' dia. x 135' t/t Material: CS shell/304 SS trays Temperature: 280°F Pressure: 40 psig Trays: 30 sieve 30 disc and donut	
First Stage Beer Preheater (HE-801)	\$29,800
Area: 2,400 square feet Type: Fixed T-S Duty: 16.1 MM Btu/hr Material: CS/CS Temperature: 160/138°F Pressure: 15/30 psig	
Second Stage Beer Preheater (HE-802)	\$16,600
Area: 1,100 square feet Type: Fixed T-S Duty: 10.2 MM Btu/hr Material: CS/CS Temperature: 230/185°F Pressure: 30/30 psig	
Trim Condenser (HE-805)	\$23,000
Area: 1,700 square feet Type: Fixed T-S Duty: 34.2 MM Btu/hr Material: CS/CS Temperature: 100/104°F Pressure: 30/150 psig Service: Start-up	
Vent Condenser (HE-806)	\$4,400
Area: 100 square feet Type: U-tube Duty: 0.5 MM Btu/hr Material: CS	
Beer Column Bottoms Pump (CP-801) - 2	<u>Items</u> \$9,200
Type: Centrifugal, AVS medium Capacity: 1,700 GPM	

Material: 304 SS Head: 50' TDH Drive: 25 HP

Temperature: 280°F

### Reflux Pump (CP-802) - 2 Items

\$7,800

Type: Centrifugal, AVS medium

Capacity: 300 GPM Material: 304 SS Head: 150' TDH Drive: 20 HP

Temperature: 235°F

### Preheater (HE-813)

\$23,000

Area: 1,700 square feet

Type: Fixed T-S Duty: 9.2 MM Btu/hr Material: CS/CS

Temperature: 200/300°F Pressure: 30/30 psig

### Final Beer Preheater (HE-803)

\$10,000

Area: 500 square feet Type: Fixed T-S Duty: 7.8 MM Btu/hr Material: CS/CS

Temperature: 250/300°F Pressure: 30/50 psig

# Beer Column Reboiler (HE-804)

\$18,800

Area: 1,300 square feet Type: Fixed T-S

Type: Fixed T-S
Duty: 52.4 MM Btu/hr
Material: CS/CS

Temperature: 386/280°F Pressure: 200/30 psig

# Reflux Drum (VT-801)

\$12,500

Type: Vertical

Capacity: 1,500 GPM

Size: 5' diameter x 10' t/t

Material: 304 SS Temperature: 230°F Pressure: 35 psig

# Dehydration Column (TW-802) \$247,000 Type: Distillation column Size: 5' diameter x 115' t/t Material: CS Temperature: 200°F Pressure: 0.05 psig Trays: 50 sieve Dehydration Column Reboiler (HE-807) \$120,000 Area: 10,000 square feet Type: Fixed T+S Duty: 29.4 MM Btu/hr Material: CS/CS Temperature: 199/230°F Pressure: 0.05/50 psig Dehydration System Overhead Cooler (HE-808) Area: 3,200 square feet Type: Fixed T-S Duty: 18.4 MM Btu/hr Material: CS/CS Temperature: 176/100°F Pressure: 15/150 psig Decanter (HT-801) Type: Horizontal drum Capacity: 8,800 gallons Material: CS Temperature: 175°F Pressure: 0.05 psig Dehydration Column Reflux Pump (CP-804) - 2 Items \$7,800 Type: Centrifugal, AVS medium Capacity: 240 GPM Material: 304 SS Head: 50' TDH Drive: 5 HP Temperature: 150°F Entrainer/Stripper Column Feed Pump (CP-805) - 2 Items \$5,000 Type: Centrifugal, AVS medium Capacity: 20 GPM Material: 304 SS Head: 50' TDH Drive: 2 HP Temperature: 150°F

Entrainer/Stripper Column (TW-803) \$80.500 Type: Distillation column Size: 3' diameter x 60' t/t Material: CS Temperature: 205°F Pressure: 0.05 psig Trays: 30 sieve Entrainer Stripper Reboiler (HE-809) \$24,400 Area: 1,900 square feet Type: Fixed T-S Duty: 4.8 MM Btu/hr Material: CS/CS Temperature: 230/205°F Pressure: 50/5 psig Entrainer Column Bottoms Pump (CP-806) - 2 Items \$6,400 Type: Centrifugal, AVS medium Capacity: 60 GPM Material: 304 SS Head: 100' TDH Drive: 2 HP Temperature: 2050F Fusel Oil Washer/Decanter (DDT-801) \$14,000 Type: Vertical Size: Top - 3' diameter x 2' t/t Bottom - 2' diameter x 10' t/t Material: CS Temperature: 100°F Pressure: 0.05 psid Fusel Oil Hold Tank (VT-802) \$4,300 Type: Vertical Capacity: 100 gallons Material: CS Temperature: 105°F Pressure: 0.05 psig Fusel Oil Cooler (HE-810) \$4,400 Area: 100 square feet Type: U-tube Material: CS Wash Water Pump (CP-809) - 2 Items \$3,000 Type: Centrifugal, AVS medium Capacity: 2 GPM

Material: CS Head: 150' TDH Drive: 0.5 HP

Temperature: 100°F

#### Fusel Oil Pump (CP-808) - 2 Items

Type: Centrifugal, AVS medium

Capacity: 2 GPM Material: CS . Head: 100' TDH Drive: 0.5 HP

Temperature: 100°F

#### Waste Water Cooler (HE-811)

Area: 100 square feet

Type: U-tube Duty: 0.1 MM Btu/hr

Material: CS

Temperature: 205°F Pressure: 50 psig

### Alcohol Cooler (HE-812)

\$10,200

Area: 300 square feet Type: U-tube

Duty: 0.6 MM Btu/hr

Material: CS

Temperature: 200°F Pressure: 5 psig

# Dehydration Column Bottoms Pump (CP-803) - 2 Items

Type: Centrifugal, AVS medium

Capacity: 60 GPM Material: CS Head: 100' TDH Drive: 2 HP

Temperature: 200°F

Subtotal Section 800 Undefined Miscellaneous Total Section 800

\$1,077,200 75,800 \$1,153,000

### SECTION 900 - FURFURAL PRODUCTION

Cost Stillage Receiver (VT-901) \$78,700 Type: Cone roof Capacity: 270,000 gallons Material: CS Temperature: 280°F Pressure: 75 psig Holdup: 8 hours Furfural Reactor Feed Pump (CP-901) - 2 Items \$11,200 Type: Centrifugal, AVS medium Capacity: 550 GPM Material: CS Head: 200' TDH Drive: 40 HP Temperature: 280°F Furfural Reactor (R-901) \$580,000 Type: Tower w/baffles Capacity: 1,000 gallons Material: Zirconium clad Temperature: 428°F Pressure: 700 psig First Flash Vessel (VT-902) \$42,000 Type: Cylinder Capacity: 1,000 gallons Material: Zirconium Temperature: 250°F Pressure: 30 psiq Second Flash Vessel (VT-903) \$28,200 Type: Cylinder Capacity: 700 gallons Material: 317 SS Temperature: 2120F Pressure: 15 psig Neutralization Tank (AOT-902) \$49,700 Type: Mixer Capacity: 5,000 gallons

Material: 317 SS

Drive: 25 HP

Temperature: 2120F Pressure: 15 psig Lime Pump (CP-902) - 2 Items \$4,000 Type: Centrifugal, AVS medium Capacity: 10 GPM Material: CS Head: 100' TDH Drive: 1 HP Temperature: 80°F Lime Tank (AOT-901) \$29,500 Type: Mixer Volume: 3,500 gallons Material: CS Drive: 20 HP Temperature: 80°F Pressure: 0.05 psig Holdup: 8 hours Thickener Feed Pump (CP-903) + 2 Items \$7,600 Type: Centrifugal, AVS medium Capacity: 450 GPM Material: CS Head: 100' TDH Drive: 20 HP Temperature: 212°F Thickener (T-901) \$34,300 Type: Gravity with rake Capacity: 450 GPM Material: CS Drive: 5 HP Diameter: 24' \$56,400 CaSO4 Centrifuge (CT-901) Type: Solid bowl Size: 10" diameter Material: CS Drive: 25 HP Overflow: 60 GPM Overflow Pump (CP-904) - 2 Items Type: Centrifugal, AVS medium Capacity: 60 GPM Material: CS

Head: 100' TDH Drive: 2 HP

Temperature: 212°F

Intermediate Furfural Receiver (VT-904)

\$32,000

Type: Cone roof

Capacity: 60,000 gallons Material: CS

Temperature: 212°F Pressure: 0.05 psig Holdup: 2 hours

Azeotrope Feed Pump (CP-905) -2 Items

\$7,600

Type: Centrifugal, AVS medium

Capacity: 500 GPM Material: CS Head: 100' TDH Drive: 20 HP

Temperature: 2120F

Azeotrope Reboiler (HE-901)

\$96,400

Area: 3,200 square feet

Type: Fixed T-S Duty: 50.8 MM Btu/hr Material: CS/304 SS Temperature: 248/2200F Pressure: 30/30 psig

Azeotrope Column (TW-901)

\$234,000

Type: Distillation column Size: 8' dia. x 140' t/t Material: Shell - CS

Trays - 304 SS

Pressure: 15 psig Trays: 70 sieve

Azeotrope Condenser (HE-908)

\$31,000

Area: 790 square feet

Type: Fixed T-S

Duty: 48.7 MM Btu/hr Material: CS/304 SS Temperature: 122/2080F Pressure: 15/15 psig

BFW Preheater (HE-903)

\$19,500

Area: 460 square feet

Type: Fixed T-S Duty: 16.6 MM Btu/hr

Material: CS/304 SS

Type: Fixed T-S

Temperature: 180/208°F Pressure: 15/15 psiq \$14,900 Decanter (HT-901) Type: Horizontal Capacity: 6,000 gallons Material: CS Temperature: 2080F Pressure: 0.05 psig Holdup: 30 minutes \$7,000 Lights Column Feed Pump (CP-910) - 2 Items Type: Centrifugal, AVS medium Capacity: 160 GPM Material: CS Head: 100' TDH Drive: 7.5 HP Temperature: 2080F Dehydration Feed Pump (CP-907) - 2 Items \$4,400 Type: Centrifugal, AVS medium Capacity: 15 GPM Material: CS Head: 100' TDH Drive: 1 HP Temperature: 2080F Lights Column (TW-902) Type: Distillation column Size: 3' dia. x 30' t/t Material: Shell - CS Trays - 304 SS Pressure: 15 psig Trays: 15 sieve Reflux Pump (CP-912) - 2 Items \$4,000 Type: Centrifugal, AVS medium Capacity: 5 GPM Material: CS Head: 100' TDH Drive: 0.5 HP Temperature: 176°F Lights Column Condenser (HE-905) Area: 200 square feet

Duty: 9.0 MM Btu/hr Material: CS/304 SS Temperature: 120/176°F Pressure: 15/15 psig

#### Reflux Drum (HT-902)

\$1,000

Type: Horizontal Capacity: 100 gallons

Material: CS

Temperature: 176°F Pressure: 0.05 psig

#### Lights Cooler (HE-909)

\$1,000

Area: 2 square feet Type: Fixed T-S Duty: 0.02 MM Btu/hr Material: CS/CS

Temperature: 120/176°F Pressure: 15/15 psig

## Lights Column Reboiler (HE-904)

\$19,600

Area: 410 square feet

Type: Fixed T-S
Duty: 9.0 MM Btu/hr
Material: CS/304 SS
Temperature: 248/212°F
Pressure: 30/15 psig

# Lights Bottoms Pump (CP-911) - 2 Items

\$7,200

Type: Centrifugal, AVS medium

Capacity: 180 GPM Material: CS Head: 100' TDH Drive: 7.5 HP Temperature: 212°F

### Azeotrope Bottoms Pump (CP-906) - 2 Items

\$8,800

Type: Centrifugal, AVS medium

Capacity: 725 GPM Material: CS Head: 100' TDH Drive: 25 HP

Temperature: 220°F

# Stillage Pump (CP-913) - 2 Items

\$8,800

Type: Centrifugal, AVS medium

Capacity: 650 GPM

Material: CS Head: 150' TDH Drive: 25 HP Temperature: 220°F Dehydration Column (TW-803) \$5,000 Type: Distillation column Size: 1' dia. x 20' t/t Material: Shell - CS Trays - 304 \$\$ Pressure: 15 psiq Trays: 10 sieve Dehydration Condenser (HE-907) \$6,600 Area: 20 square feet Type: Fixed T-S Duty: 1.2 MM Btu/hr Material: CS/304 SS Temperature: 120/2080F Pressure: 15/15 psig Reflux Drum (HT-903) \$1,000 Type: Horizontal Capacity: 100 gallons Material: CS Temperature: 208°F Pressure: 0.05 psig Reflux Pump (CP-909) - 2 Items \$4,000 Type: Centrifugal, AVS medium Capacity: 5 GPM Material: CS Head: 50' TDH Drive: 0.5 HP Temperature: 2080F Dehydration Reboiler (HE-906) \$7,000 Area: 30 square feet Type: Fixed T-S Duty: 1.2 MM Btu/hr Material: CS/304 SS Temperature: 400/3240F Pressure: 250/15 psig Dehydration Bottoms Pump (CP-908) - 2 Items \$4,500

Type: Centrifugal, AVS medium

Capacity: 25 GPM

Material: CS Head: 100' TDH Drive: 2 HP

Temperature: 3240F

# Furfural Cooler (HE-902)

\$8,000

\$1,531,000 107,000 \$1,638,000

Area: 120 square feet
Type: Fixed T-S
Duty: 1.0 MM Btu/hr
Material: CS/304 SS
Temperature: 120/3240F

Pressure: 15/15 psig

Subtotal Section 900
Undefined Miscellaneous
Total Section 900

### SECTION 1000 - HEAT GENERATION

Cost

### Multi-Effect Evaporator (E-1001)

\$2,100,000

Type: Six effect-forced circulation

Duty: Heating - 44.8 MM Btu/hr

Cooling - .67.8 MM Btu/hr
Capacity: 328,000 lb/hr distillate
Material: CS

Drive: 1,100 HP

#### Distillate Pump (CP-1006) - 2 Items

\$10,000

Type: Centrifugal, AVS medium

Capacity: 660 GPM

Material: CS Head: 150' TDH Drive: 25 HP

Temperature: 1690F

### Wood Boiler Package

Part of offsite facilities

Total Section 1000

2,110,000

# SECTION 1100 - WASTE TREATMENT

Cost

Condensate Treatment System (WTS-1101)

\$2,060,000

Package Unit

Type: Aerobic/anaerobic carrousel system

Capacity: 1,450 GPM Size: Aerobic basins - 2-177' diameter

Anaerobic clarifiers - 2-65' diameter

Gravity thickener - 1-30' diameter

Orive: 600 HP

Gypstack (WTS-1102)

\$500,000

Area: 1 acre expandable Type: Dyked/lined storage

Drives: 60 HP

Total Section 1100

\$2,560,000

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